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Part - I

Grade 7

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The National Anthem of Sri Lanka

Sri Lanka Matha
Apa Sri Lanka Namo Namo Namo Namo Matha
Sundara siri barinee, surendi athi sobamana Lanka
Dhanya dhanaya neka mal palaturu piri jaya bhoomiya ramya
Apa hata sepa siri setha sadana jeewanaye matha
Piliganu mena apa bhakthi pooja Namo Namo Matha
Apa Sri Lanka Namo Namo Namo Namo Matha
Oba we apa vidya
Obamaya apa sathya
Oba we apa shakthi
Apa hada thula bhakthi
Oba apa aloke
Apage anuprane
Oba apa jeevana we
Apa mukthiya oba we
Nava jeevana demine, nithina apa pubudukaran matha
Gnana veerya vadawamina regena yanu mana jaya bhoomi kara
Eka mavakage daru kela bevina
Yamu yamu vee nopama
Prema vada sema bheda durerada
Namo, Namo Matha
Apa Sri Lanka Namo Namo Namo Namo Matha

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ஒரு தாய் மக்கள் நாமாவோம்
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யான்று மழியாச் செல்வமன்றோ.

ஆனந்த சமரக்கோன்
கவிதையின் பெயர்ப்பு.

Being innovative, changing with right knowledge
Be a light to the country as well as to the world.

Message from the Hon. Minister of Education

The past two decades have been significant in the world history due to changes that took place in technology. The present students face a lot of new challenges along with the rapid development of Information Technology, communication and other related fields. The manner of career opportunities are liable to change specifically in the near future. In such an environment, with a new technological and intellectual society, thousands of innovative career opportunities would be created. To win those challenges, it is the responsibility of the Sri Lankan government and myself, as the Minister of Education, to empower you all.

This book is a product of free education. Your aim must be to use this book properly and acquire the necessary knowledge out of it. The government in turn is able to provide free textbooks to you, as a result of the commitment and labour of your parents and elders.

Since we have understood that the education is crucial in deciding the future of a country, the government has taken steps to change curriculum to suit the rapid changes of the technological world. Hence, you have to dedicate yourselves to become productive citizens. I believe that the knowledge this book provides will suffice your aim.

It is your duty to give a proper value to the money spent by the government on your education. Also you should understand that education determines your future. Make sure that you reach the optimum social stratum through education.

I congratulate you to enjoy the benefits of free education and bloom as an honoured citizen who takes the name of Sri Lanka to the world.

Akila Viraj Kariyawasam
Minister of Education

Foreword

The educational objectives of the contemporary world are becoming more complex along with the economic, social, cultural and technological development. The learning and teaching process too is changing in relation to human experiences, technological differences, research and new indices. Therefore, it is required to produce the textbook by including subject related information according to the objectives in the syllabus in order to maintain the teaching process by organizing learning experiences that suit to the learner needs. The textbook is not merely a learning tool for the learner. It is a blessing that contributes to obtain a higher education along with a development of conduct and attitudes, to develop values and to obtain learning experiences.

The government in its realization of the concept of free education has offered you all the textbooks from grades 1-11. I would like to remind you that you should make the maximum use of these textbooks and protect them well. I sincerely hope that this textbook would assist you to obtain the expertise to become a virtuous citizen with a complete personality who would be a valuable asset to the country.

I would like to bestow my sincere thanks on the members of the editorial and writer boards as well as on the staff of the Educational Publications Department who have strived to offer this textbook to you.

W. M. Jayantha Wickramanayaka,
Commissioner General of Educational Publications,
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2019.04.10

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Introduction

This textbook was compiled by the Educational Publications Department in accordance with the syllabus prepared by the National Institute of Education for the use of Grade seven students in the Sri Lankan school system with effect from 2016. An effort has made here to arrange the subject content to suit the national educational goals, common national competencies, the objectives of teaching science and the content of the syllabus.

The subject of science directs the student towards a more active learning process in a manner as to develop knowledge, skills and attitudes needed for a developmental scientific thought.

In the compilation of this textbook, subject content is largely arranged based on experiences of daily life. It has contributed to prove the fact that the subject of science is very much closer to the day to day life.

The compilation of this textbook based on activities is a distinctive feature. The activities are prepared based on the scientific method in order to develop knowledge, skills and attitudes. Activities that can be performed individually at home as well as in school are incorporated here. We believe that learning through activities would contribute to create a liking and an interest in the child towards learning science.

At the end of each chapter, a summary, a series of exercises and a glossary were included. It enables the student to identify the important details of the chapter as well as to self evaluate the achievement of learning outcomes.

For the purpose of directing the student to study further about the subject matter, more information is included in the "For you extra Knowledge". It is given only to broaden the subject area of the child and certainly not to ask questions at term tests.

Assignments and projects are given with the purpose of directing the student towards an explorative study. It enables the students to develop the higher order skills such as application, analysis and synthesis of the concepts achieved from the lesson.

We strongly believe that the duty of the teachers who teach science is to direct the student for self learning instead of teaching the student using traditional teaching methods. This textbook can be utilized by the teachers as a learning tool to execute their teaching role properly.

We would like to bestow our sincere thanks on the senior lecturer Asoka De Silva of the National Institute of Education and the professional writer Dr. K. Ariyasinghe.

We kindly request you to forward your comments and suggestions on this textbook to the Educational Publications Department.

Board of Writers and Editors

Contents

01 Plant Diversity

1.1 Morphological features of flowering plants

Flowers and fruits naturally come to our mind when we state about plants. Do all plants produce flowers? Pay your attention to the ornamental plants in the garden given in Figure 1.1.

Figure 1.1 ▲ View of a garden

You can observe plants with flowers and without flowers. There are different varieties of plants in our environment. Engage in Activity 1.1 to find out whether all the trees bear flowers.

Activity 1.1

Names and the pictures of several plants are given below. Although most of them are familiar to you, certain plants may not be familiar to you.

Rose

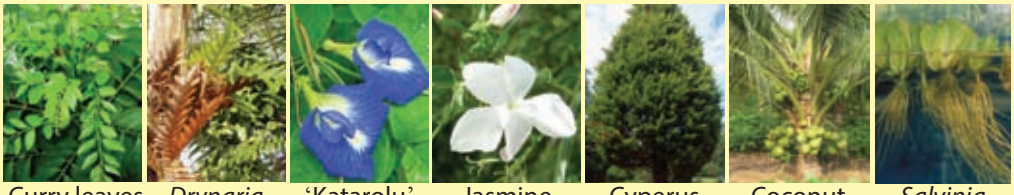
Fern

'Idda'
'Nanthiavattai'

Gardenia

'Balsam'
'Kasiththumbai'

Water Lily ('Olu' / 'Alli') | Moss | 'Sal' 'Nagalingam' | Lotus | *Cycas* | Shoe flower | Long bean



Curry leaves (Karapincha) | *Drynaria* (Beduru) | 'Katarolu' 'Kakkattan' | Jasmine | Cyperus | Coconut | *Salvinia*

Figure 1.2 ▲ Several different types of plants

Prepare a table by grouping all the plants in Figure 1.2 into plants which bear flowers and plants which do not bear flowers. Compare your table with the following table.

Table 1.1 ▼

Plants which bear flowers	Plants which do not bear flowers
Rose, 'Idda', Gardenia, Balsam, Water lily, 'Sal', Lotus, Shoe flower, Curry leaves, 'Katarolu', Jasmine, Coconut, Long bean	*Cycas*, Moss, *Salvinia*, Cyperus, *Drynaria*, Ferns

Now, it is clear to you that there are plants which produce flowers and that do not produce flowers in our environment. Plants which produce flowers are called **flowering plants** and plants which do not produce flowers are called **non flowering plants**.

Assignment 1.1

- Observe plants in your home garden or in your school premises.
- Identify them as flowering plants and non flowering plants.
- Tabulate the plants you identified as flowering plants and non-flowering plants.

1.2 Main parts of a flowering plant

Observe the flowering plants in your surroundings. They are of different sizes. You will observe small palnts as well as big trees among them. Their stems, roots, flowers and fruits are also different from each other. Therefore, plants show a great diversity in their size and morphological features, but all flowering plants have some common parts. Let's do Activity 1.2 to identify the main parts of a flowering plant.

Assignment 1.2

Identification of main parts of a flowering plant

- Select a small plant that produces flowers and fruits.

  e.g.:- 'Monara Kudumbiya', / 'Seethaviyar Selugkaluner' 'Kuppameniya' / 'Kuppaimani'
- Add some water to the root and uproot it from the soil carefully without breaking roots.
- Remove soil around roots carefully
- Press it between pages of newspapers, by adding a weight. Let it dry.
- Prepare a field book and paste the specimens in it after about one week.

Activity 1.2

The Figure 1.3 shows the parts of a flowering plant. The main parts are labelled from 1 to 11.

01. Tap root
02. Lateral roots
03. Root system
04. Stem
05. Lateral buds (Axillary buds)
06. Leaves
07. Apical bud
08. Branches
09. Fruits
10. Flowers
11. Shoot system

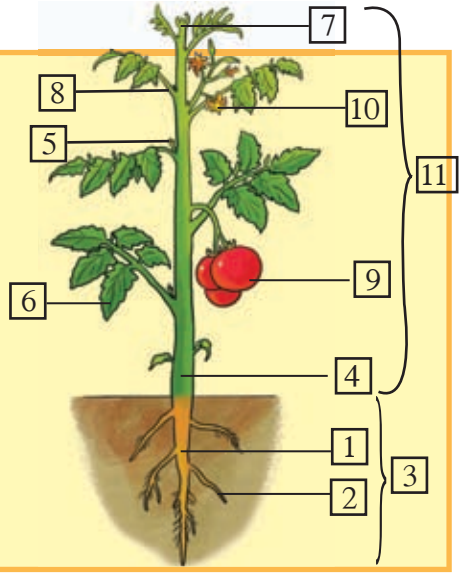


Figure 1.3 ▲ Parts of a flowering plant

Identify the main parts of a plant in Assignment 1.2 referring to Figure 1.3. Compare the pressed plant you have, with the pressed plants your friends have. Observe similarities as well as dissimilarities among the main parts of them.

1.3 Diversity of the parts of flowering plants

The characteristic feature of the flowering plant is formation of flowers. Almost all parts such as stem, leaves, roots, buds, flowers and fruits can be seen in most of the plants. Plants show a great diversity among these parts.

Diversity among roots of plants

Assignment 1.3

- Select a small grass plant and a 'Kuppameniya' / 'Kuppaimani' plant.
- After adding water uproot them carefully without damaging roots.
- Wash the soil of the roots. Press it between pages of newspaper and dry.
- Paste it in the field book after one week.

Normally root system remains underground in the soil. There are two types of root systems.

- Some plants have a single large root which originates from the base of the stem. It is called the **tap root**. Large number of roots are originated from the tap root. They are called **lateral roots**. This type of root system is called a **tap root system**.

e.g.:- 'Kuppameniya'/ 'Kuppaimani', Mango, Cashew

- Some plants have a large number of small roots which are originated from base of the plant stem. This type of root system is called a **fibrous root system**.

e.g.:- Coconut, Arecanut, Bamboo, Grass, 'Kitul'

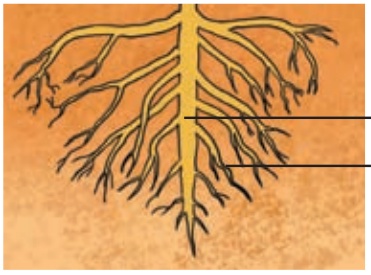


Figure 1.4 (a) ▲ Tap root system

Figure 1.4 (b) ▲ Fibrous root system

Activity 1.3

- Look at the root systems of two plants you pasted in your fieldbook.
- Observe the differences between root systems of those two plants.

You will notice that plants like 'kuppameniya', mango, cashew have a tap root system whereas grass, coconut, arecanut, and bamboo trees have a fibrous root system.

Functions of roots of plants

- Fix the plant to the soil
- Absorb water and minerals dissolved in water (absorption)
- Give rise to new plants through vegetative propagation
  e.g.:- Curry leaves, 'Beli'/ 'Vilvam', Bread fruit

In addition there are some roots adapted for different functions. Let's do Assignment 1.4 to identify such roots.

Assignment 1.4

- Uproot a *Mimosa* plant without breaking its roots and wash the roots carefully.
- Draw the root system of it on a white drawing paper.
- Paste it in the fieldbook.
- Observe the yams of beetroot, carrot, sweet potato and manioc.
- Draw them on a white drawing paper and paste it in your fieldbook.

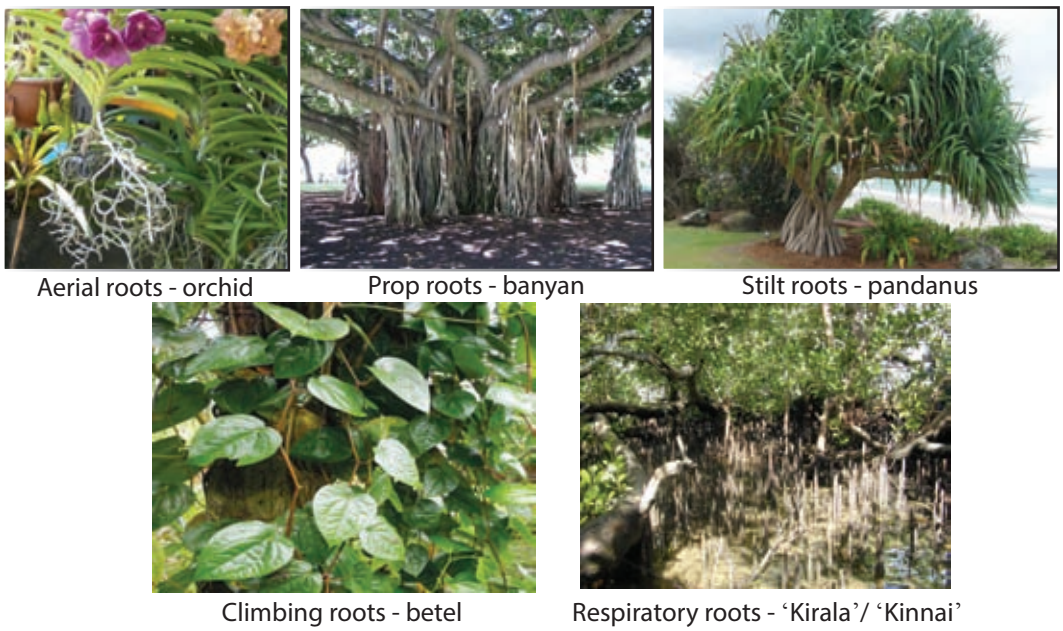
Manioc Beetroot Carrot Sweet potato *Mimosa*

Figure 1.5 ▲ Different types of roots

Small nodules like structures can be seen in the root system of *Mimosa* plant. Such nodules are known as root nodules. There are micro organisms called bacteria live inside them. These bacteria supply nitrogen nutrients required for the growth of *Mimosa* plant and root system supplies nutrients for the bacteria which live inside the nodules. Such nodules are present in plants such as 'kathurumurunga', beans and winged beans which belong to family Leguminosae.

The other roots shown in Figure 1.5 are generally called as yams because food is stored in those roots. Such roots are known as **storage roots**. Food can be stored in tap root as well as lateral roots.

Naturally, roots grow downward into the soil, but you may observe some roots that grow above the soil. Figure 1.6 shows different types of roots that grow above the soil level.



Aerial roots - orchid Prop roots - banyan Stilt roots - pandanus

Climbing roots - betel Respiratory roots - 'Kirala'/ 'Kinnai'

Figure 1.6 ▲ Roots adapted for various functions

Activity 1.4

Identify the different types of roots shown in Figure 1.6 through a field study.

Study Table 1.2 and try to understand various functions of root types.

Table 1.2 Different types of roots and their functions

Type of root	Example	Functions
Prop roots	Banyan	Support the branches
Stilt roots	Pandanus, 'Rampe'	Support the stem
Aerial roots	Orchid	Absorb water vapour from the atmosphere Some aerial roots carryout photosynthesis
Climbing roots	Pepper Betel	Helps to climb the stem by attaching to another stem
Respiratory roots	'Kadol'/'Kandal' 'Kirala' / 'Kinnai'	Exchange of air with the atmosphere
Storage roots	Carrot Beetroot Manioc Sweet potato	Store food
Roots with root nodules	*Mimosa* Legumes (eg: Beans, Long beans, Winged beans)	Bacteria live inside root nodules add nutrients to the soil. Thus make the soil fertile.

For extra knowledge

Corks used as stoppers of bottles are taken from respiratory roots of 'Kirala'/'Kinnai' plant. The porous or sponge nature of 'Kirala' / 'Kinnai' corks occur due to presence of very tiny pores and spaces in their roots.

Diversity among stems

Recall the nature of plant stems that you have observed in your surrounding. You may have observed very strong stems, stems with thick bark as well as stems with various colours. There are plants that grow up with the help of supporters and run on the ground due to their weak stems. The basic difference in any type of the stem is that whether the stem divides into branches or not.

Figure 1.7 ▲ A plant with unbranched stem

e.g.:- Coconut, Arecanut, Kitul, Palmyrah, Paddy, Bamboo

Figure 1.8 ▲ A plant with branched stem

e.g.:- Mango, Rambutan, Cashew, Guava, Shoe flower

Functions of plant stem

- Bears flowers, leaves, buds, fruits and seeds.
- Supports the plant, by keeping rigid.
- Transports water and food through plant body
- Many plants produce new plants by stems. (vegetative propogation)

  e.g.:- Shoe flower, Jasmine/ 'Saman pichcha' / 'Sadimalligai', Sweet potato, Manioc

- Some plants with green coloured stems carryout photosynthesis

  e.g.:- 'Nawahandi' / 'Kally', 'Heeressa' / 'Pirattai', 'Hathawariya'/ 'Sathavari'

- Some aerial stems store food.

  e.g.:- Sugar cane, Kitul

Eventhough most stems grow above the soil, some stems grow inside the soil. They are called **underground stems**.

Examples for underground stems are given in Figure 1.9

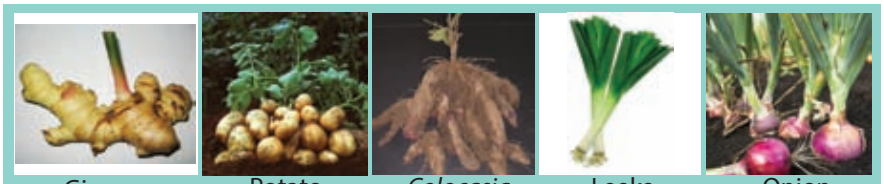


Figure 1.9 ▲ Plants with underground stems

Diversity of leaves

Plant leaves also display a great diversity similar to other parts of the plant; based on shape, size and colour. Let's engage in Activity 1.5 to learn about diversity of leaves.

Activity 1.5

- Collect about ten different types of leaves from your home garden.
- e.g.:- Manioc, Curry leaves, Jak, Grass, Croton, 'Akkapana' / 'Sadaikaraichchan' , Pumpkin, 'Hathawariya'/ 'Sathavari', 'Katurumurunga'/ 'Agathi'. Study their similarities as well as differences, by observing them.

Photosynthesis mainly occurs in a leaf of a plant. Plant leaves get energy from sunlight to do photosynthesis. Therefore, leaves are arranged on the stem in a way to get maximum amount of sunlight.


Figure 1.10 ▲ Arrangement of plant leaves in plants to absorb maximum amount of sunlight

Eventhough leaves have different shapes, sizes and different colours, they all have common parts. Those common parts are shown in Figure 1.11.

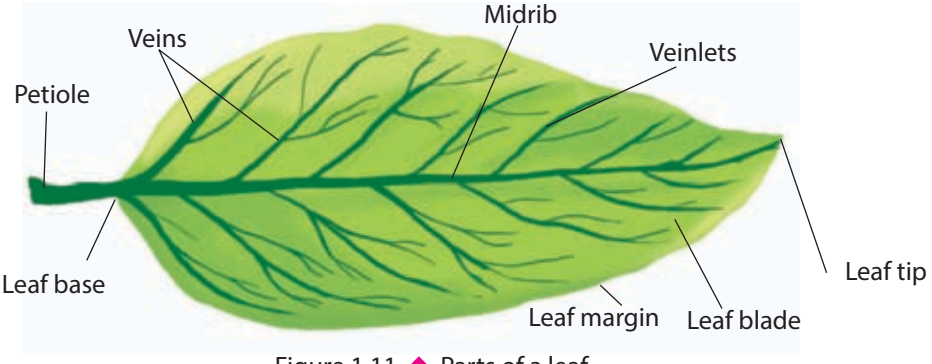


Figure 1.11 ▲ Parts of a leaf

Activity 1.6

- Select a leaf of a plant you pressed and paste it on the fieldbook.
- Identify all parts given in Figure 1.11 and label them.

Leaf venation

The arrangement of veinlets in the leaf is called leaf venation. There are two main venation patterns occur in plants.

- Reticulate venation

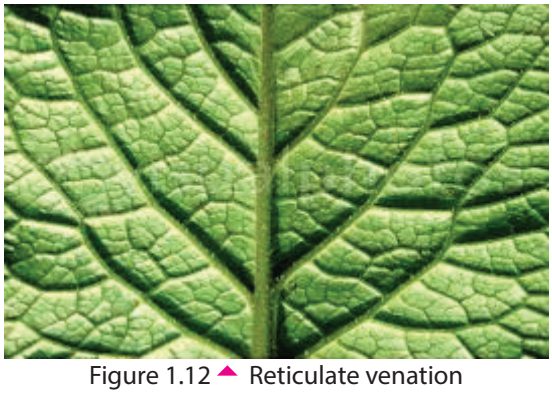


Figure 1.12 ▲ Reticulate venation

The branches initiated from the mid rib spread as a net through out the leaf.

eg:- Shoe flower, Mango, Jak, 'Gotukola'/ 'Vallarai'

- Parallel venation

Figure 1.13 ▲ Parallel venation

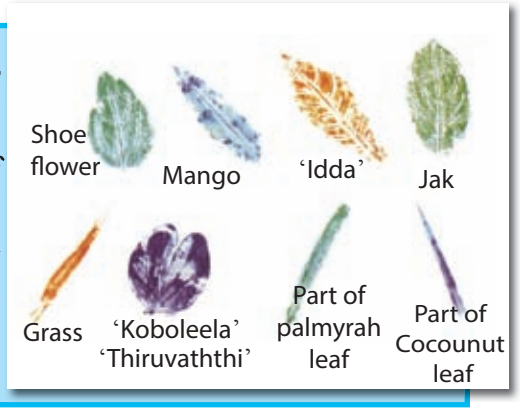
Veinlets parallel to the midrib spread in the leaf.

eg:- Grass, Bamboo, Coconut, Arecanut

You can observe leaf venation by observing lower surface of the plant leaf clearly.

Assignment 1.5

Apply paint on the lower surfaces of some leaves and get copies of them to a white paper. Identify the type of leaf venation and classify them as reticulate venation or parallel venation. Finally paste them in your fieldbook.



Simple and compound leaves

When the leaf blade of a leaf is not divided into segments it is called a **simple leaf**.
e.g.:- Shoe flower, Jak

The leaf blade of some simple leaves are partially divided into segments.
e.g.:- Manioc, Papaw

Shoe flower | Jak | Papaw

Figure 1.14 ▲ Several simple leaves

Leaf blade of a compound leaf is completly divided into small leaf like parts called leaflets. These types of leaves with leaflets are called **compound leaves**.

e.g.:- Coconut, 'Katurumurunga'/ 'Agathi', Tamarind

Coconut | Tamarind | 'Katurumurunga'/ 'Agathi'

Figure 1.15 ▲ Several compound leaves

Functions of plant leaves

- The main function of a plant leaf is photosynthesis. Through photosynthesis plants produce food in leaves.
- Some leaves are adapted to store water.

  e.g.:- *Aloe*, 'Akkapana' / 'Sadaikaraichchan'

Figure 1.16 ▲ *Aloe*

Figure 1.17 ▲ *Bryophyllum* / 'Akkapana'/ Sadaikaraichchan'

- Some leaves produce new plants (vegetative reproduction).

  e.g.:- 'Akkapana', Begonia

Assignment 1.6

- Place a leaf of Bryophyllum in between two blotting papers and keep it between the pages of a book for few days.
- Observe it after few days.
- Identify the roots arising from leaf margin. They are called adventitious roots.
- Cut the leaf into pieces and get new plants by planting those pieces.

Parts of a flower and diversity of flowers

The main function of flowers is to produce fruits. The seeds inside the fruits produce new plants. These seeds are dispersed by various methods (by animals, wind, water, explosive mechanism)

Flowers bear male and female reproductive structures. They combine and form seeds by sexual reproduction.

The most attractive part of the plant is the flower. They differ greatly in size, smell, shape and colour.

Assignment 1.7

- Collect different types of flowers.
- Draw their external appearance on a white paper.
- Write the name of the plant below the diagram.
- Identify the parts of flowers and name them, by using Figures 1.18 and 1.19.

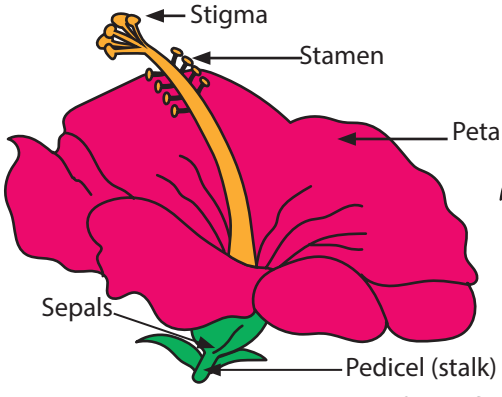


Figure 1.18 ▲ External appearance of shoe flower

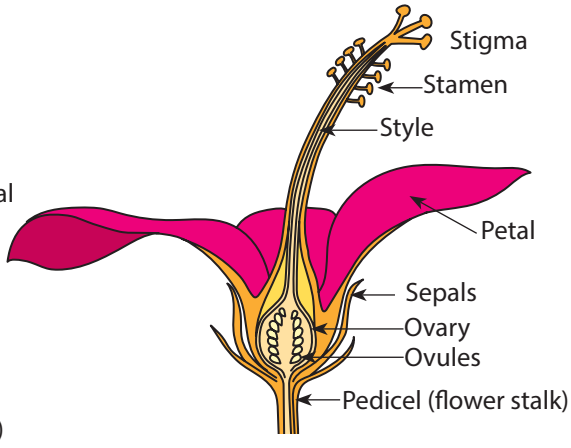


Figure 1.19 ▲ Half flower of shoe flower

Flowers show great diversity among them, but they have a common structural plan. A typical flower consists of the following three parts.

- Sepals
- Petals
- Gynoecium/ Androecium

All these parts can be easily observed in a shoe flower. Let's identify parts of a flower by observing half flower of a shoe flower. (Figure 1.19)

Activity 1.7

- Select a bigger flower. (e.g.:- Shoe flower, Thunbergia)
- Cut and separate the flower into two parts longitudinally by carefully cutting it from the pedicel using a sharp blade.
- Use Figure 1.19 to show a diagram of the longitudinal section of a flower. Identify its parts and name them.

Sepals

Sepals are generally green in colour. The main function of sepals is the protection of flower buds.

Petals

Petals are brightly coloured. They attract insects for pollination as well as protect the internal parts of the flower.

Androecium

The male part of a flower is called androecium which consists of stamen. Each stamen is made up of two parts.

- anther
- filament

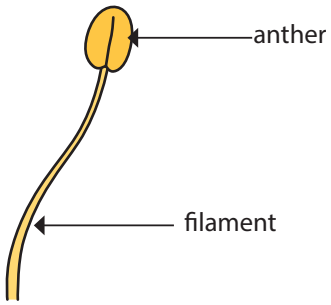


Figure 1.20 ▲ A stamen of shoe flower

The function of androecium is production of pollen.

Different shapes of stamens can be seen in flowers

Lotus | 'Sal'/ 'Nagalingam' | 'Niyagala' / 'Karthikaipoo' | Lily

Figure 1.21 ▲ Different shapes of stamens in flowers

Gynoecium

Gynoecium contains following parts.

- Stigma
- Style
- Ovary

Ovules are present within the ovary. The function of gynoecium is the production of seeds.

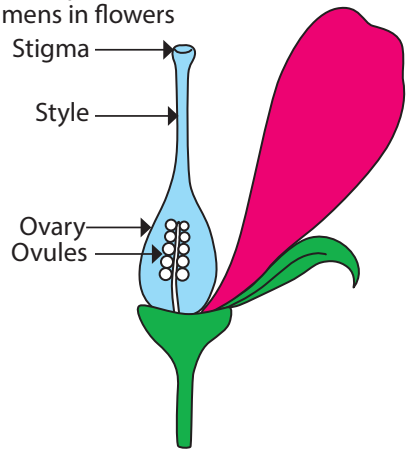


Figure 1.22 ▲ Gynoecium of a flower

Activity 1.8

Using maximum number of examples complete Table 1.3 further with the help of the features of flowers that you have observed. One example is given for each section.

Table 1.3

White coloured flowers	'Idda' / 'Nanthiavattai',
Flowers with colourful petals	Rose, ...
Flowers that bloom at night	'Sepalika' / 'Pavala mallikai'
Flowers with sweet smell (fragrance)	Jasmine/ 'Samanpichcha'/'Sadimalligai'
Flowers with nectar	'Kathurumurunga' / 'Agathi'

Diversity of fruits and seeds

Fruits are formed from the flowers of flowering plants. Seeds are found inside the fruits. Seeds produce new plants.

Assignment 1.8

- Collect fruits and seeds that are fallen near trees in the school garden. (Collect the seeds into a seed box)
- Collect fruits and seeds that are fallen under trees in your home garden.
- Collect different types of fruits and seeds that are not found regularly. (Try to find the names of those plants)

Fruits and seeds are naturally adapted for dispersion.

e.g.:- Cotton, 'Wara' / 'Eruku', 'Hora' / 'Ennai', 'Gammalu' / 'Thanakku'

Cotton 'Wara'/'Eruku' 'Hora'/'Ennai' 'Gammalu'/ 'Thanakku'

Figure1.23 ▲ Fruits and seeds which are dispersed by wind

Assignment 1.9

Collect fruits and seeds as shown in Figure 1.23. List out the adaptations shown for wind dispersal . Get assistance from your teacher.

Arecanut | Coconut | 'Kottamba', 'Kottankachchi' | 'Diya kaduru' 'Kullithy' | Lotus

Figure 1.24 ▲ Fruits and seeds which are dispersed by water

Assignment 1.10

Collect fruits and seeds as shown in Figure 1.24. List out the adaptations they show to be dispersed by water. Get assistance from your teacher.

Tomato | Papaw | Chillies | Castor | 'Nagadarana' 'Pulinagam'

Figure 1.25 ▲ Fruits and seeds which are dispersed by animals

Assignment 1.11

- Collect fruits and seeds as shown in Figure 1.25. List out the adaptations they show to be propagated by animals. Get assistance from your teacher.

1.4 Monocotyledonous and dicotyledonous plants

Flowering plants are mainly divided into two groups as monocotyledonous (monocots) and dicotyledonous (dicots).

Do Assignment 1.12 to identify differences between two types of plants.

Assignment 1.12

- Collect many seeds as far as possible from your kitchen

e.g.:-Gram, Paddy/Rice, Green gram, Tamarind, Cowpea, Beans, Long beans, Cashew, Arecanut, Maize, Jak seeds

- Take about five seeds from each type and soak them in water
- Take the seeds out of water after about 24 hours carefully
- Divide them according to number of cotyledons (seed leaves) inside the seed and add to Table 1.4

Table1.4 - monocot and dicot seeds

Seeds with single seed leaf	Seeds with two seed leaves
Arecanut, Paddy, Maize	Bean, Long Beans, Cashew, Gram, Green gram, Tamarind, Cowpea, Jak seeds

You may have seen that certain seeds can be divided easily into two seed lobes whereas other seeds cannot be divided easily like that. There are two seed lobes. These seed lobes are called **seed leaves** (Cotyledons). Seeds with two cotyledons are called **dicot seeds**.

Some seeds cannot be divided into two seed lobes because they have only one seed leaf (cotyledon). This type of seeds are called **monocot seeds**. Germination of monocot and dicot seeds are different from each other.

Assignment 1.13

- Observe the plants/ trees found in your school garden and group them as monocots and dicots.

Let's do Activity 1.9 to learn more about nature of those seeds.

Activity 1.9

- Take bean seeds and paddy seeds which were soaked for about 24 hours and keep them on a wet cloth for about three days. Add a little bit of water every day.
- When bean seeds are about to break into two, separate the seed lobes in one seed and observe them.
- Plant other seeds in pots with wet soil. After few days you can observe that there are two leaves different from the other leaves of the germinated bean seeds. They are called seed leaves.
- Seed leaf cannot be observed in paddy seeds. Because the seed leaf does not come out during germination.

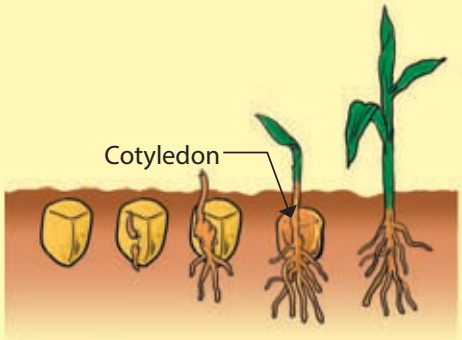


Figure 1.26 ▲ Germination of monocot seed

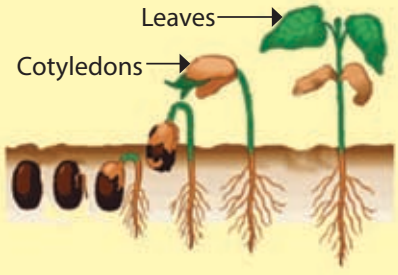


Figure 1.27 ▲ Germination of dicot seed

You have learnt about the variations of the basic parts of plant throughout this lesson. Now study the differences of basic parts of monocot plants and dicot plants that you have observed. Compare the differences you observed with Table 1.5.

Table 1.5

Part	Monocot plant	Dicot plant
Seeds	Only one seed lobe	Two seed lobes
Stem	Not branched	Branched
Leaves	Parallel venation	Reticulate venation
Flowers	Three petals or its multiples	Four/ Five petals or its multiples

Root	Fibrous root system	Tap root system
	 Figure 1.28 ▲ A monocot plant	 Figure 1.29 ▲ A dicot plant

Activity 1.10

- Create a model to show the basic differences between main parts of dicot and monocot plants that you have already observed.

Now you must have a proper understanding about the vast diversity of members of the plant world, upon observing and studying morphological variations of them.

The field book that you prepared would display the diversity of plants very well.

Summary

- Plants which produce flowers are called flowering plants whereas plants do not produce flowers are called non flowering plants.
- The main parts of the flowering plants are roots, stem, leaves, flowers, fruits and seeds.
- Although main parts of a plant usually perform one specific function, sometimes they are adapted for several other functions.
- A vast variation can be seen among main parts of the plants.
- Flowering plants can be divided into two groups as monocotyledonous (monocot) plants and dicotyledonous (dicot) plants.

Exercise

1. The table below indicate several kinds of plants with numbers identified by a group of students in a field trip to a forest.

Name of the plant	'Kitul'	Cashew	'Dan'/ 'Naval'	'Kottamba'	'Beduru'	*Cycas*	'Madu wel'	*Mimosa*
Number of plant	2	3	4	4	2	1	10	12

I. Display the data in a bar chart.

II. What is the most abundant plant found in this forest?

III. Name a plant/plants in the above forest, that found with following features

(a) A weak stem
(b) An unbranched stem
(c) No flowers/non-flowering
(d) A fibrous root system
(e) Compound leaves
(f) Fruits with fibrous outer layer
(g) Having root nodules

IV.

(a) Select a monocotyledonous and a dicotyledonous plant out of the plants given above.

(b) Write one major difference of (a) leaves (b) Stem (c) roots and (d) seeds of above two plants.

Technical Terms

Monocotyledonous	ඒකබීජපත්‍රී ශාක	ஒரு வித்திலைத் தாவரம்
Dicotyledonous	ද්විබීජපත්‍රී ශාක	இருவித்திலைத் தாவரம்ளு
Flowering plants	සපුෂ්ප ශාක	பூக்கும் தாவரங்கள்
Non flowering plants	අපුෂ්ප ශාක	பூக்காத் தாவரங்கள்
Gynoecium	ජායාංගය	பெண்ணகம்
Androecium	පුමංගය	ஆணகம்
Petals	දලපත්‍ර	அல்லிகள்
Corolla	මුකුටය	அல்லி வட்டம்
Sepals	මණ්ඩපත්‍ර	புல்லிகள்
Stigma	කලංකය	குறி
Style	කීලය	தம்பம்
Ovary	ඩිම්බ කෝෂය	சூலகம்
Venation	නාරටි වින්‍යාසය	நரம்பமைப்பு

02 Static Electricity

2.1 Charging an object

Cut several small pieces of dry paper and place them on a table. Rub a plastic pen against your dry hair and hold the pen close to the pieces of paper. What can you observe ?

You can observe how the pieces of paper are attracted to the pen.

- Are there any other objects that can be used instead of a pen to attract pieces of paper?

Figure 2.1 ▲ How the pieces of dry paper are attracted to the pen

Do you know the reason for the attraction of light things towards an object which is rubbed against some other material ?

Let's do Activity 2.1 to find the reason.

Activity 2.1

You will need :- A glass rod, a piece of PVC pipe, an ebonite rod, a sheet of polythene, a piece of silk cloth, a piece of woolen cloth, some small pieces of paper

Method :-

Hold the objects given in Table 2.1 close to the tiny pieces of paper. Observe what happens before and after rubbing them and complete the column of the table according to your observations.

Table 2.1 ▼

Instance	Object	Observation when held close to the pieces of paper
Before rubbing	• Glass rod • PVC pipe • Ebonite rod	....
After rubbing	• Glass rod rubbed with the silk cloth • PVC pipe rubbed with the polythene sheet • Ebonite rod rubbed with the woolen cloth	....

It is observed that the objects mentioned in Table 2.1, do not attract the pieces of paper before rubbing. But they do attract them after rubbing. Accordingly, it is clear that rubbing an object can make a certain change on it.

The scientist William Gilbert (1600 AD) was the first to reveal that light things are attracted to rubbed objects.

For extra knowledge

Scientist William Gilbert observed that when a piece of solid amber is rubbed with a silk cloth, light objects like tiny pieces of paper and feathers are attracted to it

Gilbert said that electric charges are generated on the surface of objects, when they are rubbed. So, light objects are attracted to them.

- When some objects are rubbed, electric charges are generated on them.
- Electric charges generated on the surface of objects by rubbing, are called static electric charges.

2.2 Types of static electric charges

Let's do Activity 2.2 to study further about the static electric charges.

Activity 2.2

You will need :- Two rods of glass, two rods of ebonite, a piece of silk cloth, a piece of woolen cloth, few pieces of thread, two stands

Method :-

- Hang one glass rod rubbed with the silk cloth, on one stand.

- Hang one ebonite rod rubbed with the woolen cloth, on the other stand.
- Rub the other glass rod with the silk cloth to charge it. Then bring the charged glass rod separately towards the rods hung.

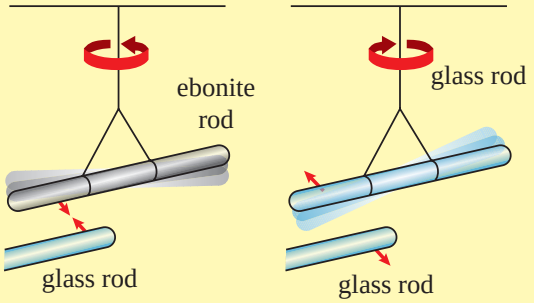


Figure 2.2 ▲ Attractions and repulsions

- Record the observations.
- Rub the other ebonite rod with the woolen cloth and charge it. Then bring it separately towards the rods hung.
- Tabulate your observations. Table 2.2 ▼

Hanging rod	Rod brought towards	Observations
Glass	Glass	Repelled
Ebonite	Glass	
Glass	Ebonite	
Ebonite	Ebonite	

Compare the observations of your group with those of the other groups. Discuss the reasons for the observations.
It is observed that there are attractions as well as repulsions between charged rods.

Attractions and repulsions between charged rods are illustrated in Figure 2.3 and 2.4.

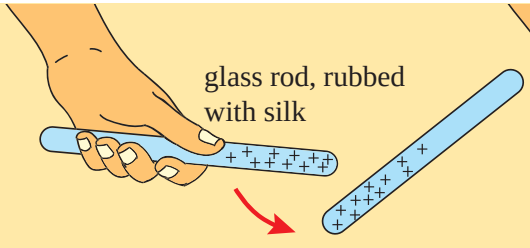


Figure 2.3 ▲ Repulsion

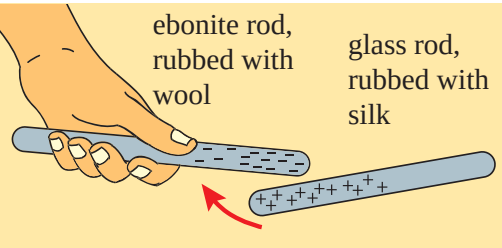


Figure 2.4 ▲ Attraction

The reason for the attraction and the repulsion between charged rods is the existence of two types of static electric charges.

They are as follows.

1. Positive (+) static electric charges.
2. Negative (-) static electric charges.

- Objects with like electric charges repel each other.
- Objects with unlike electric charges attract each other.

Therefore, a glass rod rubbed with silk gains positive (+) static electric charges and ebonite rod rubbed with wool gains negative (-) static electric charges.

Exercise

A PVC rod, rubbed with a polythene sheet is hung with a thread. When a glass rod rubbed with silk is brought towards the PVC rod, it is attracted. What is the type of static electric charge on PVC rod ?

For extra knowledge

- There are various equipment to identify static electric charges. One of them is the gold leaf electroscope.

When a charged object is brought towards the metal plate of this electroscope, the gold leaf deflects.

Let us do the Activity 2.3 to summarise what we have learnt so far.

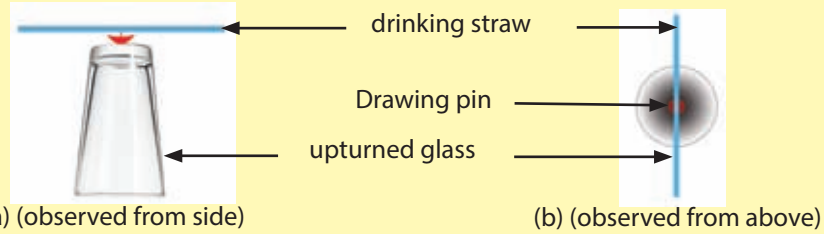
Activity 2.3

You will need :- Clean and dry drinking straws, drawing pins, a glass, sheet of polythene

Method :-

- Charge the drinking straw by rubbing it with the sheet of polythene.

- Balance the charged drinking straw on an upturned glass using a drawing pin as showing in the Figure 2.5 (a)



(a) (observed from side) (b) (observed from above)

Figure 2.5

- Keep a finger near the charged drinking straw (about 1 cm away). Figure 2.5 b, shows how it in seen when observed from above
- Now, keep the polythene used to charge the drinking straw near it.
- Then, keep another charged straw near the charged straw.
- Discuss the reasons for the observations.

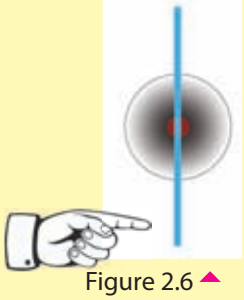


Figure 2.6

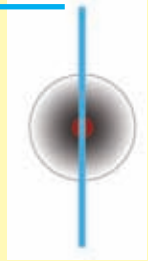


Figure 2.7

2.3 Generation of static electric charges

The amounts of positive and negative particles are equally distributed on an object before it is rubbed. Therefore, the object does not show any net charge.

When the objects are rubbed against each other negatively charged particles (electrons) remove from one object and transfer to the other object.

The object from which negatively (-) charged particles are removed, is charged positively, and the object that gained negatively (-) charged particles is charged negatively.

The process that takes place when a pair of objects are rubbed against each other can be illustrated as follows.

1) **Before rubbing**

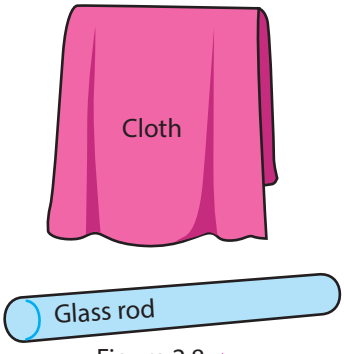


Figure 2.8

Positively (+) and negatively(-) charged particles are scattered uniformly. Therefore, objects do not show any net charge.

2)While rubbing

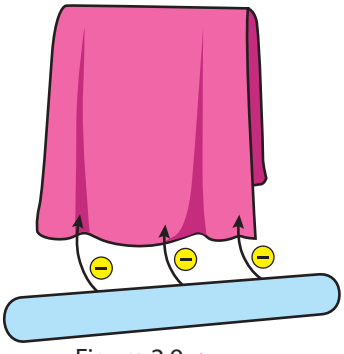


Figure 2.9

Negatively (-) charged particles removed from one object and they are gathered on surface of the other.

3) After rubbing

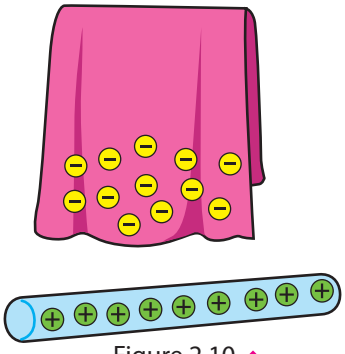


Figure 2.10

One object is negatively (-) charged, because more negetively (-) charged particles are collected on it. The other object is positively (+) charged, because nagatively (-) charged particles are removed from it.

It is clear to you when two objects are rubbed with each other, one object is positively charged while the other is negatively charged.

Accordingly, positively charged and negativly charged objects contact each other, two objects become neutral. Therefore, objects do not possess any net charge.

Let us do activity 2.4.

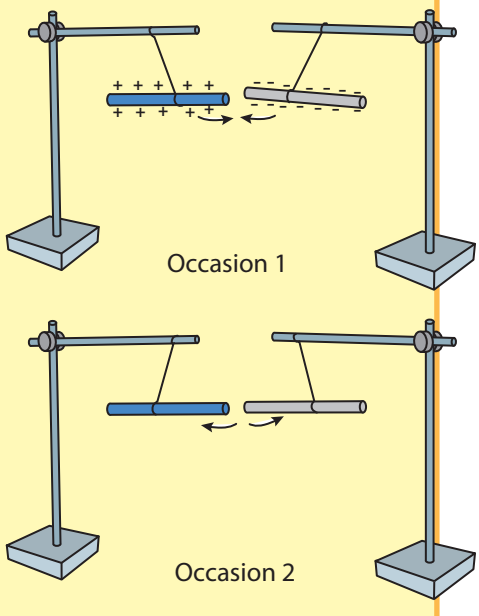
Activity 2.4

Testing the neutralization of charged objects due to the exchange of static electric charges

You will need :- A glass rod and an ebonite rod of equal size, a piece of silk cloth, a piece of woolen cloth, two stands, pieces of thread

Method:-

- Rub the silk cloth against the glass rod and rub the woolen cloth against the ebonite rod. Hang the two rods separately on two stands with pieces of thread.
- Bring two rods close to each other slowly.
- Record the observations.
- Repeat the previous step several times and observe what happens
- Can you observe the same observation as in the previous occasion?
- Discuss the reasons for your observations.



It is observed that positively (+) and negatively (-) charged rods attract at the first time when they are brought close to each other. But they neither attract nor repel when they are brought together repeatedly.
The reason for this is that they are neutralized because of exchange of charges at the first attraction.

Assignment 2.1

Briefly explain how charges are transferred when a piece of woolen cloth is rubbed against an ebonite rod.

2.4 Phenomena associated with static electricity

There are many occasions in our day-to-day life when we experience incidents associated with static electricity. Let us consider some of them.

- **Lightning**

You can recall instances of lightning with or without rain. Sometimes lightning can damage property as well as lives. Lightning occurs because of the generation of static electric charges on clouds. Scientists have not come to an exact decision on how static electricity occurs in clouds. It is believed that static electric charges occur by rubbing particles of ice and water one another. Thus, static electric charges collect on the cloud and the cloud becomes charged. A sudden spark of these static electric charges either inside the cloud or between two clouds or between a cloud and the earth is called **lightning**.

Figure 2.11 ▲ Lightning

For extra knowledge

The scientist Benjamin Franklin was the first to carry out experiments on lightning. Once he sent a kite upto a cloud, while there was lightning and static electric charges in the cloud was discharged to the earth.

Benjamin Franklin

Benjamin Franklin carrying out an experiment on lightning

- **Emitting 'tic' sound when ironing clothes**

Sometimes you may have observed that the hair on your arm is attracted to silk clothes giving a " tic" sound, when you iron them. The reason for this phenomenon is the generation of static electric charges, when iron comes into contact with the silk clothes.

- **Attraction of hair on the arm towards the TV screen**

Sometimes you may have experienced that hair on your arm attracts to the TV screen when TV is switched off. The reason for this, is the accumulated static electric charges on the TV screen.

Some other instances associated with static electricity

Static electricity is used in the electronic circuits in equipment like photocopy machines, radio and television.

2. 5 Capacitors

Activity 2.5

You will need :- Two aluminium sheets approximately of 15cmx15cm, two pieces of wires, a galvanometer, a polythene sheet, two dry cells, cellotape or rubber bands

Method :-

- Connect the two pieces of wire to each aluminium sheet.
- Place the polythene sheet between the aluminium sheets. Then hold aluminium sheets securely without letting them touch each other, using cellotape or rubber bands.
- Connect the free ends of the wires to dry cells and keep for some time.
- Remove the cells, connect the wires to the galvanometer and observe instantly.
- Discuss reasons for your observations.

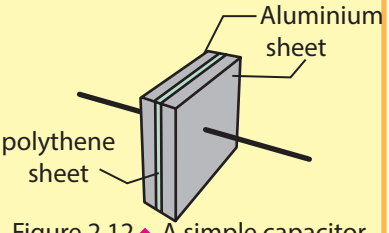


Figure 2.12 A simple capacitor

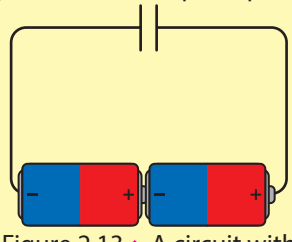


Figure 2.13 A circuit with a simple capacitor and dry cells

What you made in Activity 2.5 is a simple capacitor. When connected to dry cells, static electric charges are stored in the capacitor. When it is connected to the galvanometer, releasing of charges can be observed. The appliance that can store static electric charges is known as the **capacitor**.

Storage of static electric charges in the capacitor is known as charging, and releasing the charges from the capacitor is known as discharging.
When discharging occurs through the galvanometer, it get deflected.
The amount of electric charges that can be stored in a capacitor is measured in Farads (F). Microfarad (µF) is used as a subunit.

Figure 2.14 ▲ A capacitor

There are more efficient types of capacitors are made in various sizes.

Figure 2.15 ▲ Symbol of capacitor

Details like chargeable voltage, positive terminals (+) and negative terminals (-), maximum capacity that can be stored are mentioned on most of them.

Assignment 2.2

Observe the electronic circuits and identify the types of capacitors in radios, televisions and CFLs.

Activity 2.6

Identification of charging and discharging of a capacitor

You will need :- 1000 µF capacitor, three dry cells, a small LED, pieces of wire

Method :-

Connect pieces of wire to the terminals of the capacitor.

Connect the other ends of the wires to dry cells correctly. After few seconds, remove the dry cells and connect the LED correctly to the capacitor and observe (Positive terminal of the dry cell should connect with the positive terminal of LED). Discuss the reasons for your observations.

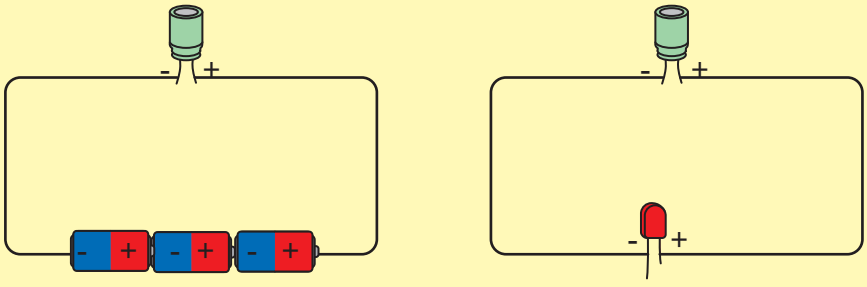


Figure 2.16 ▲ Charging and discharging of a capacitor

The reason for the lighting of the LED is the discharging of charges stored in the capacitor through LED. The capacitor can be recharged by connecting it correctly to the dry cells.

Repeat the above activity several times.

Summary

- Static electricity is generated when one object is rubbed against another.
- There are two types of static electric charges, which are positive (+) and negetive (-)
- When objects are rubbed, negatively (-) charged particles in one object are removed from it and are transferred to the other.
- Repulsions occur between likely charged objects and attractions occur between unlikely charged objects.
- Lightning is a natural phenomenon caused by static electric charges.
- Static electricity is used in televisions, photocopy machines etc.
- Capacitor is an appliance that can be used to store static electric charges.

Exercise

1. Mention two instances where static electric charges are used.

2. A student hung a PVC rod charged by rubbing with a piece of thread. He brought a glass rod rubbed with silk towards the PVC rod. The PVC rod was repelled.

