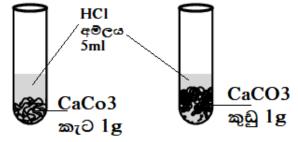


Rate of the Reaction

- 01) Given below are some of reactions around us.
 - Ripening of fruits
 - Burning a paper
 - Dissolving salt in water
 - Blast of a cracker

- Rusting of iron nail
- Breaking a rock
- Turning water into ice
- i) Select the chemical reactions from the above.
- ii) Name 2 fast chemical reactions from the above.
- 02) What are the 4 factors that affect the rate of the reaction?
- 03) Following is the apparatus related to a chemical reaction.



- i) It was taken 8 minutes to complete the reaction of CaCO3 crystals and 2 minutes to complete the reaction of CaCO3 powder. Calculate the rate of the reactions separately.
- ii) What is the conclusion of the reaction?
- 04) Pay your attention to the activity on page number 120 in your textbook.
- i) If the time to disappear the purple colour of the KMnO₄ solutions is measured, which KMnO₄ solution will get less time to disappear the purple colour?
- ii) What is the reason for the above observation?