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Province/Weekly School Department (of Education, Sabaragamuwa Province/Wei Translated by: Ms. Ms. K.P.A Menaka
De Grade -10	nuwa Province/ Weekly School Depart Kg/Dehi Sirisaman National School

Read the page numbers from **164 to 181 in Part I Science Book** carefully and write the answers for the below given Questions.

- 1) Mention the term, Chemical Bond.
- 2) Name 2 types of such Chemical Bonds.
- 3) What is the ion formed by Metals, by removing electrons?
- 4) What is the ion formed by Nonmetals, by grasping electrons?
- 5) Fill in the banks in below given table.

Element	Electronic configuration	Ionic electronic configuration	Ion symbol
Li	2,1		Li ⁺
Na	2,8,1		
Ca		2,8,8	
F	2,7		F
Cl	2,8,7		
0		2,8	

- 6) Represent the ionic bond formed in between Na and Cl using dot cross diagram.
- 7) Mention 5 examples for ionic compounds.
- 8) Explain the covalent bond formed in between H and O atoms in water molecule.

9) What are the 3 ways that a covalent bond can be represented?

10) Use the proper diagrams to represent the following covalent bonds.

- i. CH₄
- $ii. \ O_2$
- iii. CO₂

11) Explain the way of forming an ionic lattice.

12) Define the following terms,

- i. Electronegativity
- ii. Polarity
- iii. Polar Covalent Bonds
- iv. Non polar Covalent Bonds.