I. What is the reason for the repulsion of the PVC rod ?

II. What is the type of static electric charges occur on the PVC rod ?

3. An electrical circuit prepared by a student is given in Figure 2.16. The indicator of the milliammeter deflects, when terminal X is connected to point A. When X is connected to B, the indicator deflected again. Clarify the above observation. Mention another observation, that can be made in this occasion.

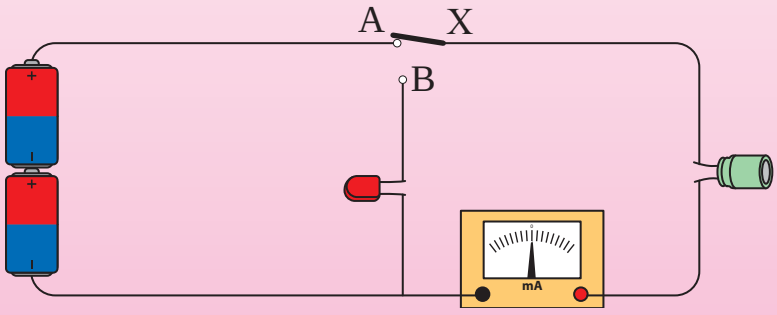


Figure 2.17

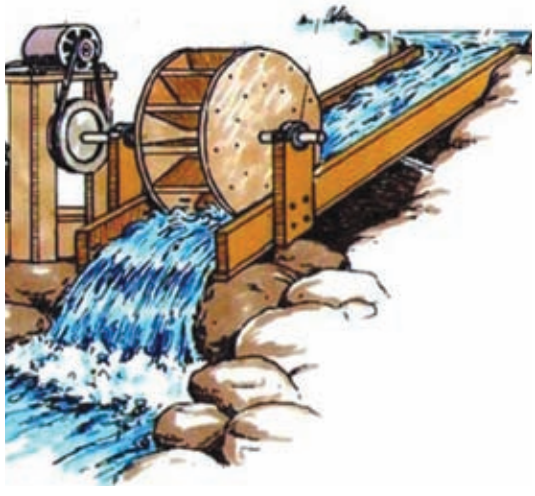
Technical Terms

Static Electricity	-	ස්ථිති විද්‍යුතය	-	நிலைமின்
Charge	-	ආරෝපණය කිරීම	-	மின்னேற்றல்
Discharge	-	විසර්ජනය කිරීම	-	மின்னிறக்கல்
Positive charges	-	ධන ආරෝපණ	-	நேரேற்றம்
Negative charges	-	ඍණ ආරෝපණ	-	மறையேற்றம்
Capacitor	-	ධාරිත්‍රකය	-	கொள்ளளவி

03 Generation of Electricity

From dawn to dusk, we are engaged in various tasks. Various equipment are used to carry out these tasks. As you know, most of these equipment that ease our work function by electricity.

Let us carry out Assignment 3.1 bearing in mind the facts we learnt on electricity in grade six.



Assignment 3.1

Complete Table 3.1, which gives information on electrical equipment used frequently.

Table 3.1

Name of the electrical equipment	Use	Method that electricity is supplied to the equipment
1. Clock	To know the time	Electric cells
2. Rice cooker	To cook rice	
3. Head lamp of bicycle		
4.		
5.		
6.		

3.1 Sources of electricity

Let us pay our attention to the last column of the Table 3.1. There are equipment that supply electricity for our day-to-day electrical needs. Equipment or appliances that generate electricity are called sources of electricity.

Let us carry out Activity 3.1 in the classroom as groups, to study further about the sources of electricity.

Activity 3.1

There are some sources of electricity given in Figure 3.1, which are important to generate electricity in various occasions.

Figure :- 3.1 ▲ Various sources of electricity

- Discuss how electricity is generated in each equipment.
- Classify those sources of electricity, based on the way of generating electricity in them.
- Present the findings of your group to the class

- Electricity is generated by a chemical process in some sources of electricity. There are various chemicals in them.

e.g.:- Dry cells, simple cells, car batteries etc.

Let us do Activity 3.2 to identify the chemicals in a dry cell.

Activity 3.2

Examine the contents in a dry cell

You will need :- Some used dry cells, hacksaw blade, a pair of pliers, a sheet of paper, a pair of gloves

Method :-

- Cut a dry cell longitudinally using the hacksaw blade.
- Observe the longitudinal section of the cell carefully.
- Remove the contents of the dry cell and place them separately on the sheet of paper.
- Note that there are chemicals in the dry cell.
- Carefully dispose the chemicals with the assistance of your teacher.

Figure :- 3.2 Contents of the dry cell

It is clear through the activity, that dry cells contain various chemicals. Thus, all the electric cells and batteries contain chemicals.

- Some sources generate electricity by rotating or moving them.
  e.g.:- bicycle dynamo, electric generators

Equipment that generate electricity can be grouped as given below, according to the way they generate electricity.

1. Electric cells and batteries :- Equipment that generate electricity by a chemical reaction.
2. Dynamo :- Equipment that generate electricity by rotating or moving.
3. Solar cell :- Equipment that generate electricity by using solar energy

Cells and batteries

Generating electricity is an easy task. You can do it yourself even at home. Let us do Activity 3.3 for this.

Activity 3.3

Generating electricity using a lime fruit

You will need :-A lime fruit thoroughly pressed, so as not to come the juice out, a plate of copper, a plate of zinc, connecting wire, musical circuit in greeting cards or a milliammeter

Method :-

- Insert the pieces of copper sheet and zinc sheet into the lime fruit (without contacting each other).
- Connect a wire to each sheet.
- Remove the cell of the musical circuit in the greeting card, and connect the free ends of the above mentioned wires to the circuit or you can connect those wire ends to the milliammeter (copper sheet and zinc sheet should be taken as positive (+) and negative (-) terminals, respectively)
- What can you observe?

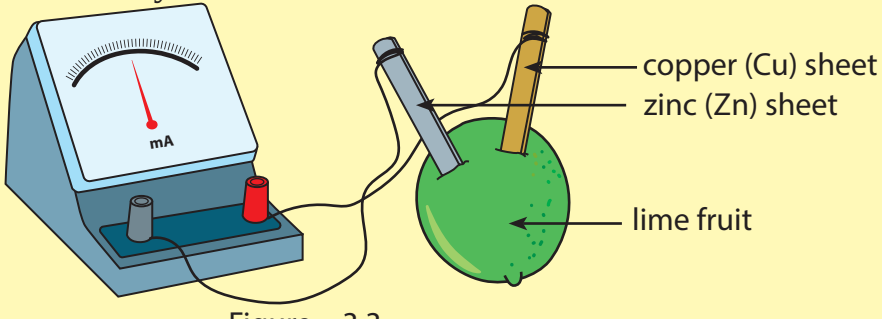


Figure :- 3.3

Now let us consider a more developed set-up.

Activity 3.4

Construction of a simple cell

You will need :- A small beaker (250 ml), copper and zinc sheets (3 cm x 5 cm), a torch bulb, a bulb holder, a small motor, an iron or nichrome wire which is about 30 cm long, a centre-zero ammeter, dilute sulphuric acid, few pieces of connection wire

Method :-

- Clean the copper and zinc sheets thoroughly and connect a piece of wire to one end of each sheet.
- Fill half of the beaker with dilute acid.
- Dip copper and zinc sheets into the acid without letting them come into contact with each other.
- Connect the torch bulb to the free ends of the wires and observe what happens.
- Connect the centre-zero ammeter to one end of the bulb as shown in the Figure 3.4 and observe.
- Connect the small motor instead of the bulb and observe.(Fig. 3.5)
- Wind the nichrome/iron wire round a plastic rod and connect this coil instead of the motor.

(Remove copper and zinc sheets from the acid and brush them before connecting each item to the set-up)

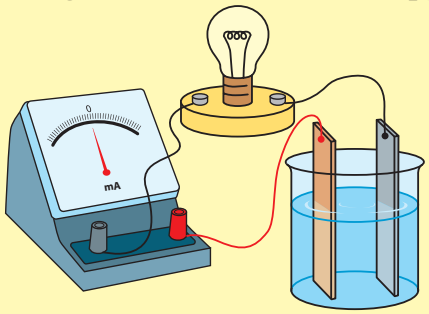


Figure 3.4

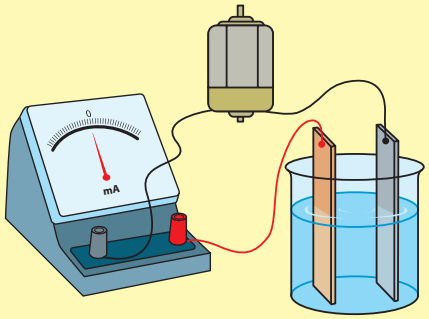


Figure 3.5

- Tabulate your observations in Table 3.2

Table 3.2

Observations when the bulb is connected	Observations when the motor is connected	Observations when the coil is connected	Observations when the ammeter is connected	Any other observations

- The set-up you have constructed is a **simple cell**.
- Illumination of the bulb and the moving of the indicator of the ammeter reveal that electricity is generated in the set-up.

Heating of the coil also indicates the flow of electric current.

Do the Activity again using any other acid instead of dilute sulphuric acid.

For extra knowledge

Centre-zero ammeter

An ammeter or a milliammeter with zero at centre, is used to detect the amount as well as the direction of electric current, flowing through a conducter.

Ampere (A) is the standard unit to measure electric current. The subunit, milliampere (mA) is also used to measure small current.

Activity 3.5

- Place all the simple cells, made by your groups on a table.
- Connect the copper sheet of one cell to the zinc sheet of the other and so on. Finally, you will get a set of cells connected together as shown in the Figure 3.6.
- Connect separately the torch bulb, the motor and a coil to the free wire ends (A and B) of the set of cells.
- Observe what happens and discuss reasons for your observations
- Suggest a name for this set-up.

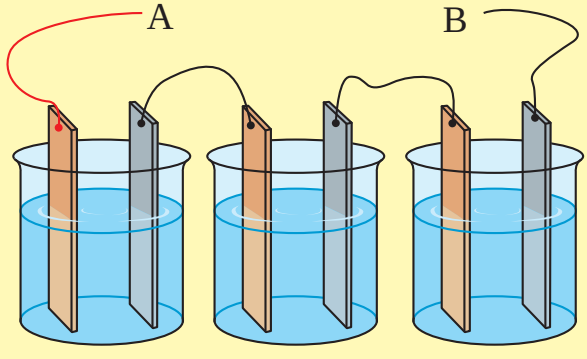


Figure :- 3.6

You will observe that the illumination of the bulb, the speed of the motor and heating effect of the coil are increased when several cells are

connected as above.

Let us make another set-up, using several dry cells.

Activity 3.6

You will need :- Four dry cells, connecting wire, a piece of card board, cellotape or rubber bands.

Method:-

- Connect four dry cells as shown in the Figure 3.7.
- Cellotape or rubber bands can be used to connect wires to dry cells.
- Wrap the set of dry cells round using a piece of cardboard, to make it a handy pack.
- Take out the terminals from the pack.

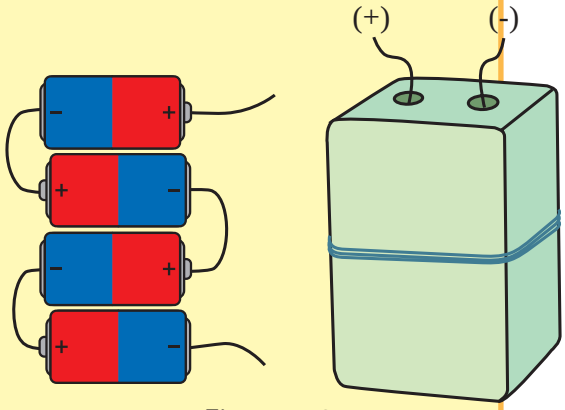


Figure :- 3.7

A set-up constructed using several cells is known as a battery. More electric current can be obtained from a battery, than from a cell.

Now, can you tell the difference between a battery and a cell?

Self-assesment

1. Mention three weaknesses of simple cells.
2. Write an advantage of a battery, compared to a cell
3. Give examples for instances when cells and batteries are used in day-to-day life.

Today, simple cells are not in use because of some weaknesses in them. Some of them are mentioned below.

- Difficulty of using them, because it contains liquids.
- Inability to obtain current for a longer period of time

Today there are cells and batteries which are easy to use and can supply more current.

For extra knowledge

Information on several types of cells and batteries, available in the market are given below.

Table 3.3 - Various types of chemical cells and batteries

Name	Material of make	Instances of uses
Dry cells 	zinc sheet, carbon rod, carbon powder and other chemicals	Electric torches, Radio sets, wall clocks etc.
Alkali cells 	Metals like nickel and cadmium, alkaline components	Telephones, cameras
Button cells 	Substances like lithium and mercury	wrist watches, calculators etc
Lead - acid accumulator (car battery) 	Lead and dilute sulphuric acid	Cars, buses, motor cycles and in rechargeable electric torches

For your special attention

Batteries and cells disposed after use should be directed for recycling, without throwing them into the environment.

Disposed batteries and cells

Terminals of an electric source

You may have heard that the terminals of dry cells should be connected correctly, when they are put to an electric torch or to a toy car.

- There are terminals on an electric source to draw out electricity
- In most electric sources, there are two main terminals.

1. (+) ve terminal
2. (-) ve terminal

Activity 3.7

- Find various types of cells and batteries.
- Observe the data mentioned at their terminals.
- Discuss how their (+) ve and (-) ve terminals are marked.

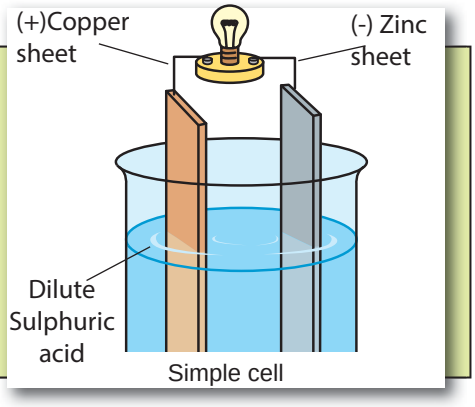
Figure :- 3.8 ▲ How the terminals of various cells and batteries are marked

Now, it may be clear to you that (+) ve and (-) ve terminals are marked in various ways on various types of cells and batteries.

It is very important that terminals of cells and batteries should be connected correctly to the electric appliances.

For extra knowledge

Terminal on the copper sheet is considered as (+) ve and the one on zinc sheet is considered as (-) ve in a simple cell.



Standard symbol to denote a cell

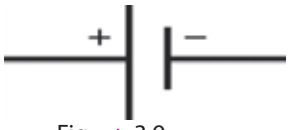


Fig:- ▲ 3.9

Direction of current flowing from an electric source

Let us connect the external wires of an electric source to an electric appliance (bulb).
Electric current flows from source through the wire and the appliance.
Then the appliance starts to work.

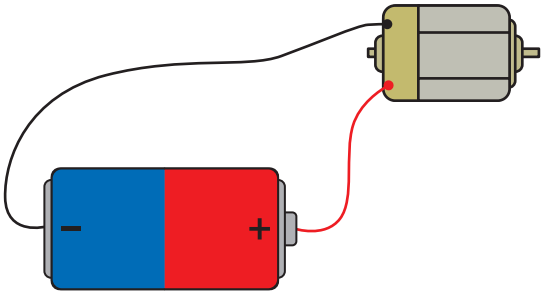


Fig:- ▲ 3.10 Running an electric motor by electric current

Activity 3.8

You will need:- Two dry cells, pieces of wire, an electric motor, a centre-zero milliammeter

Method:-

- Prepare the circuit as shown in Figure 3.11.
- Note the direction in which the motor turns, and the direction in which the indicator of milliammeter deflects.
- Change the terminals connected to the circuit and observe. Note down the observations

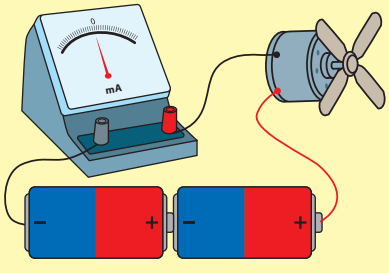


Fig:- ▲ 3.11 (a)

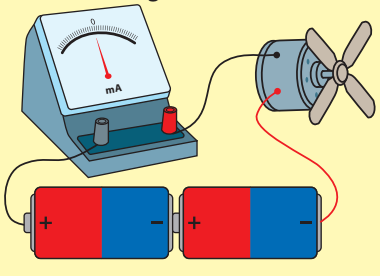


Fig:- ▲ 3.11 (b)

It is accepted that the electric current flows from the (+) ve terminal to (-) ve terminal of the source through the circuit.

When the terminals of cells were changed the direction in which the motor turns and the direction in which the indicater of milliammeter above changed. The reason for this is the change of direction of current. Figure 3.12 shows the direction of the current flowing from a dry cell. Thus, it is clear that, there is a definite direction for the current to flow.

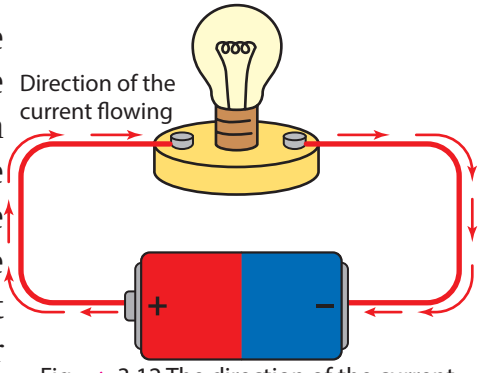


Fig:- ▲ 3.12 The direction of the current flowing from a dry cell

Solar panels

Solar energy reaches the earth as heat and light. Modern man uses solar energy for various purposes. The generation of electricity is one of those uses. Solar panel consists of several solar cells.

The equipment used to generate electricity using sunlight, is known as a solar panel. Have you ever seen wrist watches, calculators and various toys powered with solar cells?

Let us do the Activity 3.9 to study the function of a solar cell.

Activity 3.9

Studying the function of a solar panel

You will need:- A solar panel, a small electric motor, a torch bulb, connection wire

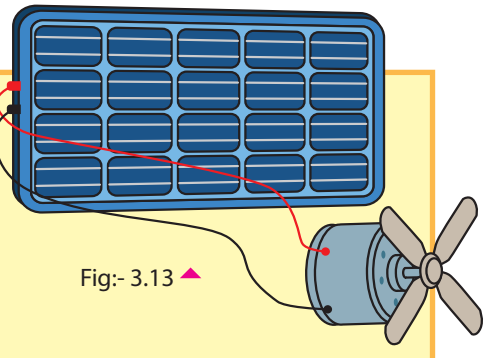


Fig:- 3.13 ▲

Method:-

- Connect the terminals of the electric motor to the terminals of the solar panel.
- Expose the solar panel to light and observe.
- Change the terminals of the solar panel, which are connected to the motor. Observe whether the direction of turning of motor changes.
- Keep the solar panel in the dark and observe the running of motor.
- Repeat the activity using the torch bulb instead of the motor.
- Tabulate the observations.

Table 3.4

Instance	Electrical Motor	Torch bulb
When solar panel is exposed to light		
When solar panel is kept in dark		
When the terminals of solar panels are changed		

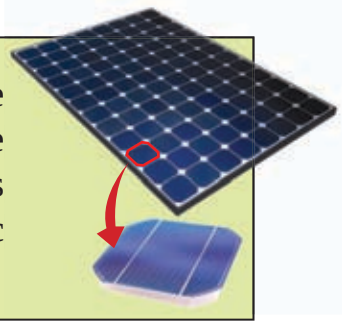
Positive (+) and negative (-) terminals are marked on solar cells also. Therefore, the terminals should be connected correctly, when solar panels are used.

Today solar panels are used for electrical needs in houses as well as in vehicles.

Electricity is generated in a solar panel, only when there is light. Electricity, thus generated should be stored in cells or batteries to use when there is no light.

For extra knowledge

Solar cells are manufactured using elements like silicon. A single solar cell can generate a minute current. Therefore, a large number of solar cells should be connected together to obtain a large electric current. Such a connection is known as a solar panel.



Dynamo

Bicycle dynamo is used in most of the bicycles to light lamps at night. Dynamo is another source used to generate electricity.

Fig:- 3.14 ▲ How a dynamo is fixed to a bicycle

Assignment 3.2

List other instances where dynamo is used to obtain electricity, other than for bicycles.

Other than in bicycles, various types of dynamos are used in fuel-driven electric generators, hydropower stations, thermal power stations and vehicles to generate electricity.

Various types of dynamos

A fuel - driven electric generator

An electric generator in a hydro-power station

Fig:- ▲ 3.15

For extra knowledge

An equipment called dynamo model is used in laboratories to study about dynamos.

Dynamo model in laboratory

Let us consider how electricity is generated in a dynamo.

Activity 3.10

Identifying how electricity is generated in a dynamo

You will need:- One metre long insulated copper wire, a bar magnet, a galvanometer

Method:-

- Wind the insulated copper wire around a cylindrical tube to make a coil.
- Clean both ends of the coil well and connect them to the galvanometer.
- Move one end of the bar magnet into and out of the coil.
- Observe how the indicator of the galvanometer moves.
- Discuss the reason for the observation in the class.

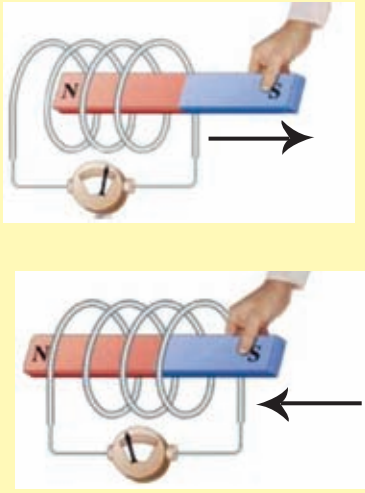


Fig:- 3.16

Generation of electricity in a conductor when magnetic field is cutting with the conductor is known as **electromagnetic induction**.

There is a conducting coil and a permanent magnet in between in a bicycle dynamo. The magnet in the dynamo rotates when the rear wheel of the bicycle rotates. Then, elecricity is generated in the conducting coil.

Thus, it will be clear to you, that electricity is generated in the bicycle dynamo, according to the principle of electromagnetic induction.

Let us carry out Activity 3.11 to study how a bicycle dynamo functions.

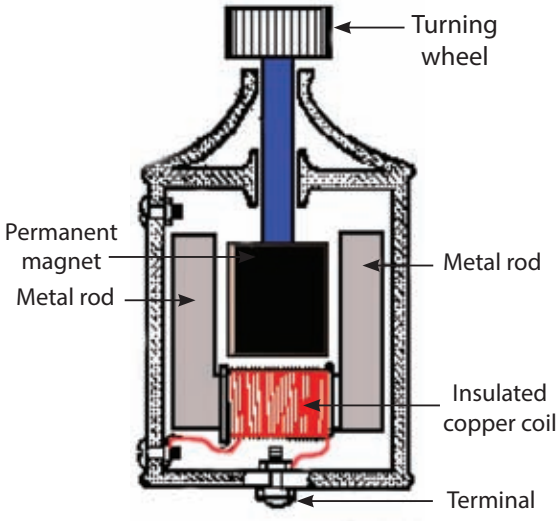


Fig:- 3.17 Inner view of a bicycle dynamo

Activity 3.11

Generation of electricity in a bicycle dynamo

You will need:- A bicycle dynamo or a laboratory dynamo model, a torch bulb, a few pieces of wire

Method:-

- Connect the torch bulb to the terminals of the dynamo.
- Turning the dynamo slowly and faster, observe the brightness of the bulb.
- Discuss the reason for the observation.

Figure 3.18

It will be clear that the amount of electricity generated increased with the rotating speed of the dynamo.

Activity 3.12

Making a simple dynamo

You will need:- About 4 m of insulated copper wire (32 SWG), a bar magnet, a large cork, about 10 pieces of iron wires (each 15 cm long), a galvanometer, cellotape, a bicycle spoke

Method:-

- Bend all the 10 iron wires in U shape, where each arm is about 2 cm long.
- Keep all the bent wires in a single pile to make a bundle of them.
- Wind the insulated copper wire around the bundle of iron rods, as shown in the Figure 3.19, to make a coil.
- Clean both ends of the coil and connect them to the galvanometer
- Place the bar magnet, fitted to the large piece of cork, near the coil and turn it.
- Observe the movement of the indicator of galvanometer.

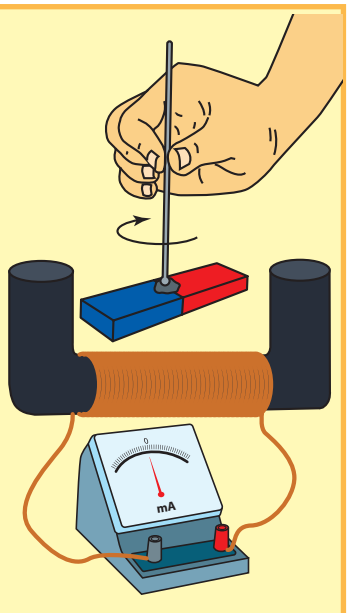


Figure 3.19

What you have made, is a simple dynamo. Let us consider how to develop it further.

Make the following changes in the dynamo you made.

1. Increase the number of turns of the coil. Turn the magnet and note down the amount of movement of the galvanometer indicator.
2. Use a more powerful magnet and repeat the activity.

Can you give reasons for your observations?

The efficiency of the dynamo can be increased by increasing the number of turns of the coil and the power of the magnet.

3.2 Direct current and alternating current

Let us do Activity 3.13 to study what happens when a motor and a dynamo is connected to a motor

Activity 3.13

You will need:- Two dry cells, a small fan, a bicycle dynamo, a few pieces of connecting wire, a small DC motor

Method:-

- Connect the dry cell to the small DC motor as shown in the Figure 3.20. Fit the small fan to the motor.
- Record the observations.
- Connect the dynamo instead of the dry cell.
- Record the observations.

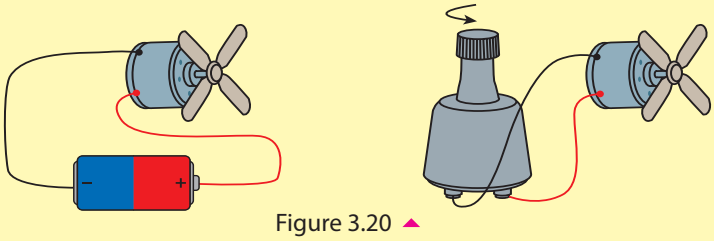


Figure 3.20

You may observe that the small fan rotates when dry cells are connected. When the dynamo is connected the fan vibrates only (does not rotate).

Activity 3.14

You will need :- Two dry cells, two LEDs of different colours, a center-zero milliammeter, a bicycle dynamo, a few pieces of connecting wire

Method:-

- Connect two LEDs and the milliammeter as shown in the figure 3.21.
- Connect two dry cells to X and Y ends as shown in the figure.
- Record the observations.
- Change the terminals of the dry cells.
- Record the new observations.
- Instead of the dry cells, connect the dynamo to X and Y ends.
- Record the observations when the dynamo is rotated.
- Discuss the conclusions that you can arrive, according to the observations.

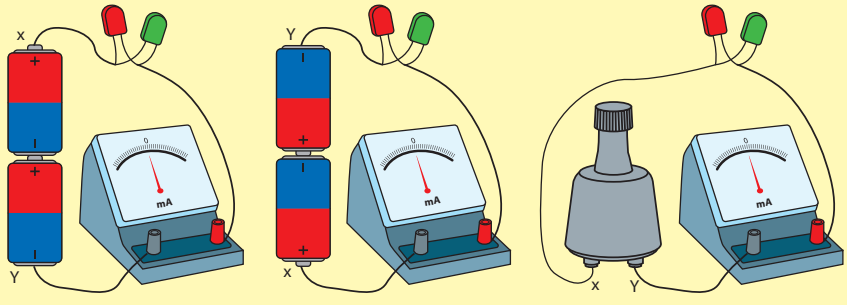


Figure 3.21

Discuss the answers for the following questions, based on the Activity 3.14.

1. Why only one LED illuminates and the indicator of milliammeter deflects only in one direction in all instances when dry cells are connected?
2. Discuss the reason, why two LEDs illuminate alternately and the direction of the movement of milliammeter indicator changes constantly, when dynamo is connected and in rotating?

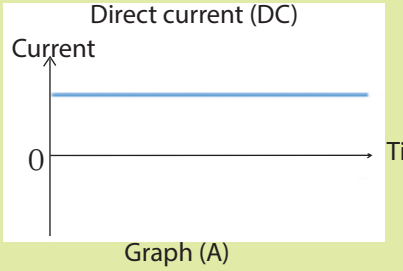
The current flows to the same direction only, when dry cells are connected. But the direction of the current has changed alternately, when dynamo is connected.

- The current that flows to one direction is known as direct current (DC).
- All types of electrical cells and batteries generate direct current (DC).
- The current that changes the direction with time is known as alternating current (AC).
- Most of the dynamos and electric generators generate alternating current (AC).

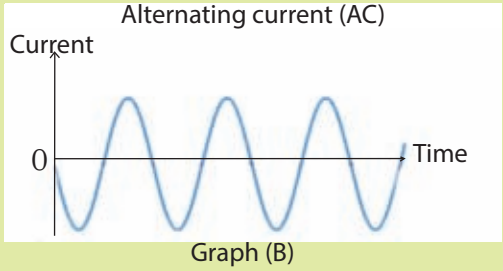
A center-zero ammeter or a galvanometer can be used to identify the direction of the current.

For extra knowledge

The pattern of the graphs, when direct current and alternating current are plotted against time are given below.



Graph (A)



Graph (B)

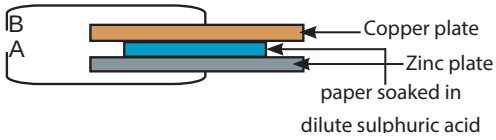
Summary

- Various types of electric sources are used in day-to- day life to obtain electricity. They can be mainly named as chemical cells (batteries), dynamos and solar cells.
- Simple cells, dry cells and alkali cells are examples for chemical cells.
- A battery can be constructed by connecting several cells.

- More current can be drawn from a battery than from a single cell.
- Terminals of an electric cell are named as (+) ve and (-) ve. Current flows from (+) ve terminal to (-) ve terminal.
- Dynamo consists of a magnet and a conducting coil.
- Current is generated in the dynamo according to the principle of electromagnetic induction.
- Various types of dynamos are used to generate electricity in motor vehicles, bicycles, electric generators and hydro-power stations.
- Electricity that flows to the same direction is known as direct current (DC) and that change direction with time is known as alternating current (AC).
- Electric cells generate direct current (DC) while dynamos generate alternating current (AC).
- The toxicity of disposed chemical cells create a number of issues to man and environment.
- Therefore, it should be directed for proper recycling.

Exercise

1. A student constructed the following set-up using copper and zinc sheets of same size. A piece of paper soaked in dilute sulphuric acid was kept between the copper sheet and zinc sheet. Pairs made thus, were placed one over the other as shown in figure above.



(I) Suggest a name for this set-up.

(II) Name the (+) ve and (-) ve terminals of this.

(III) What can be observed when a coil is connected to terminals A and B ? Give reason for the observations.

(IV) Is it a direct current or an alternating current, that can be obtained from the set-up?

(V) Describe briefly, an experiment to confirm your answer for part IV

(VI) Draw the diagram in figure above, using symbols.

2.

(I) Mention three sources of electricity that can be used to overcome present electricity crisis.

(II) Is the main electricity supplied to houses, a direct current or an alternating current?

(III) Complete the following table which is about electric sources.

	Instance	Source	Type of current supplied	
			Direct current	Alternating current
1	Lighting the head lamp of a bicycle	Dynamo		✓
2	Wall clock working on electricity			
3	Generating electricity in a hydropower station			
4	Calculator that works when light falls on it			
5	Starting a car			

Technical Terms

Cell	-	කෝෂය	-	கலம்
Battery	-	බැටරිය	-	பற்றரி
Dynamo	-	ඩයිනමෝව	-	டைனமோ
Electric current	-	විද්‍යුත් ධාරාව	-	மின் ஓட்டம்
Electric generator	-	විදුලි ජනක යන්ත්‍රය	-	மின் பிறப்பாக்கி
Direct current (D.C)	-	සරල ධාරාව	-	நேர் மின்னோட்டம்
Alternating current (A.C)	-	ප්‍රත්‍යාවර්තක ධාරාව	-	ஆடலோட்டம்
Bulb holder	-	බල්බ හෝල්ඩරය	-	மின்குமிழ் தாங்கி
Electromagnetic induction	-	විද්‍යුත් චුම්බක ප්‍රේරණය	-	மின்காந்தத் தூண்டல்
Lines of magnetic force	-	චුම්බක බල රේඛා	-	காந்த விசைக் கோடுகள்
Solar cell	-	සූර්ය කෝෂය	-	சூரியக் கலம்
Coil	-	දඟරය	-	சுருள்

04 Functions of Water

4.1 Water as a solvent

You already know that the sea water has a salty taste. Do you know the reason for this taste? It is because many salts are dissolved in sea water. All living beings need oxygen to breathe. How do fish get oxygen to breath in water? They use oxygen that is dissolved in water.

Water acts as a **solvent**. All the above phenomena are connected with this special function of water.

Figure 4.1 ▲ Sea water

Figure 4.2 ▲ Fish living in water

Let's do Activity 4.1 to find out how water is important as a solvent.

Activity 4.1

Take the following materials given in Table 4.1 in equal quantities and dissolve them separately in 5 ml of water in test tubes. Record your observations in the Table 4.1.

Table 4.1

Substances	Observations
White sugar	Sugar crystals disappear. Solution is colourless.
Glucose	
Kondis crystals	
Surgical spirit	
Coconut oil	
Laundry blue	
Paraffin wax	
Kerosene	
Common salt	
Vinegar	
Turmeric powder	
Camphor balls	
Sodium bicarbonate	

Most of the above substances dissolve in water. But some dissolve a little and some do not dissolve in water. Through this activity we examined how some solids and liquids dissolve in water. Do gases dissolve in water? Let's do Assignment 4.1 to find it out.

Assignment 4.1

- Observe and record the positions of fish in a fish tank in which the water is bubbled with air (oxygen).
- Now observe the positions of fish when the air supply is stopped.

The gases such as oxygen, carbondioxide are soluble in water. Fish use oxygen, dissolved in water for respiration.

Water is considered as a solvent because many things dissolve in water. Similarly, we can separate the things that are dissolved in water. Therefore, this solvent property of water helps us in day-to-day life as well as in industrial activities.

Imagine how "water as a solvent" is important in your day-to-day activities.

Let's do Assignment 4.2 to find out how water is important as a solvent.

Assignment 4.2

- List the difficulties that you face when there is no water supply in your kitchen.
- Record the methods of supplying nutrients for hydroponic cultivation

It is clear that this special property of water; acting as a solvent, is important not only for us but also for the aquatic organism to live and for the plants to grow.

Following are some instances where water is important as a solvent.

- To make drinks by dissolving sugar, colourings and flavours in water
- To dissolve salt and flavours to make food tasty
- To dissolve concentrated acids in water to prepare battery acid, artificial vinegar etc
- To dissolve medicine in water
- To produce vaccines and saline for purposes of health
- Aquatic animals use oxygen that is dissolved in water to breathe
- To remove dirts on clothes and body
- Use coloured water for decorating purposes.

Assignment 4.3

Prepare another list of instances where water is used as a solvent.

For extra knowledge

- Dilute sulphuric acid is prepared by dissolving concentrated sulphuric acid in water. Battery acid contains dilute sulphuric acid.
- Artificial vinegar is prepared by dissolving acetic acid in water.
- A type of saline solution is prepared by diluting sodium chloride solution to a standard concentration.

Assignment 4.4

- Prepare some coloured solutions by dissolving different colours of dye in water.
- Pour them into glass containers with different shapes.
- Prepare a list of instances where such colourful solutions are used in our day-to-day life.

Figure 4.3

Assignment 4.5

- Collect some labels of different kinds of soft drinks.
- List out the substances that are dissolved in water to prepare them.

Uses of separating materials dissolved in water

Activity 4.2

You will need :-

A common salt solution, spirit lamp, a candle, a lid of a tin

Method:-

Put some common salt solution on to the lid and heat it as shown in Figure 4.4.

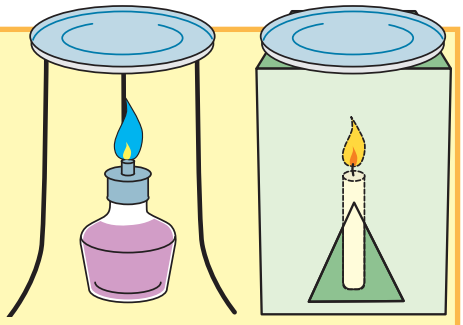


Figure 4.4

Use a tripod and a wine spirit lamp or an empty box of milk powder with a ventilator at the side and a lighted candle to heat it. Record your observations.

You will observe salt remains on the lid as a white powder.

Different types of salts of minerals are dissolved in water, when rivers, streams and water ways are flowing towards the sea. When this happen for long period of time, sea water become salty. Thus, sodium chloride is the mineral salt that dissolved most in the sea water. Salt is produced from sea water, by evaporating water exposing to solar heat.

Figure 4.5 ▲ A saltern

Figure 4.6 ▲ A sugar cane tree

The juice in sugar cane contains sucrose dissolved in water. Sugar is produced by removing the water in sugar cane.

A sugary solution can be extracted from coconut flower which is known as sweet toddy. Treacle can be produced by removing some amount of water from this sweet toddy. If water is totally removed from sweet toddy jaggery can be produced. Jaggery and treacle can also be produced from palmyra and kitul trees.

Figure 4.7 ▲ A coconut tree used to get sweet toddy

Assignment 4.6

Design a poster to illustrate the use of water as a solvent.

4.2 Water as a coolant

Why do buffaloes wallow in water on hot days?

Why do we feel cool when we wash our hands, legs and face with water at a time of sweating?

How do you explain the cooling ability of water?

Water can bear a large amount of heat. It absorbs the heat and reduces the heat of objects. Because of this feature water is considered as a good **coolant**.

Let's do Activity 4.3 to observe the coolant property of water.

Activity 4.3

You will need :- Two beakers of similar size, cotton, two thermometers

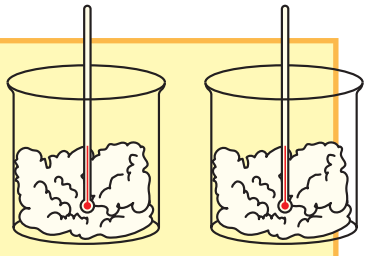


Figure 4.8

- Get two beakers of similar size. Put same amount of cotton wool into both beakers.
- As shown in the Figure 4.8 keep two thermometers inside the beakers and get the readings.
- Put a little amount of water on to cotton wool in one beaker and keep for a few minutes.
- Get the readings of the thermometers and compare the readings.

You can observe that the temperature of the thermometer in wet cotton wool has been reduced.

Uses of coolant property of water

Figure 4.9

- The energy generated by burning fuel helps a vehicle to run. At this process of burning fuel the engine of the vehicle gets heated a lot and stops functioning. The heat of the engine is absorbed by water or by a coolant in the radiator and prevents the engine from faltering due to overheat.
- Water is used as a coolant in factories to avoid overheating of machines while operating.

Assignment 4.7

List out some other instances where water is used as a coolant.

4.3 Water as a medium of life

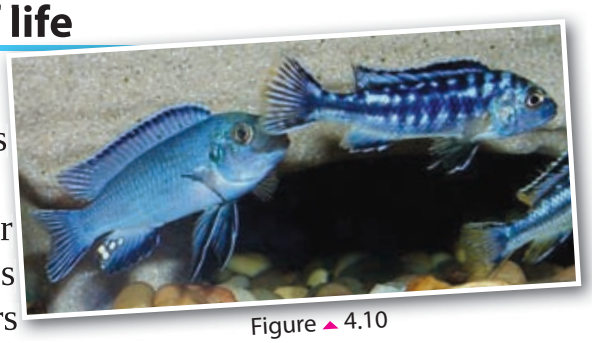


Figure ▲ 4.10

Can a fish live out of water?
It will die in a few minutes, if it is taken out of water. Why is it?
Fish use oxygen dissolved in water for respiration. When water pass through their gills, oxygen enters into blood vessels.
Fish cannot obtain oxygen if there is no water. Lots of fish die during a drought period because they cannot get oxygen without water.

- Many aquatic organisms use water as their living environment (Habitat)

e.g.:- Fish - Tilapia, Tuna
Amphibian - Toad
Reptile - Watersnake, Turtle
Mammals - Dolphin, Whale

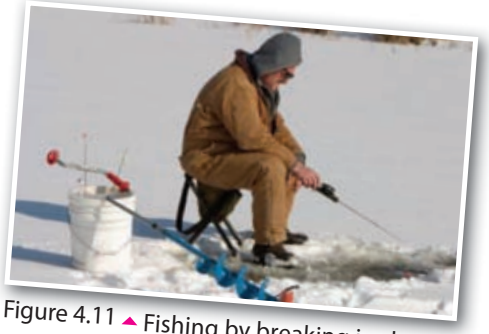


Figure 4.11 ▲ Fishing by breaking ice layers

During the Winter season, although the surface water gets frozen, water remains in liquid state under the ice layer, so that fish can survive without any harm during the Winter.

Some people break these ice layers and do fishing during this season.

Water acts as a medium for biological activities taking place in all living beings which live in water and land.

- Digestion of food and producing energy through reaction of glucose and oxygen and many other chemical reactions take place in a medium of water.
- The nutrients absorbed to the body after the digestion of food, are transported to the cells by dissolving in blood. So this process too takes place in a medium of water.
- Vitamins, minerals, medicine are transported through our body by using a medium of water in blood.

- Also the excretory materials such as urea produced in our cells, are transported to the excretory organs with the help of blood.

For extra knowledge

- The excess protein taken to our body is decomposed to urea in the liver.
- This urea is mainly excreted as urine which has a watery medium. Little amount of urea is also excreted as sweat.

Therefore, it is obvious that water plays a major role in the existence of life.

Assignment 4.8

Prepare a wall paper about the living beings that use water as a medium of life. Divide them as plants, animals and micro organisms.

Assignment 4.9

Design a poster to illustrate the use of water as a medium.

Summary

- The main functions of water are as a solvent, as a coolant and as a medium of life.
- The solvent property of water is helpful to dissolve things in water and to separate things from water.
- Heat is generated in our body during the biological processes.
- Heat is also generated in machines when they are operated. This heat can be removed by using water because water act as a coolant.
- Water is the medium of life for aquatic organisms.
- Water is an essential medium to maintain the biological processes in all living beings.

Exercise

01) Select the correct answers for the given questions.

I. What is the most soluble substance in water?

1. Laundry blue 2. Table salt 3. Sand 4. Coconut oil

II. Which property of water is used to cool the engine of a vehicle?

1. As a solvent 2. As a medium
3. As a coolant 4. As an insulator

02) Fill in the blanks using the correct answer.

I. Sea water has become salty due to dissolving ..... in it.

II. ..... can be produced by evaporating sea water.

III. ..... property of water, makes it easy to absorb minerals by plants.

IV. When we take an ice cube into our hand we feel cold. The reason for this is flowing of heat from ..... to .....

V. ..... acts as the medium for biological processes that take place in the human body.

Technical Terms

Solvent	-	ද්‍රාවකය	-	கரைப்பான்
Solution	-	ද්‍රාවණය	-	கரைசல்
Solute	-	ද්‍රාව්‍යය	-	கரையம்
Coolant	-	සිසිලනකාරකය	-	குளிராக்கி
Medium	-	මාධ්‍යය	-	ஊடகம்

05 Acids and Bases

5.1 Identification of acids and bases

Did you ever think why fruits have different tastes?

Figure ▲ 5.1 Different types of fruits

Fruits have different tastes because they contain different chemical compounds. Fruits such as oranges, pineapples, lemon, tamarind and flavours such as vinegar, lime, tomatoes are sour in taste. The reason for this sour taste is containing **acids** in them.

Sodium bicarbonate is used as a treatment for bee sting. Milk of magnesia tablets are taken for gastritis. Lime is added to reduce the acidity of soil.
Sodium bicarbonate, milk of magnesia and lime water contain chemical compounds called **bases**. Bases can be used to reduce the problems occurring due to acids. Bases have a soapy nature.

Water, alcohol, salt solution, kerosene do not show the properties of acids or bases. They are called **neutral substances**.

5.2 Acids and bases available in school laboratory and home

The substances, we use at home as well as the chemicals we use in the laboratory can be classified as acids, bases and neutral substances according to their properties.

Let's do Activity 5.1 to identify substances as acids, bases and neutral substances.

Activity 5.1

You will need :-

Some shoe flowers, lime juice, vinegar, soapy water, lime water, ash dissolved water, common salt solution, water

Figure 5.2

Method :-

- Boil the shoe flowers and prepare a solution.
- Put equal volumes (2 ml) of solutions given in the table below into separate test tubes.
- Put two drops of shoe flower boiled water into each test tube and shake well.
- Record your observations in the Table 5.1.

Table 5.1

Solution	Acid/base/neutral	Colour given with shoe flower solution
Lime juice	acid	
Vinegar	acid	
Soap water	base	
Lime water	base	
Ash mixed water	base	
Common salt solution	neutral	
Water	neutral	

You will observe that acids give one colour while bases give another colour with shoe flower solution.

The solutions/things that give different colours with acids and bases are known as **indicators**.

Litmus is such an indicator that can be found in the laboratory. There are two types of litmus. They are red and blue.

Let's do Activity 5.2 to identify acids, bases and neutral substances using litmus.

Figure ▲ 5.3 Red and blue litmus

Activity 5.2

Use red litmus and blue litmus instead of shoe flower solution with the solutions you used in activity 5.1. Record your observations in a table. Compare your observations with the following colours.

Type of litmus	Colour with acids	Colour with bases	Colour with neutral substances
Red litmus			
Blue litmus			

In the presence of,

- acids, blue litmus turns into red and red litmus does not change the colour.
- bases, red litmus turns into blue and blue litmus does not change the colour.
- neutral substances, both blue and red litmus do not change the colour.

You can prepare indicators by using some materials in the natural environment. Given below are some of them.

- Shoe flower boiled water
- Extraction of "Girithilla" / "Seendukodi" flowers
- Extraction of "Nil katarolu" / "Nela Kakkattan" flowers
- Arecanut boiled water
- Turmeric boiled water
- Red cabbage boiled water

Let's do Activity 5.3 by using some of the above indicators.

Activity 5.3

Prepare some of the above indicators. Add some drops of those indicators to the solutions given in the following table and record the colour you observe.

Table 5.2

Liquids/ Solutions	Indicators prepared			
	Turmeric boiled water			
Lime juice				
Vinegar				
Lime water				
Soap water				
Shampoo				
Soda water				
Colourless soft drink				
Common salt solution				
Sugar solution				
Glucose solution				
Kerosene				

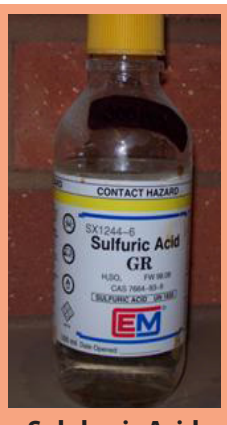
Classify the liquids/solutions as acids, bases and neutral substances depending on the colour change with the indicators.

Let's do Assignment 5.1 to identify the acidic and basic substances in the school laboratory.

Assignment 5.1

With the help of your teacher observe the labels of the containers with acids and bases. Collect the information given in the labels. Then prepare a list of acids and bases that can be seen in the laboratory. Do not touch any acid or base. If touched accidently wash yourself well with cool water.

Some acids that are used frequently in the laboratory are given below.



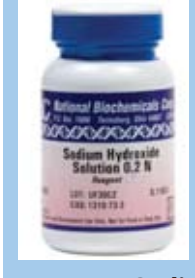
Sulphuric Acid

Nitric Acid

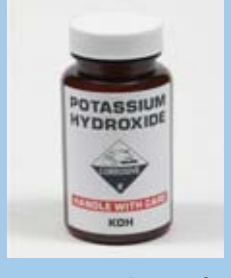
Hydrochloric Acid

Figure 5.4 ▲ Some acids

Some strong bases that are used frequently in the laboratory are given below.



Sodium Hydroxide



Potassium Hydroxide

Figure 5.5 ▲ Some bases

Apart from litmus the following indicators can be used to identify acidic, basic and neutral substances in the laboratory.

1. pH papers **2.** Phenolphthalein indicator **3.** Methyl orange indicator

Table 5.3 shows the colour changes of the indicators with acids and bases.

Table 5.3 Colour changes of the indicators with acids and bases

Indicator	Nature of the indicator	Colour with acidic substances	Colour with basic substances
Blue litmus	a kind of blue coloured paper strips	red	blue (no colour change)
Red litmus	a kind of red coloured paper strips	red (no colour change)	blue
pH papers	A kind of yellow coloured paper strips	Red, orange, yellow	Dark green, blue, violet
Phenolphthalein	A kind of white powder. This powder is dissolved in ethanol or surgical spirit. The solution is colourless.	colourless	pink
Methyl orange	A kind of yellow powder. This powder is dissolved in water. The solution is yellow in colour.	red	yellow

- **pH papers**

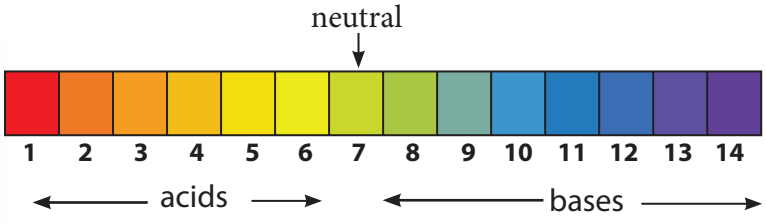


Figure 5.6

- Phenolphthalein powder

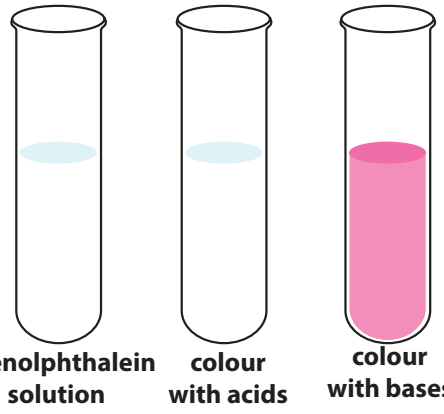


Figure 5.7

- Methyl orange powder

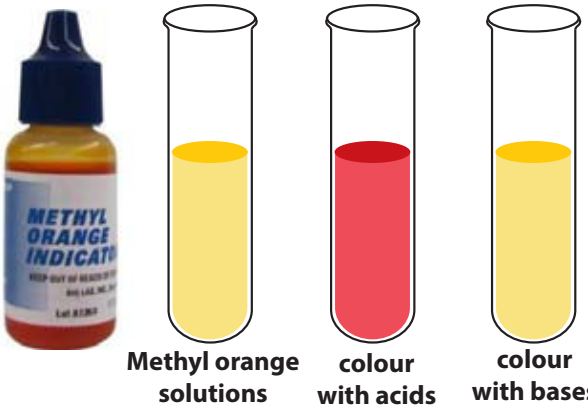


Figure 5.8

Let's do Activity 5.4 to identify acidic, basic and neutral substances using pH papers.

Activity 5.4

You will need :-

pH papers, solutions used in Activity 5.3.

Method: -

- Dip the pH papers in the solutions and compare the colours with the given code. Arrange the solutions according to the consequent numbers.

- Acidic substances show in the range of 1-6 colours
- Bases or basic substances show in the range of 8-14 colours
- Neutral substances show the colour 7 of the pH paper

Activity 5.5

You will need :-

Dilute hydrochloric acid, dilute sulphuric acid, dilute sodium chloride solution, dilute calcium hydroxide solution

Method :-

Use pH papers, red litmus, blue litmus, phenolphthalein, methyl orange indicators with above solutions to identify the acids and bases. Get the help of your teacher.

Summary

- We use acidic, basic and neutral substances regularly in our day-to-day activities and also in the laboratory.
- Identifying acidic and basic substances is useful in day-to-day activities.
- Different kinds of indicators are used to identify acidic, basic and neutral substances.
- Lemon, vinegar, tamarind, gamboge are some of the acids that can be found at home.
- Sulphuric acid, hydrochloric acid, acetic acid are some of the acids that can be found in the laboratory.
- Soap, lime water, shampoo, ash are some of the basic substances that we use at home.
- Sodium hydroxide, calcium hydroxide are some of the basic substances that can be found in the laboratory.
- Sugar, common salt, kerosene, glucose are some neutral substances that are used at home.

Exercise

- Select the correct answer for the questions given below.

(01) Which answer contains only acidic substances?

1) Lemon, soap,common salt 2) Lime water, common salt, vinegar

3) Vinegar, lemon, tamarind 4) Common salt, vinegar, lemon

(02) ……..turns red litmus into blue.

1) Common salt solution 2) Lime water 3) Orange juice 4) Water

(03) An indicator that is used in the laboratory is

1) Sodium hydroxide 2) Methyl orange

3) Sulphuric acid 4) Calcium hydroxide

(04) What is the colour of the pH papers found in the laboratory?

1) Yellow 2) Blue 3) Red 4) Violet

(05) What is the solution that turns phenolphthalein into pink?

1) Dilute Sulphuric acid 2) Sodium hydroxide

3) Dilute Nitric acid 4) Soda water

- Three containers named as A, B and C contain an acidic, a basic and a neutral solution. The following table shows the resulting colours of red litmus and blue litmus when they are dipped in these solutions.

	Solution A	Solution B	Solution C
Blue litmus	blue	blue	red
Red litmus	red	blue	red

1) Which solution shows acidic properties?

2) Which solution shows basic properties?

3) Which solution shows neutral properties?

4) If one vessel contained water, what is the letter given for the solution?

