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## **Hydrocarbons and their Derivatives**

Compounds containing carbon & hydrogen only are hydrocarbons

Ex: Petrol, Methane, LP gas, Kerosene oil, Diesel, Wax

• Study the text book & writ down the steps of a test to confirm that carbon & hydrogen are contained in wax

Based on their structures hydrocarbons are classified into three parts

- Alkane
- Alkene
- Alkyne
  - What are known as alkanes & Explain the bond type between C & H using an example

Hydrocarbons in which the number of double bonds is one more are called alkenes Ex: Ethene ( $C_2H_4$ )

$$H \subset C \subset H$$

	Alkane	Alkene	Alkynes
Introduction			
Structure			
Bond type between C & H			

•	Name four derivatives of ethene & draw their structures

Ethene & derivatives f ethene are used to produce polythene, Styrofoam & Teflon

Polythene polyme	r is given bellow		
H H H	$ \begin{array}{c ccccc} H & H & H & H \\                        $	H    -C-     H	
ii) Mention th iii) Briefly ind	e monomer of polythe	ne olythene	olyethene belongs?
	ythene, Polymers like pylene (Teflon) can be i		lyvinyl chloride),
Material Material	Special Properties	Polymer includes	Structural formula of monomer
Polythene bags			
Water pipes			
Non-Stick cooking pans			
•	lassified into two parts		
	ivided into two based o	_	
1)	r Natural & Artificial :	•	
Write examples fo			
Write examples fo Polymers can be c 1)	-		them

• What are known as polymers?