Technical Terms

Acid	-	අම්ලය	-	அமிலம்
Base	-	භස්මය	-	காரம்
Neutral substance	-	උදාසීන ද්‍රව්‍ය	-	நடு நிலைப் பதார்த்தங்கள்
Indicator	-	දර්ශකය	-	காட்டி
Medium	-	මාධ්‍යය	-	ஊடகம்

06 Animal Diversity

6.1 Vertebrates and invertebrates

Animal world consists of millions of different animals with a wide diversity.

Assignment 6.1

- Observe the school garden and identify 10 different animals and name them.
- Group them based on different features you observed in them.
- Compare the way you grouped with the way your friends in the class did.

You and your friends may have grouped animals based on different criteria. Mode of locomotion, body shape, body colour, size of the body and mode of nutrition are some of the criteria you can use to group animals. Therefore you will learn that there is a vast diversification in animals. Human is also a member of this diversified animal world.


Human


Turtle


Snail


Worm


Crow


Crab


Elephant


Frog

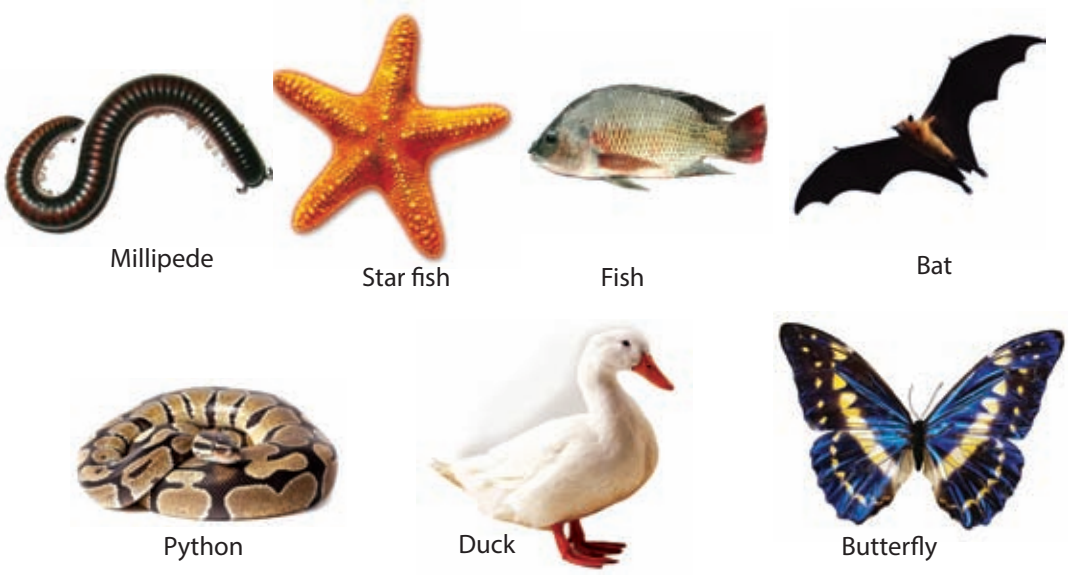


Figure 6.1 ▲ Several species of animals

As there are many different species of animals living in the animal world, they are grouped in order to make it easy for naming, identification and to study about them. Recall how you grouped the animals using different criteria.

Considering the mode of locomotion, bat, butterfly and crow can be grouped into one group as 'flying' animals, but these animals display a huge diversity with regard to other features. Therefore, a scientific way of classification is essential to group animals. Observe a human skeleton in your school laboratory.

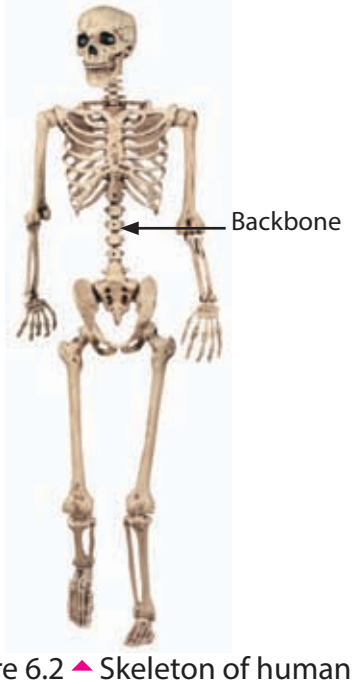


Figure 6.2 ▲ Skeleton of human

The central line of bones is known as **backbone**. Many animals including human being have a backbone. Observe the skeletons given in Figure 6.3 and identify the nature of backbone of the animals.

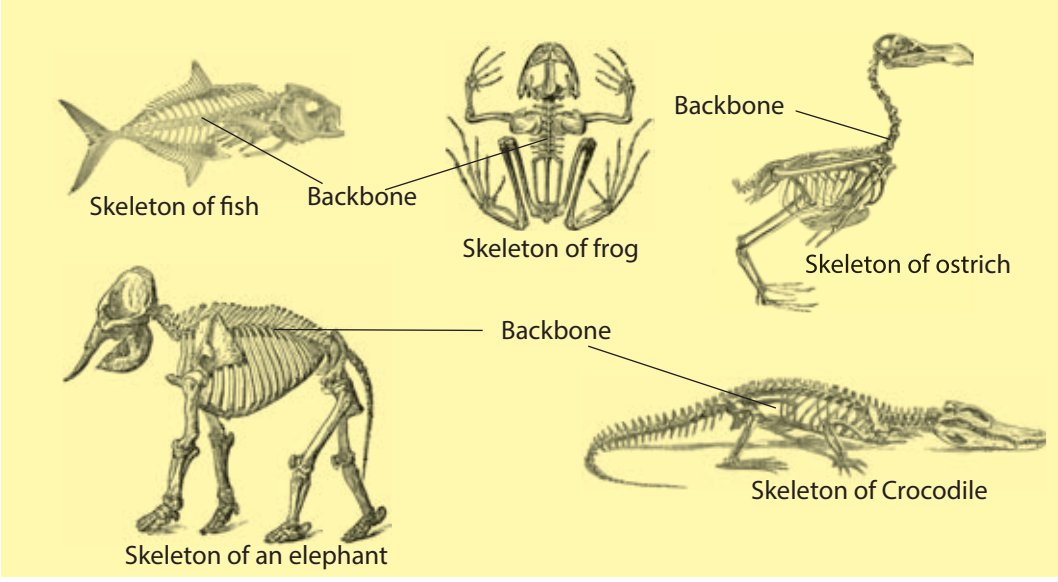


Figure 6.3 ▲ Skeletons of vertebrates

Some animals do not have a backbone. The Figure 6.4 shows several species of animals, without a backbone.

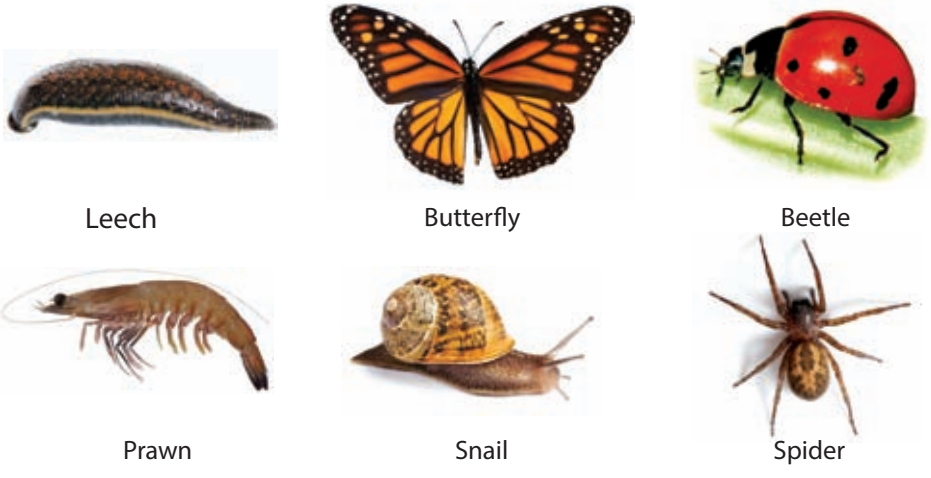


Figure 6.4 ▲ Several species of invertebrates

Animals can be divided into two groups as animals with a backbone and animals without a backbone.

Animals with a backbone are called **vertebrates**; and animals without a backbone are called **invertebrates**.

Animals
- Animals with backbone (vertebrates)
- Animals without backbone (invertebrates)

Engage in Assignment 6.2 to identify vertebrates and invertebrates in our environment

Assignment 6.2

The picture given below shows some species of animals living in sea shore. Group them into vertebrates and invertebrates.

Figure 6.5 ▲ Animals observed in the sea shore

1. Sea gull
2. Turtle
3. A species of fish
4. Prawn
5. Hermit crab
6. Bivalve
7. Star fish
8. Sea anemone
9. Snail
10. Worm

Compare the way you grouped them with the table given below.

Table 6.1

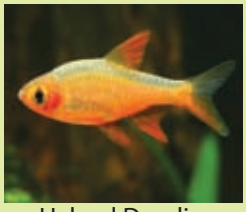
Vertebrates	Invertebrates
Sea gull Fish species Turtle	Hermit crab Starfish Bivalve Snail Prawn Worm Sea anemone

For extra knowledge

Vertebrates can be further divided into following groups;

- Fish
- Amphibians
- Reptiles
- Birds
- Mammals

A - Fish


Halmal Dandiya


'Bulath hapaya'


Shark

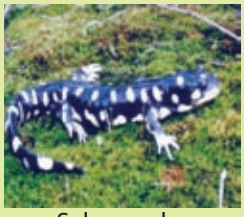

Tuna

B - Amphibians


Toad


Ichthyophis


Frog


Salamander

C - Reptiles


Turtle


Lizard


Snake


Crocodile

D- Birds


Hawk


Kingfisher

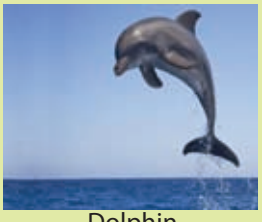

Magpie


Oriole

E- Mammals


Squirrel


Bat


Dolphin


Deer

Engage in Activity 6.1 and classify the vertebrates mentioned in it.

Activity 6.1

A picture of a forest drawn by a student is given in Figure 6.6. Identify organisms in the picture.

Figure 6.6 ▲ Several species of animals

Classify above animals as vertebrates and invertebrates.

Vertebrates	Invertebrates
.....	.....
.....	.....
.....	.....

6.2 Adaptations of organisms to environment

Organisms live in various environments like water, land, atmosphere, on other organisms and also inside other organisms. Other than that there are organisms who live in snow, deep sea, in deserts as well as hot water springs. They have conquered such difficult environmental conditions due to their ability to adapt for these environments.

The ability of organisms adapt to their environment is called **adaptation**. These adaptations are useful for them to fulfil their needs (e.g.:- food, shelter, protection). Thereby the organisms ensure their existence in the environment.

Usefulness of colour for the existence of animals

Most of the time colour of the animal blend with their living environment; thereby an animal cannot easily be identified by the predators and they will be protected.

Activity 6.2

You will need :- 100 small pieces of ekels/tooth picks, colours (red, green, white, brown)

Method :-

- Colour the pieces of ekel/tooth picks (25 in one colour)
- Spread the wrapped pieces of ekel in a lawn randomly.
- Appoint 4 students to pick up the pieces of ekel.

- What colour ekels were picked up at first? What colour was completed picking up at last?
- Next spread the above ekel on a gravel floor. Which pieces of ekel will be collected at last if they were asked to pick up as before.
- Do this activity in different environments in the above manner.

It is obvious that you might have collected the green colour ekel sticks last. Due to the similar colour of the grass and the green colour wrapped ekels sticks make difficult to separate them when picking.

Then, disperse all these ekel sticks on a gravel floor. When you make the students to pick them once again which colour of ekel sticks would be picked by them last.

Some of the animals who show camouflage are shown in the Figure 6.7.


Leaf insect


Moth


Grass hopper


Butterfly


Caterpillar


Whip snake

Figure 6.7 ▲ Camouflage among different animals

Assignment 6.3

Complete the following table using Figure 6.7.

Table 6.2

Name of the animal	Environment	Colour of the Environment	Colour of the body
Grasshopper	Plant leaves	.....	.....
Leaf insect	Leaves of guava	.....	.....
Butterfly	Flowers	.....	.....
Caterpillar	Plant leaves	.....	.....
Whip snake	Twigs of plants	.....	.....
Moth	Stem	.....	.....

The colour of these animals are properly blended with their environment. Therefore, the predators cannot identify them at once.

Presence of the same colour in the environment and the body of many animals will help them to protect themselves from the predators.

The difficulty to identify animals separately from their surroundings due to blending of body colour to particular environments is called **camouflage**.

Animals get the following benefits due to camouflage.

Figure 6.8 Leopard seeking for prey

As the skin colour of animals blend with the environment they live in, predators find difficult to catch them on sight at once. **Most of the animals do not become victims to predators due to their adaptations to the environment.**

Skin colour of the animals helped them not only to protect themselves from predators but also to find prey for them.

e.g.:- The leopard with spotted skin and dusty colour helps them to catch preys without been seen.

For extra knowledge

Biston betularia, a moth species lived in Manchester town in England. They were in two colours as white and black. The black coloured moths easily became the preys of predators as they were clearly visible. After industrial revolution the environment became grey and the black moths were safe due to their black colour.

Some kinds of lizards change their colour according to the environment they live.

Figure 6.9 ▲ Species of Lizard showing camouflage

Assignment 6.4

Find the animals who show camouflage. Prepare an album with a collection of photographs of them.

How shape helps the existence of animals

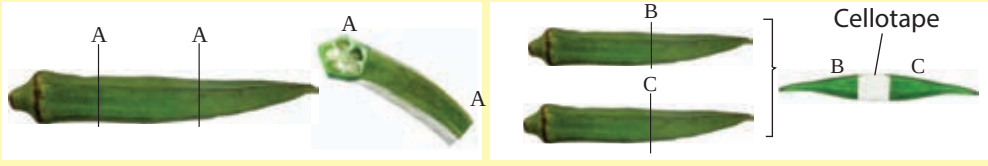
It is important to change the body colour of animals for their own protection. Similarly the body shape of them is also very important for locomotion. Let us engage in Activity 6.3 to examine that.

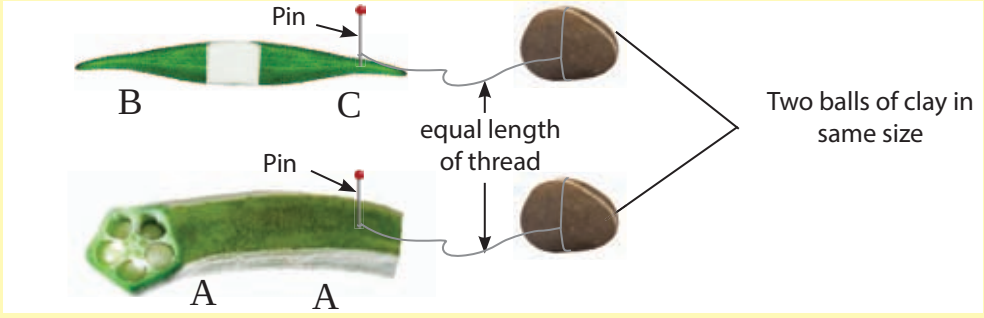
Activity 6.3

Showing how the body shape helps in locomotion

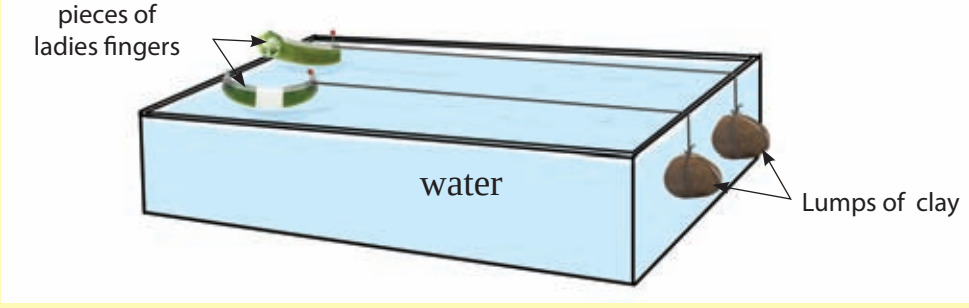
You will need :- Some pods of ladies fingers, two pieces of thread about 50 cm, two small balls of clay, pins

Method :- Cut the ladies fingers according to the illustrated picture.





Make the two forms of prepared ladies fingers to move on a tray filled with water.



Observe whether both the structures move with a similar speed or their speed differs.

Both the ends of the A-A form of ladies fingers get a circular shape whereas the B-C form gets pointed shape (Tapering ends).

When the two balls of clay lower down two forms of ladies fingers float along the tray.

You will be able to see that streamlined shaped (B-C) form reaches the end of the tray faster than the other form (A-A) of ladies fingers.


streamlined shape


Fish


Bird

Figure 6.10 ▲ Streamlined body shape of fish and birds

Name few streamlined shape animals.

You will observe several kinds of birds and fish possess streamlined shape.

The body shape of birds and aquatic animals helps to overcome the difficulties they have in their environment.

The body shape of birds and fish is mainly of streamlined shape because they need to have efficiency in their locomotion.

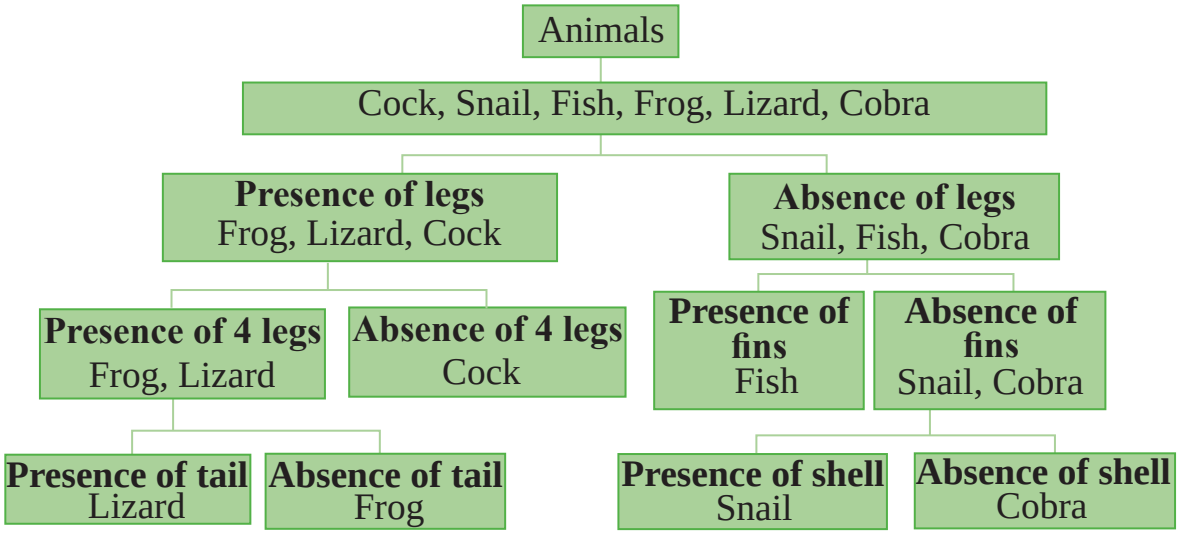
6.3 Use of dichotomous key for classification of organisms

Assume that you remember how several plant leaves were classified using the dichotomous key. Dichotomous key is used to classify living organisms based on the presence and the absence of characteristics. It is more appropriate if the characteristics chosen for this purpose is easily observable.

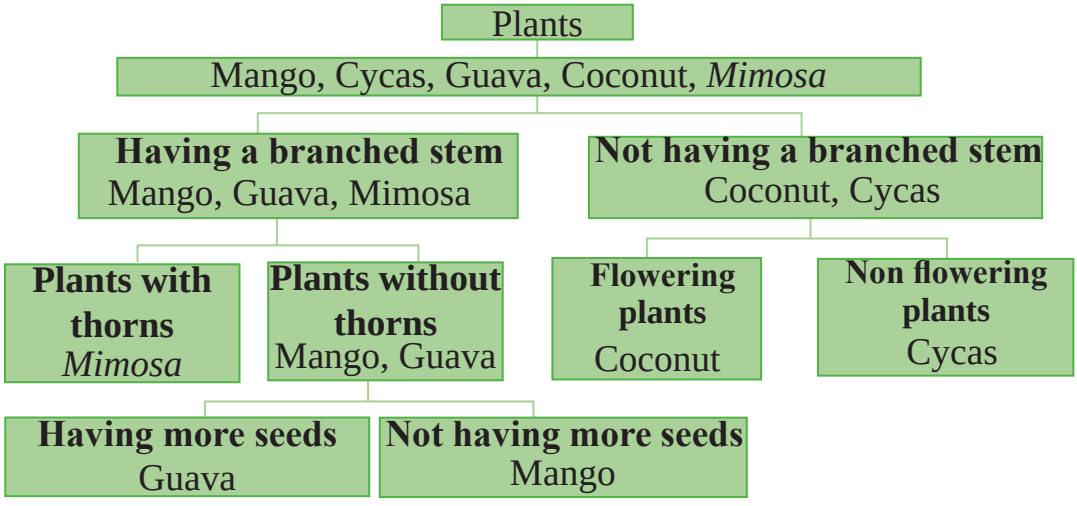
Features of dichotomous key

- Select a feature that could be differentiated easily.
- Consider one feature at a time and separate that feature as present or absent.
- Finally, separate the items so that only one item will remain at the end.

Some examples of categorisation using dichotomous key



Similarly, a dichotomous key can be prepared for the classification of plants.



You might have realised that plants and animals can be identified separately by classifying them with the use of a dichotomous key.

Assignment 6.5

- Observe the birds in your school premises/in your garden.
- Based on the various features of those observed birds, prepare a dichotomous key.

Summary

- Animals can be grouped into two as animals with a backbone and animals without a backbone.
- Animals with a backbone are named as vertebrates and animals without a backbone are named as invertebrates.
- Similarities and differences can be seen among vertebrates.
- Various changes occur in organisms to suit and survive in their habitat is called adaptation.
- Organisms are well-adapted to live in their environment with body colour and shape.
- Dichotomous keys are used to classify organisms mainly based on their external features.

Exercise

1. Choose the correct answer.

i). Select the group of animals consists of only vertebrates.

a. Bull, Snail, Crow

b. Butterfly, Sparrow, Bat

c. Gecko, Iguana, Crocodile

d. Crab, Prawn, Shark

ii). Select the invertebrate;

a. Toad b. Sea horse c. Prawn d. Rat snake

2. A list of animals is given below.

Iguana, Mosquito, Squirrel, Fish (Snake head), Whale, Crow, Bat, Crab, Bull, Butterfly, Bee, Scorpion, Millipede

i) Group the animals in the above list as vertebrates and invertebrates

ii) Make a dichotomous key for the above vertebrates

3.

i). Name three animals showing camouflage.

ii). State three advantages of camouflage with examples.

4. Write two adaptations shown by the below mentioned animals, to their living environment.

e.g.:-

Fish	-	presence of fins	-	streamlined shape
Whip snake	-	.....	-	.....
Bird	-	.....	-	.....
Millipede	-	.....	-	.....
Mantis	-	.....	-	.....
Caterpillar	-	.....	-	.....

Technical Terms

Vertebrates	-	පෘෂ්ඨවංශීන්	-	முள்ளந்தண்டுளிகள்
Invertebrates	-	අපෘෂ්ඨවංශීන්	-	முள்ளந்தண்டிலிகள்
Adaptation	-	අනුවර්තන	-	இசைவாக்கம்
Camouflage	-	වේශාන්තරය	-	பொய்க்கோலம்
Streamlined shape	-	අනාකුල හැඩය	-	அருவிக்கோட்டு வடிவம்
Dichotomous key	-	දෙබෙදුම් සුචිය	-	இரு கிளைச் சாவி

07 Forms of Energy and Uses

We do various types of work in our day-to-day life. Animals and machines also do work. Some examples for such instances are given in Figure 7.1.

Running a vehicle

Lifting a weight

Rotating a fan

Running

Figure 7.1 ▲ Different types of work

In addition to those mentioned above, people engage in different types of work in day-to-day life. Can you give some examples for such work?

Assignment 7.1

- Give five examples for instances of doing work in day-to-day life.

It is clear that man, as well as animals and machines do a lot of work. Now, let us consider what is needed for man as well as other objects to do work. Let us engage in Activity 7.1 and 7.2 for studying this.

Activity 7.1

You will need :- A piece of bicycle tube (2 cm x 30 cm) or any other stretchable thing of that size, a metre ruler

Method :-

- Give the strip of rubber tube to each student and tell to stretch it as much as possible.
- Note down the maximum length, that each student stretched.

Figure 7.2

Table 7.1

Name	Maximum length stretched

- The maximum length of the stretched rubber strip was different from student to student. Discuss the reason for this difference.
- What have they applied/used to stretch the rubber tube?

Activity 7.2

You will need :- One metre long thread, a laboratary stand, a small piece of stone

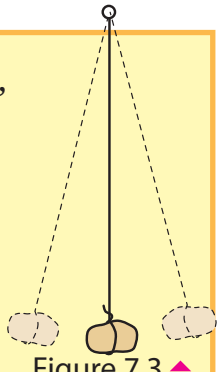


Figure 7.3 Motion of a hanging stone

Method :-

- Tie one end of the thread to the piece of stone.
- Tie the other end of the thread on the stand.
- Give the hanging stone a push.
- Observe the motion of the stone for a long period of time.
- What did you apply on the stone to move?
- Can you tell the reason for the gradual decrease of motion of the stone?
- Discuss in the classroom,
  - * What do you do to start the motion of stone?
  - * What does the stone gain to start its motion?

Work can be simply explained as a **push** or a **pull**, resulting a movement of an object. It is clear that work is done when stretching the strip of rubber tube in Activity 7.1 and moving the stone in Activity 7.2.

- Ability to do work is known as **energy**.
- International unit of measuring energy is **Joule (J)**.

Energy is necessary to perform work. To increase the amount of work done, the amount of energy applied should be increased. For example, you should supply more energy to move the stone further in Activity 7.2.

Energy exists in different forms. Now, let us consider the different forms that energy or the ability to do work can exist. Let us pay our attention to some different types of work done and different forms of energy used.

Carry out Activity 7.3 to find out the different forms of energy used to do different types of work.

Activity 7.3

Identify various forms of energy

Collect following items.

You will need :- A torch bulb, dry cells and some pieces of wire, a battery powered wall clock, a greeting card which produce music, a bicycle dynamo, a radio set, various types of winding and battery-powered toys, an electric motor, winding table clock

Method :-

- Study as groups how the given equipment work.
- Identify the main forms of energy that made each equipment work.
- Identify other forms of energy created, when functioning the equipment.
- Complete the following table according to your observations.

Figure 7.4

Table 7.2

Equipment	Main form of energy that made the equipment work	Other forms of energy Created
Electric bulb	Electricity	Light energy, Heat energy

By doing Activity 7.3 you may have identified several forms of energy found in most instances. Most of them are used in various tasks. Main forms of energy identified in Activity 7.3 can be listed as given below.

1. Kinetic energy
2. Potential energy
3. Electrical energy
4. Sound energy
5. Light energy
6. Thermal or heat energy
7. Chemical energy

Let us study further about some forms of energy you identified.

7.1 Kinetic energy

In most instances, we see moving things. Wind, a moving vehicle, flowing water and a moving pebble can be given as examples.

Let us find out whether the moving objects have some sort of energy.

Activity 7.4

You will need :- A turbine made by fixing metal blades to a cork stopper, a metal rod, a ball, a moving toy car, a piece of stone, a bowl of water

Method :-

- Hold the turbine under an opened water tap
- Keep the ball on a table and send the toy car towards it.
- Drop the piece of stone into still water.
- Lead a discussion, based on your observation.

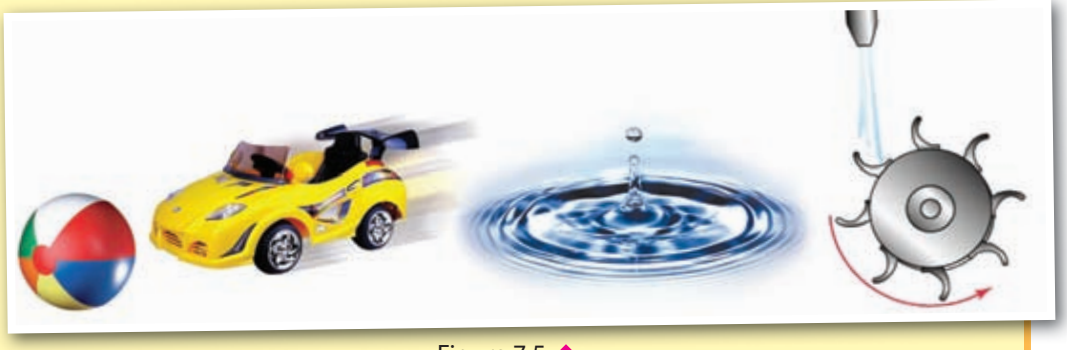


Figure 7.5

You may have observed that some work is done in each of the above activities. Can you say from where the energy was obtained for the work done?

The energy for above work is gained from the moving objects.

e.g.:- flowing water, moving stone

This activity reveals us that moving objects have energy.

Energy that a moving object contains is called the **kinetic energy.**

Electricity can be generated by kinetic energy of sea waves. grinding grains, pumping water and generating electricity can be done using kinetic energy of wind.

Electricity is generated in a hydropower station using kinetic energy of water.

Assignment 7.2

List out five objects which contain kinetic energy

Energy transformation

When work is done using different forms of energy, it is converted to another form of energy.

Let us consider the generation of electricity by kinetic energy of wind. Here kinetic energy is converted to electrical energy by the dynamo fixed to the wind mill.

Conversion of one form of energy to another form of energy is called energy transformation.

Energy transformation that occurs when electricity is generated by wind can be given as below.

Kinetic energy ⟶ Electrical energy

Self-assesment 1

Write the energy transformation associated with the following instances?

Obtaining light from an electric bulb

Generating electricity by a dynamo

7.2 Potential energy

Let us consider a water stream. We already know that it contains kinetic energy and it can do work.

e.g.:- turning a turbine

How did that water get energy for flowing? Does water in any place have the potential to flow?

Always water in a higher position can be subjected to flow.

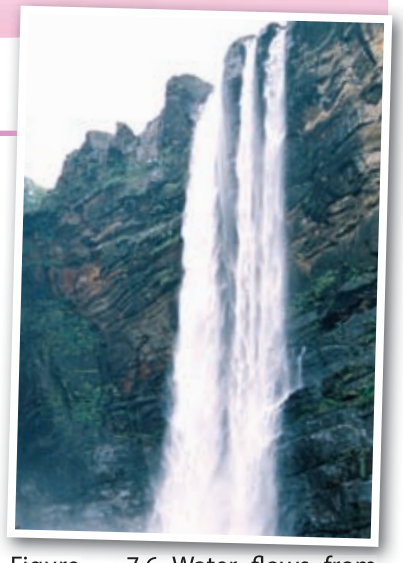


Figure ▲ 7.6 Water flows from higher elevation

e.g.:- water in a tank or reservoir at a higher elevation.

It is clear that water at a higher elevation contains energy and it has the ability of doing work.

Let us consider a toy car operates by winding a spring.

After winding Before winding

Figure 7.7 ▲ Spiral spring

When winding the spring energy is stored and the shape of the spring is also changed. The instrument operates using the stored energy of the wound spring.

Energy stored in an object because of the change of positions or the change of shape is known as potential energy.

Now, it may be clear to you that the energy stored in water of a reservoir at a higher elevation and a wound spring is **potential energy**.

When a spring is being unwound, potential energy stored in it is gradually converted to kinetic energy. Thus, unwound spring does not contain potential energy.

Figure ▲ 7.8 Inside a clock, that works by potential energy stored in a spring

Activity 7.5

Making a toy cart

You will need :- An empty tin can or a thread bobbin, a rubber band, a strong metal wire of 20 cm long

Method :-

- Make two holes in the tin can, as shown in Figure 7.9, and send the rubber band through the holes.
- Bend the metal wire and tie the rubber band to its ends.
- Turn the wheel of the toy cart you made, to wind the rubber band.
- Now place the toy cart on a table and observe. Think of a way to construct your toy cart more creatively.
- How did the toy cart gained energy to move?

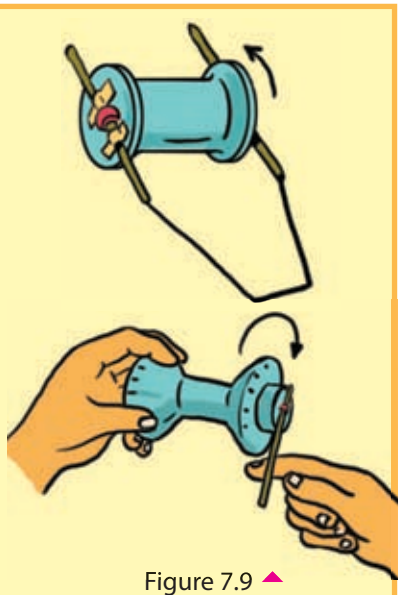


Figure 7.9

Self-assesment 2

1. Name five instances, where there is potential energy.

Potential energy and kinetic energy are commonly known as mechanical energy.

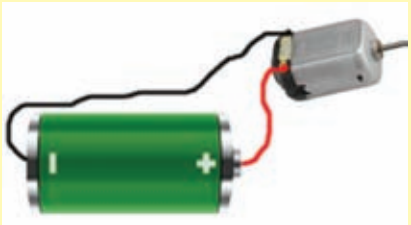
7.3 Electrical energy

You may have learnt electricity as a useful energy form. Let us do Activity 7.6 keeping what you have already learnt in mind.

Activity 7.6

- List out instances where electricity is used, recalling the work done using electricity.
- Discuss the facts that your group listed out in the classroom.

Check whether any of the points you noted down are illustrated in figures given below.



Rotating a motor

Running a toy car

Lighting a bulb

Ironing clothes

Figure 7.10

You will not hesitate to mention electricity as a form of energy used to do various types of work. Electrical energy can be used for various purposes like rotating electric fans, pumping water, lighting and heating.

Some appliances that work using electricity are given in Table 7.3 below. Complete the table.

Table 7.3

Appliance	Usage of appliance
Small electric motor	
Computer	
Electric iron	

Activity 7.7

Making a bell functioned by electricity

You will need :- A cork or rubber stopper, a piece of wooden plank (10 cm x 10 cm), an electric motor, a bell cover, a piece of metal sheet, two dry cells, iron nails, pieces of wire

Method :-

- Fix the bell cover on the wooden plank using nails.
- Fix the stopper to the end of motor.
- Fix the electric motor to the wooden plank using the piece of metal sheet, in such a way that the stopper just touches the bell cover.
- Supply electricity to the motor using dry cells.
- Adjust the positions of motor and bell cover to make the bell ring
- Write down the energy transformation that occurs when the bell is working
- Discuss the ways you can follow to develop this set-up you made.

Figure 7.11

7.4 Sound energy

You may have experienced that, doors and windows of your house vibrate when thundering. High tones of vehicles are unbearable to ears. We like the sounds like chirps of birds and sweet melody of music. But we do not like to hear unbearable sounds. Energy contains in pleasant and unpleasant sounds also. Let us do Activity 7.8 to understand this property of sound.

Activity 7.8

You will need :- A radio set, a sheet of paper, small pieces of polystyrene

Figure 7.12

Method:-

- Place the front side of the radio set upside and increase the volume.
- Put few pieces of polystyrene on the sheet of paper and hold it above the radio.
- Note down the observations.

- You can observe the vibration of the sheet of paper and the motion of polystyrene particles. Thus, it is clear that sound also contains energy.
- Energy contained in sound is scientifically known as **sound energy.** Energy transformation that occours when a radio set operates is given below.

Electrical energy ⟶ Sound energy

For extra knowledge

When addressing a mass gathering, sound energy produced by human voice is not sufficient. Therefore, sound is amplified using electricity. Appliance used for this purpose is called loud speaker set.

Assignment 7.3

Name four instances where sound energy is used

7.5 Light energy

Light is very important for vision. Therefore, since ancient days various methods were used to produce light. Green plants use light energy for the process of manufacturing food, called photosynthesis.

Assignment 7.4

Tabulate the materials used in various light sources to produce light.

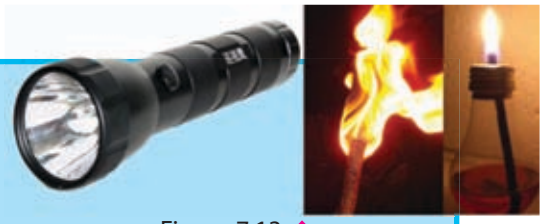


Figure 7.13

Table 7.4

Source of light	Materials used to produce light
Electric torch	Dry cells
Candle	.....
Torch made of dried coconut leaves/ ('Hulu aththa')	.....
Kerosene lamp	.....

In addition to get sight there are other purposes received from light. One of them is the generation of electricity.

Activity 7.9

You will need:- A solar panel, an electric motor, a torch bulb

Method:-

- Connect the electric motor to the solar panel and expose to light.
- Note down the observations.
- Remove the electric motor and connect the torch bulb.
- Discuss what is revealed by the observations.

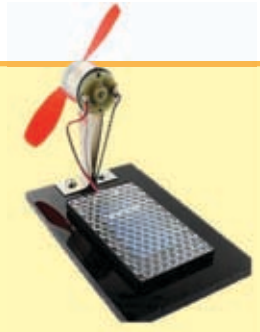


Figure 7.14

The electric motor and the torch bulb were opearated by light energy. Thus, it is clear that light energy can be used for various purposes.

The main source of light to the earth is the sun.

Though a large amount of light energy, falls on the earth from the sun, only a very small amount of light energy is utilized. So, there is an increasing interest to use light energy of sunlight (solar energy) in large scale today.

Energy transformation that occurs when electricity is generated in a solar cell is given below.

Solar energy ⟶ Electrical energy

Do you know?

Green plants maufacture food using light energy. This process is known as photosynthesis. Here, light energy is stored in plant as food. Energy is supplied to the whole living world by this food, produced by plants.

Self-assesment 3

List out some other instances where light energy is used

7.6 Heat/thermal energy

What do you feel when you bring your hand close to a flame? What you feel is the **heat energy**.

Heat is a form of energy, which is very important in doing various types of work like cooking food.

Let us do Activity 7.10 to understand more about heat energy.

Activity 7.10

You will need :- A glass bottle (750 ml), vessel to dip the bottle, a balloon

Method :-

- Fix the balloon to the empty bottle.
- Take hot water to the vessel and dip the bottle into it slowly.
- Note down the observations.
- Discuss the reasons for the observations.

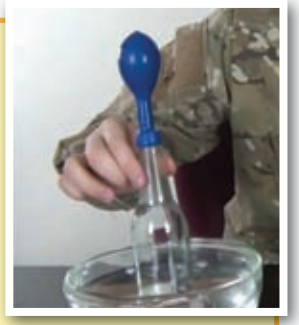


Figure 7.15

The balloon is inflated because of the increase of the volume of air inside the balloon due to heat energy.

More things can be done by heat. This will be clear to you by carrying out Activity 7.11.

Activity 7.11

You will need :- A candle, PVC tube/pipe, a piece of paper, a boiling tube with water, a test tube holder, a pair of crucible tongs

Method:-

- Light the candle.
- Hold the piece of paper to the flame of the candle. Observe what happens.

Figure 7.16

- Heat and fold the PVC tube.
- Heat the boiling tube with water.
- Note down the observations of each instance.
- Discuss the reasons for the observations.

Deformation of things, melting, burning and evaporation can be caused by heat. The reason is the energy contained in heat.

Today heat energy is used to generate electricity, to drive steam engines and many other purposes. Let us construct a set-up that works by energy of heat.

Activity 7.12

Making a steam turbine

You will need :- A small tin can, a sheet of aluminium, a cork stopper, few pieces of thick wire, a tripod, a burner

Method:-

- Bore a small hole at the middle of the lid of tin can.
- Put a small amount of water into the can and close the lid.
- Make a turbine fixing pieces of aluminium blades to the cork stopper. Using the pieces of thick wire, hold the turbine above the hole of the tin can.
- Place the set-up on the tripod and heat it using the burner.
- Discuss the reasons for what you observe.

Figure 7.17

Electricity is generated in thermal power stations by rotating dynamos, connected to large turbines, which are driven by steam.

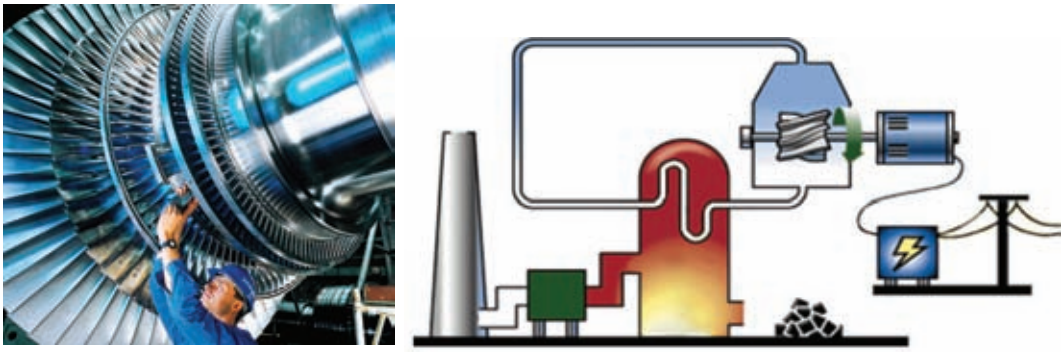


Figure ▲ 7.18 :- Parts in a thermal power station

Self-assesment 4

Write down the energy transformation that occurs in a thermal power station

It is heat energy, that is responsible for occuring winds, driving water cycle, drying of clothes and so on.

Assignment 7.5

List out five occasions where heat energy is used.

7.7 Chemical energy

Chemicals can exist as solids, liquids or gases. Large amount of energy is stored in most of the chemicals.

Energy stored in chemicals is known as **chemical energy**.

Let us carry out Activity 7.13 to understand facts about chemical energy.

Activity 7.13

You will need :- A candle, a box of matches, dilute hydrochloride acid, a piece of magnesium ribbon, a test tube

Method:-

- Light the candle and fix it on the table. Observe it for a few minutes and note down the observations.
- Put the piece of magnesium ribbon into the test tube with dilute hydrochloric acid. Note down the observations.

Figure 7.19

The candle, dilute hydrochloric acid, the piece of magnesium ribbon are all chemical substances. There are chemical substances in the dry cell also. What is released, in the Activity 7.13 is the energy stored in chemicals.

Assignment 7.6

Write down instances where chemical energy converts to other forms of energy.

It is the chemical energy, that is stored in the food, kerosene oil, fuels like fire wood, fire crackers and match sticks. It is also the chemical energy, that is stored in destructive objects such as bombs.

The energy transformation that occurs in a dry cell is given below.

Chemical energy ⟶ Electrical energy

There are more forms of energy, other than those we have studied so far. They may be considered in the future.

Summary

- Energy is necessary to do work.
- There are various forms of energy, which are utilized to do various types of work. Some of them are mechanical energy (potential and kinetic), electrical energy, light energy, heat energy, sound energy and chemical energy.
- Conversion of one form of energy to another form is called energy transformation.
- Most of the appliances we use, work by various forms of energy. One form of energy is transformed to other forms during the operation of various appliances.

Exercise

1. Chandrakantha comes to school by bus. Its horn has a high tone. There are electric bulbs to illuminate the bus. When the bus runs for a long time, the engine is heated.

I. Write four forms of energy you identified in the bus.

II. What is the basic form of energy used for generation of various forms of energy in the bus?

III. List out three uses of each form of energy you mentioned above.

2. Complete the table given below.

Equipment	Form of energy given to the equipment	Other forms of energy generated in the equipment
Electric bulb	Electrical energy	.....
TV set	.....	.....
Electric bell	.....	.....
Winding clock	.....	.....
Hydropower station	Kinetic energy of water	.....

Technical Terms

Work	-	කාර්යය	-	வேலை
Energy	-	ශක්තිය	-	சக்தி
Heat energy	-	තාප ශක්තිය	-	வெப்ப சக்தி
Electrical energy	-	විද්‍යුත් ශක්තිය	-	மின் சக்தி
Kinetic energy	-	චාලක ශක්තිය	-	இயக்க சக்தி
Potential energy	-	විභව ශක්තිය	-	அழுத்தச் சக்தி
Light energy	-	ආලෝක ශක්තිය	-	ஒளிச் சக்தி
Chemical energy	-	රසායනික ශක්තිය	-	இரசாயனச் சக்தி
Sound energy	-	ශබ්ද/ ධ්වනි ශක්තිය	-	ஒலிச் சக்தி

08 The Nature of the Earth

8.1 The structure of the earth

The earth we live on is the third planet from the sun in the solar system.

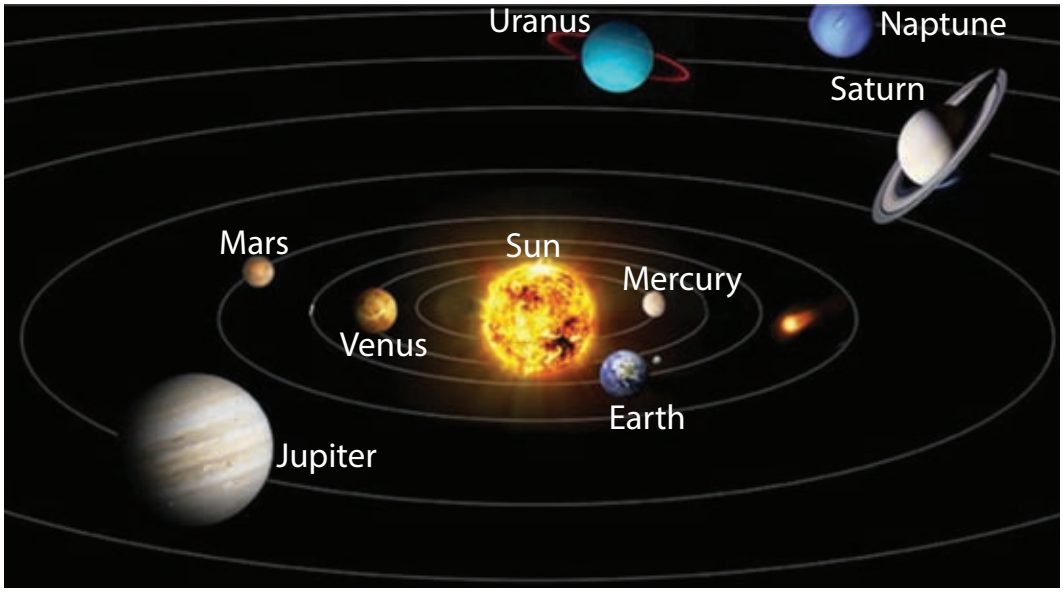


Figure 8.1 ▲ Solar System

When compared to other planets of the solar system, the earth looks more beautiful because there are mountains, rivers, oceans, flora and forests on it. Because of these favourable environmental conditions, there is life on earth. The other planets are lifeless because of the non-existence of such environmental conditions.

Figure 8.2 ▲ Different types of environments on the earth

What is the nature of the inner part of this beautiful earth?

Geologists obtain information about the nature of the inner part of earth using different methods.

In times of volcanic eruptions, different types of rocks are brought to the surface of the earth. By examining them, geologists obtain information about the inner part of the earth.

Geologists receive more information about the nature of the inner part by examining earthquakes. During earthquakes huge rock movements within the earth give rise to **seismic waves**. These waves reach the surface of the earth by running through various layers of the earth.

Seismometers are installed in different stations of the earth. Seismic waves are automatically marked by these seismometers. (A Seismometer is installed in Pallekelle, Sri Lanka.) The speed of seismic waves running through various layers of the earth are different. Information about internal layers of the earth can be obtained by measuring those speeds.

That information has helped to discover that the inner part of the earth consists of several layers different from each other.

As shown in Figure 8.3, the inner part of the earth can be divided into three layers. They are **core**, **mantle** and **crust**.

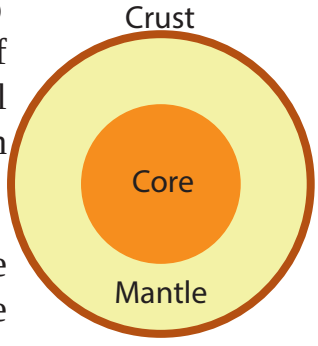


Figure 8.3 ▲ Cross section of the Earth


Figure 8.4 ▲ Cross section of a hard boiled egg

It is a replica of a hard boiled egg cut across from top to bottom. The inner parts of earth can be compared to this replica of an egg.

Egg york	→	Core
Egg white	→	Mantle
Egg shell	→	Crust

Information on the structure of the earth from its surface to the core are given below.

Crust

The surface of the earth where life exists is the crust. When compared with the size of the earth, it is a thin layer. Mountains, plains and oceans are found on the crust. Its thickness varies from place to place. At the bottom of the oceans its thickness is about 5 km. On land its thickness is about 35 km. The earths crust consists of rocks and soil. It is made up of basic elements like oxygen, silicon and aluminium.

The earths crust provides most of the elements necessary for our sustenance.

e.g.:- Construction materials
Metals
Fossil fuel
Soil for agriculture

Mantle

Underneath the earths crust lies the mantle. Its thickness is about 2900 km. It consists of rocks. These rocks contain oxygen, silicon, magnesium and iron. The upper part consists of solid rocks. The lower part is made up of molten rocks due to extreme hot environment.

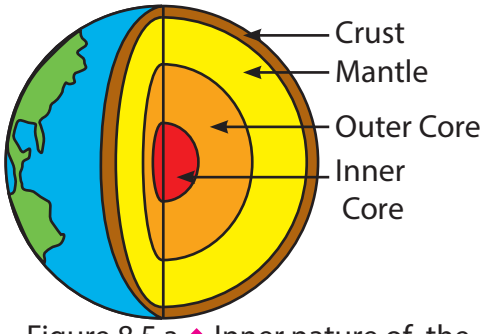


Figure 8.5 a ▲ Inner nature of the earth

Core

The innermost part of the earth is the core. Its thickness is about 3500 km. The upper part of it is made up of molten iron and nickel metals. The temperature of this area is between 4400 °C - 5000 °C.

Due to high pressure, the lower part of the core remains hard. Its temperature is more than 5000 °C which is as high as the temperature on the surface of the sun.

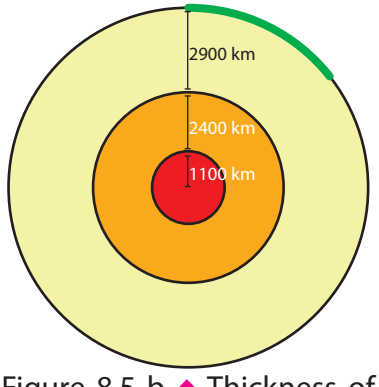


Figure 8.5 b ▲ Thickness of the inner layers of the earth

Table 8.1 Layers of the earth

Part of the earth	Thickness	Composition	Elements present	Special Information
Crust	Deep bottom of oceans 5km Land 35km	Rocks Soil	Oxygen Silicon Aluminium	Very thin layer
Mantle	2900km	Solid Rocks and Molten Rocks	Oxygen Silicon Magnesium Iron	Upper part consists of solid rocks due to high temperature Molten rocks are in the lower part
Core	3500km	Upper part consists of molten iron and nickel metals	Molten iron and nickel	Temperature between 4400-5000 °C

Activity 8.1

Preparation of a replica of the internal structure of the earth

You will need :- 35 x 35 cm piece of cardboard, saw dust, paint in three colours, glue

Method :-

- Draw a circle with a radius of 1 cm at the centre of the cardboard.
- Draw a circle with a radius of 7 cm which is concentric with the above circle.
- Draw another circle with a radius of 13 cm which is concentric of above two circles.
- Now you will get a Figure like 8.5 b.
- Put saw dust in three colours.
- Paste them on three layers and name them.
- Now show your creation to the classroom.

Activity 8.2

Preparation of a three dimensional replica of the internal structure of the earth

You will need :-

Clay in three colours, a sharp knife

Method :-

- Make a globe of the size of a lime using clay of one colour.
- Taking a half of the diameter of the previous globe, paste a separate coloured layer of clay on the top of the globe.
- On the top of the second clay layer, paste a different colour of clay layer which is thin as much as possible.

Figure 8.6 ▲ Model of the internal structure of the earth

- Use a sharp knife and cut the clay globe you made into two equal halves.
- The above cross section of the clay globe shows how inner layers are placed.

Assignment 8.1

Prepare the model made in Activity 8.2 using clay/ Polystyrene/ paper pulp. Cut them into two equal halves and examine the cross section of the models.

8.2 Tectonic plate and plate tectonics

Figure 8.7 shows an earth map of a strong earthquake on 25th April 2015 in Katmandu, Nepal. Thousands of people died and thousands of

people got injured, while thousands of people were left homeless. This earthquake occurred on the Eurasian and Indian plate margins. Two destructive earthquakes have been recorded in the year 1905 and 1934 in this area.

-News on Internet-

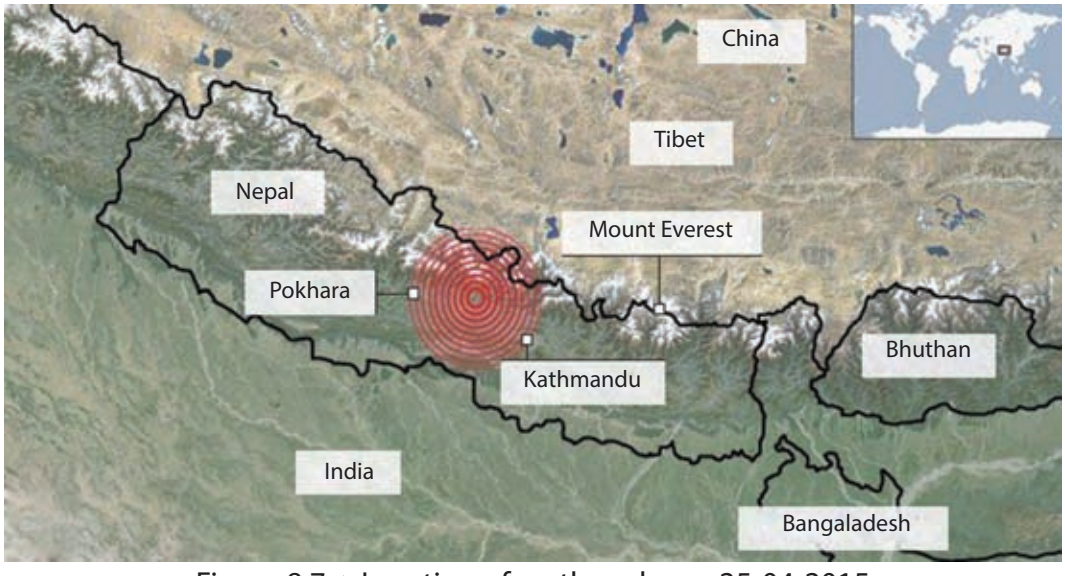


Figure 8.7 ▲ Location of earthquake on 25-04-2015

The surface layer of the earth is the crust. The crust is divided into sections called tectonic plates. These tectonic plates move relatively to each other.

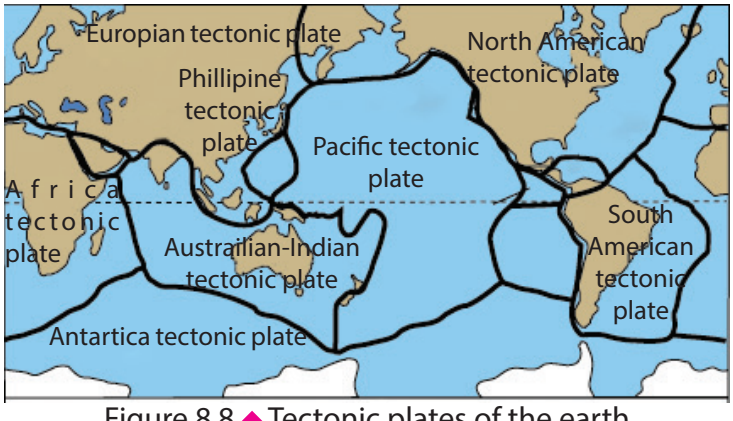


Figure 8.8 ▲ Tectonic plates of the earth

Assignment 8.2

Study the map in Figure 8.8. It shows the tectonic plates of the earth. Find tectonic plates near Sri Lanka, India and Nepal.

Tectonic plates move relatively to each other in three ways.

a) According to the Figure 8.9 **a** two plates move apart. As a result, a deep gulf could occur.
b) According to the Figure 8.9 **b** one plate moves upward by pressing down the other. An earthquake could occur in this instance.
c) According to Figure 8.9 **c** two plates slide past each other. Earthquakes could occur in this instance too.

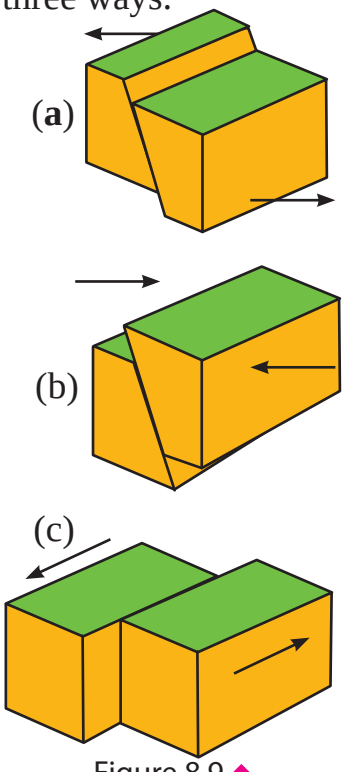


Figure 8.9

Tectonic plates move very slowly. It is 1 - 2 cm per year.

Plate movements can be recognized on plate margins. Most of the plate boundaries or margins are at the bottom of the sea. When they occur on land they can be easily recognized.

One such plate margin is located in North America. It is known as San Andreas Fault. It is visible in a birds eye view/an aerial view as a fracture of 1000 km in length in the state of California.


Figure 8.10 San Andreas Fault

The North American Plate and the Pacific Plate are situated on either side of this tectonic boundary. These two plates slide past each other for about 2.5 cm per year. On these tectonic boundary earthquakes occur frequently.

Tectonic plates float on the molten rocks of the lower mantle. Because of the movement on the molten rocks plate, tectonics also move.

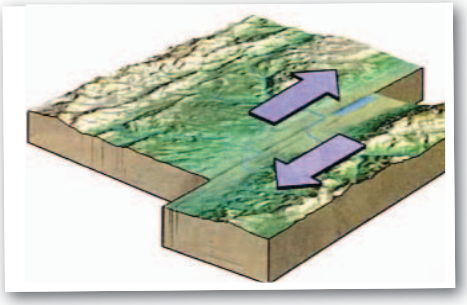


Figure 8.11 ▲ How tectonic plates slide

Assignment 8.3

Do a literature search to find out the geological history of the earth and prepare a report. Use the internet, reports on geological findings and geographical books.

Activity 8.3

Demonstrating the activity of plate tectonics

You will need :-

A yellow coloured orange with a thick skin, a knife

Figure 8.12 ▲ Replicate plate tectonics using an orange

Method :-

- Cut the orange with the skin into different shapes.
- Keep the orange between the two palms.
- Then slowly press the orange while observing the movement of parts of the peels.

For extra knowledge

In 2004, Tsunami brought severe destruction to the coastal areas of Sri Lanka. This happened because of an earthquake occurred near the Sumatra Island due to collision of tectonic plates.

Now you can understand that the earth's crust where we live is not an immovable thing.

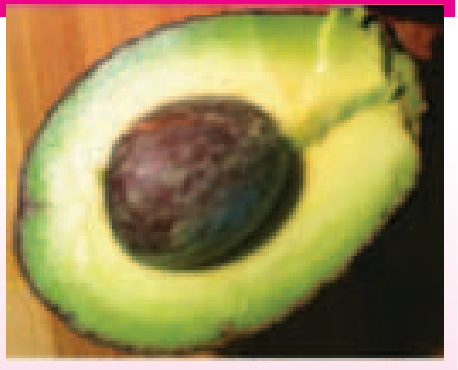
Summary

- In the solar system, the earth is the most suitable planet for the living beings.
- The inner part of the earth consists of three layers; the core, the mantle and the crust.
- The earths crust consists of tectonic plates which move relatively to each other.
- Tectonic plate margins consist of volcanoes and constant earthquakes occur near them.

Exercise

1. A cross section of an avocado is shown here.

   A cross section of the earth consists of core, mantle and crust. Which sections of the avocado corresponds with above three areas?



2.

I) Name four resources obtained by man from the earth crust.

II) On which tectonic plate Sri Lanka is situated? Why severe earthquakes do not occur in Sri Lanka?

III) State two methods used by geologists to obtain information about the internal structure of the earth.

IV) State three countries where earthquakes occur frequently.

Technical Terms

Core	-	හරය	-	அகணி
Mantle	-	ප්‍රාවරණය	-	மென்மூடி
Crust	-	කබොල	-	ஓடு
Rocks	-	පාෂාණ	-	பாறைகள்
Tectonic plates	-	භූ තැටි	-	புவித்தகடுகள்
Plate tectonics	-	භූ තැටි චලනය	-	புவித் தகட்டியக்கம்
Earthquakes	-	භූ කම්පන	-	நிலநடுக்கம்
Seismic waves	-	භූ කම්පන තරංග	-	நிலநடுக்க அலைகள்
Seismometer	-	භූ කම්පනමානය	-	நிலநடுக்கமானி
Volcanoes		ගිනි කඳු		எரிமலைகள்

09 Light

9.1 Formation of umbra and penumbra

Pay your attention to Figure 9.1. It shows some instances where umbrae or shadows are formed.

Figure :- 9.1 ▲ Some instances where umbrae / shadows are formed

In our day-to-day life, umbrae or shadows can be seen frequently. How do umbrae or shadows form? Let us do the Activity 9.1 to investigate this.

Activity 9.1

You will need :- A candle, a white screen, a small ball

Method :-

- Place the lighted candle on a table, keep the ball infront of it and obtain the umbra of the ball on a screen or on a wall.

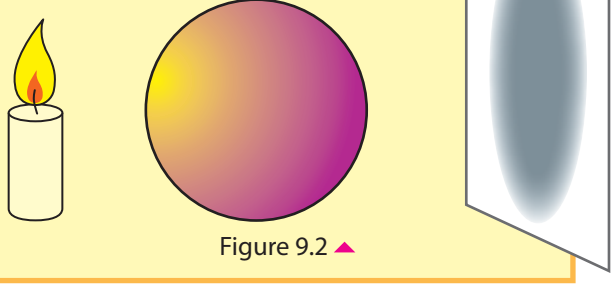


Figure 9.2 ▲

Here the shadow or the umbra of the ball can be clearly observed on the screen.

An umbra of the ball is formed on the screen, because the light emitted from the candle does not pass through the opaque ball.

Let us do Activity 9.2 to study about umbrae further.

Activity 9.2

You will need :- An electric torch, a screen, a small ball, a translucent glass sheet, a cardboard sheet, a transparent glass

Method :-

- Place the lighted electric torch, the transparent glass sheet and the ball as shown in the Figure 9.3 and obtain the umbra of the ball on a screen or on a wall.
- Observe the nature of the umbra.
- Then replace the transparent glass sheet with the translucent glass sheet and observe the umbra formed.
- Finally replace the translucent glass sheet with the cardboard sheet and see whether an umbra of the ball can be obtained.

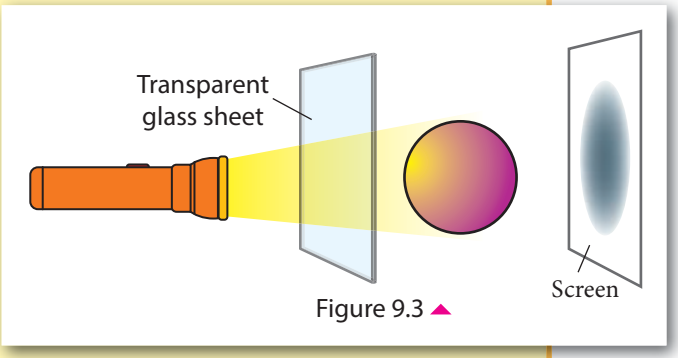


Figure 9.3

Compare your observations with those given below.

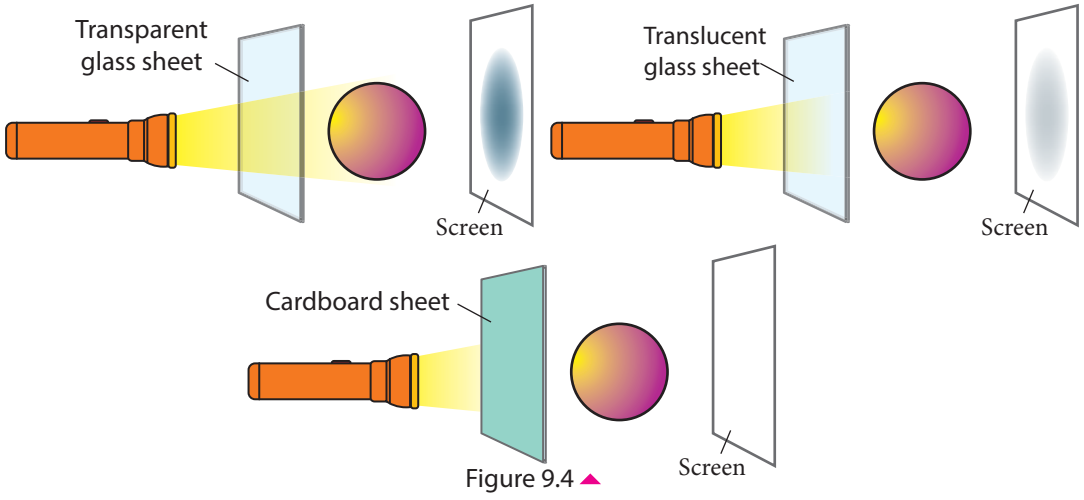


Figure 9.4

A sharp umbra is formed on the screen when transparent glass sheet is placed before the ball. A blurred umbra is formed when translucent glass sheet is placed.

No umbra is formed when cardboard sheet is placed, because light does not travel through a cardboard sheet.

The length and the direction of shadows differ due to the angle and the direction of sunlight falling on objects. In ancient times, the length of umbrae or shadows were used as arbitrary units to measure time.

e.g.:- Sun dial

Figure :- 9.5 ▲ Sun dial

Various artistic creations can be done with umbrae or shadows. Some such creations done with hand and fingers are shown below.

Figure 9.6 ▲ Some creations done with umbrae

Assignment 9.1

Test whether you can do creations like those shown in Figure 9.6. Present a collection of such creations with your friends.

In concerts creations done using shadows are popular in the field of art today. Some such instances are shown in Figure 9.7.

Figure 9.7 ▲ Artistic creations of shadows

Assignment 9.2

Enjoy watching artistic creations of shadows on the internet or in video clips.

Let us do Activity 9.3 to study further about umbrae.

Activity 9.3

You will need:- An electric torch, a screen, a small ball

Method:-

- Place the ball infront of the lighted electric torch and obtain its umbra on a screen or on a wall.
- Identify the dark brown/ light brown umbra and penumbra of the ball.
- Change the distance between the ball and the electric torch and observe how the umbra and penumbra are changing.

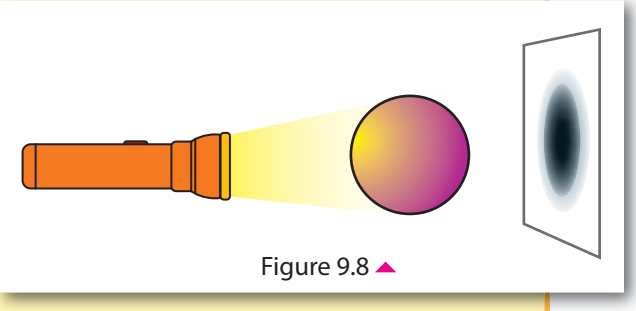


Figure 9.8 ▲

Compare your observations with those given in Figure 9.9

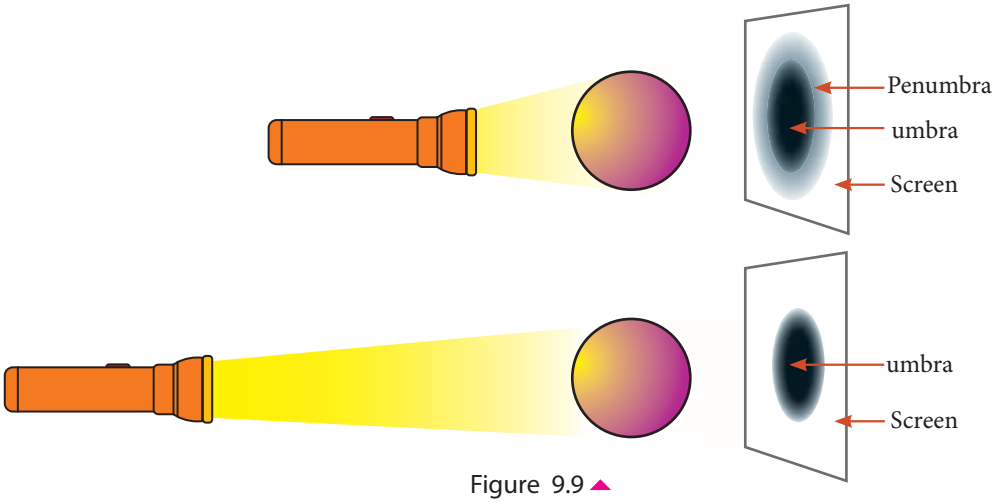


Figure 9.9

When the ball and the electric torch are closer, the umbra on the screen is not sharp and clear. There is a penumbra also around the umbra. When the distance between the ball and the electric torch is increasing, the penumbra gradully disappears. When the electric torch is kept at a considerable distance from the ball, only the umbra can be observed.

So, the light source should be far from the object to obtain a clear and sharp umbra.

Let us study further about penumbra

Activity 9.4

You will need:- A piece of polythene, red and blue cellophane sheets, a torch, red and blue marker pens, platignum, a small ball, a screen

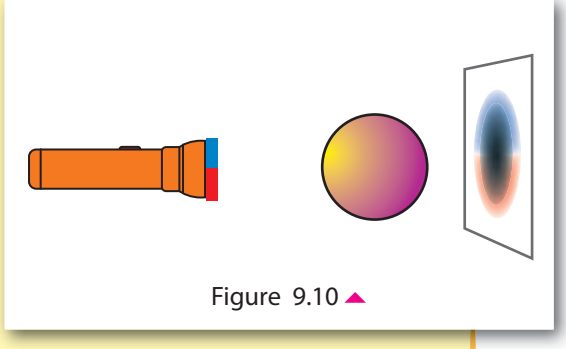


Figure 9.10

Method:-

- Tie the piece of polythene on to the face of the torch and divide it into two semi-circles.
- Colour one of the semi-circles in blue and the other in red or use red and blue cellophane to cover face of the torch.
- Light the torch and focus it until you observe mixed colours on the screen.

Compare your observations with the following.

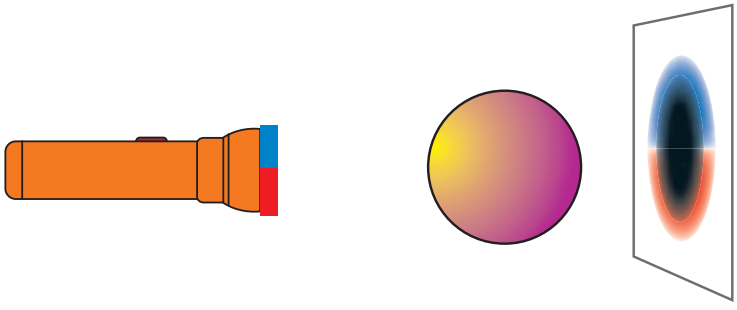
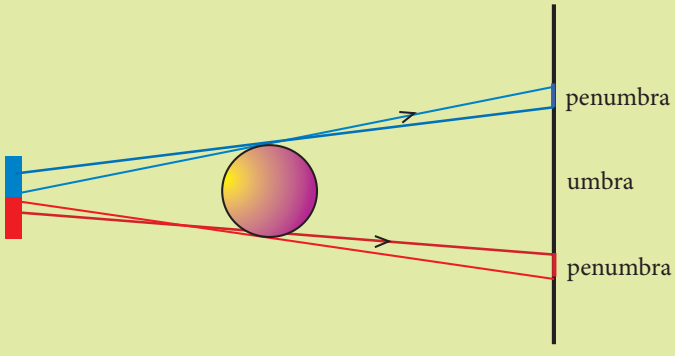


Figure 9.11

The upper part of the penumbra is seen in one colour (blue) and the lower part in the other colour (red).

For extra knowledge

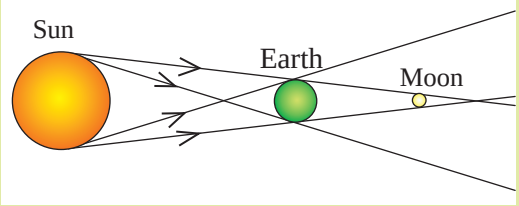
To describe Activity 9.4 further.



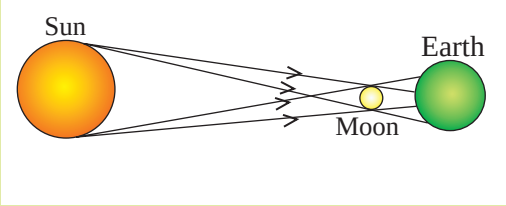
Umbra is formed on the screen, because light does not pass through the ball. Penumbra is formed because of the light rays emitted from the edges of the light source.

Thus, it is clear that penumbra is formed by the light emitted from one part of the light source.

Solar and lunar eclipses occur due to umbrae or shadows.



When the earth is between the sun and the moon, and all three are in a straight line, the umbra of the earth falls on the moon, causing a lunar eclipse.



When the moon is between the sun and the earth, and all three are in a straight line, the umbra of the moon falls on the earth. So, the sun cannot be seen from the earth. This incident is known as a solar eclipse.

9.2 Images formed by plane mirrors

Look at your face through a mirror. You can see your image inside the mirror. How is it formed? Can you recall instances where sunlight is directed on to a wall in the house from outside using a mirror?

Figure 9.12 ▲ Reflection of light

Here, the light that falls on the mirror is turned back to the house.

Figure 9.13 ▲ Reflection of light

The phenomenon of returning the light rays in the same medium after falling on a surface is called reflection of light.

Light reflects well from smooth polished shining surfaces.

Polished shining surfaces act as mirrors. Forming images by mirrors is a result of the reflection of light. The Figure 9.14 shows an image formed by a mirror.

Figure 9.14 ▲ Reflection seen in a plane mirror

A polished shining plane surface is known as a plane mirror.

A plane mirror is drawn in a diagram in the following manner (Figure 9.15).

Figure 9.15 ▲ plane mirror

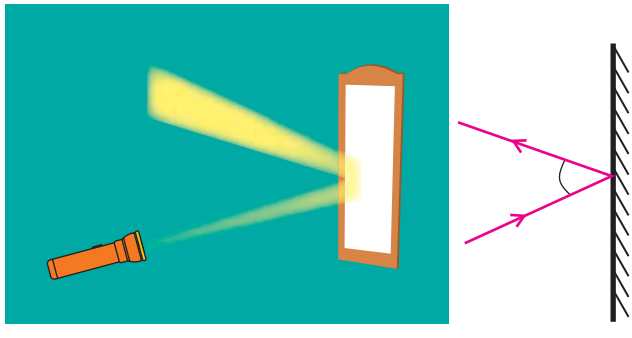


Figure 9.16 ▲ How light is reflected by a plane mirror

The image of a candle, placed in front of a plane mirror is shown in Figure 9.17.
Let us do Activity 9.5 to find out about images formed by plane mirrors.

Figure 9.17

Activity 9.5

You will need:- A plane mirror, a candle, a ruler

Method:-

- Place the ruler perpendicular to the mirror as shown in Figure 9.18. Place the lighted candle at the far end of the ruler.
- Observe the image of the candle formed through the plane mirror.
- Record your observations of the nature of the image in the following table.

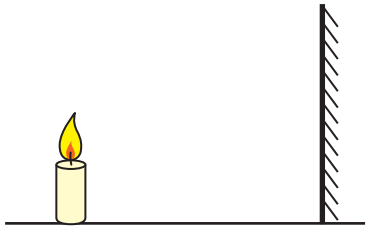


Figure 9.18 An image seen through a plane mirror

Table 9.1

Properties of the image	Observation
Can/Cannot be obtained on to a screen	
Upright / Inverted	
Size of the image	

If the image can be obtained on to a screen the image is termed as "real". If it cannot be taken on to a screen, then the image is known as " virtual".

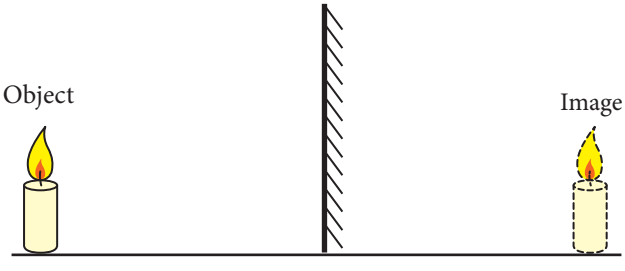


Figure 9.19

Let's do Activity 9.6 to find out about images formed by a sheet of glass.

Activity 9.6

You will need :- A sheet of glass, two candles which are same in size and shape, a ruler, a screen

Method :-

- Place the ruler perpendicular to the sheet of glass and place a lighted candle at the far end of the ruler as shown in figure 9.20

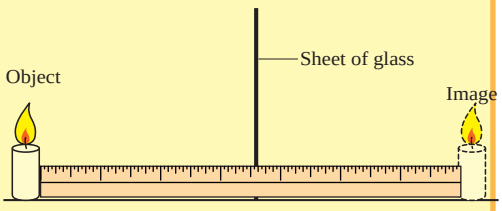


Figure 9.20

- Observe the image of the candle formed through the sheet of glass (It is more suitable to carry out this activity in a dark place)
- Place the other candle, where you see the image as shown in the figure.
- Compare the size of image with the candle.
- Measure the distances between the first candle and the sheet of glass (object distance) and between the second candle and the sheet of glass (image distance)
- Record your observations in the following table.

Table 9.2

Properties of the image	Observation
Size of the image	
Distance between the first candle and the sheet of glass	
Distance between the second candle and the sheet of glass	

Features of an image formed by a plane mirror are given below.

- Cannot be obtained on to a screen (virtual), erect.
- Size of the image is equal to the size of the object.
- Image is formed behind the mirror (Distance from the mirror to the object is equal to the distance from mirror to the image.)
- Left and right sides of the image are inter changed. (lateral inversion)

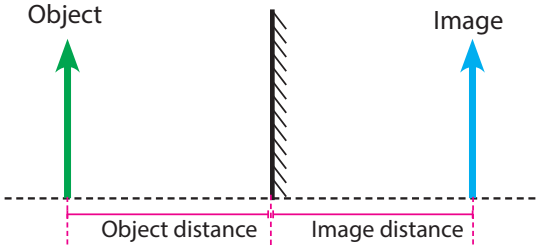


Figure 9.21

Lateral inversion

Activity 9.7

You will need :- A plane mirror, letters O, B, D and P (Write letters on a paper).

Method :- Place each letter cut from a cardboard sheet in front of the plane mirror and observe.

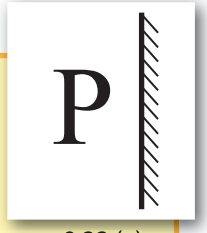


Figure 9.22 (a)

Figure 9.22 (b)

It is observed that the right and left sides of the letters B, D and P are inversed. The right and left side of the letter O is also inversed, but it is not observable because the letter O is symmetrical.

This phenomenon of inversing right and left sides of an object, when observed through a plane mirror is known as **lateral inversion.**

Figure 9.23 Lateral inversion of an image

The phenomenon of lateral inversion is further illustrated in Figure 9.23.

Think of the reason for painting the word AMBULANCE on ambulances as shown in this figure.

Assignment 9.3

Tabulate the letters of english alphabet as those which can be identified as laterally inverted and cannot be identified as laterally inverted.

Multiple images

To show the number of items as increased in jewellery shops, plane mirrors are kept behind the items and parallely on either sides. Light is reflected by those mirrors and a large number of images can be observed.

Figure 9.24 ▲ Jewellery shops where multiple images are formed

When two or more plane mirrors are kept at an angle or parallel to each other and an object is kept in between more than one image are formed. Such a formation is known as multiple images.

Let us carry out Activity 9.8 to find out more about this phenomenon.

Activity 9.8

You will need :- A candle, two plane mirrors, a protractor

Method :-

- Keep two plane mirrors at an angle of 90^0 and place a lighted candle between them.
- Count the number of images formed.
- Change the angle between the plane mirrors as 60^0, 45^0 and 30^0 and count the number of images formed in each instance
- Tabulate your observations as indicated below.

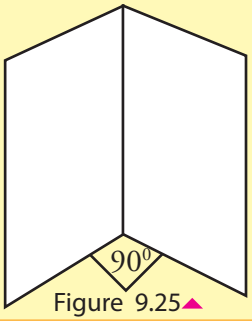


Figure 9.25 ▲

Table 9.3 ▼

Angle between the two plane mirrors (degrees)	Number of images formed
90	
60	
45	
30	

Compare your observation with those in the table below.

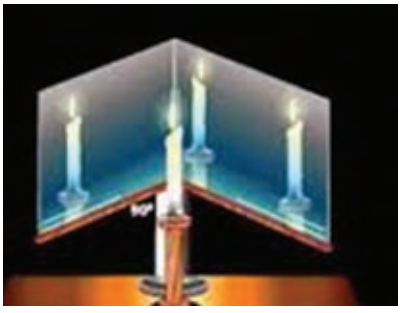


Figure 9.26 ▲ How multiple images are formed

Table 9.4 ▼

Angle between the two plane mirrors (degrees)	Number of images formed
90	3
60	5
45	7
30	11

The number of images formed are increased when the angle between the two plane mirrors decreases.

Assignment 9.4

- Observe the number of images formed, when the angle between two plane mirrors are decreased.
- Mention the number of images formed/the nature of images when an object is kept between the two parallelly placed plane mirrors.
- Discuss your answer with your teacher.

Assignment 9.5

Multiple images formed by plane mirrors are used in day-to-day life. Find out and report several such instances.

Instances where plane mirrors are used

- As mirrors in dressing rooms
- To illuminate various objects in cinematography
- To direct light on to the slide of microscopes
- To construct kaleidoscopes
- To construct periscopes

Kaleidoscope is an instrument in which the formation of multiple images is used. Let us carry out Activity 9.9 to construct a kaleidoscope.

Activity 9.9

You will need :- Three strips of plane mirrors with equal dimensions (6 cm x 2 cm), black paper, gum tape

Method :-

- Keep the strips of plane mirrors triangularly as shown in Figure 9.27. Cover one end with the piece of tissue paper.
- Cover the mirrors with black paper and stick securely with gum tape.
- Put small coloured items like beads, petals into this and observe through the open end. Turn round the instrument to see how the coloured patterns change.

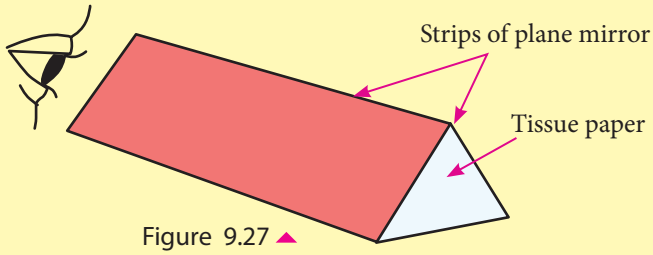


Figure 9.27

You will be able to observe various patterns. These patterns are formed by the reflection of light from several mirrors. Vivid patterns formed in a kaleidoscope are used to create designs for textile, floor tiles and wall tiles.

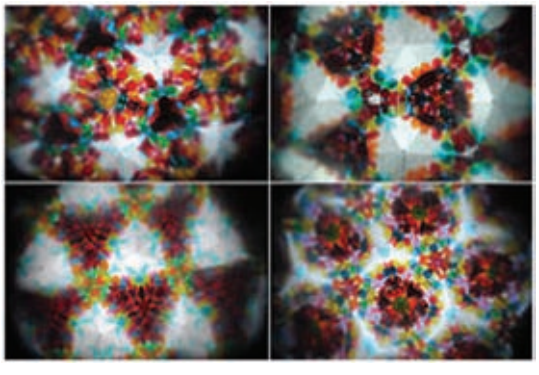


Figure 9.28 Vivid patterns formed in a kaleidoscope

Periscope is another instrument made using the phenomenon of reflection of light through plane mirrors. Let us carry out the Activity 9.10 to make a periscope.

Activity 9.10

You will need:- Two identical plane mirrors, pieces of cardboard sheet, gum tape

Method:-

- Make tubes of appropriate size, using the piece of cardboard sheet construct the equipment as shown in the Figure 9.29. Place the pieces of mirrors at an angle of 45^0 at each bend. Get the assistance of your teacher for this construction.
- Observe objects through the instrument you made.

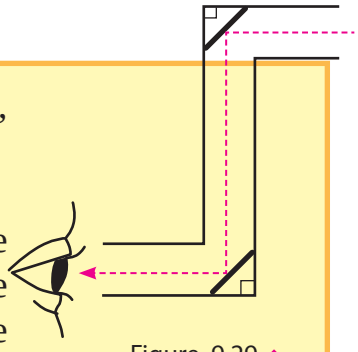


Figure 9.29

An observer positioned in a low level, can use a periscope to observe incidents that occurs in a higher level. This is commonly used in submarines and bunkers.

Assignment 9.6

Find out and record other instances where periscope is used.

Assignment 9.7

Mention other instances where plane mirrors are used in day-to-day life.

9.3 Images formed by curved mirrors

Figure 9.30 Some objects with curved surface

Have you ever seen the image of your face on a metal spoon? The image is seen differently on inner surface and on outer surface of the spoon. This happens because the surfaces of spoon act as curved mirrors.

Mainly there are two types of curved mirrors, named as convex mirrors and concave mirrors. Reflecting surface of a concave mirror is curved inwards and that of a convex mirror is curved outwards.

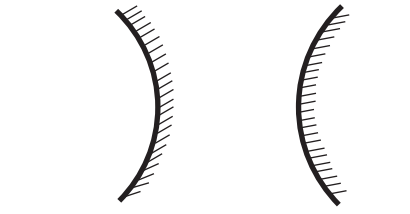


Figure 9.31

A ray of light is denoted by a straight line and the direction is indicated by an arrow head drawn on the straight line.

A beam of light is made up of a bundle of light rays.

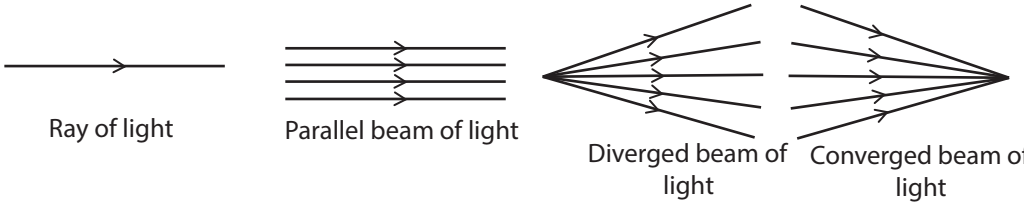


Figure 9.32 light rays and light beams

Concave mirrors

What happens when a narrow, parallel beam of light falls on a concave mirror? Let us do Activity 9.11 to find out this.

Activity 9.11

You will need :- A concave mirror, a plane mirror

Method :-

- Direct a narrow, parallel beam of light on to the reflecting surface of a concave mirror. (This can be done by using a plane mirror)
- Record your observations.

You can observe that the light beam collects to a spot infront of the concave mirror.

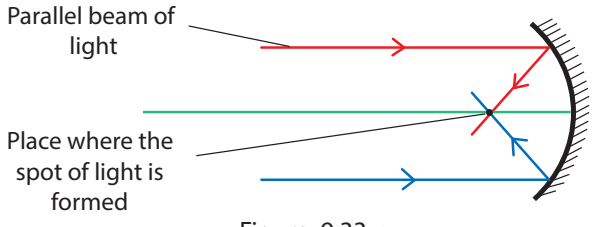


Figure 9.33

This collection of a parallel light beam to a single spot in front of the mirror is known as **convergence** of light. Therefore, concave mirrors can be used to converge light.

What happens when a narrow, parallel beam of light falls on a convex mirror? Let us do Activity 9.12 to find out this.

Activity 9.12

You will need :- A convex mirror, a plane mirror

Method :-

- Direct a narrow, parallel beam of light onto a convex mirror. (A plane mirror can be used for this purpose.)
- Record your observations.

It can be observed that light which falls on a convex mirror, " spreads out" after reflection. This spreading out of light after reflection is known as **divergence** of light. Therefore, convex mirrors are considered as divergent mirrors.

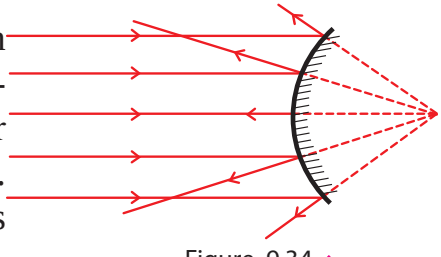


Figure 9.34

Images formed by concave mirrors

Let us do Activity 9.13 to observe the nature of images formed by concave mirrors.

Activity 9.13

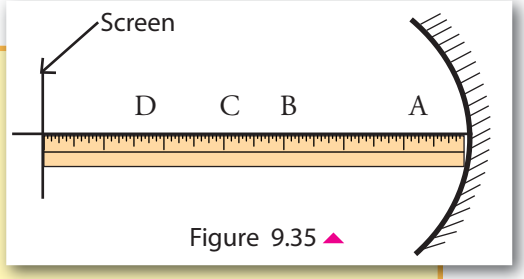


Figure 9.35

You will need :-

A concave mirror, a mirror holder, a candle, a screen, a metre ruler

Method :-

- Place a lighted candle on point A, in front of a concave mirror. The candle should be placed close to the mirror.
- Try to obtain the image formed, onto a screen.
- Observe the nature of the image, with the assistance of your teacher.
- Then, place the lighted candle on points B, C and D respectively and observe the nature of the images formed.
- Tabulate your observation as follows.

Table 9.5

Point where candle is kept	Image can/ cannot be obtained onto a screen	Upright/ Inverted	Size of the image
A			
B			
C			
D			

Compare your observations with those given below.

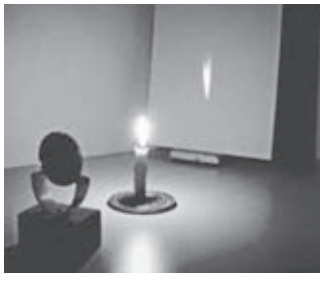


Figure 9.36 Instance of forming images from a concave mirror

Table 9.6

Point where candle is kept	Image can/ cannot be obtained onto a screen	Upright/ Inverted	Size of the image
A	Impossible	Upright	Larger than the object
B	possible	Inverted	Larger than the object
C	possible	Inverted	Equal to the object
D	possible	Inverted	Smaller than the object

Instances where concave mirrors are used in day-to-day life

- As mirrors used for shaving
- For dentists to examine teeth of patients
- Used in reflecting telescopes
- To make solar cookers
- To direct light to the glass side of microscope.

Figure 9.37

Assignment 9.8

Make a list of other instances where concave mirrors are used in day-to-day life.

Convex Mirrors

Let us do Activity 9.14 to observe the nature of images formed by convex mirrors.

Activity 9.14

You will need :-

A convex mirror, a mirror holder, a candle, a screen, a metre ruler

Method :-

- Place a lighted candle on point A, infront of convex mirror.

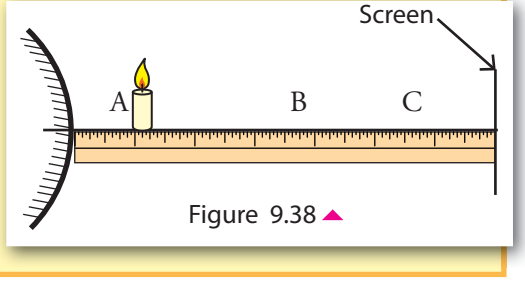


Figure 9.38

- Place the candle close to the mirror.
- Try to obtain the image formed, onto a screen.
- Observe the nature of the image with the assistance of your teacher.
- Get the assistance of the teacher to select points B and C.
- Then place the lighted candle on points B and C respectively and observe the nature of the images formed.
- Tabulate your observations as follows.

Table 9.7

Point where candle is kept	Image can/ cannot be obtained onto a screen	Upright/ Inverted	Size of the image
A			
B			
C			

Compare your observations with those given below.

Figure 9.39 Images formed by a convex mirror

Table 9.8

Point where candle is kept	Image can/ cannot be obtained onto a screen	Upright/ Inverted	Size of the image
A	Cannot	Upright	Smaller than the object
B	Cannot	Upright	Smaller than the object
C	Cannot	Upright	Smaller than the object

It is clear that the nature of the image does not change when the distance between the object and convex mirror changes.

Let us engage in the Activity 9.15 and study about the images formed by curved mirrors.

Activity 9.15

You will need:- Two candles which are same in size and shape, watch glass, a screen, mirror holder

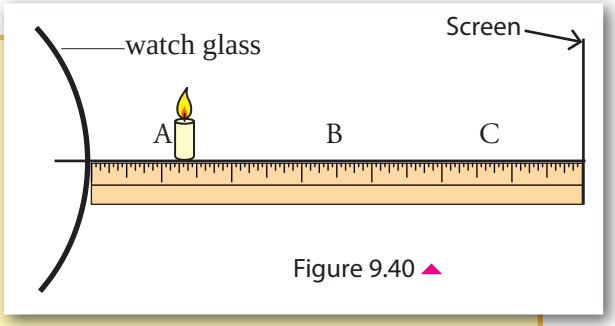


Figure 9.40

Method:-

- Place a lighted candle on point A, infront of convex surface of the watch glass as shown in Figure 9.40.
- Observe the image formed.
- Try to obtain the image on to a screen (it is appropriate to use a dark place for the above activity)
- Place the other candle, where you see the image and compare the size of the image with the candle.
- Observe the changes of the images placing the candle at B and C.
- Tabulate your observations

Instances where convex mirrors are used in day-to-day life

- As side mirrors of vehicles

The driver can view a large rear area of a vehicle in the side mirror because convex mirror forms smaller, upright images.


Figure 9.41 How rear side of a vehicle is viewed in a side mirror

Assignment 9.9

Make a list of other instances where convex mirrors are used in day-to-day life.

Summary

- Umbrae or shadows can be observed frequently in day-to-day life.
- Umbrae are formed when light does not travel through opaque objects.
- Clear and sharp umbrae are formed, when the object is far from the light source is kept close to the object.
- Smooth, shining surfaces act as mirrors.
- Returning light rays back in the same medium, after striking on a surface is known as reflection of light.
- Images are formed in mirrors due to the reflection of light.
- Images formed in plane mirrors are always upright, equal to the size of object and cannot be obtained on to a screen.
- Parallel beam of light can be converged by a concave mirror, and can be diverged by a convex mirror.
- The nature of images formed by concave mirrors differs according to the distance between the object and the mirror.
- The nature of the images formed by convex mirrors does not differ according to the distance between the object and the mirror.
- Mirrors are used for various purposes in day-to-day life.

Exercise

1) Select the appropriate word from the brackets and fill in the blanks.

1. Clear umbra can be obtained by ..... (a candle/ an electric torch bulb)
2. A ..... mirror is used to diverge a parallel beam of light. (Convex/ Concave)
3. Images formed by plane mirrors are always ..... the object. (equal to/ smaller than)
4. Images formed by convex mirrors are always ..... (upright/ inverted)
5. ..... mirrors should be used to obtain inverted images. (concave/ convex)

2) Select the correct answer.

1. Which one of the following is not a property of an image formed by a plane mirror?

i. Ability to take onto a screen ii. Upright

iii. Equal to the size of object iv. Laterally inverted

2. A property of an image, formed by a convex mirror is;

i. Inverted ii. Ability to take onto a screen

iii. Smaller than object iv. Larger than object

3. The angle between two plane mirrors, to obtain three images, should be;

i. 60^0 ii. 45^0 iii. 90^0 iv. 30^0

4. The type of mirrors that should be used to form an inverted image on a screen is;

i. Convex ii. Concave iii. Plane iv. All given above

Technical Terms

Umbra	-	ඡායාව	-	நிழல்
Penumbra	-	උපඡායාව	-	அயல்நிழல்
Image	-	ප්‍රතිබිම්බය	-	விம்பம்
Reflection	-	පරාවර්තනය	-	தெறிப்படைதல்
Convex mirror	-	උත්තල දර්පණය	-	குவிவாடி
Concave mirror	-	අවතල දර්පණය	-	குழிவாடி
Plane mirror	-	තල දර්පණය	-	தளவாடி
Convergent	-	අභිසාරී	-	விரிகற்றை
Divergent	-	අපසාරී	-	குவிகற்றை
Periscope	-	පරීක්ෂය	-	கலையுருக்காட்டி
Kaleidoscope	-	බහුරූපේක්ෂය	-	சூழ்காட்டி

10 The Correct Use of the Microscope

Many measures have been taken from ancient times to observe small things by magnifying them. Different equipment have been produced and used for this purpose.

Do Activity 10.1 by using instruments or materials that can be supplied from your home environment easily.

Activity 10.1

You will need :-

A closed glass bottle filled with water, a glass sphere, a glass slide with a drop of water, a sealed transparent polythene bag filled with water, a water filled filament bulb sealed with wax

Method :-

Observe small letters by using each of the above instruments or substances.

You can observe that small letters are magnified in size.

One common feature of above mentioned instruments is the presence of water. Did you find out another feature common to the all instruments?

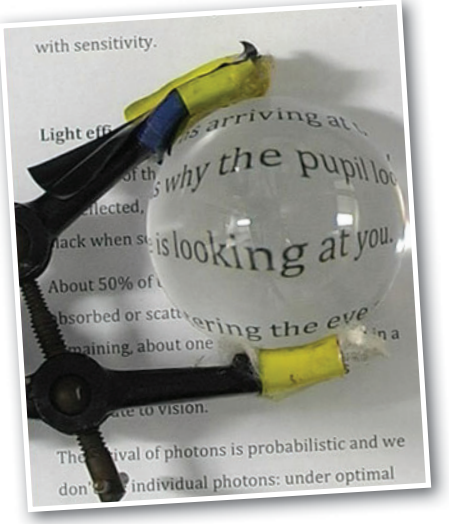


Figure 10.1 ▲ Letters are magnified when seen through a glass sphere

You must have observed that the other common feature of the above instruments was the curved surface or the **protruded** surface. You might now understand that these curved surfaces can be used to magnify small items. **Convex lens** is an instrument with these features, which is found in school laboratory.

10.1 The simple microscope

A hand lens/simple microscope can be produced by fixing a handle with a frame to a convex lens.

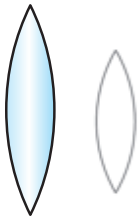


Figure 10.2 ▲ Convex lens

Figure 10.3 ▲ Hand lens made by using convex lenses

Activity 10.2

Observe the letters that you used in the above activity using a hand lens. Make sure to adjust only the object (letters) and do not change the distance between the lens and the eye.

Figure 10.4 ▲ Observing an object through a lens

You will observe that when the distance between hand lens and letters are increased gradually the letters will appear more and more larger. At a particular limit the letters will be seen with a maximum size. When the distance is increased further, the letters will become unclear.

Therefore, you will realize that there should be a constant distance kept between the object and the lens when an object is observed by a hand lens.

10.2 Magnification of a microscope and resolving power

Magnification

You might have observed that the size of the letter will be increased by several times. The number of times a specimen is magnified is known as the **magnification** of the lens or the magnifying power.

Activity 10.3

Observe different things by using a hand lens. Use a table to record your observations. Try to separate things which cannot be seen through your naked eye clearly.

Table 10.1

Materials observed	Observation
1. Soil sample	Sand particles with different sizes, gravel and several insects are observed.
2.	
3.	

Resolution

Resolution is the minimum distance by which two points must be separated in order to be seen as two distinct points.

There should be at least a minimum distance of 0.1mm between two adjacent points to distinguish between them by the naked eye. Now you will realize that the resolution of the naked eye is around 0.1mm.

Activity 10.4

You will need :- Colourful picture from a paper, a hand lens

Method :-

Observe a picture from the paper by a hand lens.

Present your observations to the class.

You will observe that the respective image is composed of a large number of small dots seen by the naked eye. You will see these dots very clearly by a hand lens. Now you will realize that the resolving power of the hand lens is higher than the resolving power of the naked eye.

Activity 10.5

You will need :- A white paper, a pencil or a pen, two hand lenses

Method :-

- Mark two dots with a minimum distance of not touching each other.
- Observe the distance between the two dots and their magnification by using a hand lens.

- Now use two hand lenses to observe the distance between two dots and their magnification.

It must be clear to you that the distance between the two dots is maximum and magnification is maximum when two lenses are used for the observations. Accordingly two hand lenses or two convex lenses can be used to obtain a higher magnification and high resolution.

Similarly, the compound light microscope is produced using two convex lenses with a high magnification power.

10.3 Compound light microscope

The compound light microscope is made of at least two convex lenses. Therefore, it is called the compound microscope and it is also called the light microscope because light is used.

Although micro organisms cannot be observed through a hand lens, they can be observed through the compound light microscope.
The maximum magnification of a developed compound light microscope is 2000 times and the maximum resolution is about 0.2µm (0.0002mm).

(The resolving power of the compound light microscope is 500 times than the naked human eye.)

Activity 10.6

Let us identify the parts of a compound light microscope

- Observe the compound light microscope in your school laboratory.
- Identify its main parts and functions with the assistance of your teacher.

Check whether you have identified all the parts of it.

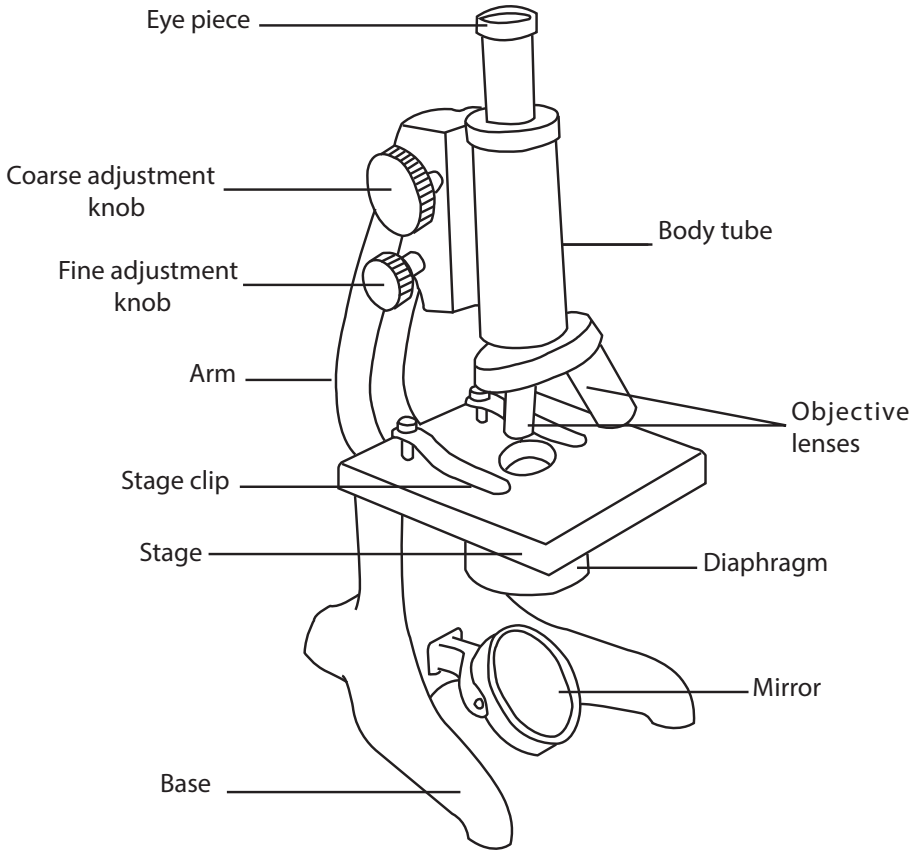


Figure10.5 ▲ Parts of a compound light microscope

Table 10.2 ▼ Parts of a microscope

Part of the microscope	Description
Eye piece	The closest lens to the eye. It is a convex lens. Magnification is shown as x5, x10 or x15. The specimen is magnified by using them.
Objective lens	The closest lens to the specimen. They are made of convex lenses. It magnifies the specimen. Mostly there are 3 objective lenses that can be identified. They are 1. Low power objective lens The lens with the least magnification (x4,x5 or x8) 2. Mid power objective lens The lens with the medium magnification (x10) 3. High power objective lens The lens with the highest magnification (x40)
Coarse adjustment knob	To adjust the body tube to get a clear image of the specimen
Fine adjustment knob	To adjust the body tube finally to get a more clear image of the specimen
Stage	To keep the slide of the specimen
Clip	To keep the slide of the specimen on the stage
Diaphragm	Control the amount of light obtained by the specimen
Mirror	Contains concave and flat surfaces. Focuse light on to the diaphragm
Base	Keeps the microscope on the supporting surface steadily

For extra knowledge

A bit from the history.....

The compound light microscope was first made by Zacharias Janssen. The magnification power of it was 9 times.

Zacharias Janssen and his microscope

But the honour of making a microscope first goes to the Dutch scientist Anton Van Leeuwenhook (1632-1723). The reason is that a world not seen from a naked eye was revealed by this instrument.

Anton Van Leeuwenhook and his microscope

Assignment 10.1

Prepare a report on the history of the microscope.

The correct method to use a compound light microscope.

1. First, keep the microscope on a horizontal table steadily (in a place where strong sunlight will not fall).
2. Use the coarse adjustment knob to bring the low power objective lens down.
3. Adjust the mirror and the diaphragm to obtain a clear spot of light by keeping the eye open on the eye piece.
4. Fix the prepared slide on the stage using clips.
5. Bring the low power objective lens upward using the rough adjustment until a clear image is obtained. Make sure that both the eyes are open during the process, keep the comfortable eye at a distance of 1 cm to the eye piece.
6. Try to obtain a clear image using fine adjustment.
7. A sharp and clear image could be obtained by adjusting the middle and high power objective lens if necessary.
8. Use the rough adjustment to bring the low power objective lens upward after the observation.
9. Remove the specimen from the stage. Use a dry piece of cotton cloth to wipe the stage.

Factors to be considered in the use of microscope

1. When the microscope is carried from one place to another, the arm of the microscope should be held by your familiar hand and the other hand should be kept below the base of the microscope. Then hold the microscope towards your body.

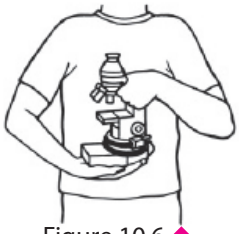


Figure 10.6

2. Wipe the stage using a clean piece of cotton after using the microscope.
3. If the microscope is not used for a long period of time, the lenses should be removed and place inside a desiccator with silica gel or anhydrous calcium chloride.
4. The microscope should be stored in an upward position to minimise the accumulation of dust.
5. Use clear slide and coverslips in the observation.
6. Do not change the lenses of one microscope to another.

Calculation of the magnification power of a microscope

The magnification power means the number of times that the speciman is subjected to be magnified. The number of times is obtained by multiplying the magnification of the eye piece with the magnification of the objective lens.

Magnification of microscope	=	Magnification of eye piece	×	Magnification of objective lens

Question :-

The magnification of eye piece was x10 and the magnification of the objective lens was x40 in a particular microscopic observation.

What is the magnification of the microscope?

Magnification of the microscope = Magnification of the eye piece × Magnification of the objective lens

= 10 × 40

= 400

The magnification should be stated in the following way when linear (line) diagrams are drawn upon observing a specimen.

Magnification of the eye piece	×	Magnification of the objective lens	×	The number of times the visual image is magnified when drawn

Observation of plant and animal tissue with a compound light microscope.

Activity 10.7

You will need :-

A compound light microscope, slides, cover slips, a leaf of betel, a leaf of Rhoeo, a bulb of onion, a slide made of cheek cells

Method :-Observe above mentioned types of plant and animal cells/ tissues by various magnification powers. Draw line diagrams and mention the magnification of each of them.

The Figure 10.8 show the linear diagrams of the Rhoeo lower epidermal tissue observed through a light microscope.

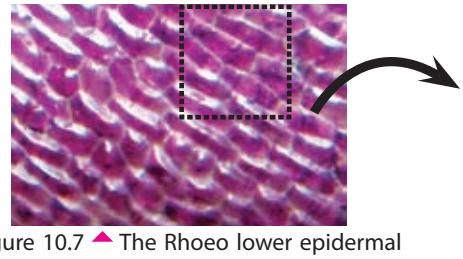


Figure 10.7 ▲ The Rhoeo lower epidermal tissue under low power of microscope (10 x 4)

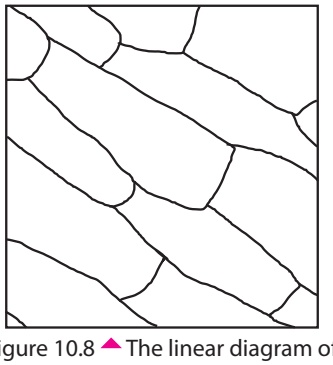


Figure 10.8 ▲ The linear diagram of the Rhoeo lower epidermal tissue (10 x 4 x 3)

The Figure 10.10 show the linear diagrams of the human cheek cells observed through a light microscope.

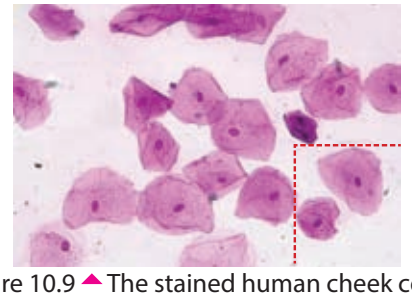


Figure 10.9 ▲ The stained human cheek cells under high power of light microscope (10x40)

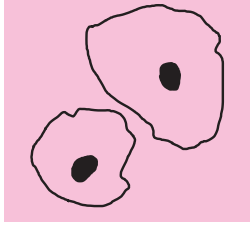


Figure 10.10 ▲ The linear diagram of the human cheek cells (10x40x2)

Assignment 10.2

List the differences between the simple microscope and the compound light microscope.

10.4 The electron microscope

An electron beam is used in the electron microscope instead of light rays.
The maximum magnification power of an electron microscope is 500 000 times. The maximum resolution of it is about 0.0005µm. This is 200 000 times than a healthy human eye. Although virus cannot be observed by a light microscope, they can be observed by an electron microscope.

Figure10.11 ▲ The electron microscope

Instances where the electron microscope is used,

- To observe the activity of pathogenic microorganisms (virus, bacteria) during research.
- To learn about internal structure of the cell in detail.
- To use in genetic research.
- To get information on crime investigation.

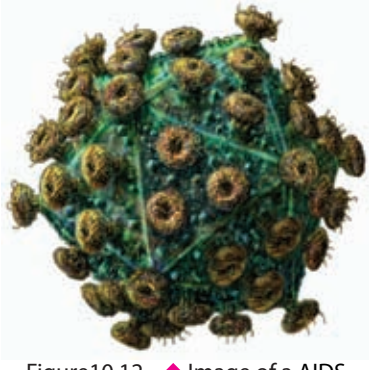


Figure10.12 ▲ Image of a AIDS virus under the electron microscope

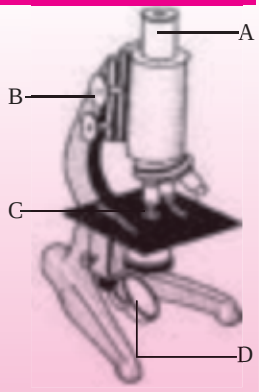
Figure10.13 ▲ Image of a Bacteria under the electron microscope

Summary

- Transparent objects with a convex nature can be used to observe small things in a large scale.
- Hand lens, light microscope, and electron microscope are three optical instruments used to observe small things in a large scale.
- The magnification of a light microscope is obtained by multiplying the magnification of the eye piece with the magnification of the objective lens.
- The minimum distance that should be present to clearly distinguish between two adjacent points or objects is called resolution.
- The electron microscope has a higher magnification and a higher resolving power than the light microscope.

Exercise

1. Name the parts labelled as A to D in the picture of the microscope.



2. The magnification of the eye piece was x5 and the magnification of the objective lens was x40 in a particular microscopic observation. What is the magnification of the microscope?

3. Write two factors to be considered when a microscope is used.

Technical Terms

Simple microscope	-	සරල අණ්වීක්ෂය	-	எளிய நுணுக்குக் காட்டி
Magnification	-	විශාලනය	-	உருப்பெருக்கம்
Resolution	-	විභේදනය	-	பிரிவலுத் தன்மை
Resolving power	-	විභේදන බලය	-	
Compound light microscope	-	සංයුක්ත ආලෝක අණ්වීක්ෂය	-	ஒளி நுணுக்குக் காட்டி
Eye piece	-	උපනෙත	-	பார்வைத் துண்டு
Objective lens	-	අවනෙත	-	பொருள் வில்லை
Electron microscope	-	ඉලෙක්ට්‍රෝන අණ්වීක්ෂය	-	இலத்திரன் நுணுக்குக் காட்டி

SCIENCE

Part - II

Grade 7

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The National Anthem of Sri Lanka

Sri Lanka Matha
Apa Sri Lanka Namo Namo Namo Namo Matha
Sundara siri barinee, surendi athi sobamana Lanka
Dhanya dhanaya neka mal palaturu piri jaya bhoomiya ramya
Apa hata sepa siri setha sadana jeewanaye matha
Piliganu mena apa bhakthi pooja Namo Namo Matha
Apa Sri Lanka Namo Namo Namo Namo Matha
Oba we apa vidya
Obamaya apa sathya
Oba we apa shakthi
Apa hada thula bhakthi
Oba apa aloke
Apage anuprane
Oba apa jeevana we
Apa mukthiya oba we
Nava jeevana demine, nithina apa pubudukaran matha
Gnana veerya vadawamina regena yanu mana jaya bhoomi kara
Eka mavakage daru kela bevina
Yamu yamu vee nopama
Prema vada sema bheda durerada
Namo, Namo Matha
Apa Sri Lanka Namo Namo Namo Namo Matha

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ආනන්ද සමරකෝන්

ஒரு தாய் மக்கள் நாமாவோம்
ஒன்றே நாம் வாழும் இல்லம்
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Be a light to the country as well as to the world.

Message from the Hon. Minister of Education

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This book is a product of free education. Your aim must be to use this book properly and acquire the necessary knowledge out of it. The government in turn is able to provide free textbooks to you, as a result of the commitment and labour of your parents and elders.

Since we have understood that the education is crucial in deciding the future of a country, the government has taken steps to change curriculum to suit the rapid changes of the technological world. Hence, you have to dedicate yourselves to become productive citizens. I believe that the knowledge this book provides will suffice your aim.

It is your duty to give a proper value to the money spent by the government on your education. Also you should understand that education determines your future. Make sure that you reach the optimum social stratum through education.

I congratulate you to enjoy the benefits of free education and bloom as an honoured citizen who takes the name of Sri Lanka to the world.

Akila Viraj Kariyawasam
Minister of Education

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The educational objectives of the contemporary world are becoming more complex along with the economic, social, cultural and technological development. The learning and teaching process too is changing in relation to human experiences, technological differences, research and new indices. Therefore, it is required to produce the textbook by including subject related information according to the objectives in the syllabus in order to maintain the teaching process by organizing learning experiences that suit to the learner needs. The textbook is not merely a learning tool for the learner. It is a blessing that contributes to obtain a higher education along with a development of conduct and attitudes, to develop values and to obtain learning experiences.

The government in its realization of the concept of free education has offered you all the textbooks from grades 1-11. I would like to remind you that you should make the maximum use of these textbooks and protect them well. I sincerely hope that this textbook would assist you to obtain the expertise to become a virtuous citizen with a complete personality who would be a valuable asset to the country.

I would like to bestow my sincere thanks on the members of the editorial and writer boards as well as on the staff of the Educational Publications Department who have strived to offer this textbook to you.

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Introduction

This textbook was compiled by the Educational Publications Department in accordance with the syllabus prepared by the National Institute of Education for the use of Grade seven students in the Sri Lankan school system with effect from 2016. An effort has made here to arrange the subject content to suit the national educational goals, common national competencies, the objectives of teaching science and the content of the syllabus.

The subject of science directs the student towards a more active learning process in a manner as to develop knowledge, skills and attitudes needed for a developmental scientific thought.

In the compilation of this textbook, subject content is largely arranged based on experiences of daily life. It has contributed to prove the fact that the subject of press; science is very much closer to the day-to-day life.

The compilation of this textbook based on activities is a distinctive feature. The activities are prepared based on the scientific method in order to develop knowledge, skills and attitudes. Activities that can be performed individually at home as well as in school are incorporated here. We believe that learning through activities would contribute to create a liking and an interest in the child towards learning science.

At the end of each chapter, a summary, a series of exercises and a glossary were included. It enables the student to identify the important details of the chapter as well as to self evaluate the achievement of learning outcomes.

For the purpose of directing the student to study further about the subject matter, more information is included in the "For extra knowledge". It is given only to broaden the subject area of the child and certainly not to ask questions at term tests.

Assignments and projects are given with the purpose of directing the student towards an explorative study. It enables the students to develop the higher order skills such as application, analysis and synthesis of the concepts achieved from the lesson.

We strongly believe that the duty of the teachers who teach science is to direct the student for self learning instead of teaching the student using traditional teaching methods. This textbook can be utilized by the teachers as a learning tool to execute their teaching role properly.

We would like to bestow our sincere thanks on the senior lecturer Asoka De Silva of the National Institute of Education and the professional writer Dr. K. Ariyasinghe.

We kindly request you to forward your comments and suggestions on this textbook to the Educational Publications Department.

Board of Writers and Editors

Contents

11 Sound

11.1 Production of sound

We hear various sounds frequently in our day-to-day life. Sounds of birds, sounds of radio and television are some of them. Let us do Activity 11.1 to find out more about the sounds we hear.

Activity 11.1

You will need :- A speaker, some pieces of polystyrene, connecting wire, dry cells

Method:-

- Connect the wires to the terminals of the speaker. Place the pieces of polystyrene on the speaker.
- Connect one of the two free ends of wires to the dry cell.
- Touch the other free end of the wires with the other terminal of the dry cell and remove at once. (Do not keep the wires connected to the dry cell)
- Keep on touching and removing one end of the wires to the dry cell and record your observations.

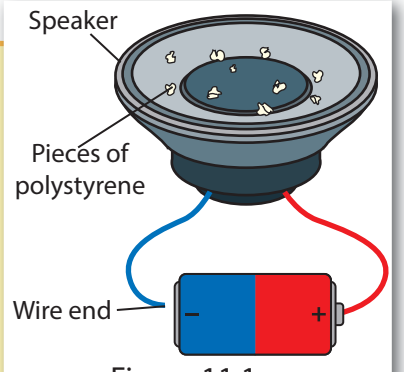


Figure 11.1

It is observed that pieces of polystyrene are thrown up, only when the speaker is connected to the dry cell.

It means that pieces of polystyrene are thrown up, only when the speaker produces sound. This happens because of the to and fro movement of the speaker cone. This type of a speedy to and fro movement is known as **vibration.**

It is clear that sound is generated due to the vibration of an object.

Let us do Activity 11.2 to find out more about the generation of sound.

Activity 11.2

You will need :- A tuning fork, a laboratory stand, a piece of thread, a small polystyrene ball

Method:-

- Hang the small polystyrene ball (using a piece of thread), as shown in the figure.
- Vibrate the tuning fork and bring it close to the polystyrene ball.
- Observe what happens. (Repeat this several times.)

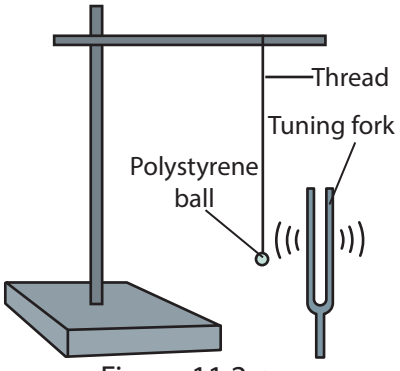


Figure 11.2

It is observed that the polystyrene ball moves towards the vibrated tuning fork, when it is brought closer. The polystyrene ball moves only when there are vibrations in the tuning fork.

Let us do Activity 11.3 to find out how our vocal sound is produced.

Activity 11.3

Method :-

- Place your fingers on the middle of your throat, as shown in the figure.
- Speak slowly and record what your finger feels.
- Speak louder and record the feeling to your fingers.


Figure 11.3

A light trembling is felt to the fingers when speaking slowly. This trembling increases when the voice is increased. This type of trembling is felt because of the vibration of membranes in the throat which are known as vocal cords. That means, our vocal sound is produced because of the vibration of vocal cords in the throat.

Assignment 11.1

- Design and carry out simple activities to show that sound is produced by vibrations.

Instruments that produce sound

Activity 11.4

You will need :- A guitar, a drum, a flute, a tabla

Method :-

- Produce sound using the sources of sound that you are provided with.
- Observe the part of sound source that vibrates when sound is produced.
- Tabulate your findings.

Figure 11.4

Compare your observations and findings with Table 11.1.

Table 11.1

Sound source/ Instrument	Part that vibrates
Guitar	Strings
Drum	Membranes
Flute	Air
Tabla	Membranes

Accordingly, it can be concluded that various musical instruments produce sound in various ways. Musical instruments can be categorized according to the part that vibrates to produce sound.

Figure 11.5 ▲ Instruments that produce sound by vibrating strings

Figure 11.6 ▲ Instruments that produce sound by vibrating membranes

Figure 11.7 ▲ Instruments that produce sound by vibrating air

Assignment 11.2

- Listen carefully to some natural and artificial sounds heard in the environment and list them out. Tabulate them as given below, with the part that vibrates when producing sound. (method of producing sound)

Sounds heard	Part that vibrates
School bell	

Assignment 11.3

- Make a list of musical instruments commonly used. Mention the method of producing sound in each listed instrument.

Assignment 11.4

- You may have heard of the sounds of some animals like mosquitoes, crickets and honey bees. Find out and record how they produce those sounds.

11.2 Propagation of sound

In classroom, you hear the voice of your teacher. In the space, two astronauts cannot hear each other's voice despite being closer to each other. What is the reason for this?

Let us do Activity 11.5 to find about it.

Figure 11.8 ▲ Astronauts

Activity 11.5

You will need :- A bell jar, a vaccum pump, an electric bell, few dry cells, connecting wire, a switch

Method :-

- Fix the electric bell in the bell jar as shown in the figure. Connect it to dry cells and to a switch to supply electricity.
- Connect a vacuum pump to the bell jar so that the jar can be evacuated.
- First, switch on the electric bell and record your observations.
- Then switch on the electric bell again while evacuating the air in the bell jar using the vacuum pump. Record your observations.

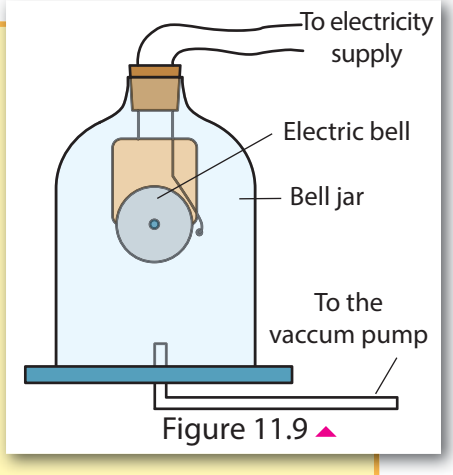


Figure 11.9 ▲

- Finally switch on the electric bell again when the bell jar is completely evacuated.
- Record your observations.

Ringing of the electric bell can be heard, when the bell jar is filled with air. When the air in the bell jar is gradually evacuating, the ringing of the bell gradually fades. When the air in the bell jar is completely evacuated, ringing completely fades off.

It is clear that ringing of the bell is heard only when there is air in the bell jar. Sound is not heard in a space where there is no air (in a vacuum). Thus, it is clear that a medium is essential for sound to propagate.

We hear the voice of the teacher in the classroom, because sound travels through air towards our ears. But two astronauts in the space cannot hear each other because there is no air in the space for sound to travel through. Thus, it is clear that sound does not travel in an empty space and sound needs a medium to travel. Sound which is produced from a sound source, travels from where it is originated know as propagation of sound. We hear different sounds when sound propagate through air to our ears.

Does sound travel only through air? Let us do Activity 11.6 to find out whether sound propagates through liquid and solid medium.

Activity 11.6

You will need :- Three plastic funnels, pieces of rubber tubes, membrances of balloons, a basin, Y tube, water

Method :-

- Assemble the equipment as shown in the figure. Hold the funnels A and B on your ears and dip the funnel C in the basin of water.
- Make a sound near the funnel in water.
- Record your observations.

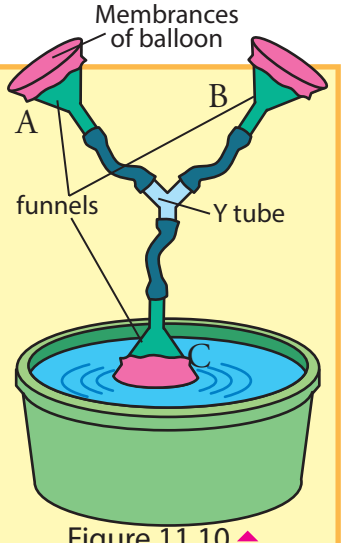


Figure 11.10

Sound, made in water, can be heard well through funnels A and B. Thus, it is clear that sound travels through liquids like water.

When shallow grounds are flooded, the sound of the vehicles travelling, the nearby roads can be heard louder than other days. The reason for that is sound travels through liquids faster than through air.

Let us do Activity 11.7 to find out whether sound propagates through solid medium.

Activity 11.7

You will need :-

Two empty yoghurt cups, a long thin thread (about 10 m)

Figure 11.11

Method :-

- Connect the bottoms of the two yoghurt cups to the thread as shown in the figure.
- Get the support of one of your friends for this activity and keep one of the yoghurt cups to your ear. Tell your friend to go as far as possible (Keeping the thread well stretched) and speak into the other cup.

You can hear your friend clearly. Thus, it is clear that sound propagates through the thread.

Let us do Activity 11.8 to test whether sound travels through solid medium.

Activity 11.8

You will need :- A clock giving 'tic' sound, a long table

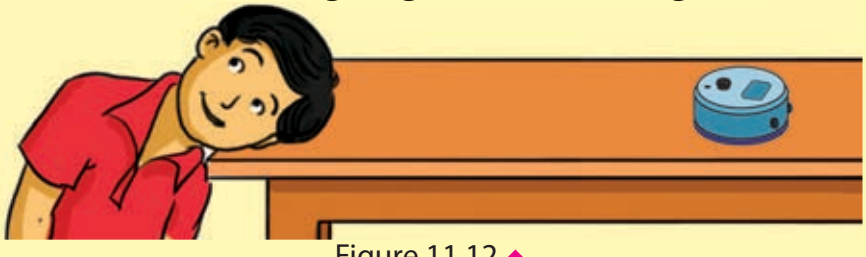


Figure 11.12

Method :-

- Keep the clock at one end of the table as shown in the figure.
- Stand at the other end of the table and test whether you can hear the 'tic' sound of the clock through air.
- Then place your ear on the table (as shown in the figure) and listen to the 'tic' sound of the clock.

The ‘tic’ sound of the clock can be heard through the table more clearly than through the air.

It is clear by the above activities, that sound propagates faster through solid medium than gaseous medium.

Assignment 11.5

- Plan and do different experiments to show that sound travels through solid and liquid medium.

For extra knowledge

- Stethoscope is an instrument used by physicians to diagnose patients. It functions in such a way that membrane of it vibrates accordingly to the sounds emitted by the heart, lungs and pulses.
- Sound pipes are used in the past in large buildings, ships and graphite mines to propagate sound to far away places.

The sound of a train coming from far away can be heard clearly when the ear is kept on the rail line. (This is a very risky activity) The reason for this is that sound travels faster through solids than through air.

Figure 11.13 ▲ Dolphins

We can hear the sound emitted by dolphins more intensely when we are in under water. But, the same sound emitted by them above water is heard very lightly. The reason for this is that sound travels faster through water than through air.

The speed of propagation of sound through solid, liquid and gaseous media is different from each other. Sound propagates faster through solids than through liquids and gases. Speed of sound is least in gaseous medium.

Table 11.2 ▼ Speed of sound according to medium

Medium	Speed of sound (Meters per second)
Air (gas)	330
Water (liquid)	1500
Steel (solid)	4500

During lightning, thundering sound and the light emit at the same time. But we hear the sound after few seconds of seeing light. The reason for this is that the speed of sound is slower than that of light.

For extra knowledge

When a canon is fired, the sound and the emission of smoke occur at the same time. But, a far away observer hears the firing sound after few seconds of seeing the smoke.

Summary

- Sound is generated due to vibrations.
- Sound can be produced by the vibration of membranes, strings or air.
- A medium is necessary for the propagation of sound.
- The speed of sound is different in different media.
- The speed of sound is faster in solid media than in liquid or gaseous media.

Exercises

1. Fill in the blanks using the words given below.
(speed, solid, a medium, vibrations, air)

I. Sound is produced by .....
II. ..... is necessary for the travelling of sound.
III. Sound travels more faster through ..... medium.
IV. Sound is produced in a flute by the vibration of .....

2. Select the correct answer.

I. The musical instrument that produces sound by vibrating strings is
i. Drum ii. Violin
iii. Pair of tabla iv. Flute

II. Consider the following statements.
A. Sound travels through a vacuum.
B. The speed of sound in air is higher than the speed of light.
C . A medium is essential for the travelling of sound.
The correct statements out of the above is/are;
i. A only ii. B only iii. C only iv. A and B only

Technical Terms

Sound	-	ධ්වනිය	-	ஒலி
Propagation of sounds	-	ධ්වනි සම්ප්‍රේෂණය	-	ஒலியின் கடத்துகை
Vibration	-	කම්පනය	-	அதிர்வு
Speed of sound	-	ශබ්දයේ වේගය	-	ஒலியின் கதி
Vacuum	-	රික්තය	-	வெற்றிடம்
Tuning fork	-	සරසුල	-	இசைக்கவர்
Vacuum pump	-	රික්ත පොම්පය	-	வெற்றிடப் பம்பி

12 Biological Processes

12.1 Organisational levels of life

The brick wall is made up of numerous bricks. In the same manner, a bee hive is composed of numerous hexagonal units

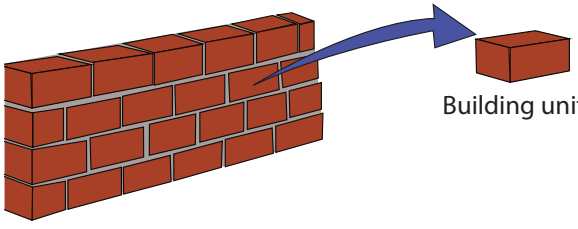


Figure 12.1 ▲ Wall made of bricks

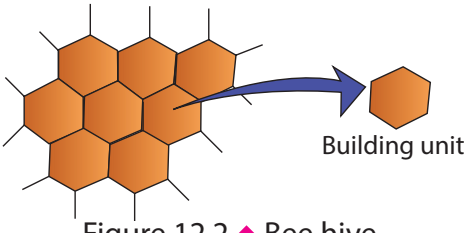


Figure 12.2 ▲ Bee hive

The living body contains large number of small building units. This basic building unit is called as the **cell**. Cell is the structural unit of living organisms.

Structural unit of plant is referred to as plant cell and structural unit of animal is referred to as animal cell.

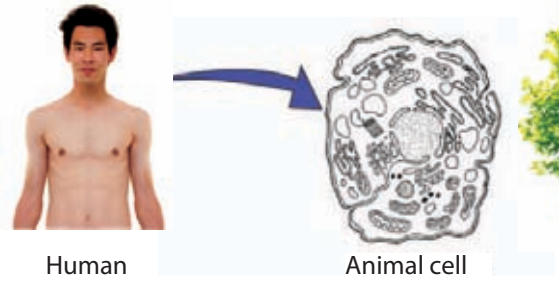


Figure 12.3 ▲ Animal cell

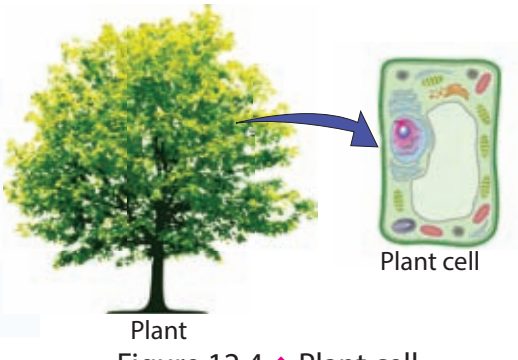


Figure 12.4 ▲ Plant cell

Recall the plant and animal cells observed in a previous lesson.

Activity 12.1

Observe the cells in lower epidermis of betel leaf and cheek cells, with the help of your teacher, using a light microscope.

Then try to identify plant cells and animal cells.

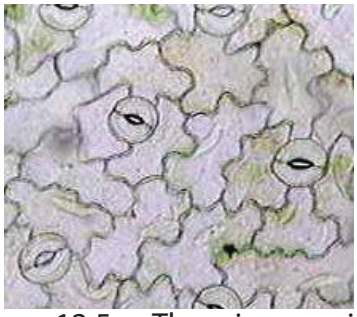


Figure 12.5 ▲ The microscopic view of cells in lower epidermis of a plant leaf

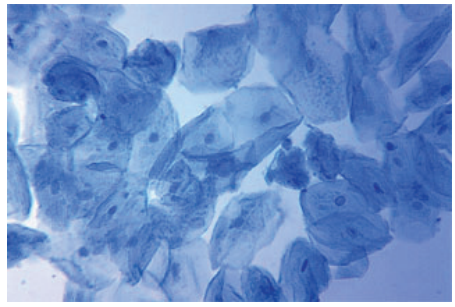


Figure 12.6 ▲ The microscopic view of stained human cheek cells

Assignment 12.1

- Collect historical information regarding discovery of cell and present them to the classroom.

Some of the organisms are represented by a single cell. Therefore, they show a simple organization. The basic functional unit of organism is called the **cell**. Some of the unicellular organisms are shown in Figure 12.7.

Amoeba *Euglena* *Paramecium*

Figure 12.7 ▲ Unicellular organisms (microscopic view)

Activity 12.2

Try to identify unicellular organisms by observing a sample of water taken from fresh water pond and hay stagnated in water.

The organisational levels in living body

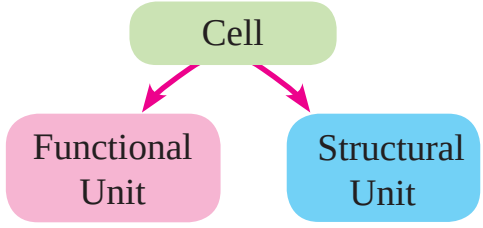
Except unicellular organisms, all other organisms are made up of few cells or large number of cells. They are called multicellular organisms. They show variations in levels of organisations. The level of organisation from simple level 'cell' to complex level 'organism' is given below.

Cells → Tissues → Organs → Systems → Organism

Each and every oragnisational level of above is discussed in this chapter.

Cell

The cell is the basic structural and functional unit of the life. Cell is the building unit of living organisms and the simple functioning unit too.



Tissue

A **tissue** is a group of cells, organised for a specific functions.

Activity 12.3

You will need :- A permanent slide or a picture of a muscle tissue, A Slide with a lower epidermis from a plant leaf

Method :-

Observe those specimens using a light microscope. Identify different kinds of cells in them.

Tissues have following features;

- Contain same shaped cells or different shaped cells
- Perform a common function

The following diagrams show different types of tissues found in living organisms.

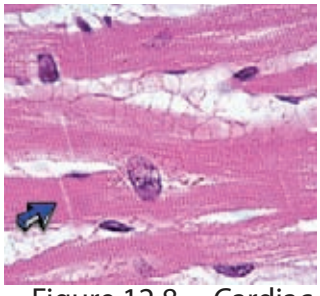


Figure 12.8 ▲ Cardiac muscle tissue

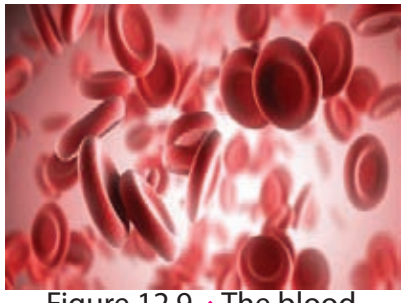


Figure 12.9 ▲ The blood tissue

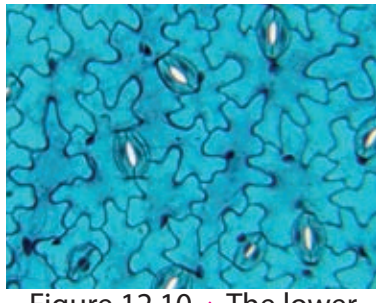
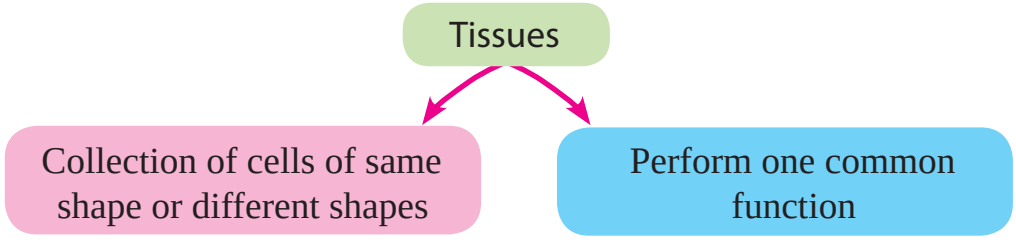


Figure 12.10 ▲ The lower epidermal tissue of the plant leaf



Organ

A collection of different tissues worked together to perform a specific function or functions is called an **organ.**

Human stomach is an organ and it is made up of different types of tissues.

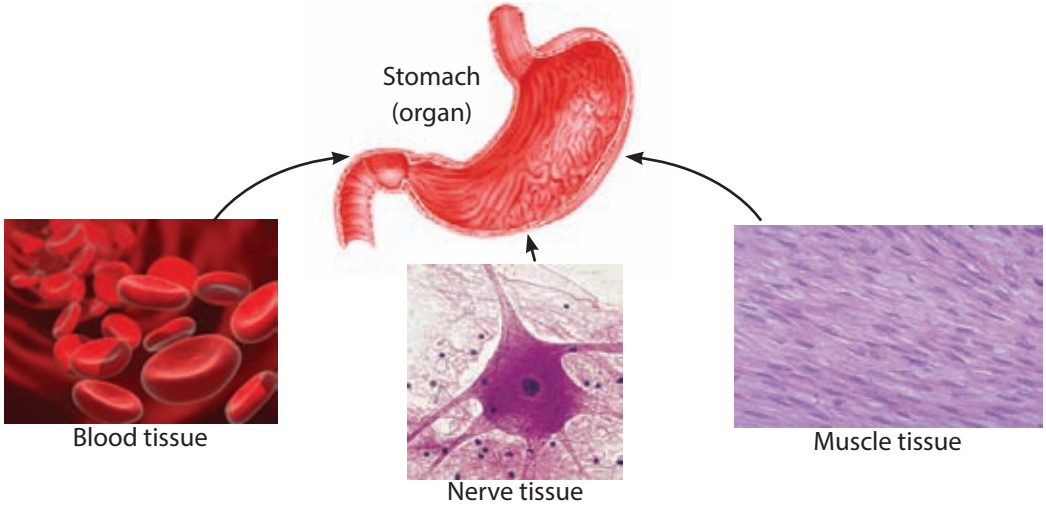


Figure 12.11 ▲ Different types of tissues in human stomach

Activity 12.4

You will need :- A plant like 'balsam' with a transparent stem, vessel filled with water, dye soluble in water (Kukul sayam)

Method :-

- Put the red dye into water vessel and prepare a red coloured solution.
- Put the balsam plant into the vessel in such a way that root system is submerged.
- When the stem becomes red, take a thin section of the stem and observe the cross section through light microscope.

- The tissue which transports water and mineral is coloured in red. It is called the **xylem tissue**.
- **Phloem tissue** located outside the xylem tissue, transports food in the plant body.
- In addition to xylem and phloem, there are many other tissues in stem.

Now, it is clear to you that the plant stem is composed of many tissues.

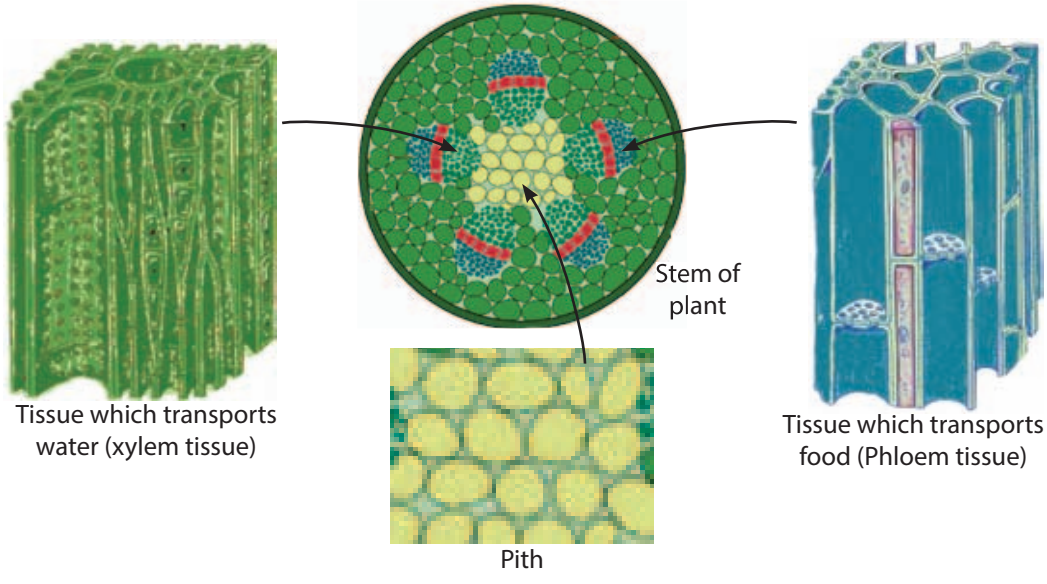


Figure 12.12 ▲ Different types of tissues in plant stem

Activity 12.5

Observe a cross section of the plant leaf using a permanent slide and study different types of tissues.

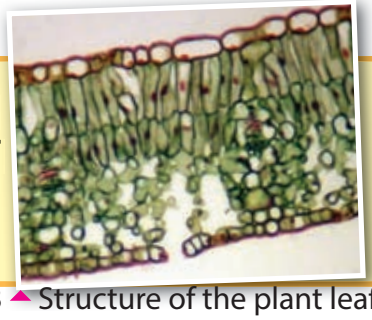
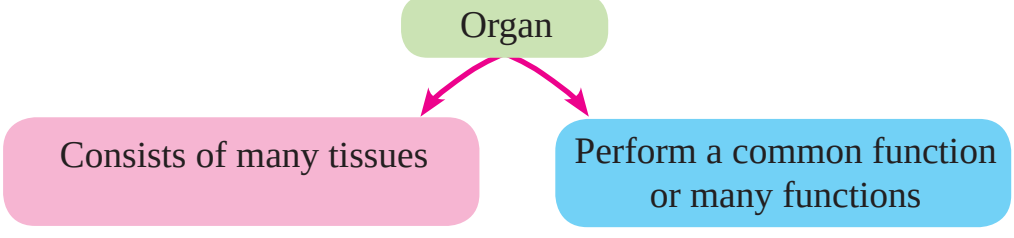


Figure 12.13 ▲ Structure of the plant leaf



System

A group of organs that perform a special function or functions is called the **system.** Figure 12.14 shows the arrangement of digestive system of man by combination of different organs.

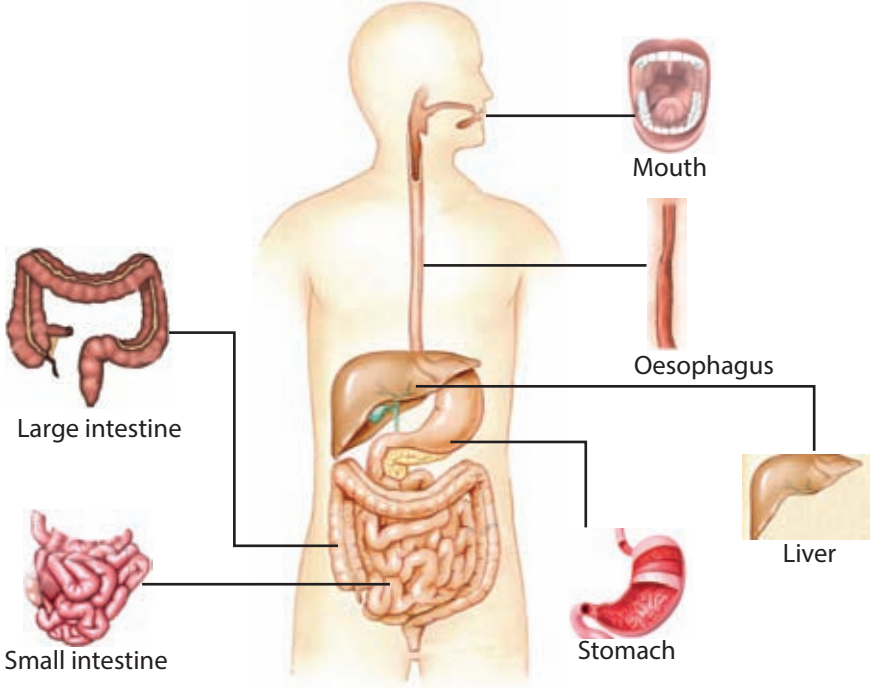


Figure 12.14 ▲ The digestive system of man

System

Consists of many organs

Perform a common function or many functions

Activity 12.6

- Uproot a plant like "kuppameniya"/"kuppaimani"
- Observe different organs and systems of it
- Draw a line diagram and try to identify the organs and two systems in the plant body

Compare your findings with the Figure 12.15.

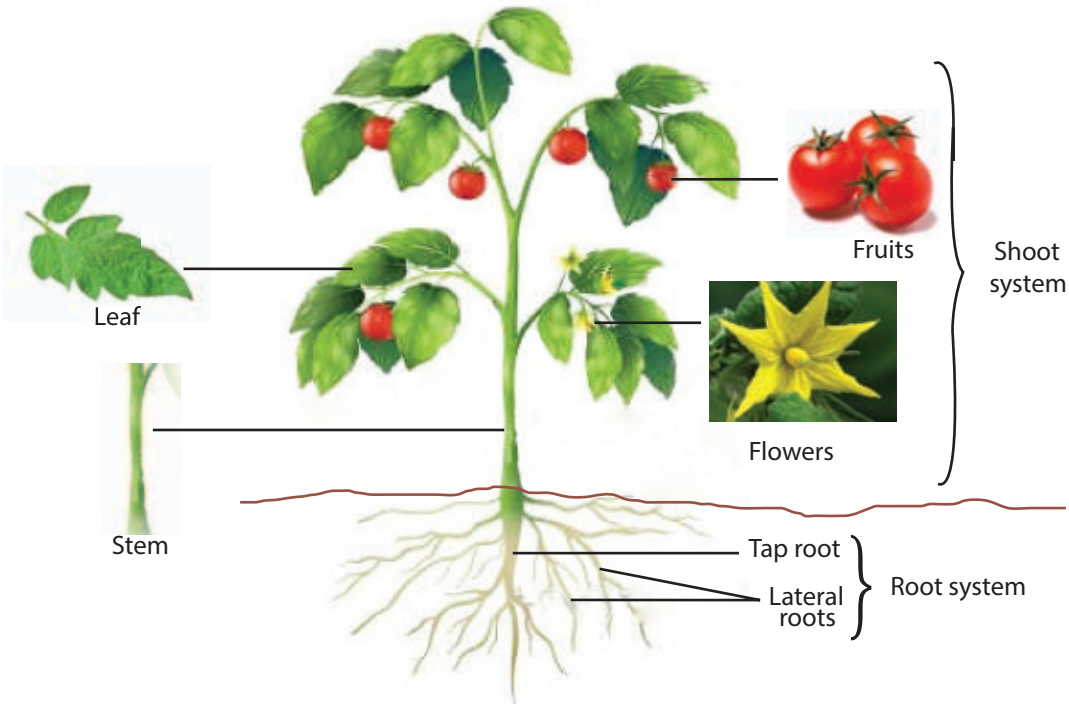


Figure 12.15 ▲ Organs and systems in the plant body

You may notice that there are two main systems in the plant body. They are;

1) Root system
2) Shoot system

Assignment 12.2

Try to identify main organs and systems in grass. Draw a line diagram and name them.

Organism

All systems of the body join together and form the body of organism. Figure 12.16 clearly display that the way of forming organism by combination of all systems in the body.

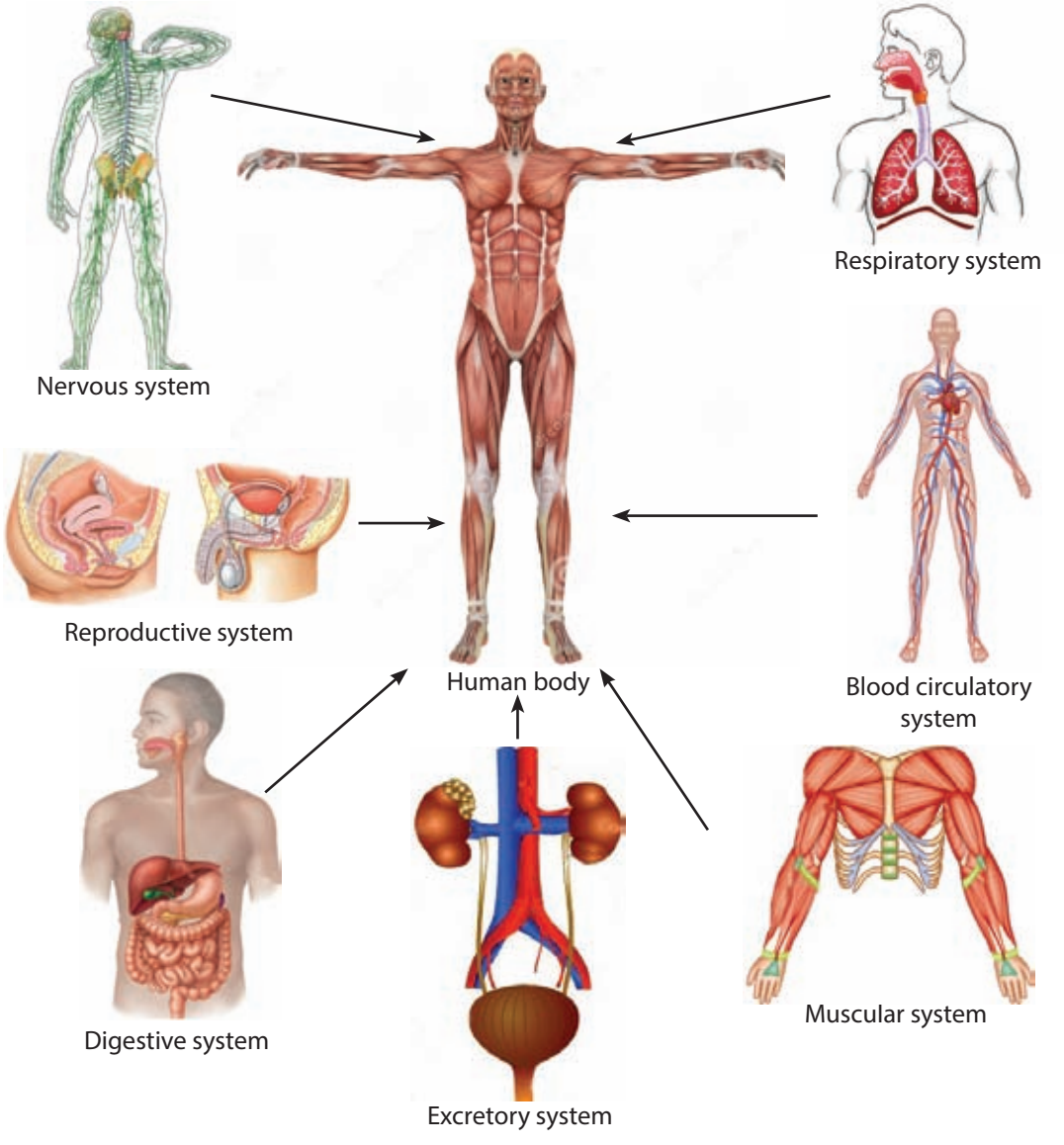


Figure 12.16 ▲ Organism consists of organ system

You have studied that the plant body is made of two systems

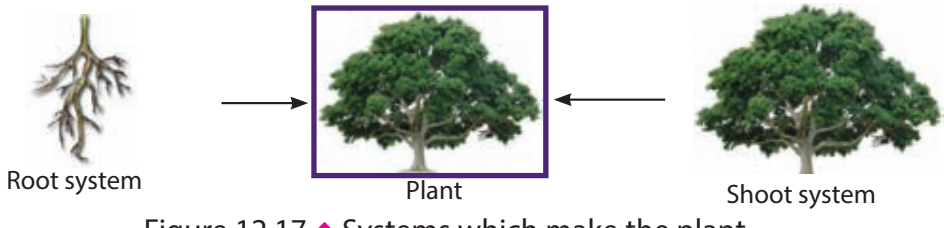


Figure 12.17 ▲ Systems which make the plant

Activity 12.7

- Make a model to show the organisational levels of life.
- Present it in the classroom.

12.2 Systems of the human body

There are various biological processes like respiration, food digestion, movement, excretion taken place in the human body. Energy is needed for the function of them.

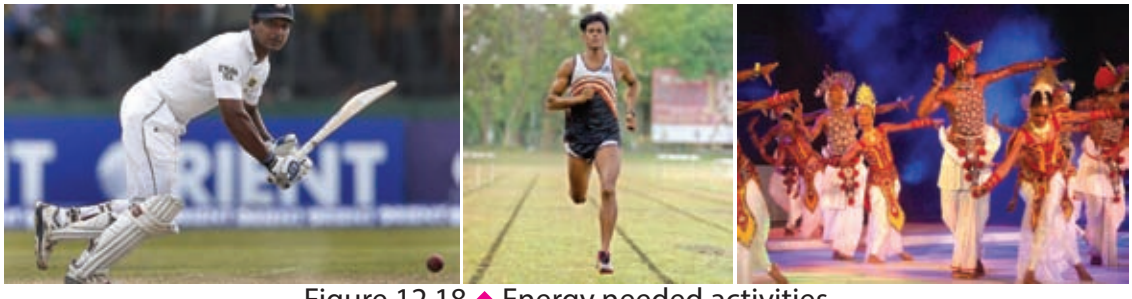


Figure 12.18 ▲ Energy needed activities

Several reactions take place in our body when we are active with activities like playing or dancing and even when we are not doing anything. Energy needed for such activities is taken by burning or reacting the food we take, with oxygen in the cells.

Consider how oxygen and simple food is supplied to man for obtaining energy. Complex food are broken into simple components by the digestive system. Respiratory system supplies oxygen for our body. Let us study about these systems further.

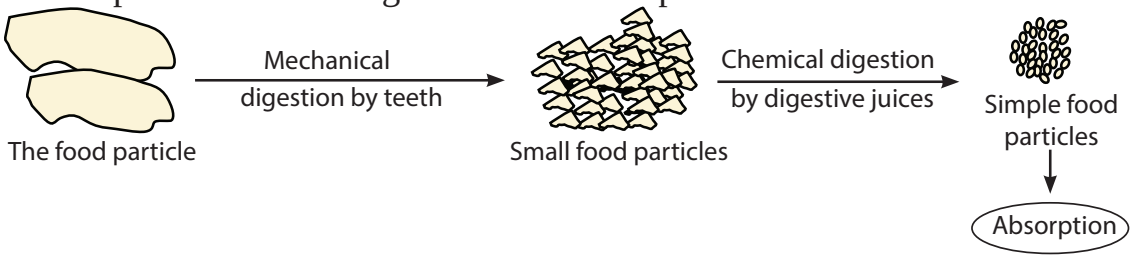
Digestive system of man

The food we eat are normally big pieces and complex. Therefore, that cannot be absorbed to the body. The breaking down of food into simple substances is called digestion. Then it can be absorbed easily.

Teeth break down food into smaller particles inside the mouth. Then, it makes easier to digest food (by digestive juices) when pass along

digestive tract. Then, those small food pieces are digested to simple pieces by the digestive juices secreted from different parts of the digestive system. Finally they are absorbed to the body.

The process of food digestion can be simplified as follows



It is not easy to digest all types of food that we take in a single organ. Therefore, the digestion takes place step-by-step in several organs.

Two main functions of the human digestive system are mentioned below.

1) Conversion of complex food into simple substances.

2) Absorption of simple food to the body.

Activity 12.8

Try to identify main parts of the human digestive system by using a model or a diagram.

Compare your findings with the following Figures.

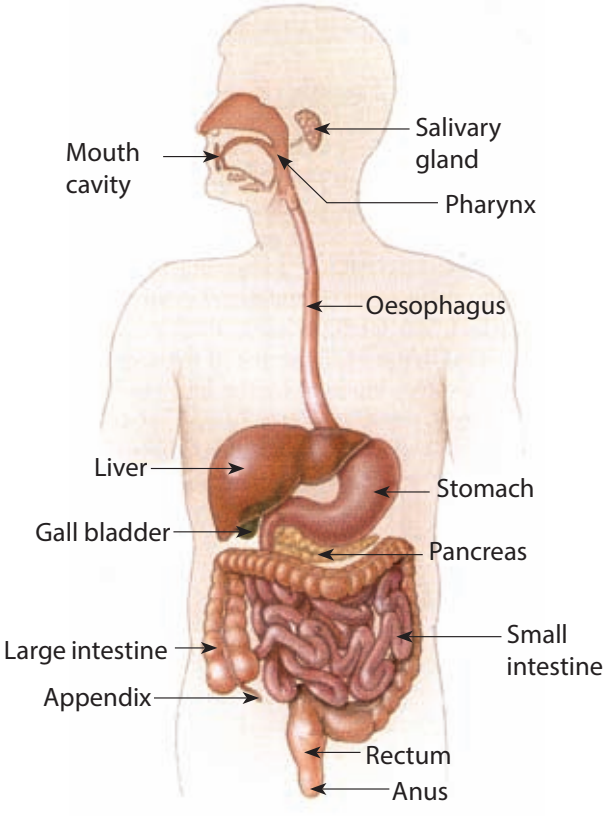


Figure 12.19 ▲ Digestive system of man (Three dimensional structure)

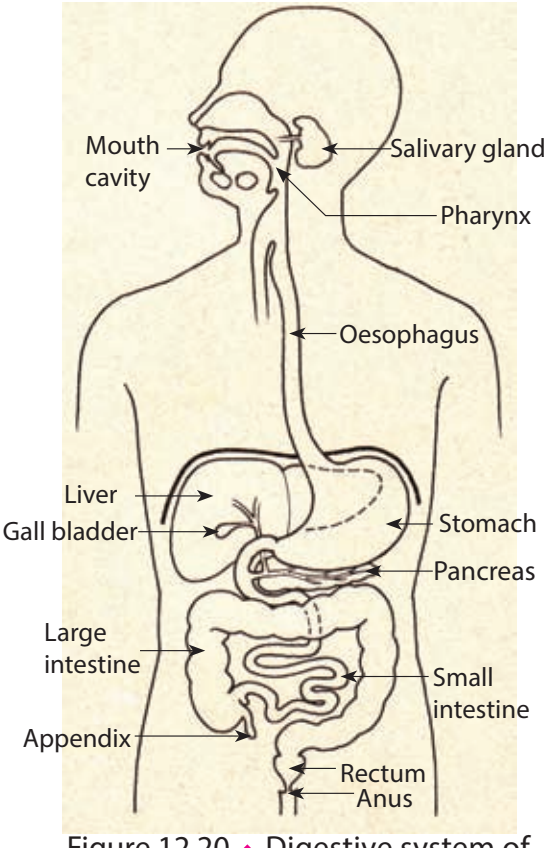


Figure 12.20 ▲ Digestive system of man (Line diagram)

The structure and functions of the organs in the digestive system is given in Table 12.1.

Table 12.1 ▼ Functions of various organs of the human digestive system

Organ	Structural features & functions
Mouth cavity	• Breakdown of food particles into small pieces by teeth (mechanical digestion) • Secretion of saliva by salivary glands to aid digestion of food • Beginning the chemical digestion of food by saliva • Mix food with saliva by tongue (Chemical digestion)
Pharynx	• It is the common cavity of the digestive tract and the larynx • Push the food in the mouth cavity into oesophagus
Oesophagus	• Push the food in mouth cavity to the stomach.

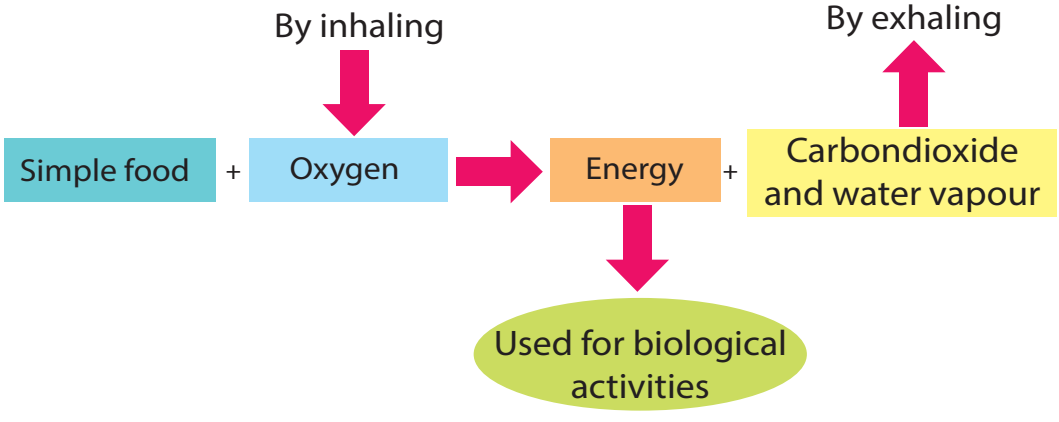
Stomach	• It is a muscular sac. • Digestion is efficient due to acidic nature of digestive juices. • Food is stored in the stomach for about three hours. • Food is digested further by mixing of digestive juices in the stomach.
Small intestine	• It is about 6 m long folded tube. • Different types of food digest with digestive juices. • Digestion is completed and absorb digestive food to the body. • There are finger like projections called villi in the small intestine to increase the surface for efficient digestion.
Large intestine	• It is shorter than the small intestine, but considerably broad. • Absorption of water.
Anus	• It is the terminal end in the digestive system. • Semisolid faeces is released through anus.

Activity 12.9

Make suitable models to demonstrate the human digestive system.

Human respiratory system

Why do animals breath? You can find out the answer by studying following flow chart.



It shows that as a result of respiration, energy is generated and that energy is used for biological activities in the body.

The process where energy is produced by reacting or burning of simple foods with oxygen is called respiration.

The intake of air needed for respiration is called **inspiration** (inhalation or breathing in).

During respiration carbon dioxide and water vapour are formed and the elimination of these byproducts is called **expiration** (exhalation or breathing out).

Accordingly, the air exchanging system in the body is called the respiratory system. Engage in Activity 12.10 to identify the major parts of the respiratory system.

Activity 12.10

Try to identify the main parts of the human respiratory system, using a model or figure in the laboratory

Use the following diagram to identify the parts of it.
Compare your findings with Figure 12.21.

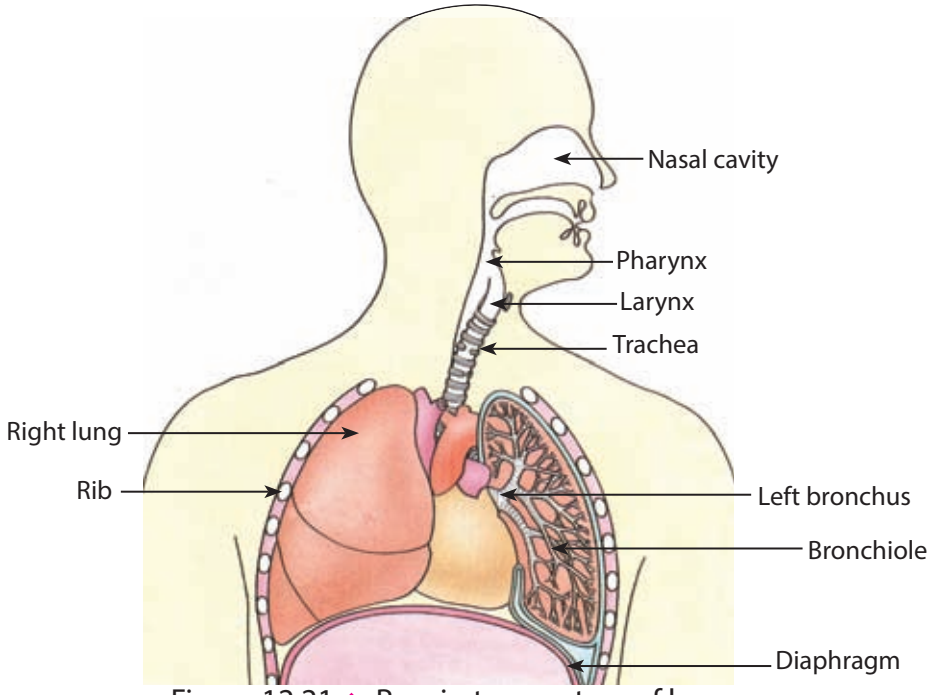


Figure 12.21 ▲ Respiratory system of human

The following flow chart explains the activities of human respiratory system.

Intake of inhaled air into the nasal cavity

↓

Air gets warmed and filtered while passing along the nasal cavity, add wetness

↓

Air moves from parynx to trachea through larynx

↓

Air passes through the right and left bronchus

↓

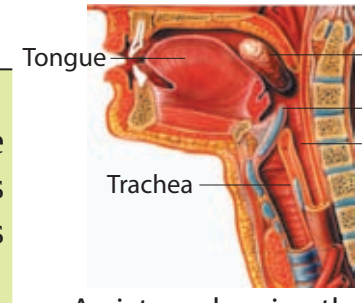
Air reaches the lungs and exchange of air. O_2 in air is absorbed to the blood then CO_2 in blood is moved to lungs

↓

Then exhaled air passes back through the same path and released out through the nasal cavity

For extra knowledge

The food bolus that comes to the pharynx passes to the oesophagus without going into the trachea as the door of the trachea is closed.



A picture showing the action of epiglottis

Activity 12.11

Make a suitable model to demonstrate the human respiratory system. Present it to the class.

The diagram given below shows the model of human respiratory system.

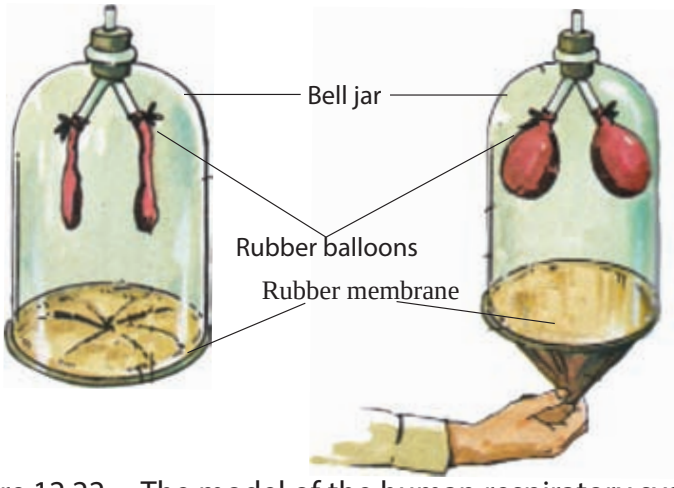


Figure 12.22 ▲ The model of the human respiratory system

Activity 12.12

The rubber membrane, rubber balloons, the bell jar and Y tubes in the figure 12.22 are represent the different parts of the human respiratory system. Name them.

Summary

- The cell is the basic structural and functional unit of the living beings.
- Order of the organisational levels of the body is,
  Cells → tissues → organs → systems → organism
- Food is converted to a simple state by the digestive system so that food could be absorbed to the body.
- The respiratory process contributes to obtain oxygen needed to produce energy and excretion of carbondioxide and water vapour.
- The process where energy is produced by reacting simple foods with oxygen is called respiration.

Exercise

(01) Select the correct answer.

(i) The structural and functional unit of living being is called

1. Cell 2. Tissue 3. Organ 4. System

(ii) The tissue reponsible to transport water and mineral through the plant body is

1. Xylem 2. Phloem 3. Cortex 4. Pith

(iii) The organ that does not belong to the digestive system is

1. Liver 2.Trachea 3. Large intestine 4. Anus

(iv) The organ common to the respiratory system and the food digestive system is

1. Larynx 2. Bronchioles 3. Oesophagus 4. Pharynx

(v) Which of the following process does not occur in the small intestine?

1. Collection of pancreatic juice and bile
2. Completion of digestion
3. Absorption of digested food
4. Food is broken down to small pieces

(02) Complete the following table

Organ	Function
1. Stomach	
2. Small intestine	
3. Large intestine	
4. Nasal cavity	
5. Larynx	

Technical Terms

Cell	-	සෛලය	-	கலம்
Tissue	-	පටකය	-	இழையம்
Phloem tissue	-	ප්ලෝයම පටකය	-	உரிய இழையம்
Xylem tissue	-	ශෛලම පටකය	-	காழ் இழையம்
Stomach	-	ආමාශය	-	இரைப்பை
Lungs	-	පෙනහැලි	-	சுவாசப்பை
Liver	-	අක්මාව	-	ஈரல்
Bronchus	-	ශ්වාස නාළය	-	வாதனாளி
Root system	-	මූල පද්ධතිය	-	ஆணிவேர்
Pharynx	-	ග්‍රසනිකාව	-	தொண்டை
Intestine	-	අන්ත්‍ර	-	சிறுகுடல்
Root system	-	මූල පද්ධතිය	-	ஆணிவேர்
Shoot system	-	ප්‍රරෝහ පද්ධතිය	-	அங்குரத் தொகுதி
Digestion	-	ජීරණය	-	உணவுச் சமிபாடு
Respiration	-	ශ්වසනය	-	சுவாசம்

13 Atmosphere

13.1 Layers of atmosphere

What can you see when you look at the sky? If it is the afternoon you will see the clouds or the blue sky. If it is the night time you will see stars, planets and sometimes the moon. You see all these objects through the atmosphere. But you cannot see the atmosphere.

Atmosphere is a thin layer of gases that surrounds the earth. It spreads up nearly 700 km from the earth surface.

In grade six you have learnt that gases have a mass. The weight of the gases above us make a pressure on us and all the things around us. This pressure is known as **atmospheric pressure.** The atmospheric pressure is measured in milli bars (mb) to forecast the weather.

The height of a certain place from the sea level is known as **altitude.** According to the altitude the pressure and the temperature change in different levels of the atmosphere. Based on these differences the atmosphere is divided into five layers.

The five layers of the atmosphere, spread out from the earth surface respectively are;

1. Troposphere
2. Stratosphere
3. Mesosphere
4. Thermosphere
5. Exosphere

The Figure 13.1 shows the layers of the atmosphere.

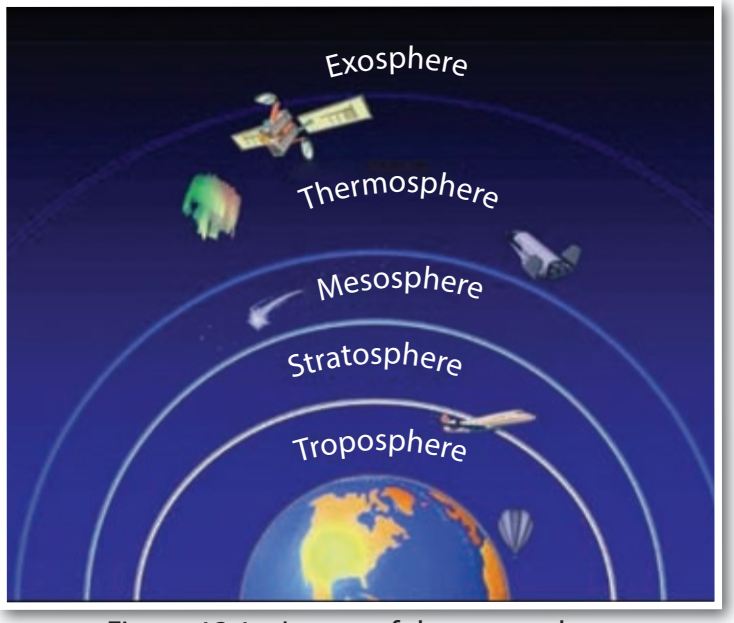


Figure 13.1▲ Layers of the atmosphere

Troposphere

This is the lowest layer of the atmosphere. Near the equator it spreads up to about 15 km from the sea level, but near the poles the height of the troposphere is about 8 km.

Nearly 75% of the air in atmosphere is in the troposphere. Most of the water vapour and dust particles are found in this layer. All the weather changes take place in the troposphere.

Helicopters, parachutes and aeroplanes travel through this layer.

Stratosphere

This layer is 15 - 50 km high above from the sea level. The air is dry as there is very little water vapour in this layer. There are no storms or turbulences in stratosphere. Therefore, jets fly through this layer.

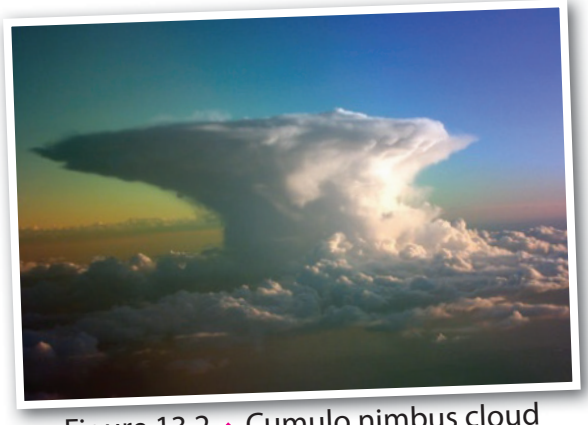


Figure 13.2 ▲ Cumulo nimbus cloud shaped as an anvil

The ozone layer lies in the stratosphere. This is a special layer which prevents the ultra violet rays (UV rays) of the sun falling on the earth.

The Cumulo nimbus closer to the stratosphere takes the shape of an anvil. The reason for this shape is the blowing wind in one direction. Rain with thundering and lightning can be expected after forming these type of clouds.

Activity 13.1

Go out on a day with a clear sky. Identify a Cumulo nimbus cloud. Continue looking at that cloud. You will see that the height of the cloud increases and the top of the cloud gets flat. Then, check whether the cloud has got a shape of an anvil.

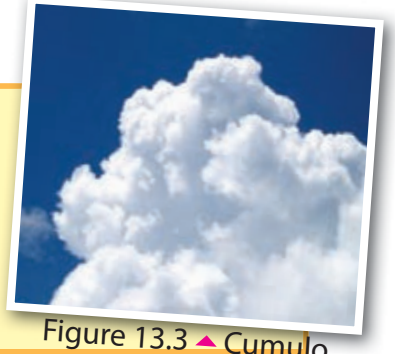


Figure 13.3 ▲ Cumulo nimbus cloud

Mesosphere

Mesosphere extends from 50-80 km up from the sea level. This is the coldest layer among the five layers of the atmosphere. In this layer water vapour gets frozen into ice clouds. When the sun sets the rays fall on these clouds and you can see these clouds during the night time.

Figure 13.4 ▲ Clouds in mesosphere

Thermosphere

Thermosphere lies from 80-120 km up from the sea level. The air particles in this layer absorb the sun's rays. So, that the temperature in this layer is very high. International Space Station is situated in this layer.

Figure 13.5 ▲ Aurora borealis

Both the special sceneries called Aurora borealis and Aurora australis occur in the thermosphere. **Aurora borealis** can be seen near the northern pole and **Aurora australis** can be seen near the southern pole.

Exosphere

The thinnest layer in the atmosphere is the exosphere. This layer is 120 km high above from the sea level. There is no certain border between the exosphere and the space.

Activity 13.2

Building up a model of the layers in atmosphere

You will need :- A4 sheets, a pair of scissors, a drawing pin, a picture of the earth

Method :-

- Cut a big circle from a A4 sheet, in maximum size.
- Cut another circle with a radius 2 cm less than the big circle.
- Cut three more circles and each circle should have a radius 2 cm less than the other circle.
- Paste the picture of the earth on the smallest circle and write the word troposphere on it.
- Keep the circles on the big circle and fix them together using the drawing pin as shown in the figure.
- Write the names of the layers on the circles.
- Show your model to the teacher.

Figure 13.6 ▲ Layers of the atmosphere

Differences of temperature and pressure in layers of the atmosphere

We live in the troposphere. What happen to the temperature and the pressure when you go up the troposphere?

The Table 13.1 shows the altitude, annual average temperature and average atmospheric pressure in some cities of Sri Lanka.

Table 13.1 ▼ Weather report in some cities

City	Altitude (m)	Average temperature (°C)	Average pressure (mb)
Colombo	01 m	27.4	1110
Kandy	500 m	24.6	956
Nuwara Eliya	1868 m	15.9	813

(Source - Meteorological department)

According to the Table 13.1 it is clear that when the altitude is increased the temperature and the pressure are decreased.

Assignment 13.1

Select some cities in Sri Lanka with different altitudes. Listen to the weather reports in media and record the temperature of those cities for a week. Show the collected details using a graph.

The Figure 13.7 shows the changes of temperature in different layers in the atmosphere.

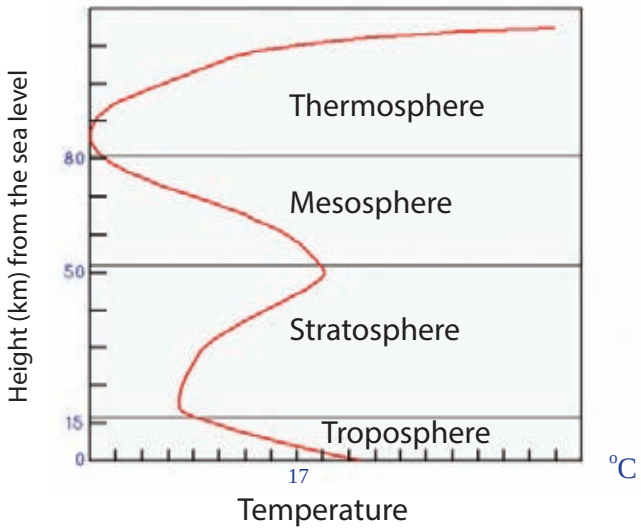


Figure 13.7 ▲ Changes of temperature in different layers

These are the details that can be taken from the Figure 13.7.

- Temperature decreases with increasing height up to the troposphere. This is how it happens. The land and the sea get heated from the sun's heat. So, the temperature near the land is high. The temperature of the troposphere gradually decreases away from the land.
- Temperature increases in the stratosphere when go up. The reason is the presence of the ozone layer which absorbs the ultra violet rays of the sun. Then, stratosphere gets heated and temperature increases.

For extra knowledge

- Temperature in the mesosphere decreases from bottom to top. It is because there is very less amount of ozone gas to absorbs sun's ultra violet rays. The lowest temperature is recorded in the top of the mesosphere. It is about - 90 °C. This temperature is less than the temperature in the Antarctic which has the lowest temperature on the earth.
- Temperature in the thermosphere increases form bottom to top again. The reason is the air particles in this layer highly absorb the sun's heat.
- Temperature increases more and more when it is close to the exosphere.

13.2 Air and its components

Troposphere is the most important atmospheric layer for the living beings on the earth. The Table 13.2 and the Figure 13.8 show the component of the troposphere.

Table 13.2 ▼ Gases in the troposphere

Gas	Volume as percentage
Nitrogen (N_2)	78%
Oxygen (O_2)	21%
Argon (Ar)	1%
Carbon dioxide (CO_2)	
Water vapour	
Other gases	

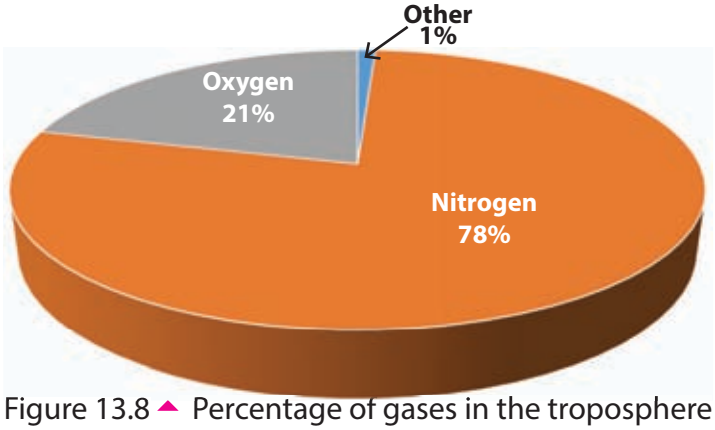


Figure 13.8 ▲ Percentage of gases in the troposphere

Nitrogen and Oxygen are the main components in the troposphere.

For extra knowledge

Nitrogen

Nitrogen is the most abundant gas in the air. Under atmospheric conditions it has low reactivity.

Oxygen

The most important gas in the air. It is essential for respiration of organism. Oxygen is necessary for combustion. Therefore, it is introduced as a **supporter of combustion.** Oxygen gas can be separated from air. Oxygen gas has many uses;

e.g.:- • To aid patients with respiratory difficulties
- For welding and cutting metals using oxy-acetylene flame
- For divers and astronauts

Argon

Argon is the third most common atmospheric gas. This is an inert (noble) gas. It does not have any chemical reactions with any other elements. Therefore, it is used for many purposes.

e.g.:- • To fill filament bulbs
- To produce electric lamps, which emit orange colour

Carbon dioxide

Carbon dioxide is essential for photosynthesis in green plants. This gas is used to extinguish fire and it helps to maintain the temperature on the earth. Carbon dioxide composition is higher in exhaled gas than in inhaled gas.

Functions of the atmosphere

- Atmosphere provides oxygen needed for the respiration of all animals and plants.
- Plants get carbon dioxide from the atmosphere for photosynthesis.
- Nitrogen, which is a very important element for the growth of the plants is provided to soil by the atmosphere.

- Heat, light and ultra violet rays of the sun fall on earth through the atmosphere. The ozone layer protects the earth from expose to these harmful rays.
- Atmosphere prevent emission of heat of earth to the space. So atmosphere keep our earth in warmer condition. The moon has no atmosphere. Therefore, the moon is very hot during the day time and very cold during the night time.
- Hydrological cycle is very important for the existence of living beings. Water vapour in the atmosphere is very essential for the maintenance of the hydrological cycle.
- Atmosphere helps for birds and some insects to fly.
- It supports for communication among people.
- Atmosphere protects the earth from meteors. Meteors fallen on to the earth from the space get heated and burnt in the atmosphere. If meteors fall on to earth it will damage many lives.

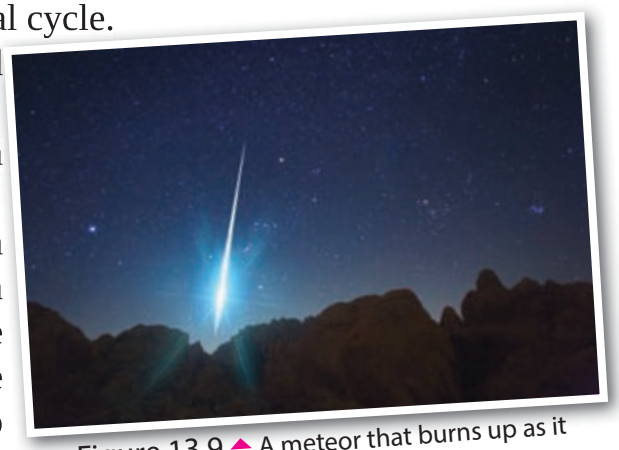


Figure 13.9 ▲ A meteor that burns up as it enters the atmosphere

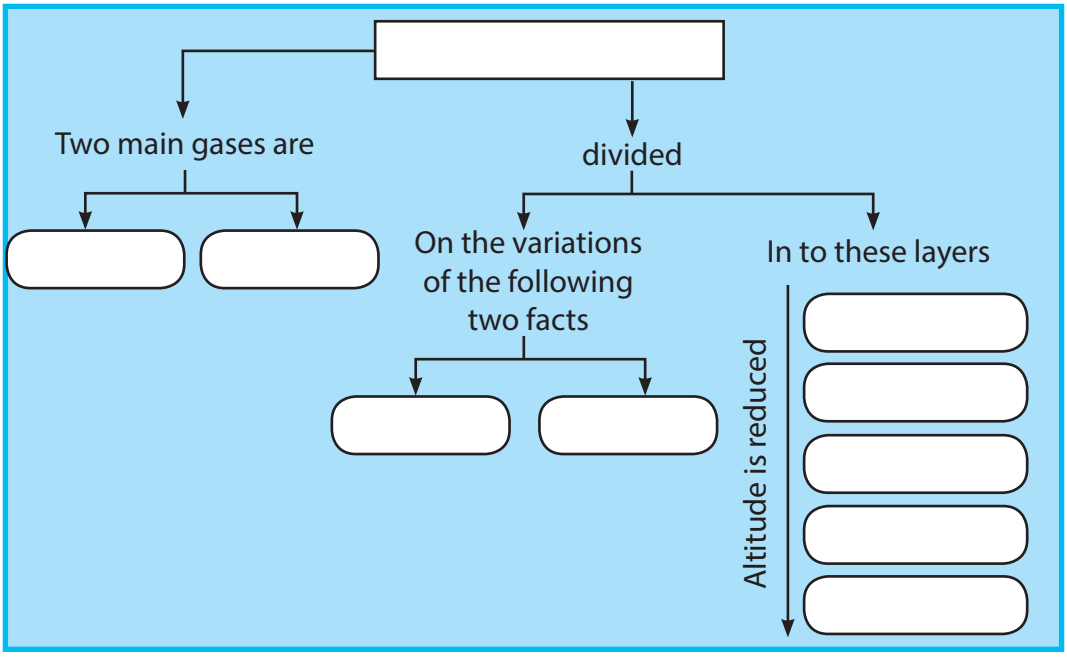
Assignment 13.2

Use the above facts and make a report of the problems that may occur on the earth "If the atmosphere is disappeared instantly".

Assignment 13.3

Following is a concept map prepared by some grade seven students about the atmosphere. Fill in the blanks using the given words.

pressure, stratosphere, mesosphere, atmosphere, nitrogen, troposphere, oxygen, thermosphere, exosphere, temperature



Air pollution

Air pollution is "adding substance to the atmosphere, changing it's composition and causing harmful effects on organism".

The substances that cause air pollution can be divided into two groups. They are;

1. Gaseous pollutants
2. Particulate pollutants

Some gaseous pollutants are carbon monoxide, sulphur dioxide, nitrogen dioxide etc.

Some particulate pollutants are carbon particles, lead particles, cement dust, insecticide droplets, unburnt fuel droplets and asbestos particles.

Activity 13.3

After travelling in a bus clean your face with a white handkerchief. What can you see?

The reason for the things in the handkerchief is the particulate pollutants in the air.

Some ways of air pollution are given in Figure 13.10.


Combustion of fuel in vehicles


Combustion of fuel in factories


Exhaust from thermal power stations


Burning of waste materials


Volcanic eruptions


Destroying forests

Figure 13.10 ▲ Ways of air pollution

Assignment 13.4

A green test must be done for a vehicle to get it licensed. In this test the particles in the emitted smoke of the vehicle is tested. Find out what are the particles tested in the green test and record your details.

Followings are some adverse effects of air pollution.

- Climatic changes on the earth
- Wide variety of health problems related to respiratory system
- Acid rain
- Increase of temperature in the atmosphere
- Reduction of the clarity and the transparency of the air

Here are some precautions that can be taken to minimize the air pollution.

- Tune the vehicle engines
- Minimize the combustion of fossil fuel
- Use eco-friendly energy sources
- Release the fumes of factories through a filter

- Re-cycle the waste without burning
- Protect the forests
- Re-forestation

There are artificial respiratory chambers in some cities to fascilitate respiration, where there is an air pollution.

Therefore, it is our duty to protect the atmosphere.

Summary

- The temperature and the pressure changes from the surface of the earth to the upper atmosphere. Depending on these changes the atmosphere is divided into five layers.
- Troposphere, stratosphere, mesosphere, thermosphere and exosphere are the five layers of the atmosphere.
- Atmosphere protects the earth and the living beings on it from the external dangers.
- Nitrogen and Oxygen are the main components in the troposphere.
- The gases in the atmosphere help man in many ways.
- Air pollution is a problem that affects all of us.

Exercise

1. State whether the following statements are True (√) or False (x).

i) Temperature decreases from bottom to top of the troposphere ( )

ii) The earth is the only planet with an atmosphere that can sustain life. ( )

iii) Oxygen comprises the highest percentage in the atmosphere. ( )

iv) Temperature increases from bottom to top of the stratosphere. ( )

v) Climatic changes take place in the troposphere. ( )

vi) Stratosphere contains the most amount of ozone gas. ( )

vii) Mesosphere has the lowest temperature. ( )

Technical Terms

Atmosphere	-	වායුගෝලය	-	வளிமண்டலம்
Altitude	-	උන්නතාංශය	-	குத்துயரம்
Pressure	-	පීඩනය	-	அமுக்கம்
Temperature	-	උෂ්ණත්වය	-	வெப்பநிலை
Cumulo nimbus clouds	-	කැටි-වැහි වලාකුළු	-	திரண்முகில்
Aurora borealis	-	උත්තරාලෝකය	-	வடமுனைவுச் சோதி
Aurora australis	-	දක්ෂිණාලෝකය	-	தென்முனைவுச் சோதி
Recycling	-	ප්‍රතිචක්‍රීකරණය	-	மீள் சுழற்சி
Air pollution	-	වායු දූෂණය	-	வளி மாசடைதல்
Air pressure	-	වායු පීඩනය	-	வளி அமுக்கம்
Ozone layer	-	ඕසොන් ස්තරය	-	ஓசோன் படை
Ultraviolet rays	-	පාරජම්බුලකිරණ	-	கழியூதாக் கதிர்கள்

14 Heat and Temperature

14.1 Measuring temperature

Environment becomes warm during the day time due to the solar heat and gets cool in the night. Warmness and coldness are two sensations that we feel.

Let us do Activity 14.1 to find out more about warmness and coldness.

Activity 14.1

You will need :- Luke warm water and cold water, two equal size beakers

Method :-

- Get equal amounts of luke warm water and cold water into two beakers of same size (Get the help of your teacher when handling warm water).
- Dip your fingers into the water of the beakers.
- Record what you feel.

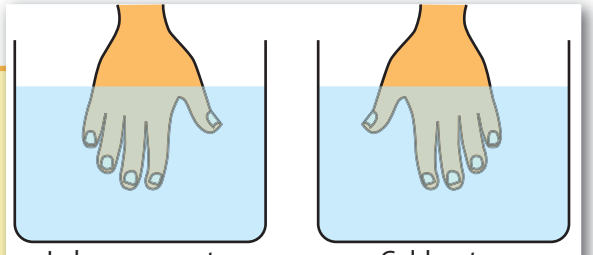


Figure 14.1

You may feel that there is a difference of warmth of the water in the two beakers. What is the difference between cold water and luke warm water? Water was heated on a hearth or on a burner. Then, water was warmed by receiving thermal/heat energy.

The measurement of warmness or coldness of a substance is known as its **temperature.**

Temperature of a substance decreases (cools) because of losing heat from that substance.

For extra knowledge

We feel warm when a heated object is being touched, because heat transfers from that object to our hand. Coldness is felt when a piece of ice is touched, because heat transfers from our hand to the piece of ice.

Measuring temperature by touching is not correct. Let us do Activity 14.2 to find out how to measure temperature correctly.

Activity 14.2

You will need :- Two small glass bottles with rubber stoppers, two empty tubes of ball point pens, two beakers, water, red ink

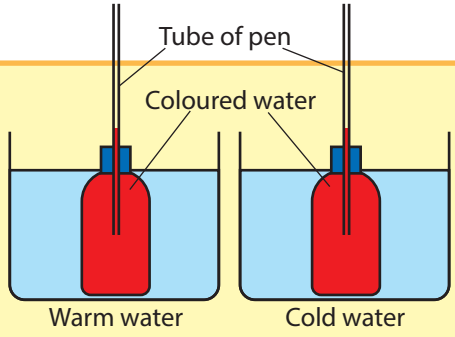


Figure 14.2

Method :-

- Fill the small glass bottles with coloured water and fix the empty tubes of ball point pens.
- Keep one of those bottles in a beaker filled with warm water and the other in a beaker filled with cold water.
- Observe what happens (Take care when using hot water).

Thermometers are used to measure temperature accurately and quantitatively.

It can be observed that;

The liquid column in the tube of the bottle, kept in warm water rises up.

The liquid column in the tube of the bottle, kept in cold water falls down.

Accordingly, heat transfers from warm water in the beaker to the water in the bottle. Then, the volume of water in the bottle increases to rise up along the tube.

Water in the bottle kept in the beaker of cold water, cools to decrease the volume (Contracts). Then the liquid column in the tube of that bottle falls down.

Increase of the volume of a liquid, by gaining heat is called the **expansion of the liquid**.

It is the property of expansion of a liquid, that is used in making thermometers.

14.2 Thermometers

A simple thermometer can be made and calibrated as indicated in the Figure 14.3 below.

A small glass bottle is filled with coloured water and the empty tube of a ball point pen is fixed into it. A paper strip on which a scale is marked, is glued to the tube.

The bottle, thus made is dipped in a beaker of water and the beaker is heated slowly.

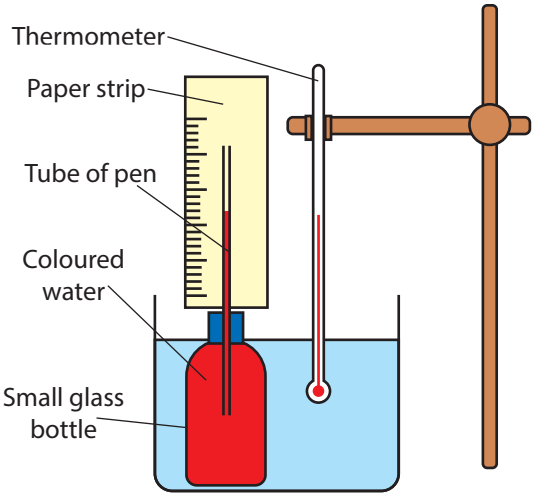


Figure 14.3

A mercury thermometer is dipped in the beaker of water. When the water is being heated, the temperature rises and the coloured water column in the pen tube goes up.

The position of upper end of the coloured water column is marked on the paper strip, for some temperature readings indicated by the thermometer. The relevant temperature also should be indicated at the mark. After marking several temperatures on the paper strip, a simple scale can be made. The water-bottle-thermometer thus made can be used to measure unknown temperatures in a short range. Here you can get an approximate value.

Assignment 14.1

- Mention the short comings of a thermometer as mentioned above. Give suggestions to overcome those shortcomings.

Mercury is used as the liquid of most thermometers used today. In addition alcohol is a liquid used in thermometers. Alcohol used in thermometers is coloured, for clear observation of the thin alcohol column.

Scales of thermometers

There are several scales used in present thermometers. Those are;

- Celsius scale
- Farenheit scale
- Kelvin scale

Units of measuring temperature in each scale is given in Table 14.1.

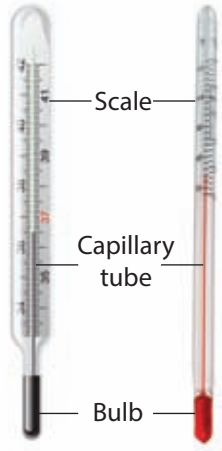


Figure 14.4 ▲ A mercury thermometer and an alcohol thermometer

Table 14.1 ▼

Temperature scale	Unit
Celsius	^{o}C
Farenheit	^{o}F
Kelvin	K

The international standard unit of temperature is Kelvin (K).

The equivalence between principal temperature scales are given in Table 14.2 for some temperatures.

Table 14.2 ▼

Temperature	Celsius scale (^{o}C)	Farenheit scale (^{o}F)	Kelvin scale (K)
Boiling point of water	100	212	373
Freezing point of water	0	32	273
Mean temperature of human body	36.9	98.4	309.9

Melting point and boiling point

There is a constant temperature, at which a soild substance changes to its liquid state .

Let us do Activity 14.3 to find out the constant temperature at which ice changes to water.

Activity 14.3

You will need :- A few pieces of ice cubes, a mercury thermometer, a glass funnel, a beaker

Method :-

- Put some piece of ice into the glass funnel and place the bulb of the thermometer in ice.
- Keep the funnel on the beaker and find the temperature of melting ice.

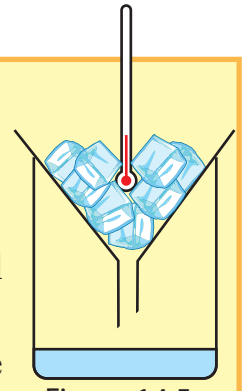


Figure 14.5

Here, it can be observed that the temperature at which ice is melting is 0 °C. Therefore, the constant temperature at which ice melts is called the melting point of ice.

Ice (solid) —Absorbing heat→ Water (liquid)

Melting point

The constant temperature at which a solid substance changes to its liquid state is known as the melting point of that substance.

When water is cooled, ice (solid) forms. The temperature at which water solidifies is also 0 °C and is known as the freezing point of water.

Melting points of some substances are given in Table 14.3.

Table 14.3

Substance	Melting point (°C) at 1 atm
Ice	0
Paraffin wax	60
Lead	317
Iron	1539

Boiling point

The boiling point can be simply explained as the constant temperture at which a liquid substance changes to its gaseous state.

Let us do Activity 14.4 to find out the boiling point of water.

Activity 14.4

You will need :- A boiling tube, a thermometer, some water, a burner, a laboratory stand

Method :-

- Take some water into a boiling tube, and prepare the set-up as shown in Figure 14.6.
- Heat the water for few minutes till it boils.
- Record the reading of the thermometer.

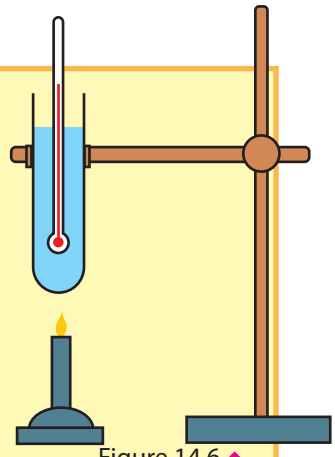


Figure 14.6

It can be observed that the temperature at which water boils (with bubbles evolving steam) is 100 °C. It can be concluded that there is an exact temperature of boiling water.

The constant temperature at which a liquid substance changes to it's gaseous state is known as the boiling point of that substance.

Table 14.4 Boiling point of some substances

Substance	Boiling point (°C) at 1 atm
Alcohol	77
Paraffin wax	370
Water	100
Lead	1744
Iron	2900

For extra knowledge

Celsius scale

This scale is prepared by dividing the range between the melting point of ice (0 °C) and the boiling point of water (100 °C) into 100 equal parts.

Farenheit scale

This scale is prepared by dividing the range between upper fixed point (212 °F) and lower fixed point (32 °F) into 180 equal parts.

The values of melting point and boiling point vary according to the atmospheric pressure. The values given in table 14.3 and 14.4 are measured at 1 atm of atmospheric pressure.

Using thermometer correctly

1. Thermometer should be held vertically. So, that the bulb of the thermometer is well in contact with the substance / liquid of which the temperature should be measured.
2. When taking the readings the thermometer should be adjusted to the eye level.
3. Eye should be kept correctly in line with the mercury column as shown in Figure 14.7.
   (Observe below or above is incorrect.)

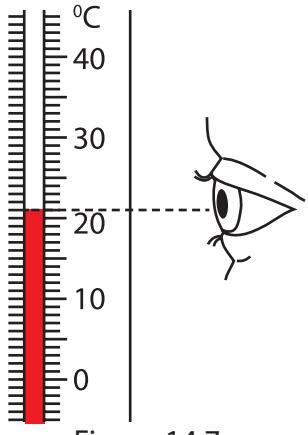


Figure 14.7

Assignment 14.2

Find out the factors, that should be considered when using a thermometer and make a report.

For extra knowledge

For the protection of the thermometer, it should be selected in such a way that the measuring temperature should be within the temperature range of that thermometer.

Engage in Activity 14.5 and Activity 14.6 to take the reading correctly.

Activity 14.5

You will need :- A beaker, water, a thermometer, a laboratory stand

Method :-

- Place the bulb of the thermometer at the middle of water and clamp it vertically.
- Observe the level of the mercury column and take the reading accurately as shown in Figure 14.8.

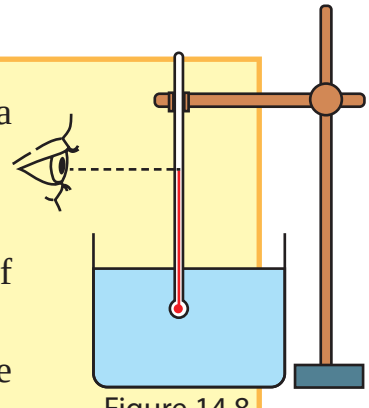


Figure 14.8

Activity 14.6

You will need :- A thermometer

Method :-

- Hold the thermometer vertically.
- Observe the mercury column accurately and take the reading.

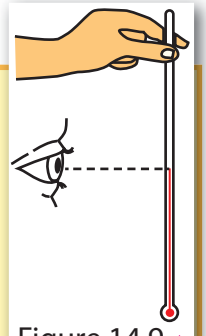


Figure 14.9

Assignment 14.3

Measure the temperature of air in some places like under a large tree, open air with lot of sunlight, near a water body.

Tabulate your readings

Activity 14.7

You will need :- A thermometer, a beaker full of soil

Method :-

- Fill the beaker with soil and dip the bulb of the thermometer well into the soil.
- After some time take the reading of the thermometer.

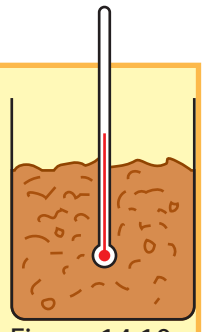


Figure 14.10

Soil temperature can be measured by the soil thermometer, in the natural environment as shown in Figure 14.11.

Figure 14.11 Measuring soil temperature

Assignment 14.4

Measure the soil temperature in the following places and tabulate the readings

- Under a large tree
- In a dry place
- In a place with sandy soil
- In a place with high moisture content

Clinical thermometer

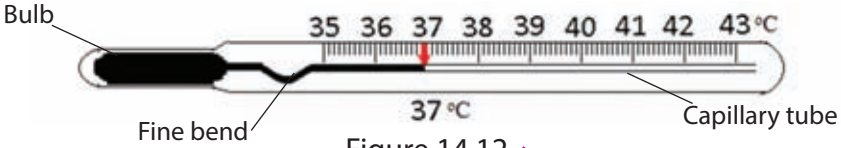


Figure 14.12

Special features of a clinical thermometer;

- There is a fine bend in the capillary tube containing mercury.
- The temperature range is short (35 °C - 43 °C)

Clinical thermometer is used to measure the body temperature.

Measuring body temperature, by using clinical thermometer;

- First wash the bulb of the thermometer with an antiseptic solution.
- Keep the bulb of the thermometer under the tongue of the patient for about two minutes as shown in Figure 14.13.
- Remove the thermometer from the mouth and take the reading accurately while holding it vertically.

Figure 14.13

(Body temperature of small babies can be measured by keeping the bulb of the thermometer under their arm pits for few minutes.)

Figure 14.14 Bend in the mercury column of clinical thermometer

There is a fine bend in the capillary tube of the clinical thermometer. It prevents the mercury column rising up or falling down before taking the measurements. Therefore, the reading of the temperature can be kept unchanged, even after the thermometer is removed from the mouth of the patient. Thermometer should be shaken well to send the mercury column down the bend before it is used for the next time.

Assignment 14.5

Measure and record the body temperature of your family members and some of your friends.

For extra knowledge

- Body temperature of a healthy person is 36.9 °C or 98.4 °F

- New types of thermometers are invented for measuring body temperature accurately.

Assignment 14.6

Collect and record information about modern equipment used to measure temperature.

14.3 Heat transfer

Heat is a type of energy. The sun is our largest heat source. Though the sun is some millions of kilometres away from the earth, we get solar heat very soon. This indicates that heat has travelled from the sun to the earth very quickly.

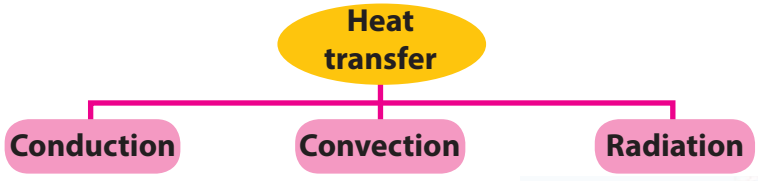
Figure 14.15 ▲ Students near a fire

Figure 14.16 ▲ Touching a hot cup of tea

We feel warm, though we are few metres away from a fire. Hand will be burnt, if a heated object is touched.

In the above instances heat has travelled from one place to another place. Travelling of heat from one place to another place is called **heat transfer.**

There are three methods of heat transfer.



Conduction

A metal spoon in a cup of tea gets heated soon.

The far end of a metal spoon gets heated when it is put into a pan on a cooker.

Figure 14.17 ▲ Heated pan on a cooker

Figure 14.18 ▲ A hot cup of tea

Let us do Activity 14.8 to find out how heat transfers through a solid substance.

Activity 14.8

You will need :- A metal (iron) rod of about 20 cm long, few pins, a candle

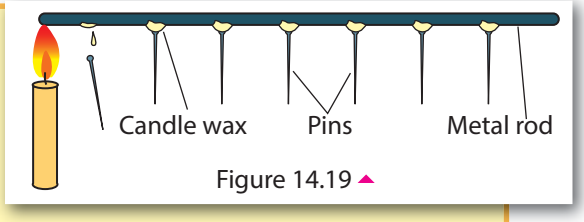


Figure 14.19 ▲

Method :-

- Take the metal rod and fix pins on it, in 2 cm intervals, using candle wax, as shown in Figure 14.19
- Heat one end of the rod using the candle
- Record your observations after sometime.

You can observe that pins are dropping down gradually, starting from the heating end of the rod, because of the melting of wax. Thus, it can be concluded that heat has transferred through particles of the rod, starting from heating end. Here, the heat transfers from one particle to the other.

This method of transferring heat from particle to paricle without the motion of particles through a solid, is known as conduction of heat.

Most of the metals conduct heat well. They are known as heat conductors.

e.g.:- Iron, Copper, Aluminium, Gold, Silver

Assignment 14.7

Make a report on the substances/ metals that conduct heat.

Substances that do not conduct heat well are known as heat insulators.

e.g.:- glass, wood, plastic, cloth, air, water

Heat conductors as well as heat insulators are important in day-to-day activities.

Figure 14.20 ▲ Instances where heat conductors and heat insulators are used

Assignment 14.8

Find out some other instances where heat conductors and heat insulators are used in day-to-day life.

For extra knowledge

People living in cold countries use woolen clothes to maintain their body temperature (in winter). As woolen clothes are good heat insulators, they prevent losing body heat to the environment.

Convection

You may have seen that small twigs of trees above a large fire are waving. What can be the reason for this?

Heated air near the fire rises up and cool air from downwards flow towards the fire. Heated air currents rising up like this is called convectional currents. When these currents strike with the twigs of trees, they start to wave.

Figure 14.21 ▲ A fire under a tree

Figure 14.22 ▲ Convectional currents of air near a fire

Let us do Activity 14.9 to study further about the way that heat travels through air.

Activity 14.9

You will need :- A tall beaker, a piece of cardboard, a candle, some joss sticks

Method :-

- Cut the piece of cardboard to the shape as shown in the figure.
- Place the piece of cardboard, at the middle of the beaker to divide the inside of it into two chambers.
- Then, place the lighted candle in the side B of the beaker as shown in the figure.
- Light some joss sticks and hold them at the mouth of the beaker on the other side (side A) of the piece of cardboard.
- Observe what happens.
- Now blow out the candle and hold the joss sticks in the side A.
- Observe what happens again.

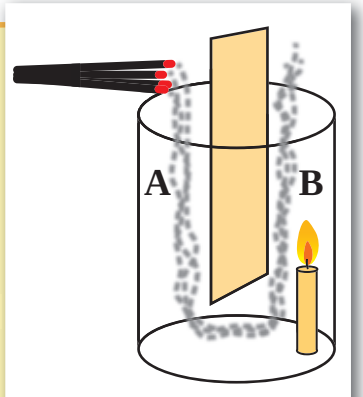


Figure 14.23

It can be observed that the smoke of joss sticks enter into the beaker from side A and comes out from side B.

When heated air rises up from chamber B, where the lighted candle is placed cool air flows down into chamber A. It is clear that the smoke of joss sticks also flow with air currents. Thus, it is clear that heat travels through air as convectional currents.

Let us do Activity 14.10 to find out how heat travels through liquids.

Activity 14.10

You will need :- A few condis crystals, a candle, a burner, a round-bottomed flask, a laboratory stand

Mehtod :-

- Place some condis crystals at the bottom of the flask and cover them with wax.
- Pour water into the flask and heat it.
- Observe what happens.

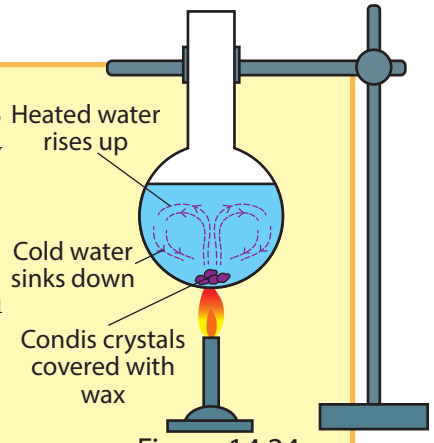


Figure 14.24

It can be observed that purple colour rises up as currents from condis crystals in water, and sinks down near the wall of the flask. Here, water at the bottom of the flask is heated and rises up because of the reduction of its density. Mean while, cold water at the top sinks down because of its higher density.

Rising heated water currents and sinking cold water currents are known as convectional currents. Water in the flask heats because of these convectional currents.

The method of transferring heat through liquids and gases by convectional currents is known as convection.

Assignment 14.9

Make toys operate by convectional currents and display them in classroom.

14.4 Application of convectional currents

Occuring of sea breeze and land breeze

Sea breeze

Wind that blows from the sea towards the land is known as sea breeze. Sea breeze occurs in day time.

During day time land area heats faster than the sea water because of the solar heat. This cause the layer of air contacted with the land to heat and rise up as convectional currents. To fill the low pressure area created on the land, air currents flow from the sea towards the land. This is known as sea breeze.

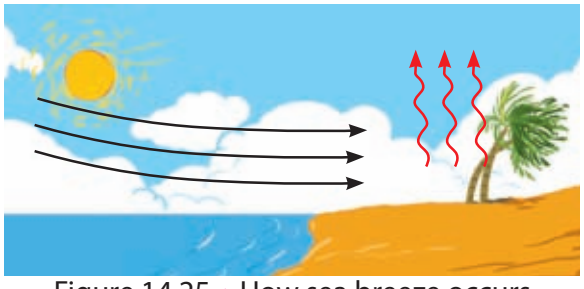


Figure 14.25 ▲ How sea breeze occurs

Land breeze

Wind that blows from land towards the sea is known as land breeze. Land breeze occurs at night.

During night time the temperature of land area decreases faster than the sea water. Therefore, land area cools faster. Because of the high temperature of sea water, the layers of air contacted with sea water get heated and rise up as convectional currents. To fill the low pressure area created on the sea, air currents flow from the land towards the sea. This is known as land breeze.

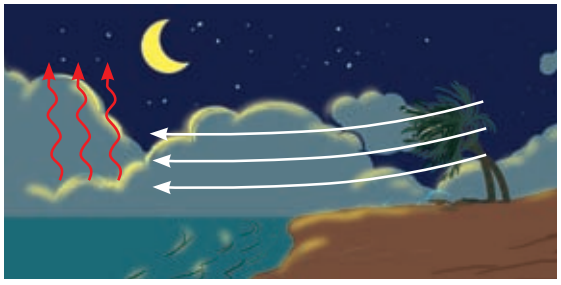


Figure 14.26 ▲ How land breeze occurs

For extra knowledge

It is with the support of the land breeze that fishermen launch their sail boats to the sea in the night time. They return back to the shore in day time with the support of sea breeze.

Assignment 14.10

Prepare a list of other applications/ important occasions associated with convectional currents.

Radiation

The method of heat transfer, without participation of the particles of a medium is known as radiation. Heat travels from the sun to the earth by radiation.

We feel warm when we are near a fire or a heated object, because heat travels towards our body by radiation.

Any heated object radiates heat.

Figure 14.27 ▲ Radiation near fire

Let us do Activity 14.11 to investigate more about radiation.

Activity 14.11

You will need :- Three equal tin cans, three thermometers, cork stoppers, black and white paint, cold water, few pieces of cardboard, stop clock

Figure 14.28 ▲

Method :-

- Keep one tin can as it is, with the shining outer surface. Paint one of the other tin cans with black paint and final one with white paint (See Figure 14.28).

- Add equal volume of cold water into each tin. Fix a thermometer to each tin can as shown in the figure and measure the initial temperature
- Place all the three cans in the same place in the sun. Measure the temperature in every five minutes time and tabulate the readings.

Time (min)	Temperature of water in the cans (^{0}C)		
	Black can	White can	Can with shining surface
0			
5			
10			

After some time, it can be observed that the temperature of water in black can has risen higher than that of both the other cans. Also it can further be observed that the temperature of water in the can with shining outer surface has risen very less.

Water in the tin cans is heated by the solar radiation. It is clear that black colour absorbs radiated heat very fast and polished shining surfaces do so very slowly. White surfaces also absorb radiated heat less than black surfaces.

Black surfaces lose heat very fast while polished, shining surfaces do so very slowly. Therefore, hot water in containers with polished, shining surfaces can be kept hot for a long time.

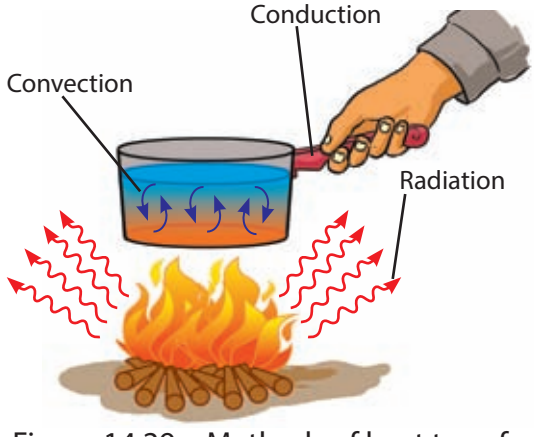


Figure 14.29 ▲ Methods of heat transfer

Countries like Sri Lanka gets more sun light through out the year. Therefore, it is more suitable to use light colours to paint outer walls of the buildings rather than dark colours. It is because light colours absorb radiated heat less, that prevents the interior of the house from heating.

Assignment 14.11

Find and report the outside colour of the vehicle radiators. Discuss the reason for painting radiators in that colour with your teacher.

Summary

- The measurement of coldness or hotness of an object is known as temperature.
- Heat is a type of energy.
- Temperature of an object increases when heat is supplied and it decreases when heat is removed from the object.
- Thermometers are used to measure temperature.
- The property of expansion of a liquid is used in making liquid thermometers.
- Units of measuring temperature are degrees celsius, degrees farenheit and Kelvin.
- Kelvin is the international unit of measuring temperature.
- Clinical thermometer is used to measure body temperature.
- The boiling point (100 ^{o}C) and the freezing point (0 ^{o}C) of water are the fixed points of a liquid thermometer at 1 atm.
- Heat transfers by means of conduction, convection and radiation.

Exercise

1) Select the suitable word to fill the blanks from the words given in the brackets.

i. International unit of measuring temperature is ..... (degrees celsius/Kelvin)
ii. Heat transfers from the sun to the earth by ..... (conduction/radiation)
iii. Aluminium pot on a hearth, heats mainly by getting heat by .... (conduction/convection)

iv. Temperature, at which liquid water converts to ice is called the ... . of water at atmospheric pressure. (boiling point/ freezing point)

v. Sea breeze and land breeze occur because of the phenomenon of ... . in the air. (convection/radiation)

2) Select the correct answer out of those given.

1. The body temperature of a healthy man is;
   1. 0 ^{0}C 2. 37 ^{0}C 3. 98 ^{0}C 4. 100 ^{0}C

2. A substance that conducts heat well is;
   1. Water 2. Air 3. Glass 4. Iron

3. A heat insulating substance is;
   1. Aluminum 2. Copper 3. Paper 4. Lead

4. A liquid that conducts heat well is;
   1. Water 2. Alcohol 3. Mercury 4. Kerosene

5. Warm is felt in a house, where the roof is covered with metal sheets. What is the method of heat transfer into this house?
   1. Expansion 2. Conduction 3. Convection 4. Radiation

Technical Terms

Temperature	-	උෂ්ණත්වය	-	வெப்பநிலை
Heat	-	තාපය	-	வெப்பம்
Freezing point	-	හිමාංකය	-	உறைநிலை
Melting point	-	ද්‍රවාංකය	-	உருகுநிலை
Boiling point	-	තාපාංකය	-	கொதிநிலை
Thermometer	-	උෂ්ණත්වමානය	-	வெப்பமானி
Heat transfer	-	තාප සංක්‍රාමණය	-	வெப்ப இடமாற்றம்
Conduction	-	සන්නයනය	-	கடத்தல்
Convection	-	සංවහනය	-	மேற்காவுகை
Radiation	-	විකිරණය	-	கதிர்ப்பு

15 Soil

15.1 Different types of soil

Soil, the topmost layer of the earth crust is one of the most important resources that we have been gifted from the nature. It plays a very important role to maintain life on the earth. The nature of soil is different from place to place on the earth.

Let us do Activity 15.1 to observe the nature of soil in different locations.

Activity 15.1

You will need:- Samples of soil taken from different locations (Close to a big tree, from a flower bed, from your compound, under a rock), a white sheet of paper, hand lens

Method:-

- Spread the soil sample on the white paper.
- Observe components of it using a hand lens.
- Take a sample of soil between thumb and forefinger and feel the texture of soil.
- Use the following table to record your observations.

Figure 15.1
▲ Testing the texture of soil

Place where soil is collected	Observed components	Texture (soft/hard)	Other facts
1.			
2.			
3.			
4.			

Important : After doing this activity wash your hands well.

You may have learnt from the Activity 15.1 that soil consists of different particles and their texture is also different.

Mineral particles in soil can be divided into 3 groups according to their size. They are sand, clay and silt. Clay contains very small particles with a soft texture. Sand particles are bigger than clay, with a hard texture. Silt is medium sized particles with a soft texture.
You can learn the relative size of them referring to the Figure 15.2.

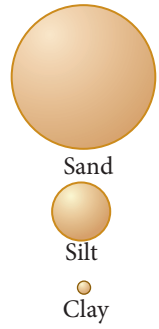


Figure 15.2 ▲ Relative size of different particles in the soil

There are three types of soil which can be identified according to the abundance of those particles in the soil. They are;

- Clay soil
- Sandy soil
- Loamy soil

Some features of those soil types are given in Table 15.1.

Table 15.1 ▼ Some features of soil types

Kind of soil	Nature of soil	Properties	Uses
Clay soil	• Clay is more abundant • Sticky	• Retain water and minerals • Retention of air is less	• Used to make bricks, tiles and ceramics
Sandy soil	• Sand is more abundant • Particles are arranged loosely	• Retention of air high • Retention of water is less	• Used in building industries • Used to make glass items
Loamy soil	• Contains sand, silt and clay • Organic materials are abundant	• Water and air retained sufficiently • Soil is rich in soil organisms and plant nutrients.	• Most suitable for agriculture

Loamy soil is more suitable for the growth of plants. Therefore, a flower pot should be filled with loamy soil.

Sieves are used to separate different sized particles in soil. These sieves are prepared in sets. The topmost sieve has biggest holes. The size of the holes decrease from top to bottom. A soil mixture is put into the topmost sieve and the set of sieve is shaken fast. Then, in each sieve, different sized soil particles remain.


Figure 15.3 ▲ Set of sieves

Activity 15.2

You will need :- A set of sieves, soil samples

Method :-

- Separate soil particles using different sized sieves
- Mix them in different ratios and form different types of soil

Sandy soil or clay soil can be converted to loamy soil by adding organic materials (decayed animal and plant parts) in suitable amounts.

15.2 Composition of soil

Activity 15.3

Separation of components in the soil

You will need :- A long polythene tube, threads, soil, water

Method:-

- Cut a polythene tube about 50 cm long and tie one end of it. Fill ¾ of the tube with water.
- Dig a hole in your school garden and take a sample of soil from it.
- Mix the soil sample with water.
- Then, add it to the water already present in the polythene bag and leave to settle.
- Now observe.
- You can observe the separation of components of soil as shown in Figure 15.4

Then, take a polythene tube of 1m long. Fill ¾ of it with water and put twice the amount of soil taken in the above activity into it

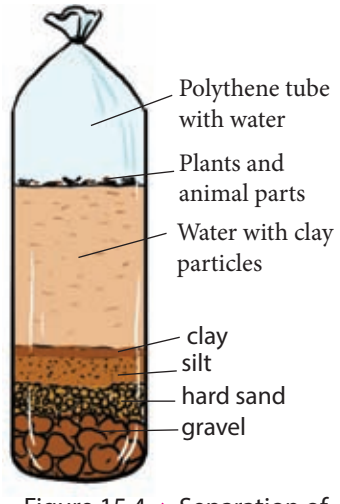


Figure 15.4 ▲ Separation of components of soil

Soil is a mixture of different components. Ratio of these mixtures are different in various types of soil. Figure 15.5 shows results of a study carried out on the composition of different types of soil.

Figure 15.5 ▲ Composition of different types of soil

Components of soil can be listed as follows.

- Solid components → Soil minerals, Soil organic materials (humus)
- Soil air
- Soil water
- Soil organisms

Soil minerals

Solid components in soil such as clay, silt and sand are considered as soil minerals. When you study the composition of soil, you have studied about these particles.

Functions of soil minerals

- Minerals that is obtained from soil to plants are provided by soil minerals.
- Water and minerals are retained by the clay particles in soil.

Soil organic materials (Humus)

Decayed parts of animals and plants in soil are called soil organic materials.

Activity 15.4

By adding humus to sandy soil, the water retention capacity is increased. Plan an experiment that can be conducted in the school laboratory to find the accuracy of this statement.

Functions of soil organic materials (Humus)

- As a storage of nutrients needed for the growth of plants
- Increase the capacity of retaining air and soil
- Prevents the cracks in soil during dry seasons
- Increase retention of water in soil

You may have understood the mineral components and organic materials in soil through the above activities.

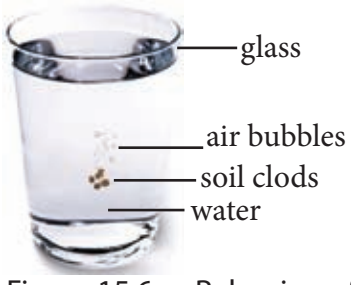


Figure 15.6 ▲ Releasing of water bubbles when a soil clod is put into water

Soil air

What happens when a soil clod is put into a water beaker?

You can observe air bubbles emitting from the soil clod. It is clear that air in the soil is coming out as bubbles.

Let us do Activity 15.5 to prove that soil contains air.

Activity 15.5

You will need :- Two glass vessels of the same size, soil, water

Method :-

- Put soil into one vessel up to 4 cm.
- Put water into the other vessel up to 4 cm.
- Add water in the water vessel into the vessel with soil.
- Then, measure the height from the bottom to the water level.
- Repeat the above experiment using soil samples collected from different locations.

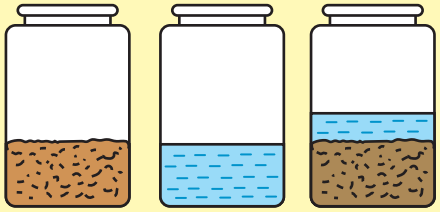


Figure 15.7 ▲ Learn about air in soil

When, water and soil are mixed the height to the water level appears to be less than 8 cm. Imagine the height after water and soil are mixed is 7 cm. We can assume that the height of 4 cm of soil contains about 1 cm of air. That means about ¼ of the volume of soil contains air. The soil air amount in different soil samples is different.

In the above activity water has moved into the space among soil particles. When, water moves into those spaces air present in them is released. Air which retains in soil particles is called soil air.

Functions of soil air

- Provides air needed for the respiration of soil organisms and plant roots
- Needs in germination of seeds
- Creates porosity in soil

Soil water

Though we think that soil is dry, there is water in it.
Let us do Activity 15.6 to identify that soil contain water.

Activity 15.6

Let us prove that soil contains water

You will need :- A sample of dry soil, a test tube, a bunsen burner or a sprit lamp

Method :-

- Put the sample of soil into the test tube and heat it.
- Write down your observations.
- Identify whether the liquid droplets deposit on the walls of the test tube are water.

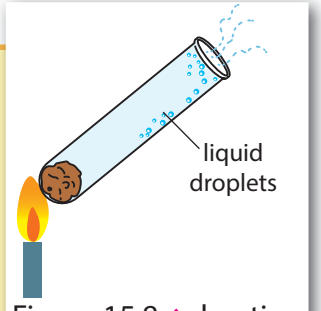


Figure 15.8 ▲ heating dry soil

How can we identify that liquid droplets formed inside the test tube are water?

A blue colour chemical called **copper sulphate** is present in the laboratory. When it is heated, it turns **white**. White coloured **copper sulphate** is called anhydrous copper sulphate. When water is added to anhydrous copper sulphate it turns again into **blue** colour.

Functions of soil water

- Helps soil organisms to maintain their function.
- Helps plants to absorb nutrients from soil.
- Controls the temperature of soil
- As a raw material for the photosynthesis of plants

Soil organisms

Do the Activity 15.7 to identify organisms in soil.

Activity 15.7

You will need :- Several soil samples taken from under a tree, a flower bed and under a large stone, a sheet of white paper

Method:-

- Spread each soil sample on the sheet of white paper. Observe soil organisms using a hand lens and draw them.

Several soil organisms observed by a student are given in Figure 15.9.

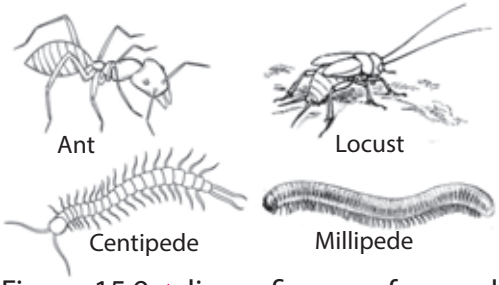


Figure 15.9 ▲ linear figures of several soil organisms

But organisms who are not visible to our naked eye also may be present in soil. They are called soil microorganisms.

Let us do Activity 15.8 to observe whether microorganisms are present in the soil.

Activity 15.8

You will need :- Two test tubes, milk, cotton, a soil sample

Method:-

- Sterilize the two test tubes in boiling water (Microorganisms in them will be destroyed).
- Put equal amounts of boiled milk into the test tubes and let them cool.
- Take a soil sample and divide them into two equal portions.
- Take one sample and heat it for about five minutes keeping on a metal plate.
- Add heated soil sample to one test tube with milk and add the non-heated soil sample into the other test tube
- Fix cotton stoppers to both test tubes. Both test tubes will gain air but microorganisms cannot enter them.
- Observe the tube hourly.
- Record your observations.

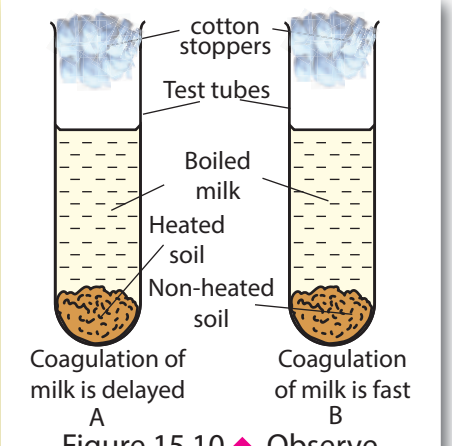


Figure 15.10 ▲ Observe whether microorganisms are present in the soil

Milk in the test tube with unheated soil will coagulate faster than the other one. Coagulation of milk is a microbial activity. Milk in the test tube with heated soil will coagulate slowly.

When cotton stoppers are fixed to the test tubes, air enters but not microorganisms. Accordingly, the above experiment shows that microorganisms are present in the soil.

The diagram given below shows the organisms who live in soil.

Figure 15.11 ▲ Different organisms who live in soil

Functions of soil organisms

- When earthworms dig holes, soil gets loosened and it gets air.
- Microorganisms like bacteria decay plants and dead bodies so that minerals get into soil.

Soil profile

Assignment 15.1

Observe an occasion where earth is dug in a slope of a mountain using a backhoe loader. Identify the soil layers that can be seen.

A vertical section of the different layers of the soil from the earth crust is called soil profile. It spreads from top layer to the bed rock. There are mainly three layers that can be identified in the soil profile.

- Top soil
- Sub soil
- Bed rock

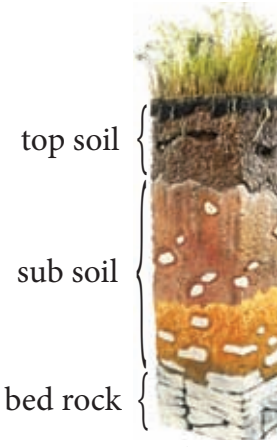


Figure 15.12 ▲ A soil profile

When moving from top to bottom in the soil profile, the size of the particles will gradually increase. The bottom of the profile is the bed rock. Sub soil is formed by weathering of bed rock and forming small particles. The top soil is formed by further weathering of the particles in the sub soil.

Roots of the most plants are spread in the top soil. But, roots of some large plants can penetrate in to the sub soil layer. The soil profile is very important in studying about soil.

Activity 15.9

Let us make a model of a soil profile

You will need:- A set of soil sieves, cardboard, soil, glue

Method:-

- Dig a hole of 30 cm deep in your home garden and collect a soil sample from it.
- Separate the soil sample by using the set of sieves. (Follow the instructions from your teacher)
- Get a piece of cardboard (similar in size to a file cover) and separate it into three areas by drawing two horizontal lines.
- Paste the soil particles remaining in the second sieve from top to bottom on the lowest region on the cardboard.
- Then, paste the soil particles remaining in the second sieve from the top on the lower area.
- Paste the soil particles in the third sieve from the top.
- Paste the sieved soil particles from all the sieves on the topmost area.

Now you have created a model of a soil profile.

15.3 Soil erosion

What is the colour of water flowing above the soil during a rainy day? You have observed that the colour of water changes based on the fact of soil being covered with a covering or being exposed. The reason is that soil particles flow into another location with water. **Soil erosion is the removal of the top soil layer of a certain location by water, wind and animals.**

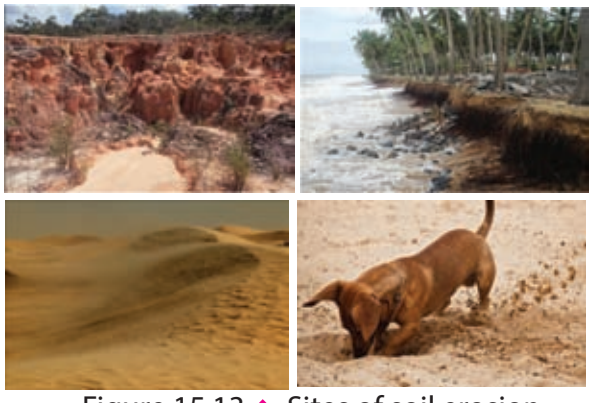


Figure 15.13 ▲ Sites of soil erosion

The most fertile part of soil is the top layer of soil. The removal of the top soil layer can happen in the following manner

- Top soil is washed away mainly by rain and flowing water.
- It is carried away by wind during the dry season as dust.
- Sand in coastal areas are washed away by sea waves.
- Soil erosion mainly occurs during agricultural and development activities.
- Removal of soil by legs, hoofs and horn of animals is also a type of soil erosion.

Soil erosion takes place often in lands with slopes. When there is no plant covering in such a land, soil erosion takes place fast.

Let us do Activity 15.10 to find it out.

Activity 15.10

Finding how the covering of earth affects soil erosion

You will need :- A tin with holes at the bottom, three large plastic bottles, three small plastic bottles, soil, grass clod, dried/decayed leaves

Method :-

Figure 15.14 ▲ Comparing soil erosion

- Cut three large bottles as shown in Figure 15.14. Remove their lids.
- Cut three small bottles as shown in the Figure. Do not remove the lids.
- Add the same type of soil in equal amounts to each large bottle. Then press them well.
- Plant the grass clod in one bottle and keep dried/decayed leaves on the surface of the other bottle. Keep the soil in the third bottle open.
- Make holes in small bottles and tie them to the mouth of the large bottles, as shown in the figure.
- Keep those bottles on the edge of the table in a manner that the bottom of the large bottles is little tittled up.
- After the grass grow well, pour water to the soil samples by using three taps. The speed of the flowing should be equal. Or else, pour equal amounts of water on to the soil samples in the same height. Use the tin with holes at the bottom to pour water.
- Observe the colour of water collected in small bottles.

The water flowing from the open bottle is darker than the other two bottles. The water flowing from the bottle containing water with decaying dried leaves is less darker than the above. The water flowing from the grass grown bottle is the clearest.

This Activity reveals us when there is a cover on soil, soil erosion is less. Let us see the devices used to minimize soil erosion in a slopy land.

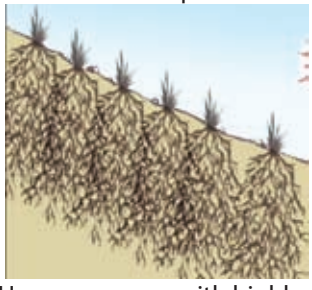
Following are the methods that can be applied to minimize the soil erosion in a land with slopes.

- Make stone ridges against the slope
- Make drains according to contours
- Cultivate according to the contour method
- Use cover crops with highly rooted plants
  e.g. :- grow "savandera"/"vettivear"
- Prepare fields according to terraced system (for paddy cultivation)
- Use wind breakers


Make stone ridges against the slope


Make drains according to contours


Cultivate according to the contour method


Use cover crops with highly rooted plants


Prepare fields according to terraced system


Use wind breakers

Figure 15.15 ▲ Strategies used to minimize soil erosion

Substances like polythene, used batteries, chemical fertilizer and used electronic appliances pollute the environment.

Soil is one of the most valuable resource on earth. Therefore, it is our duty to protect it.

Assignment 15.2

Identify the locations in the school garden that have undergone soil erosion. Discuss the strategies that can be used to prevent soil erosion in these places. Make a report including these facts.

Summary

- The outermost layer of the earth crust is soil. Soil is important in different ways for the existence of plants and animals.
- The components of soil are minerals, water, air and soil organisms. There are various advantages of them to the soil.
- Soil is formed by weathering of bed rock due to various factors for a long period of time.
- Soil erosion is the removal of top soil layer of a particular location by water, wind or animals.
- Soil erosion takes place due to natural reasons as well as human activities.
- Soil conservation is essential for the existence of all life on Earth.

Exercise

01. Select the most suitable answer.

I. What is the type of soil most suitable for cultivation?

1) Clay soil 2) Sandy soil 3) Loamy soil 4) Gravel soil

II. A component /components of soil is/are

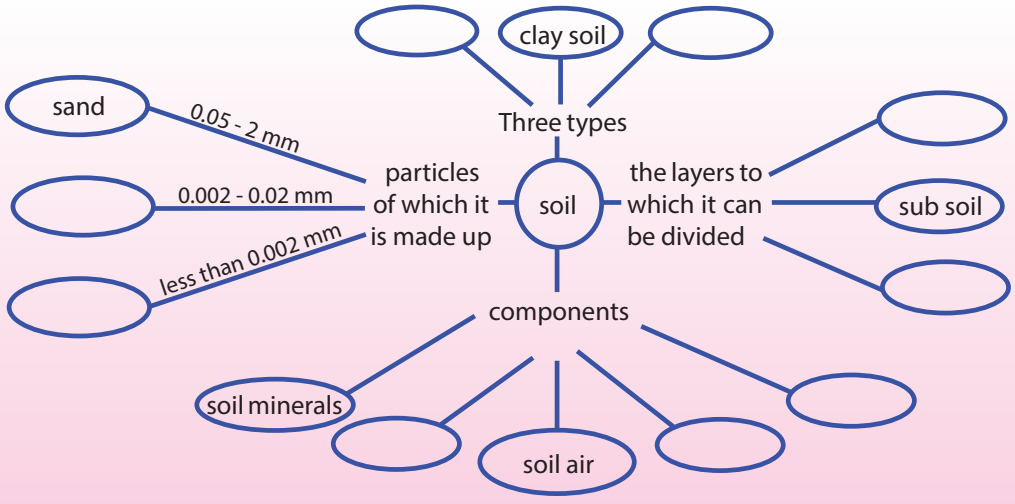
1) Air 2) Water 3) Minerals 4) All of the above

02. Fill in the blanks using the given words.
(Humus, clay, rocks, loamy soil, silt)

The most important part of the earth crust is soil. As a result of weathering ....., soil is formed. The amount of sand,..... ..... and clay particles present in the soil is different from place to place. Soil is classified into three groups according to the composition of particles in the soil. They are clay soil, sandy soil and ..... The inorganic component which has the ability to retain water in the soil is ..... The organic material present in the soil is ..... .

03. Select words from the given list and complete the concept map.

(sandy soil, clay, soil organisms, outermost soil, soil water, organic materials, bed rock, sand, loamy soil, clay soil, silt, sub soil)



4) 'Earthworm is a friend of the farmer' explain the above idea with your own words.

5) List four activities used to prevent soil erosion.

Technical Terms

Sand soil	-	වැලි පස	-	மணல் மண்
Clay soil	-	මැටි පස	-	களி மண்
Loamy soil	-	ලෝම පස	-	இருவாட்டி மண்
Silt	-	රොන් මඩ	-	அடையல்
Soil sieve set	-	පස් පෙනේර කට්ටලය	-	மண் அரிதட்டுத் தொகுதி
Top soil	-	මතුපිට පස	-	மேல்மண்
Sub soil	-	උප පස	-	கீழ்மண்
Bed rock	-	මව් පස	-	தாய்ப்பாறை
Soil erosion	-	පාංශු ඛාදනය	-	மண்ணரிப்பு
Soil conservation	-	පාංශු සංරක්ෂණය	-	மட்காப்பு
Soil texture	-	පාංශු වයනය	-	மண்ணின் இழையமைப்பு
Soil air	-	පාංශු වාතය	-	மண் வளி
Soil water	-	පාංශු ජලය	-	மண் நீர்
Soil organisms	-	පාංශු ජීවීන්	-	மண் அங்கிகள்

16 Force and Motion

16.1 Distance and displacement

Imagine that you have a chance of going on a trip during your school vacation. Can you suggest a method to find the distance to the places, that you expect to visit during that trip? Let us pay attention to Activity 16.1.

Activity 16.1

- Find a road map of Sri Lanka, drawn to scale and identify the scale.
- Place a thin thread along the main road and measure the length between two cities.
- Calculate the distance to travel from Colombo to Kandy, Galle, Matara, Kurunegala, Puttalam, Trincomalee and Jaffna.
- Measure the straight length from Colombo to each city mentioned, and calculate the straight distance.
- Tabulate the findings as below.

Table 16.1

Two cities	Distance to travel between the two cities	Straight distance between the two cities
Colombo to Kandy		
Colombo to Galle		
Colombo to Matara		
Colombo to Kurunegala		
Colombo to Puttalam		
Colombo to Trincomalee		
Colombo to Jaffna		

It will be clear to you that the straight distance between the two cities is always less than the distance to travel between them.

Recall an instance where there is a staircase and an escalator go to the

top floor of a multi-storeyed building. It may be clear to you that the path of the escalator is linear and is shorter than the path of the staircase.

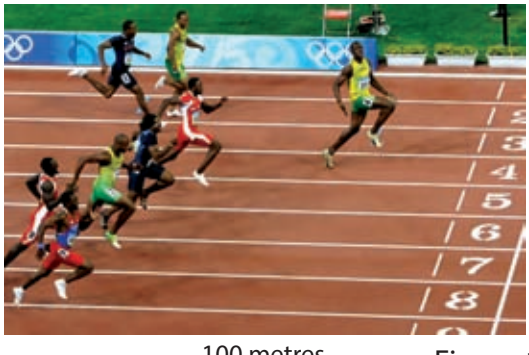
You may travel from one place to another when you go on a trip. A person may go from one floor to another in a multi-storeyed building. An escalator goes up and down in a straight path. The change of position, with time, of living or non living objects is known as motion.

Distance is the total length of the path of a motion. Here, the direction of motion may or may not change from time to time. Therefore, distance has not a definite direction.

Displacement is the linear length between the starting and ending positions of a motion. When mentioning displacement, the direction of motion should always be indicated.

Recall Activity 16.1 you did above. You may have understood that the length of the road between two cities is known as the distance and the straight distance is known as the displacement. Geographical directions or the direction from one city towards the other can be used to mention the direction between the two cities. The same way, motion along a staircase is the distance and motion of an escalator is the displacement. The direction of the displacement is the direction of the motion of the lift.

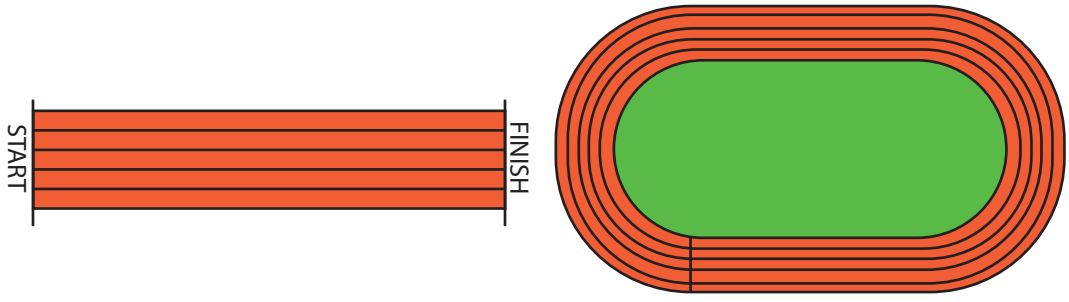
To find out more about distance and displacement, let us consider the following facts. Recall the 100 metre and 400 metre running events of a sportsmeet.



100 metres

400 metres

Figure 16.1a



100 metre track 400 metre track

Figure 16.1b

Both events are short distance races. But, there are some clear differences between them.

Table 16.2

Differences	**100** metres	**400** metres
01. Direction of motion	Direction does not change	Direction changes from time to time
02. Total length of the path	100 metres	400 metres
03. Linear length between the points of start and end	100 metres	0 metre (zero)

Thus the distance (distance of the motion) of the runner in the 100 metre event is 100 metres, and the distance (distance of the motion) of the runner in the 400 metre event is 400 metres. The distance between start and end points of the 100 metre runner is 100 metres. Therefore, his displacement is 100 metres. The 400 metre runner ends his event at the same point where he started. So, his displacement is zero.

As both the distance and displacement are lengths, the unit of measuring distance and displacement are the units of measuring length.

You already know that milimetre, centimetre, metre and kilometre are used to measure length. Out of those, the international unit of measuring length is metre. Therefore, the international unit of measuring both distance and displacement is metre (m). Displacement has a definite direction, therefore it has to be mentioned.

Let us solve a simple problem to clarify distance and displacement further.

The route of the motion of an object from A to B is shown in Figure 16.2.

The distance travelled from A to B or the total length of the route passed is 120 m.

The displacement from A to B or the linear length between the initial and final points of motion is 40 m towards the direction AB (South-east)

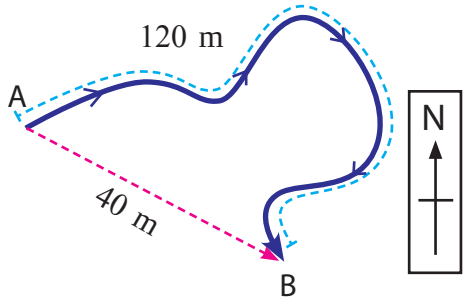


Figure 16.2

Self assessment -1

A route for a child to travel from his home to a tank is shown in Figure 16.3.

i. What is the distance travelled by the child?

ii. What is the displacement of the child?

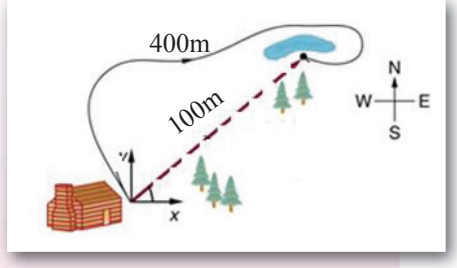


Figure 16.3

Answer

i. 400 m

ii. 100 m towards north-east

02. Let us consider a man travelling from P to R via Q by a motor vehicle as shown in Figure 16.4.

iii. What is the total distance travelled during this motion?

iv. What is the displacement of the vehicle during this motion?

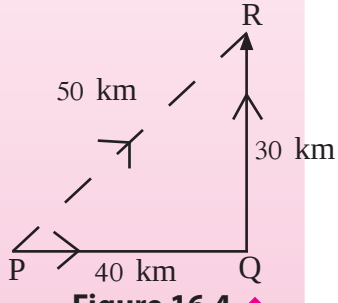


Figure 16.4

Answer

i. 70 km

ii. 50 km towards north-east

Now, it is clear to you that distance is a quantity with a magnitude, and displacement is a quantity with both magnitude and definite direction.

16.2 Force

We engage in various actions in our day-to-day life. Some of them are given below.

- Pushing a table
- Riding a bicycle
- Drawing water from a well
- Hitting a ball
- Erasing pencil lines drawn in a book
- Writing with a pen
- Pulling a table
- Applying brakes of a bicycle
- Turning a tap
- Brushing teeth
- Lifting a bag

What do we do during these actions? You may have understood that a pull or a push is applied during all these actions.

Pulling a table | Riding a bicycle | Pushing a table

Figure 16.5

Lifting a bag is an upward pulling. When drawing water from a well, a pull is applied. When riding a bicycle, push is applied by the feet on the paddle.

Let us do Activity 16.2 to find out more about pulls and pushes.

Activity 16.2

Engage in the following activities. Copy table 16.3 in your notebook and mention whether the action done is a pull or a push.

Table 16.3

Action done	Is it a pull or a push
Pushing a table	Push
Drawing water from a well	Pull
Lifting a bag	
Riding a bicycle	
Turning a tap	
Hitting a ball	
Writing with a pen	
Brushing teeth with a tooth brush	
Erasing pencil lines drawn in a book	
Applying brakes of a bicycle	

Pulling or pushing is known as applying a force.

Pulling Pushing

Figure 16.6

When pulling a car, the rope attached is pulled and a force is applied on the car. What is done by the child when pushing the trolley, is the forward application of a force. A large force should be applied to push a heavier table than to push a lighter one.

Let us do Activity 16.3 to find out about the amount of force applied on an object.

Activity 16.3

You will need :- Two similar size wooden blocks, a rubber band, a Newton spring balance

Method :-

- Place a wooden block on a table and pull it using a rubber band attached.
- Place the other wooden block on the first block and pull again using the same rubber band.
- When the motion starts the rubber band will stretch longer. That means the force applied later is larger.
- Repeat the activity using a Newton spring balance instead of the rubber band.
- Record the readings of the Newton balance in both occasions. The amount of force applied can be obtained numerically.

The rubber band is stretched less when there is one wooden block and it is stretched more when there are two wooden blocks. That means, more force is applied at the second occasion. When you compare the readings in the spring balance, it can be understand properly.

Figure 16.7 Newton spring balance

International unit of measuring force is newton. Symbol of newton is N. The equipment used to measure force is newton spring balance.

Activity 16.4

Take a balance calibrated in grams or kilograms and a spring balance calibrated in newton. Weigh the same object using both balances and record the readings in a table as given below. Develop a relationship between weight and mass with the help of your teacher

Table 16.4

Object weighted	Mass (kg)	Weight (N)

Activity 16.5

You will need :- A spring balance, a piece of thread, a wooden block

Method :-

- Attach a newton spring balance using a piece of thread, to a wooden block kept on a polished table. Pull the block using the newton balance.

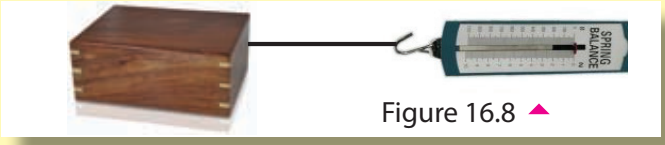


Figure 16.8

- Observe the direction of the motion of the wooden block and the direction of pulling the thread.
- Push the wooden block, while it is on the table. The direction of pushing is the direction of applying the force.

Figure 16.9

- Observe the direction of pushing the wooden block and the direction of the motion of it.

The thread in Figure 16.8 shows the direction of the pull. Newton spring balance shows the magnitude of the force.The wooden block has moved to the direction of the force applied in the above pulling and pushing. Thus, it is clear that a **force has a magnitude (amount) and a definite direction.**

There are various effect of a force. Let us find out some of them.

- **An object can be moved by applying a force. At the same time moving object can be stopped by applying a force.**

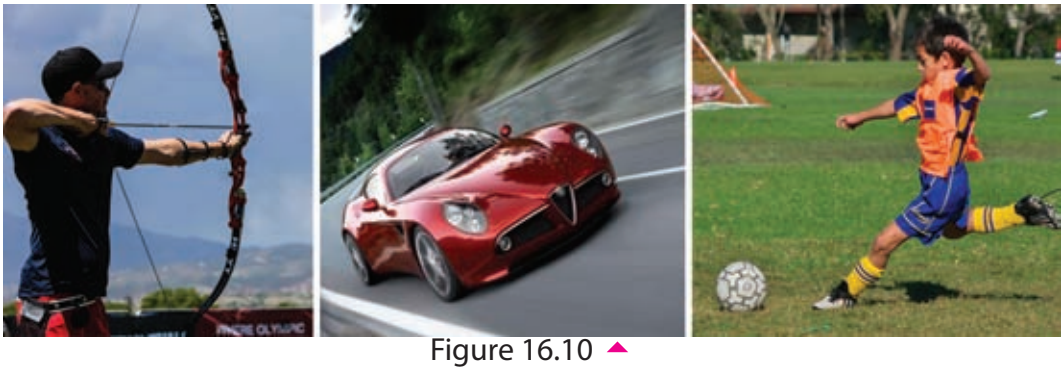


Figure 16.10

An arrow at rest can be moved by applying a force to a bow. Vehicle at rest can be moved by applying a force by its engine. Ball at rest can be moved by hitting it with a bat. Thus, living as well as non living objects, at rest, can be moved by applying a force.

• **The speed of an object can be changed by applying a force.**

Figure 16.11

The speed of motion of a bicycle can be increased by increasing the force applied on its pedal. The speed of it can be decreased by applying brakes. An animal can change its speed of moving by changing the force applied on its limbs. Speed of a vehicle can be increased by treading on the accelarator and it can be decreased by applying brakes.

• **Direction of motion can be changed by applying a force.**

Figure 16.12

In Figure 16.12 a cricketer can change the direction of the ball by hitting it. In the event of pulling rope, each team tries to change the direction of motion of the rope. Two puppies are applying force to drag the bone towards itself.

• **Rotation or a turning effect can be resulted by applying a force**

Figure 16.13

When a force is applied to a tap, it turns. A steering wheel also turns when a force is applied. How do you turn a bicycle when you are riding? It is done by applying a force on the bicycle.

• **The shape of an object can be changed by appling a force.**

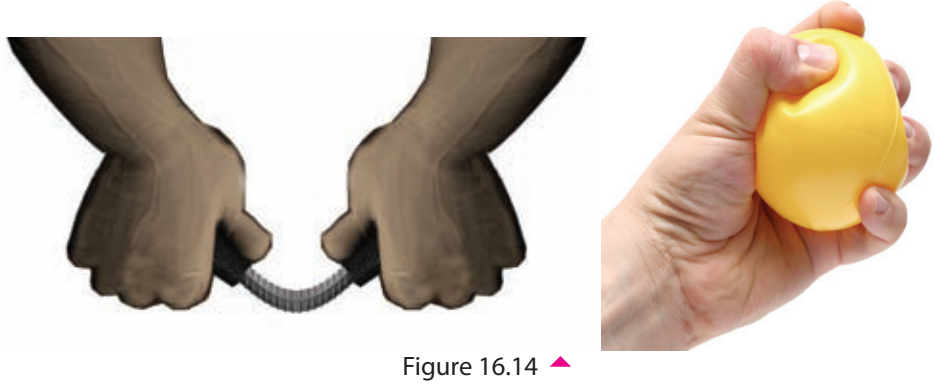


Figure 16.14

Bending iron bars is done in various displays. Then the shape of the iron bar changes.

Various goods are shaped with clay by applying force. The shape of plastic goods can be temporarily changed by pressing them.

Changing the shape due to the force applied on it is occurred in such occasions.

Assignment 16.1

Recall instances of applying forces by you today. Prepare a list of uses that you get by applying those forces.

For extra knowledge

A force is exerted by the centre of the earth on all the objects on the earth. This force is known as gravitational force.
The weight of an object is the gravitational force exerted on the mass of the object. Though the mass of an object is always constant, its weight may slightly change from place to place because of the change of gravitational force.
The gravitational force of the moon is about one sixth of that of the earth. Therefore, the weight of an object on moon is one sixth of the weight of that object on the earth.
e.g.:- The weight of a man whose mass is 60 kg, is 600 N on the earth. The weight of the same man on the moon will be 100 N. But, his mass will remain unchanged as 60 kg on the moon.

Assignment 16.2

Collect pictures, photographs and diagrams of the instances of applying forces in day-to-day life and make a collection of information on forces.

Summary

- The total length of an object in motion is the distance.
- The straight length between the starting and end points of the motion of an object is the displacement. Displacement has a definite direction.
- A pull or a push is known as a force. Force has a magnitude and a definite direction.
- The standard unit of measuring force is newton.
- An object at rest can be moved by applying a force on it.
- The speed of a moving object can be changed by applying a force.
- The direction of motion of an object can be changed by applying a force.
- Rotation or a turning effect can be done by applying a force.
- The shape of an object can be changed by applying a force.

Exercise

- Select the correct or the most suitable answer, for the questions below.

01. What is the international unit of measuring distance?

1. Millimetre 2. Centimetre 3. Metre 4. Kilometre

02. A moving bicycle is pushed from behind by a child. What will be the change of motion?

1. Speed increases
2. Speed decreases
3. It will rotate
4. Direction changes

03. What is the instrument used to measure force?

1. Pan balance
2. Table balance
3. Newton balance
4. Electronic balance

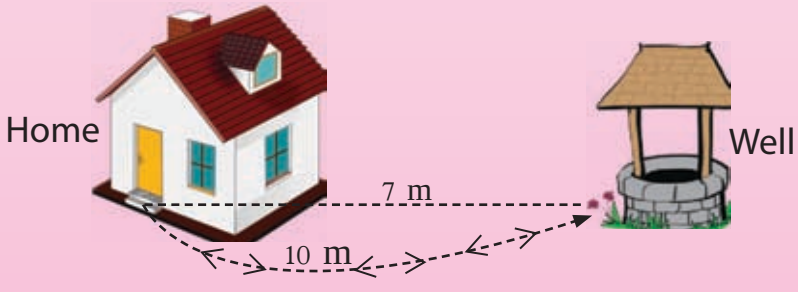
04. Which measurements, have a magnitude and a definite direction?

1. Distance and displacement
2. Displacement and force
3. Force and distance
4. Force only

05. Which is the correct statement about force?

1. Force has a magnitude only
2. Pulling only can be resulted by applying a force
3. Pushing only can be resulted by applying a force
4. Pulling and pushing can be done by applying a force

- A child goes to the well, which is in the east from his home and returns back along the same route.



01. Write down in the table, the distance travelled and the displacement for each instance given below.

Table 16.5

Motion	Distance travelled (m)	Displacement (m)
Going from home to the well		
Going from well to the home		
Going from home to the well and returning back		

02. There are some instances given below, where a force is applied. Mention whether it is a pull or a push for each instance.

....

Technical Terms

Force	-	බලය	-	விசை
Motion	-	චලිතය	-	அசைவு
Distance	-	දුර	-	தூரம்
Displacement	-	විස්ථාපනය	-	இடப்பெயர்ச்சி

17 Nutrients in Food

17.1 Food and nutrients

Air, water as well as food are the most essential for the existence of life.

The Table 17.1 shows the diet taken by a student of grade seven for during 3 days.

Table 17.1 ▼ Types of food taken for main diet

Date	Diet		
	Breakfast	**Lunch**	**Dinner**
Wednesday	A glass of milk, String hoppers, Dhal, Coconut sambol	Rice, Green gram seeds, Fried dry fish, A fruit	Rottie with "Lunumiris", Papaw
Thursday	A glass of milk, Bread, "Kiri hodhi"	Rice, Dhal, Fried potato, "Pala malluma"	Noodles, Potato curry, Egg, Mango
Friday	Rice, Fish, Coconut sambol, "Kiri hodhi"	Rice, Winged bean, Pumpkin, Egg	Rice, Fish, Vegetable salad, Luffa, Banana

There are different types of food mentioned in the above table. Those foods contain five main nutrients in different amounts.

There are five main nutrients required by our body and should be present in the food we eat.

1. Carbohydrates
2. Proteins
3. Lipids
4. Vitamins
5. Minerals

In addition to above mentioned nutrients, certain amount of water and fibre should be present in the diet. Let us consider about type of nutrients and food which they contain.

Carbohydrates

The carbohydrates we take are in the form of sugar and starch. They give energy for our body. Identify the food rich in carbohydrates.

Bread

Rice

Bread fruit

Noodles

Potato

Sugar

Figure 17.1 ▲ Several food rich in carbohydrates

Other food rich in carbohydrates - Hoppers, String hoppers, Jak, Types of yams, Types of grains

Proteins

Animal food (meat, fish, milk, eggs) as well as plant foods contain proteins. They are used mainly for growth and repair.

For extra knowledge

The growth of 80% of the human brain takes place from the conception of the child in the womb upto 2 years from birth. Therefore, a diet of pregnant woman should contain adequate amount of protein.

Figure 17.2 shows some food items which are rich in proteins.

Eggs

Soya seeds

Fish

Dry fish

Meat

Figure 17.2 ▲ Several food items which rich in proteins

Other food rich in proteins - Winged beans, Gram, Milk, Dhal, Mushroom

For extra knowledge

The table given below shows amount of protein recommended daily for different types of age limits.

The minimum amount of protein essential per day

Age (year)	Protein grams per one kilogram of body mass
2	1.2
4	0.9
8	0.7
16	0.77
18	0.45
21	0.35

Lipids

Nutrients which suppy high amount of energy for body is lipids. Lipids include fats and oils.

Oils which are liquid in nature are mostly found in plants. Fats which are solid in nature are highly abundant in animals. The main function of the lipid is supplying energy for body.

Assignment 17.1

Inquire about the advantages of having storage of large amount of body lipids, in animals like Camel and Polar bear.

Some food items rich in lipids are shown in Figure 17.3.


Egg yolk


Curd


Butter


Coconut


Gingerly

Figure 17.3 ▲ Different types of food rich in lipids

Vitamins and minerals

Vitamin and minerals are required in small amounts but they are essential nutrients for our body. This group of food protects us from diseases. Therefore, they do protective function and are important in maintaining good health.

Vitamins and their sources are listed in Table 17.2.

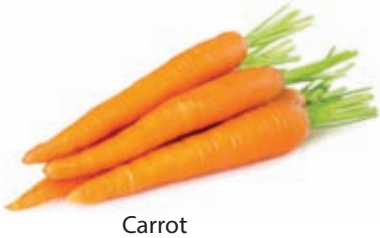
Table 17.2 ▼ Vitamins and sources of them

Vitamin	Sources
A	Cod liver oil, Liver, Milk, Butter, Carrot, Yellow coloured fruits
B	Cereals, Liver, Red-rice, Green leaves, Meat, Fish, Milk, Eggs, Green vegetables
C	Lime, Orange, 'Naran', 'Nelli', Guava, Papaw, Tomato, Vegetables
D	Butter, Cod liver oil, Eggs, Milk, Fish
E	Wheat, Green leaves, Dark green vegetables, Cereals
K	Cabbage, Cauli flower, Spinach, Tomato

For extra knowledge

Vitamins and functions of them

Vitamin	Functions
A	Imporve the vision, Maintain healthy skin and hair
B	Development of memory power, Reduce lethargy
C	Maintain healthy gum, Strength the immunity
D	Development of bones, Prevent decaying of teeth
E	Maintain cell division
K	Involve in blood clotting mechanism


Carrot


Red-rice


Papaw


Cabbage

Milk


Fish

Figure 17.4 ▲ Food rich in vitamins

Table 17.3 shows different minerals, their sources and functions of them.

Table 17.3 ▼ different minerals and their sources

Mineral	Sources
Calcium	Milk, Green leaves, Small fish (sprat)
Phosphorus	Milk, Eggs, Cheese, Meat, Garlic, Small fish, Cowpea, Carrot
Iron	Gingerly, Liver, Meat, Dhal, Spinach, 'Sarana', 'Gotukola'
Sodium	Salt used for cooking, Meat, Milk, Eggs
Iodine	Iodized salt, Sea food

For extra knowledge

Different minerals and their functions

Mineral	Functions
Calcium	Maintain healthy teeth and bones, Blood clotting during injuries
Phosphorus	Maintain healthy teeth and bones, Maintain strong muscles
Iron	Formation of haemoglobin which is needed for transportation of oxygen in blood
Sodium	Transmission of nerve impulse
Iodine	Development of intelligence and memory power, Synthesis of the hormone thyroxine

Some food items contain minerals and salt are given in Figure 17.5.

Garlic

Dhal

Milk

Spinach

Sprats

'Gotukola'

Figure 17.5 ▲ Food rich in minerals

Assignment 17.2

Prepare an article on discovery of vitamin B and C by collecting related historical information.

Activity 17.1

Study Table 17.1. Design and display graphs or tables based on food items and nutrients contain in them. Get the guidence of following chart for your creations.

Figure 17.6 shows the daily food requirments for an individual.

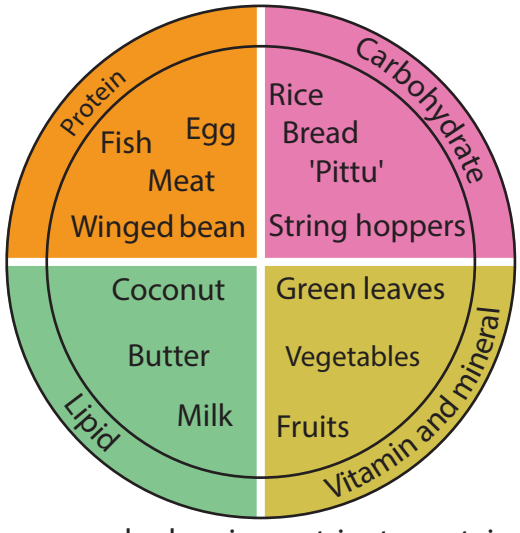


Figure 17.6 ▲ An example showing nutrients contain in a daily meal

Activity 17.2

Observe lables on different types of food containers. Study the types of nutrients contain in them and tabulate.

Most of food items rich in different nutrients.
e.g. :- Dhal contains carbohydrates, proteins and minerals.
Egg contains Proteins, lipids, vitamins and minerals.

Importance of fibre

The fibrous nature of the food is called fibre. Some fibre present in food can be observed by your naked eye. But some are microscopic.

Food rich in fibre

1. Unpolished food
2. Fruits (Ambaralla, Mango, Guava, Banana, Papaw)
3. Vegetables (Carrot, 'Kohila', Cabbage, Potato, drumstics, Bean, Radish, Luffa)
4. Cereal/Grains ('Kurakkan')

Advantages of having food rich in fibre

1. Reduce the absorption of fat in the diet to the body
2. Decrease the absorption of sugar (Glucose) in the diet to the body
3. Prevent the constipation and reduce the risk of causing disease like piles.

Importance of water

We get water with our meals to a certain extent. But, we should drink enough amount of water to maintain good health. Water is mainly used in to cool the body surface, supply medium for cellular reactions, remove excretory products efficiently from the body and prevent the constipation.

17.2 Tests to identify food

Let us do Activity 17.3 to identify main nutrients in food.

Glucose (simple sugar) test

Activity 17.3

You will need :- A test tube, Glucose solution, Benedicts solution, water bath

Method :-

- Add 2 ml of glucose solution to the test tube.
- Add 2 ml of benedict's solution and boil it in the water bath.
- Observe the colour change.

You can observe the colour change in following way.

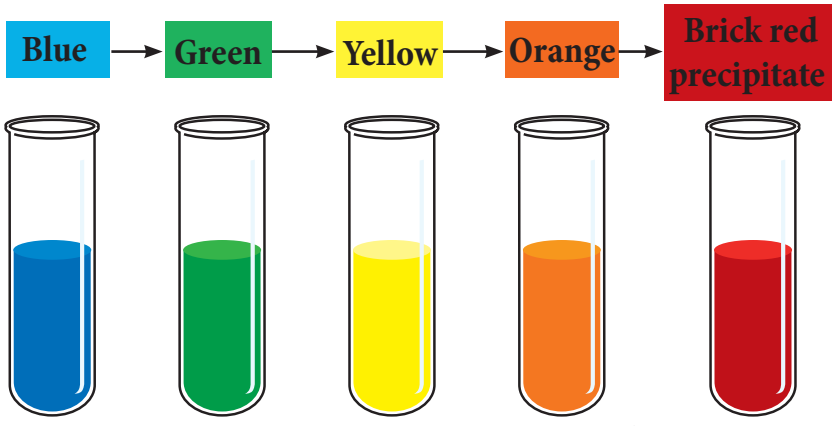


Figure 17.7 ▲ Colour change in the presence of simple sugars

Glucose can be identified by getting brick red precipitate

Starch test

Activity 17.4

You will need :- Mixture made dissolving a small amount of boiled flour

Method :-

- Boil some flour in water and prepare a solution.
- Add drops of iodine solution to the flour solution.

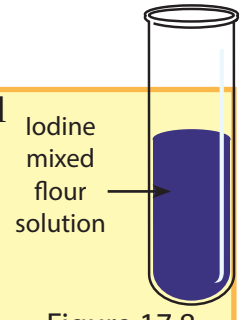


Figure 17.8 ▲

Starch can be identify by getting the colour of purple-blue

Protein test (Biurette test)

Activity 17.5

You will need :- Egg white, water, a test tube

Method :-

- Make a mixture by dissolving egg white in water.
- Put 2 ml of it, to a test tube
- Add 2 ml of sodium hydroxide solution first and then add a few drops of copper sulphate solution to the test tube.

The solution will turn to purple colour which proves proteins occur in mixture.

Figure 17.9 ▲ Biurette test and its observations

Lipid test

Activity 17.6

You will need :- A food contains lipid, white paper

Method :-

Rub some lipid of food on a piece of paper and remove.

If lipids are present, a translucent oily patch can be observed on the paper.

Activity 17.7

Collect different types of food items and test for each nutrients. Tabulate your observation.

The balanced diet

The following descriptions are about food patterns of three children as the description.

Student A

I always eat food which are shown on T.V. They are very tasty. My mother prepares them for me always. Because they are easy to be cooked. I hate taking foods like "Kola kadha" and different types of leaves "Pala". We have our dinner from outside probably 2 or 3 days per week. My father brings food for us from outside. I feel sleepy after having those food.

Student B

My mother prepare a glass of fresh milk for me early morning and all my food are home made foods. Mostly my diet contains of rice, fish, vegetables and fruits. Some days I take cereals like cowpea, gram or green gram. "Kolamalluma" or salad is a frequent essential item in my diet. "Kola kadha" is compulsory on holidays. I rarely take instant foods. I am clever both in study as well as extra curricular activities.

Student C

I normaly don't eat anything after breakfirst until the interval. During the interval I eat from my friends and drink cool drinks out of money given to me from home. I feel sleepy when eating rice. Therefore, I eat bread or buns, for my lunch. I feel very tired after returning home. I eat a little bit of rice as my mother forces me to eat. I feel sleepy.

Activity 17.8

Form groups and present your ideas regarding diet patterns of above three children.

Compare your findings with following analysis.

A = Use more sugar, starch and fat, artificial flavours, colourings and preservatives are highly collected in the body.

Results:- Can be victims of non infectious diseases like obesity, diabetes, hypertension (high blood pressure), cancer, paralysis and heart attacks in future.

B = Do not consume high amount of sugar, starch and fat, arificial flavours, colourings and preservatives are not collected in the body.

Results:- Healthy. Protect from infectious and non infectious diseases.

C = Poor immunity due to lack of enough vitamins and minerals in the diet. Further, poor growth due to lack of adequate proteins from the diet.

Results:- Can be easily caused infectious as well as non infectious diseases. Growth is not up to the age.

Student "B" has the correct pattern of diet among above three children. Because he gets a balanced diet.

What is a balanced diet?

The balanced diet contains the essential nutrients and fibre in correct proportions.

Diagrams given below illustrate three balanced diets prepared for a day.

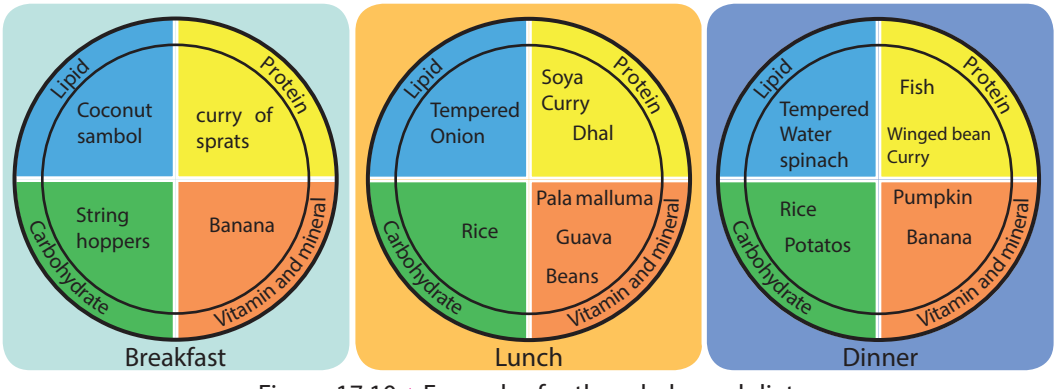


Figure 17.10 ▲ Examples for three balanced diet

Assignment 17.3

Prepare a balanced diet for breakfast, lunch and dinner for your family per day.

Unfavourable conditions due to lack of balanced diet

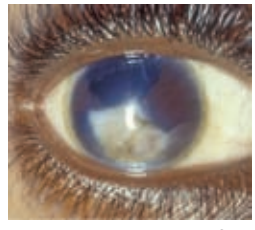
1. The body does not show adequate growth (poor growth) or show over growth.
2. Infectious diseases and non infectious disesases are easily caused.
3. Become weak and lethargic.
4. Suffering from deficiency diseases.

Rickets due to lack of vitamin D

Pellagra due to lack of vitamin B

Scurvy (bleeding gums) due to lack of vitamin C

Goitre due to lack of Iodine



Bitos' spot due to lack of vitamin A

Figure 17.11 ▲ Bad conditions due to lack of having balanced diet

Assignment 17.4

Prepare a booklet on nutrient deficiency diseases.

Summary

- Carbohydrates, proteins, lipids, vitamins and minerals act as main nutrients.
- Carbohydrates and lipids supply energy for body.
- Proteins involve in maintain tissues and development of body.
- Vitamin and minerals protect body from diseases.
- Fibre in food play many functions in the body.
- A diet that contains the essential nutrients and fibre in the correct proportion is called a balanced diet.
- Main nutrients can be identified by food tests.

Exercise

(01) Fill in the blanks with suitable words.

(i) Soya bean is rich in .....

(ii) ..... is the nutrient which supports the growth of the body.

(iii) ..... and ..... are involved in protecting body from diseases.

(iv) ..... in food, prevent the constipation.

(02) Match with the suitable answer.

Nutrient	Most abundant foods
Vitamin A	Yellow coloured fruits
Vitamin D	Red-rice
Vitamin B	Small fish
Vitamin C	Sea food
Iodine	Lime

(03) Discuss the advantages of having a balanced diet

Technical Terms

Nutrients	-	පෝෂක	-	போசனைக்கூறுகள்
Carbohydrates	-	කාබෝහයිඩ්රේට්	-	காபோவைதரேற்று
Proteins	-	ප්‍රෝටීන්	-	புரதம்
Lipids	-	ලිපිඩ	-	இலிப்பிட்டு
Vitamins	-	විටමින්	-	விற்றமின்கள்
Minerals	-	ඛනිජ ලවණ	-	கனியுப்புக்கள்
Fibre	-	තන්තු	-	நார்ப்பொருள்கள்
Balanced diet	-	තුලිත ආහාරය	-	சமநிலை உணவு

18 Minerals and Rocks

A group of grade 7 students collected some stones when they went on a field trip with their teacher. The teacher said that there are rocks as well as minerals among those stones. It was a problem for students to classify them as rocks and minerals.

The figures 18.1a and 18.1b show a piece of granite and a piece of quartz they collected.

Figure 18.1 a ▲ A piece of granite

Figure 18.1 b ▲ A piece of quartz

Let us do Activity 18.1 to inspect the nature of granite and quartz.

Activity 18.1

You will need :- A piece of granite, a piece of quartz, a hammer, a hand lens

Method:-

- Wrap the piece of granite in a cloth. Keep it on a big stone and hit it with the hammer until it is broken into small pieces.
- Check the nature of the pieces with the hand lens.
- Follow the same method with the piece of quartz as well.
- Discuss whether there are any differences according to your observations.

Granite is a type of a rock. Quartz is a mineral. Let us find out the differences between rocks and minerals.

18.1 Features of minerals and rocks

A mineral is a solid made up of one component. Rock is a mixture of several components. Minerals exist naturally in the earth in a crystalline form with a definite geometrical shape. Graphite, dolemite, feldspar, ilmenite, mica and quartz are some minerals found in Sri Lanka.The Figure 18.2 shows some crystalline shapes.

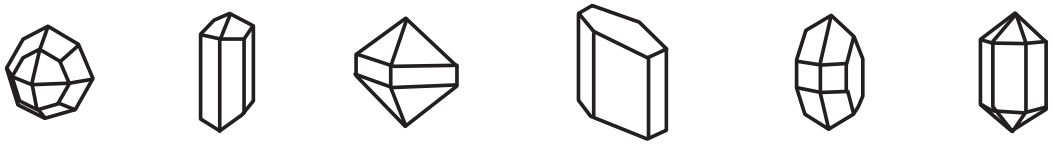


Figure 18.2 ▲ Minerals with different crystalline forms

But, rocks do not have definite geometrical shapes. Granite and gneiss are examples for rocks.

Assignment 18.1

With the help of your teacher get a collection of minerals in your school and fill in the following grid.

Name of the mineral	Colour	Special features

Show the grid to your teacher.

18.2 Kinds of rocks and minerals

You have learnt about the structure of the earth in Unit 08. The outermost surface of the earth which is called crust is made up of rocks.

Types of rocks

Rocks can be divided into three categories according to the way they had been naturally formed on the earth.

- Igneous rocks
- Sedimentary rocks
- Metamorphic rocks

Igneous rocks

About 30 km below the earth's crust, the temperature is more than 5000 °C. Because of this high temperature, the rocks get molted and exist in liquid form. This liquid form of rocks is called magma. During a volcanic eruption magma flows out. This magma is called "lava". Lava gets cool and solidifies to form igneous rocks. Igneous rocks are hard. Granite and Basalt are examples for igneous rocks.

Figure 18.3 a ▲ Flow of lava during a volcanic eruption

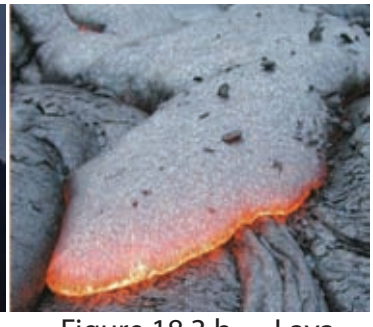


Figure 18.3 b ▲ Lava solidifies forming rocks

Figure 18.3 c ▲ A mountain of Basalt

Sedimentary rocks

Sun light, rain, and wind affect rocks and they break into small pieces. This process is called rock weathering. The rain and wind carry these pieces of rocks to other places and they settle on land, in rivers or sea as layers. Many other things get deposited on these layers. Because of the weight of the upper layers, the lower layers get tighten and the sedimentary rocks are formed.

e.g. :-

- Mudstone
- Conglomerate
- Siltstone
- Sandstone

Figure 18.4 ▲ Sandstone

Figure 18.5 ▲ Limestone

Sedimentary rocks are not as hard as igneous rocks. The skeletons of marine animals such as oyster deposit on bottom of the sea. These get subjected to pressure and limestone is made. Limestone is also a sedimentary rock.

Metamorphic rocks

Because of many reasons such as earthquakes igneous rocks and sedimentary rocks may deposit deep down the earth. These igneous and sedimentary rocks get subjected to extreme pressure and temperature and turn into metamorphic rocks.

e.g. :-

- Limestone which is a sedimentary rock undergoes metamorphism and marble is formed.
- Granite, an igneous rock undergoes metamorphism and gneiss rock is formed.

Figure 18.6 ▲ shist rock

The figure shows a metamorphic rock called “shist”. An igneous rock has undergone metamorphism to form the shist. The change of the layers is an evidence for it.

Types of minerals

A lot of types of minerals can be found in Sri Lanka. The following figure shows some minerals.

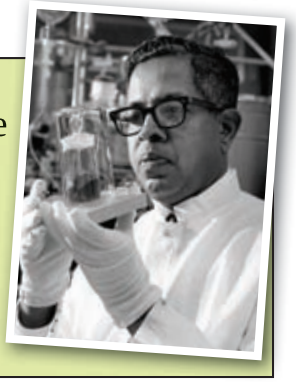
Figure 18.7 ▲ A collection of minerals

e.g. :- 1. Feldspar 2. Dolomite
3. Quartz 4. Ilmenite
5. Mica 6. Graphite

Assignment 18.2

Collect some rocks and minerals. Compare them with the collection of rocks and minerals in your school. Name them and show the collection to your teacher

For extra knowledge



There are rocks not only on earth but also on the moon and on the planets such as Mars and Venus.
The Sri Lankan scientist Dr.Siril Ponnamperuma was the person in charge in checking the rocks and soil brought from the moon during the space voyages of Apollo.

18.3 Rock weathering

When the earth is drilled at one point you will find that it is difficult to drill further because of a rock. It is the bedrock. Therefore, we can imagine that this rock has participated in creating soil.

Due to many reasons the rocks on earth break into pieces and form soil. This process is known as **rock weathering**.

Rock weathering occurs in three ways.

- Physical/mechanical weathering
- Chemical weathering
- Biological weathering

Physical/mechanical weathering

Physical weathering is the process of breaking the rocks into small pieces due to physical factors like heat, wind and flowing water. Physical weathering occurs in many ways.

During the daytime rocks get heated because of the heat of the sunlight and during the night time they get cooled. Also, sudden rains will cool down these heated rocks. Rocks break into pieces because of this heating and cooling process. Let us do Activity 18.2 to understand it.

Activity 18.2

You will need :- A glass marble, forceps, a burner, a container with water

Method :-

- Use the forceps and heat the glass marble.
- When it gets heated put it into the water basin.
- Take out the marble and check it.

You will see that the marble has been cracked.

Because of heating and cooling minerals in rocks get expanded and contracted. These expansions and contractions do not happen in to a same extent. Therefore, the pieces in the rock get loosen and removed.

When water flows through rocks they get weathered. Beacuse of this, the rough edges of stones in flowing water streams are softened and shaped.
Also, when sand which flows with wind hits the rocks, they get weathered.

There might be water inside the cavities of rocks. In countries where the atmospheric temperature is less than the freezing point 0 °C, this water turns into ice. Then the volume increases and rocks can break into pieces.

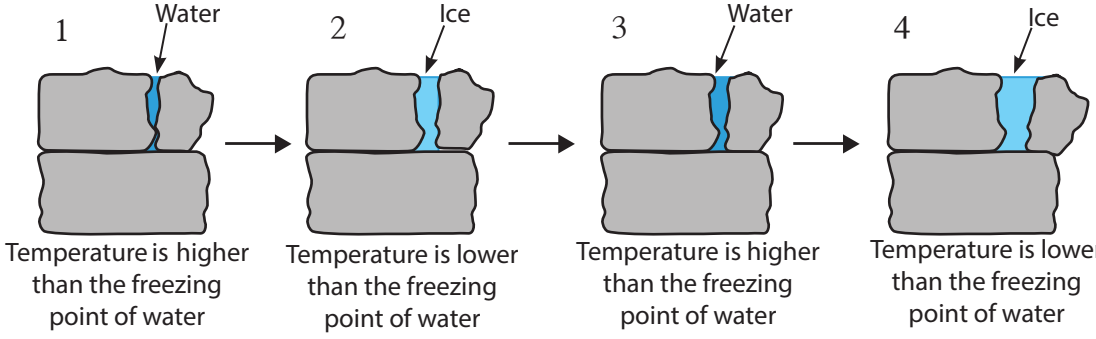


Figure 18.8 ▲ Rocks breaking due to water turning into ice

Chemical weathering

The chemicals in the environment cause rock weathering. Let us do Activity 18.3 to find out about it.

Activity 18.3

Test whether acids cause rock weathering

You will need :- A limestone, vinegar, a glass container

Method :-

- Fill the glass container with vinegar.
- Put the piece of limestone into that container.
- Record your observations.

You will see that the piece of limestone in the container with vinegar dissolves emitting air bubbles.

The conclusion is, limestone got weathered because of acids.

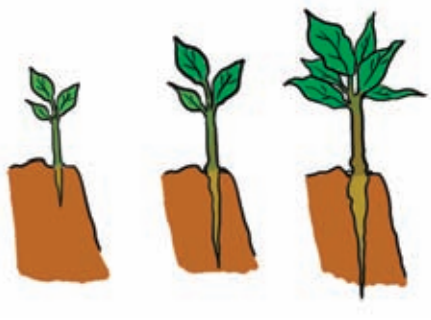
Chemical weathering is the transformation of rocks into other forms by reacting with water, acids and oxygen.

When carbondioxide dissolves in water it becomes acidic. At present the percentage of sulphur dioxide in atmosphere has been increased. When **sulphur dioxide** dissolves in water it becomes very acidic. Rocks get weathered when they react with acidic rain water.

Biological rock weathering

Rocks get weathered because of plant and animal activities.

When a root of a tree enters inside an opening of a rock it grows bigger and bigger and the rock can get cracked.



a

b

Figure 18.9 ▲ Rock weathering by a plant root

Activity 18.4

Observing how a rock weathers naturally

Method :-

- Find a big rock near your house or school.
- Find a place where a lichen can be seen on it.
- Check the texture of things near that lichen (check with your fingertips) once in two weeks for about six months.
- Observe them with a hand lens.

Figure 18.10 ▲ Lichen on a rock

With time you can observe that the substances tested have small rock particles. Thus, you can conclude that the rock is weathered.

On big rocks you can see white and light green patches which are called lichens. Lichens are a combination of algae and fungus. The chemicals and acids produced by lichens are also reasons for rock weathering.

Man uses different methods to break rocks. This is another reason for rock weathering. Due to thudding of hoofs and horn fighting, rock weathering occurs.

The soil on the earth's crust is made of these physical, chemical and biological weathering of rocks.

18.4 Rock cycle

Continuous process, in which the three types of rocks are created, changing from one form to another is known as the rock cycle.

The steps of the rock cycle are given below,

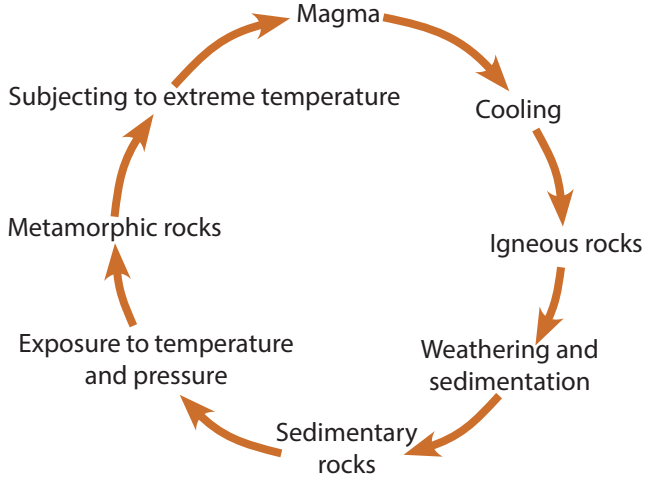
1) Magma, released from volcanic eruptions gets cool and igneous rocks are formed.

2) Igneous rocks weather and get deposit in many places of the earth to form sedimentary rocks.

3) The sedimentary rocks go deep down the earth because of earthquakes and they become metamorphic rocks.

4) Metamorphic and igneous rocks are subjected to extreme temperature and turn into magma.

Following is the summary of the rock cycle



A picture depicting the rock cycle is given below.

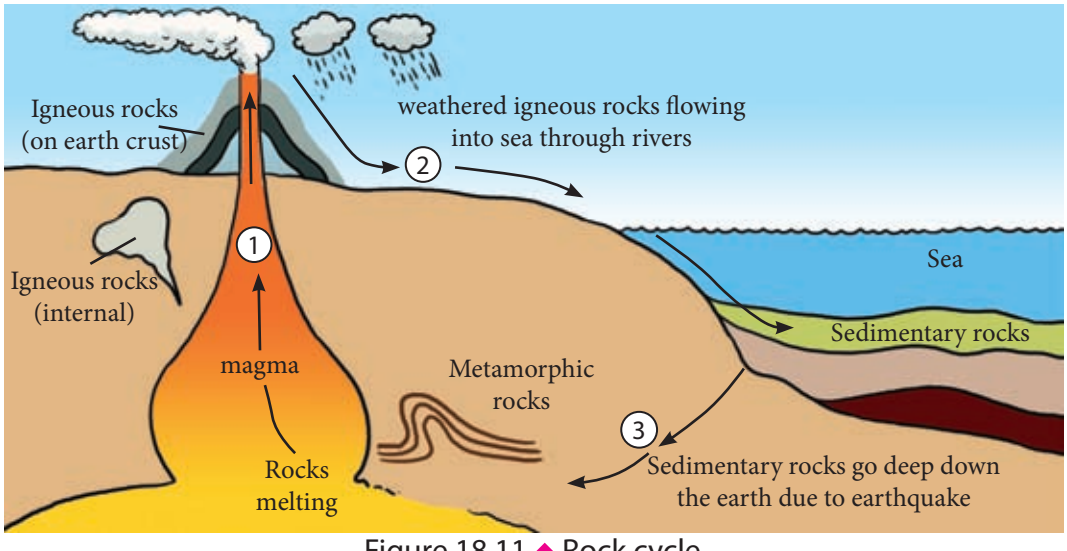


Figure 18.11 ▲ Rock cycle

It takes millions of years for a rock cycle to get completed.

Figure 18.12 explains more about the rock cycle.

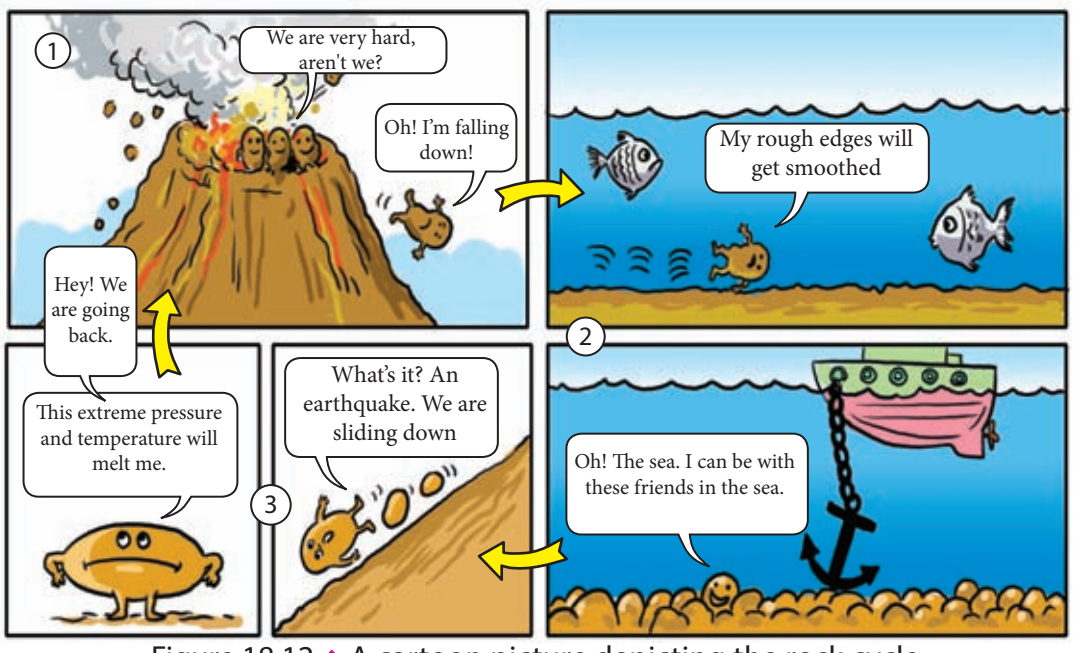


Figure 18.12 ▲ A cartoon picture depicting the rock cycle

Assignment 18.3

Making a model which depict the rock cycle

Create a model of a volcano using clay. Use paint and saw dust and make the outflaw of lava and creation of igneous rock. After making it, use gravel, sand and clay to represent the formation of soil through rock weathering from below the igneous rocks.

Sustainable usage of rocks and minerals

Rocks and minerals are very valuable resources of a country. Therefore, minerals and rocks should be sustainably utilized in a clever and environmental friendly manner. So, that the future generations may use them too.

There are a lot of minerals that can be found naturally in Sri Lanka.

Table 18.1 shows some minerals that can be found in Sri Lanka.

Table 18.1 ▼ Minerals that can be found in Sri Lanka

Mineral	Areas it can be found	Uses
Apatite	Eppawala	Produce Phosphate manure
Dolomite	Kandy, Matale, Badulla, Habarana, Rathnapura	Produce lime Produce manure
Feldspar	Koslanda, Thalagoda	For ceramic industry Produce glass
Ilemenite	Pulmude	Produce paint To get Titanium
Graphite	Bogala, Kahatagaha, Kolonna	Produce pencils, crucibles and lubricants
Mica	Wariyapola, Haldummulla, Madampe	For electric and electronic equipment
Kaolin	Boralasgamuwa, Mitiyagoda	For ceramic industry

Some minerals are named by Sri Lankan names.

For extra knowledge

Ekanite mineral is named by the name of Sri Lankan scientist F.L.D. Ekanayake.

Exporting minerals brings a lot of foreign exchange to our country. But, somehow if we are able to use these minerals as raw materials to produce different products and exporting those products may bring more foreign exchange to the country.

Rocks and minerals are non-renewable resources. Therefore, it is our responsibility to use them very carefully and save for future generations.

Assignment 18.4

Prepare an article about the minerals that can be found in Sri Lanka and their uses. Exhibit it on your wallpaper.

Summary

- Minerals are formed with a single components while rocks are made of several components.
- Rocks can be categorized into three groups as igneous rocks, sedimentary rocks and metamorphic rocks.
- Soil is formed from physical, chemical and biological weathering of rocks.
- The process in which three types of rocks changing from one form into other form for a very long time period is known as the rock cycle.
- As rocks and minerals are very valuable resources they have to be conserved.

Exercise

1. Select the correct answer.

I. Which factors cause rock weathering from the following?

(1) Physical factors (2) Biological factors

(3) Chemical factors (4) All of the above

II. Limestone is,

(1) A mineral (2) An igneous rock

(3) A sedimentary rock (4) A metamorphic rock

2. Match A with B

A	B
Rocks	made with minerals and some other materials. Parts of dead animals and plants can be contained.
Igneous rocks	has been changed from the original form.
Sedimentary rocks	made up with magma deep in the earth.
Metamorphic rocks	made up with one material.
Minerals	made up by tightening the rock pieces.

3. State whether the following statements are True (√) or False (x).

a) Granite is a mineral ( )

b) Fossils can be seen in sedimentary rocks. ( )

c) Rocks are found near the earth's crust. ( )

d) Granite is used to build stairs because granite not easily wear away. ( )

e) Extreme pressure and temperature cause the formation of metamorphic rocks. ( )

Technical Terms

Rocks	-	පාෂාණ	-	பாறைகள்
Minerals	-	ඛනිජ	-	கனியங்கள்
Weathering of rocks	-	පාෂාණ ජීරණය	-	பாறைகள் வானிலையாழிதல்
Physical weathering	-	භෞතික ජීරණය	-	பௌதிக வானிலையாலழிதல
Chemical weathering	-	රසායනික ජීරණය	-	இரசாயன வானிலையாலழிதல்
Rock cycle	-	පාෂාණ චක්‍රය	-	பாறை வட்டம்
Crystals	-	ස්ඵටික	-	பளிங்குகள்
Acid rain	-	අම්ල වර්ෂාව	-	அமிலமழை

19 Sources of Energy

Do you know that to engage in every day-to-day activity you need energy ? We get energy from the sources of energy.

The Table 19.1 shows some day-to-day activities and their sources of energy.

Table 19.1 ▼ Different activities and energy sources

Activity	Energy source
Boiling water, cooking	Firewood. L.P. gas, electricity
Transportation	Petrol, diesel
Drying clothes and different things	Solar energy (Sun)
Using the T.V, radio, computer	Electricity
Operating machines in factories	Electricity, diesel

Energy sources can be divided into two groups.

- Renewable energy sources
- Non-renewable energy sources

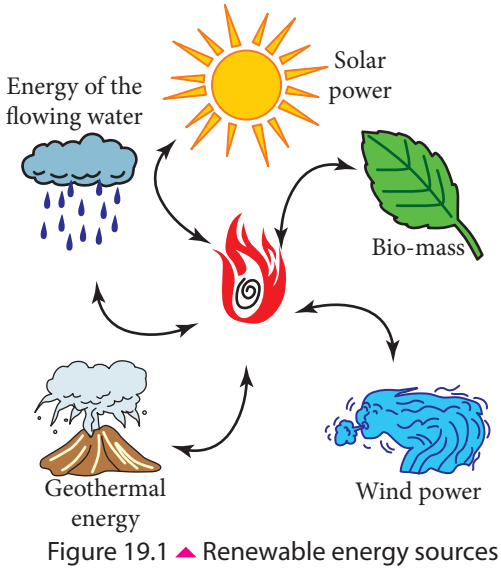


Figure 19.1 ▲ Renewable energy sources

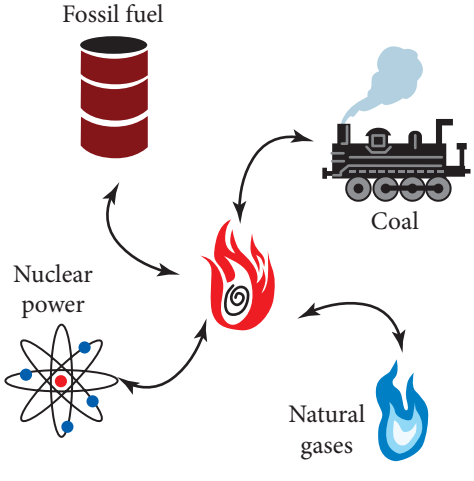


Figure 19.2 ▲ Non-renewable energy sources

To understand the concepts of **renewable** and **non-renewable** energy sources engage in Activity 19.1. Do it in groups.

Activity 19.1

Understanding the concepts of renewable and non-renewable energy sources

You will need :- Two same sized glass fish tanks, three same sized cups, another cup half in size than the other cups, four buckets

Method :-

- Name the two tanks as "A" and "B".
- Ask two of your friends to be near the tank "A" and another two near the tank "B".
- Fill ¾ of each tank with water.
- Give two same sized cups to the friends near tank "A". Give the other two cups to the friends near tank "B".
- Ask one friend near the tank "A" to take out a cup of water from the tank and ask the other to put a cup of water into the tank. Notice the level of water in the tank.
- Ask the friend with the big cup who is near the tank "B" to take out a cup of water from the tank and ask the other friend with the smaller cup to put a cup of water into the tank. Notice the level of water in the tank "B".

The tank "A" represents renewable energy sources while the tank "B" represents non-renewable energy sources.

Renewable energy sources are energy sources that are reproduced during use or within a short period of time after use.

e.g. :- ● Sun ● Wind ● Geothermal energy
● Flowing water ● Bio-mass

Non-renewable sources are not reproduced after use. Otherwise it takes a long period of time to be reproduced.

e.g.:- ● Fossil fuel ● Natural gas
● Coal ● Nuclear power

19.1 Renewable energy sources

Solar power

Solar power gives us light as well as heat. The reason for the wind in the atmosphere and waves in the sea is also the solar power. In some houses solar heaters can be seen on roofs. These heaters are used to boil water for purposes such as bathing.

Figure 19.3 ▲ A solar heater on roof

Activity 19.2

Constructing a simple solar heater

You will need:- A water tank, a rubber pipe, plastic pipes, a black coloured coverlet

Method:-

- Prepare the set up as shown in the figure. It is a model of a solar heater.
- Record the temperature of the water in the tank.

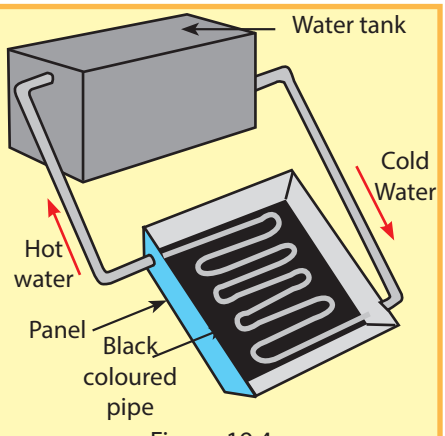


Figure 19.4 ▲

There is a longer pipe of which the surface is black in the panel. It is fixed on a black coloured surface too. It should be fixed in a manner as to allow solar heat to fall on the panel. As the black coloured surface absorb more solar heat. The water inside the pipe is heated by solar heat.

The heated water flows the water tank through the pipe above the water tank. Cool water is at the bottom of the tank. That water flows into the coiled pipe through the pipe at the bottom. When that water is also heated, it flows to the tank through the pipe above. As this is a continuous process water can be heated by this process.

Activity 19.3

Find the effect of the size of the solar heat cooker on its function

You will need :- Several torch reflectors of different sizes, several matchsticks, clay

Method :-

- Fix a matchstick in the hole of the torch reflector with the help of clay. Turn the reflector towards sunlight. Place it in a position in which the top of the match is set at the point where sunlight is collected. Measure the time it takes to light the match.
- Do this experiment by using torch reflectors of different sizes. Measure the time at each occasion. (Do this activity as a instructive model)

Figure 19.5 ▲ A torch reflector

You will observe that the time taken to light the match is decreased when the size of the reflector increases. A similar process takes place in the solar heat cooker too.

When the area of a solar heat cooker is increased, the amount of heat that can be obtained from it also increased.

Solar energy is transformed to electricity by solar cells. Solar cells were first produced to provide electricity to artificial satellites. At present, they are used in calculators, experimenting electric cars, street lamps and aeroplanes.

Activity 19.4

Generating electricity from solar batteries

You will need :- Solar batteries, multimeter

Method :-

- Find a solar panel if possible.
- Connect it to a motor and direct the solar panel towards the sunlight. If not remove the solar panel in a calculator with the help of an adult.

- Connect the connective wires to the multimeter.
- Adjust the multimeter to the scale of milliampere (mA).
- Direct the solar batteries to the sunlight and check the index of the multimeter.

Figure 19.6 a ▲ A street lamps operating from solar batteries

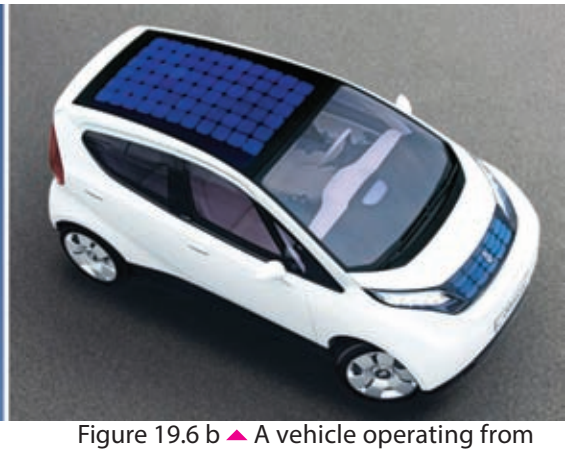


Figure 19.6 b ▲ A vehicle operating from solar batteries

Eventhough solar power is renewable and profitable, there are some disadvantages of it.

Figure 19.7 ▲ A bus stop which lightness up in the night using solar power stored during the day time

- Solar batteries are very expensive.
- Cloudy sky reduces the effectiveness of solar batteries.
- The electricity generated by solar power can be stored in batteries. However, only small amount of electricity can be stored in batteries and they should be disposed in a proper method after discharging. Improper disposal will cause environmental pollution.

Wind power

Solar power heats different places of the earth unevenly. When the air is more heated, it rises up generating a low pressure area in the atmosphere. The cool air from other places flows into the gap. This

circulation of cool air is called the wind. Ancient man has used wind for many purposes.

e.g. :- ● for cleaning paddy ● for grinding grain using windmills
● for transportation of sail boats

Electricity can be generated using wind. Do Activity 19.5 to get an idea about it.

Activity 19.5

Generating electricity using wind power

You will need:- A car cassette motor of 12 V, a LED bulb, a biscuit tin of which the diameter is 14 cm and the height is 10 cm, three plastic water bottles of one litre each (without water), three mega bottle lids, three bolt nails, a small pulley which can be fixed to the axis of the motor, twine or any other thread.

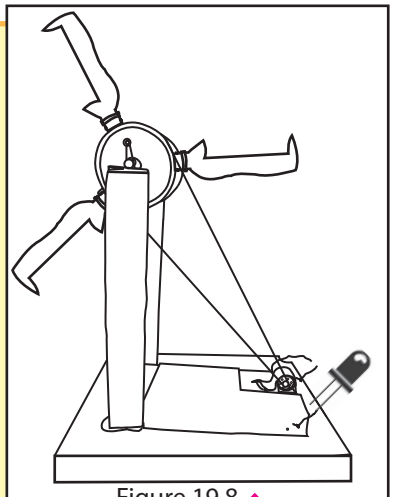


Figure 19.8

Method:-

- Fix the three mega bottle lids at a similar distance to the biscuit tin with the bolt nails.
- Fix three wind blades made by cutting and removing a part of the bottle, to those three bottle lids.
- Do not forget to fix the bottle lids towards corner of the biscuit tin and not in the middle.
- Fix the small pulley to the axis of the motor.
- Connect the car cassette motor of 12 V with the twine thread in a manner as to be able to rotate it.
- Connect a LED to the motor and rotate the turbine with the help of a fan.
- Record your observations.
- Increase the speed of the wind flow and observe the lighting of the LED.

It can be decided that electricity is produced thare as the LED is lighting. It is visible that electricity can be generated with the help of wind power. There are advantages as well as disadvantages in the wind power. The

advantages are that it does not cause any air pollutions, can be obtained free of charge and less environmental damage.
But, wind sufficient to generate electricity can be found in limited areas. Sound pollution takes place due to electricity generators powered by wind.

Geothermal energy

Thermal energy generated and stored in the earth is known as geothermal energy. Below earth's crust there is a hot and molten rock called magma. Because of this high temperature in this layer the water gets heated and turns into steam. This steam is pumped out to turn a turbine. This turbine operates a dynamo and generates electricity.

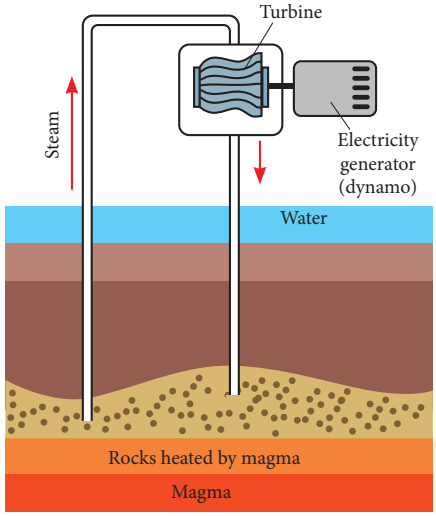
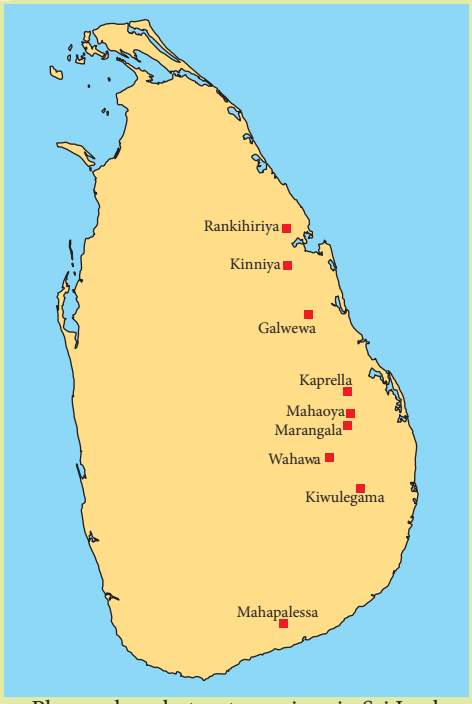


Figure 19.9 ▲ Generating electricity using geothermal energy

For extra knowledge



Places where hot water springs in Sri Lanka

The hot thermal springs (hot water springs) in different areas in Sri Lanka is an example for an instance where geothermal energy is used.
More experiments should be carried out about generating energy from these thermal springs.
Using geothermal energy seems very profitable because energy in deep down the earth's crust is used.
But, there are many drawbacks of using geothermal energy.
The places of earth with a very

high temperature are very deep in the earth's crust. Therefore, to use the geothermal energy the earth has to be drilled very deep. It needs a lot of energy and a high cost. Also many hazardous gases and many chemicals can seep up from the under ground and finding a way to safely dispose of them is very difficult and dangerous.

The energy of flowing water

From ancient time people used the energy of flowing water to rotate water turbines. Those wheels were used to grind grain.

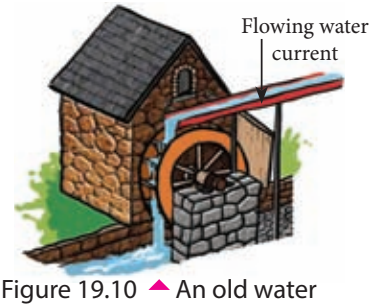


Figure 19.10 An old water wheel

Assignment 19.1

- Use a cork, a bicycle spork, yoghurt spoons, papers, clips and prepare a water wheel
- Hold the wheel to a flow of water and make a thread to coil in the cork. Prepare a method to lift a weight using the thread

Activity 19.6

Generating electricity with a water current

You will need :- 12V car cassette motor, a LED, a biscuit tin of which the diameter is 14 cm and the height is 10 cm, eight mega bottle lids, eight bolt nails, a small pulley which can be fixed to the axis of the motor twine or any other thread.

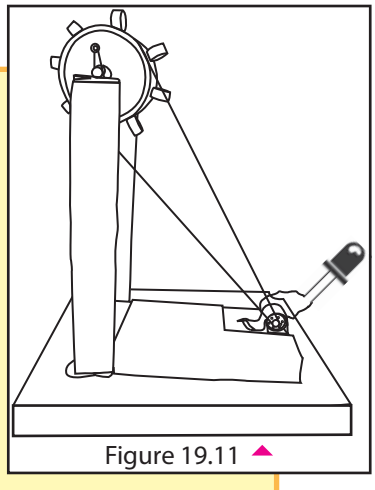


Figure 19.11

Method :-

- Fix the eight mega bottle lids at a similar distance to the biscuit tin with the bolt nails.
- Do not forget to fix the bottle lids towards corner of the biscuits tin, not in the middle.
- Fix the small pulley to the axis of the motor.
- connect the car cassette motor of 12V with the twine thread in a manner as to be able to rotate it.

- Connect a LED to the motor and rotate the turbine with the help of water current.
- Record your observations.
- Increase the speed of the water current and observe the lighting of the LED.

Electricity can be generated by using the energy of the flowing water. A location where electricity is generated using water is known as a hydropower station.

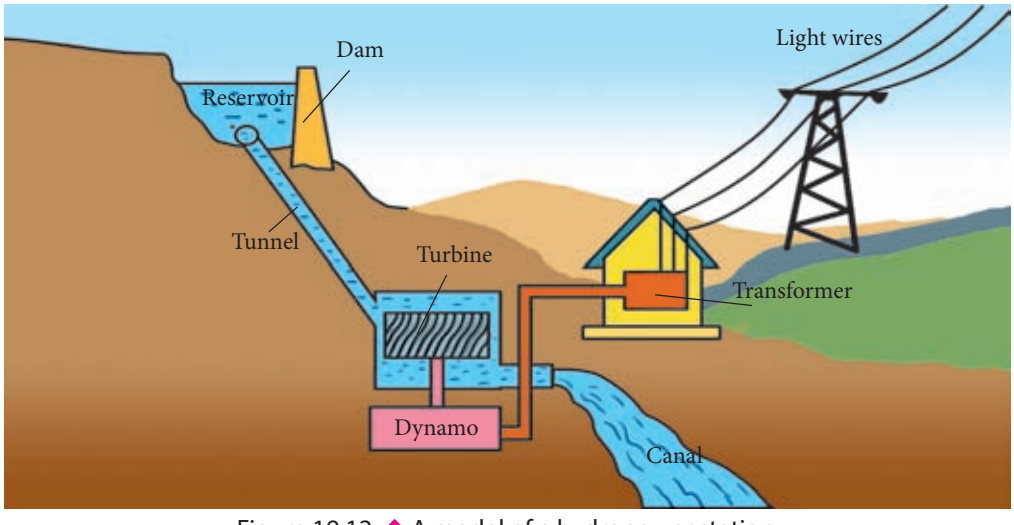


Figure 19.12 ▲ A model of a hydropower station

A concrete dam is constructed by obstructing the flow of water to generate hydro electricity. An artificial reservoir is created by that. The water thus collected in the reservoir is carried to the power station located in a lower level through tunnels and pipes. That water currents used to rotate the turbines. Dynamos (electricity generators) produce electricity by rotating turbines.

Hydropower is an environmental friendly and profitable source of energy. However, the initial cost of constructing a hydropower station is high. Due to the artificial reservoirs, animals lose their habitats. These reservoirs are filled with sediment which cause an issue as well.

In sri lanka, it is not possible to construct dynamo hydropower stations. All the possible locations have been used so far. If the expected amount of rain is not obtained, it is not possible to operate hydropower stations.

Energy of biomass

Biomass energy is the energy which is contained inside the animals and plant materials. Firewood is also considered as biomass. The amount of heat generates depends on the type of firewood.

In India, dried cow dung is used as a fuel to prepare food. Indian government encourages people to use cow dung to make biogas and use it as a fuel. Fuels produced using plant materials are considered as renewable energy sources. As long as we continue to plant new trees to replace those cut down we will always be able to get those fuels.

For extra knowledge

In some countries Gasohol which is a mixture of petrol (gasoline) and alcohol is used as a fuel for vehicles.

Gasohol supply station

Electric rice cooker is used in many homes to cook rice.
NERD institute has introduced a cooker that needs pieces of coconut shells as fuel. It can be used instead of the electric rice cooker.

Assignment 19.2

- List out the advantages and disadvantages of renewable energy sources.

19.2 Non-renewable energy sources

Mineral oil

Mineral oil is a fossil fuel. Fossils are made out of plant and animal matter, subjected to different reactions in underground. The fuels extracted from these fossils are called fossil fuels. Mineral oil is trapped among the rocks in the earth's crust. Metal tubes should be drilled inside these rocks to take the mineral oil. Mineral oil is a very common fuel used in factories and in transportation.
The advantages of mineral oil are easy to use (handle) and can be

converted to energy by modern machines easily.
Limited supply and emitting gas pollutants when combust, are the disadvantages.

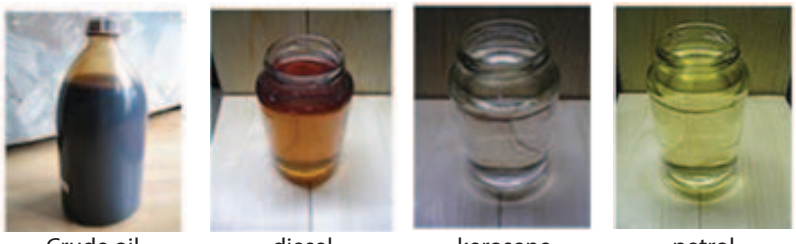


Figure 19.13 ▲ Crude oil and fuels extracted from it

Coal

Coal is also a fossil fuel similar to mineral oil. Coal mines are found among the inner rock layers of the earth.
A lot of heat can be generated by combustion of coal and also it can be handled easily.
It is easy to excavate coal mines from earth and easy to combust. There are more coal (mines) occur than mineral oil (mines) in the world. The gases emitted after combustion of coal cause environmental pollution.

Natural gases

The methane gases trapped within the rocks of the earth's crust are known as natural gases. Natural gases also are a type of fossil fuel. Metal tubes are drilled into the rocks and these gases are taken out. Natural gases are used to cook food and in cold countries they are used to heat the houses.

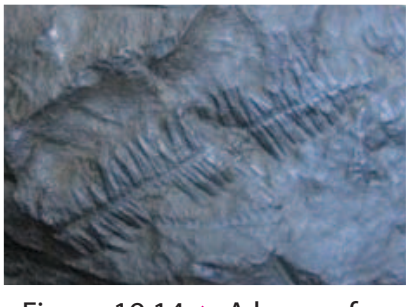


Figure 19.14 ▲ A lump of coal with marks of leaves


Figure 19.15 ▲ An instance of burning natural gases emitted from ground

Natural gases have many advantages; Cheap and easy to use. Pollution is less than coal and mineral oil. But, the supply is limited.

Nuclear Power

Energy is generated in the sun by nuclear reactions. During these reactions one type of atoms convert into another type of atoms.

Nuclear energy is generated by using Uranium which is a radioactive atom. In nuclear power plant electricity is generated using nuclear power.

Very little amount of radioactive atoms such as uranium, plutonium are sufficient to generate a vast amount of energy. This is an advantage of nuclear power.

There are mineral ores located in Sri Lanka, which contain uranium.

There are some disadvantages occurred when using nuclear energy. High expenditure of installing and maintaining nuclear power plants. Highly toxic materials are released as byproducts. Hence, need to be stored for thousands of years without exposing to the environment. That is again a highly expensive task.

Figure 19.16 ▲ A nuclear power plant (This is located in Kudankulum in India close to Sri lanka)

It causes hazardous conditions if radioactive substances are leaked by accidents. It will be harmful to man and the environment.

The discharge from these reactors are very toxic. They should be stored in a way that they would not expose to the environment for a very long time. Storing these discharge costs a lot. If accidently these radioactive substances expose to the environment it will affect people and the environment which will lead to a tragic end.

e.g. :- Explosion at the Chernobyl nuclear plant in Russia
Explosion at Fukushima nuclear plant in Japan.

Assignment 19.2

- Prepare a table that shows the advantages and disadvantages of non-renewable energy sources.

For extra knowledge

The time period of existence of some non-renewable energy sources are given below.

Mineral oil	- about 50 years
Natural gases	- about 70 years
Coal	- about 250 years

19.3 Sustainable usage of energy sources

It is our responsibility to use the non-renewable energy sources in a thrifty manner, as they will disappear one day. So, the future generation will not be able to use these energy sources.

Some strategies are given below for sustainable usage of these energy sources.

- Minimize using domestic electricity
- Walk or use a bicycle for travelling short distances (This will be good for your health too.)
- Use common transportation methods instead of using personal vehicles
- Tune the engine of vehicles to maximize the efficiency of fuel
- Use hybrid or electric vehicles
- Use optional energy sources whenever possible
  e.g. :- Solar water heaters can be used to boil water for bathing
- Use fuel that can be found from the nearby environment
  e.g. :- Using firewood, coconut shells for purposes such as cooking
- Avoid busy hours for travelling
- Minimize using plastic (a lot of fuel is needed to produce plastic)
- Make aware the public the necessity of minimizing the use of fossil fuels
- Fix catalytic converters to vehicles

Activity 19.7

Conduct a debate in the classroom on the topic "using renewable energy sources/non-renewable energy sources help for the country to get long term advantages."

Assignment 19.3

Make a booklet containing measures that can be taken to reduce the electricity cost at home.

Summary

- Energy sources can be divided into two groups as renewable energy sources and non-renewable energy sources.
- Renewable energy sources must be used as much as possible. Then non-renewable energy sources can be protected.
- There are advantages as well as disadvantages of both renewable and non-renewable energy sources.
- Sustainable usage of energy sources will help for the future generation to use them.

Exercise

01) Select the correct answer.

1) "The price of L.P. Gas has been increased." This is a news. As a remedy for this problem what is the best renewable energy source that can be used by Sri Lankans ?

a) Biomass **b)** Solar power
c) Wind power **d)** Tidal energy

2) A non-renewable energy source is

a) Solar power **b)** Coal
c) Energy in flowing water **d)** Wind power

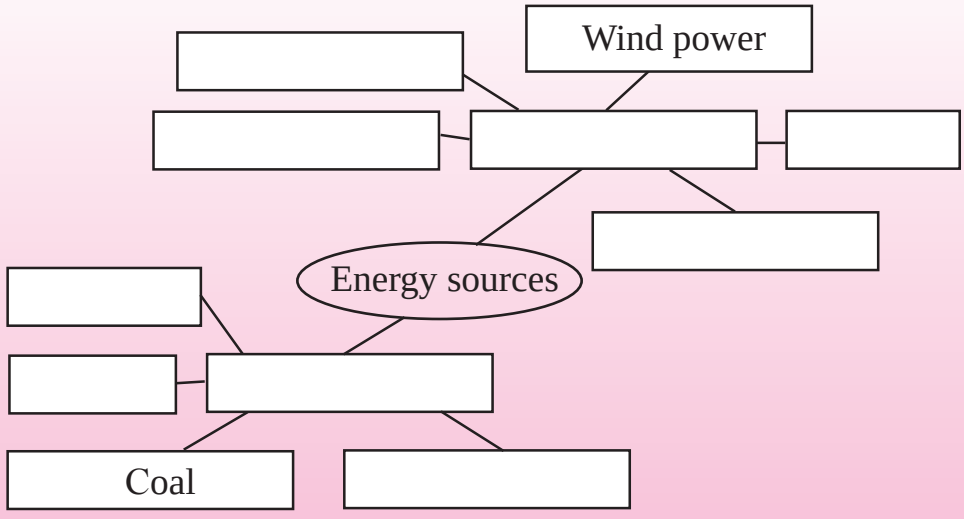
3) An energy source that does not fall into biomass category is

a) Dry cow dung **b)** Oil of castor seeds

c) Geothermal energy **d)** Chaff

02) Use the given words/ phrases and fill in the chart.

Hydro power, natural gases, solar power, mineral oil, biomass, renewable energy sources, geothermal heat, nuclear power, non-renewable energy sources



Technical Terms

Energy sources	-	ශක්ති ප්‍රභව	-	சக்தி முதல்
Renewable	-	පුනර්ජනනීය	-	புதுப்பிக்கக்கூடிய
Non renewable	-	පුනර්ජනනීය නොවන	-	புதுப்பிக்க முடியாத
Solar water heater	-	සූර්ය ජල තාපකය	-	சூரிய நீர் வெப்பமாக்கி
Solar cells	-	සූර්ය කෝෂ	-	சூரிய கலம்
Thermal springs	-	උණු දිය උල්පත්	-	வெந்நீரூற்று
Geothermal energy	-	භූ තාප ශක්තිය	-	புவி வெப்பம்
Biomass	-	ජෛව ස්කන්ධ	-	உயிர்த் திணிவு
Nuclear plants	-	න්‍යෂ්ටික බලාගාර	-	அணுக்கருச்சக்தி நிலையம்